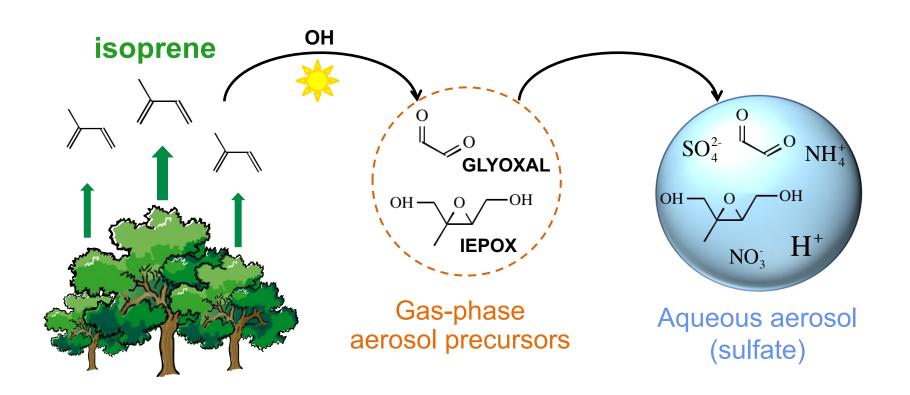
Dual air quality benefit of SO₂ emission controls by decreasing sulfate and secondary organic aerosol



Coauthors: D. J. Jacob, J. L. Jimenez, P. Campuzano-Jost, D. A. Day, W. Hu, J. Krechmer, L. Zhu, P. S. Kim, C. C. Miller, J. A. Fisher, K. Travis, K. Yu, T. F. Hanisco, G. M. Wolfe, H. L. Arkinson, J. R. Turner, L. J. Mickley, H. O. T. Pye, K. D. Froyd, J. Liao, V. F. McNeill





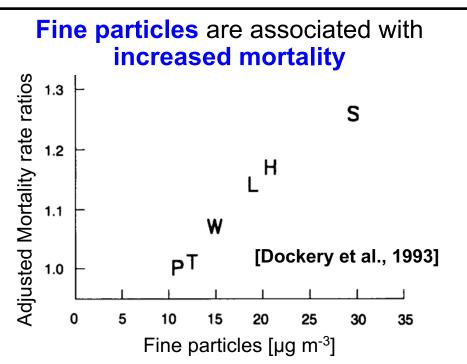


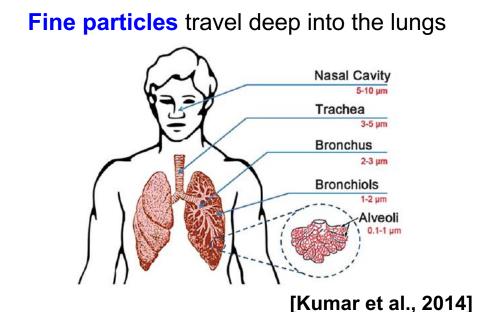




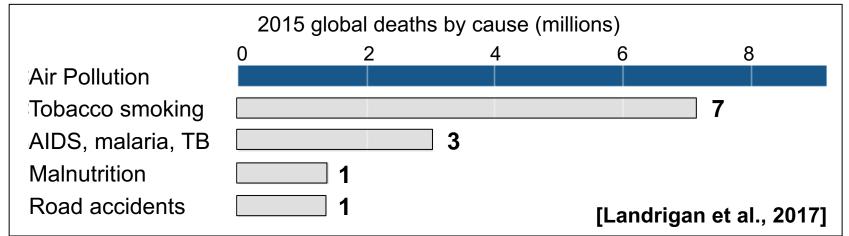


Aerosols Impact Health



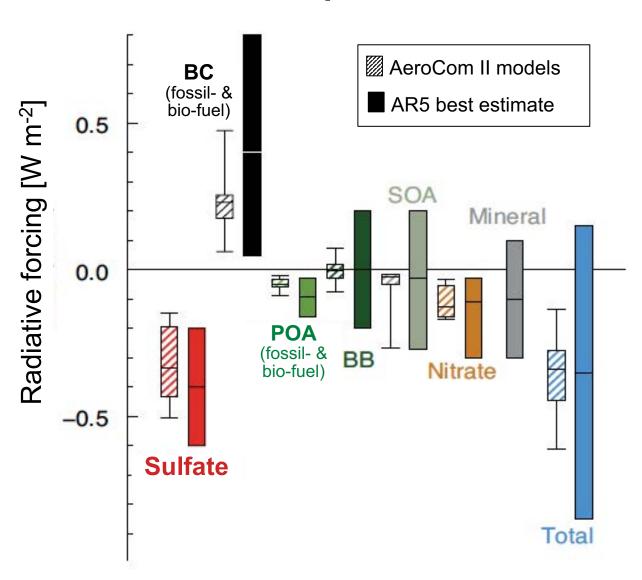


Air pollution (mostly $PM_{2.5}$) is now the greatest health risk



Aerosols Impact Climate

Climate impact is uncertain



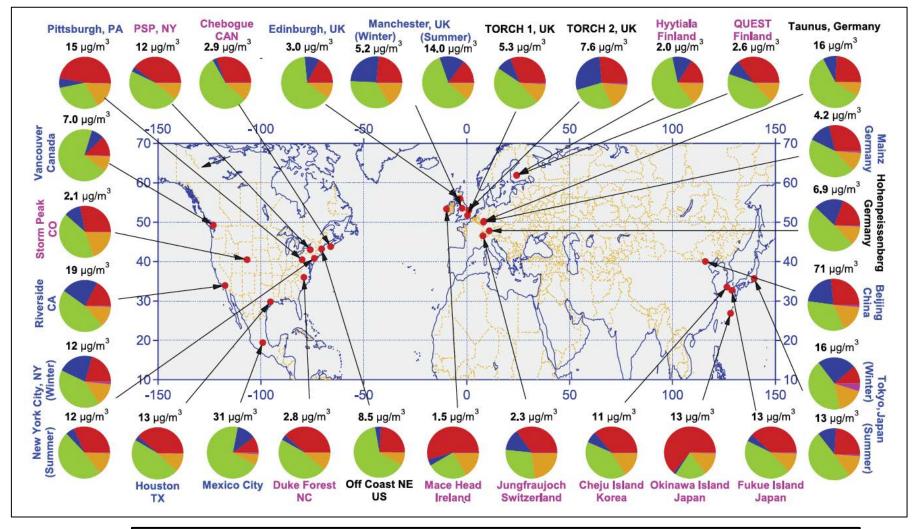
Aerosols Impact Visibility



[The Guardian, 23 Feb 2016]

Organic Aerosol is Ubiquitous in the Atmosphere

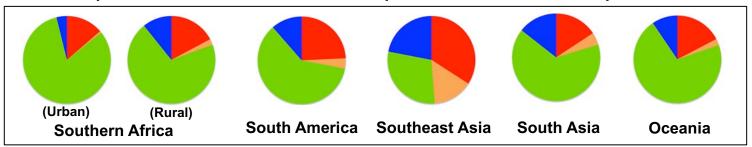
Northern hemisphere aerosol composition



Sulfate Nitrate Ammonium Organics (OA)

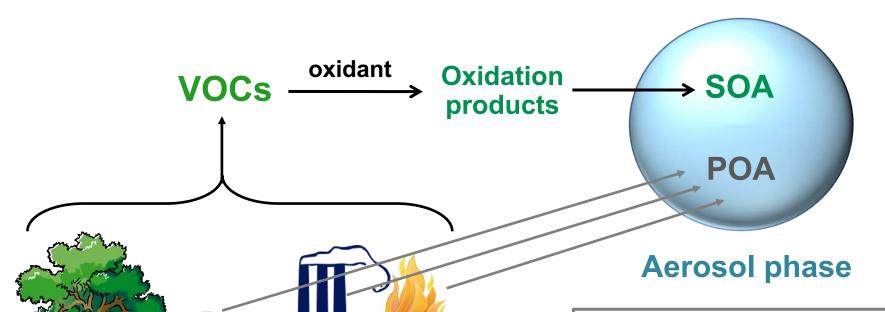
Organic Aerosol is Ubiquitous in the Atmosphere

Tropics and southern hemisphere aerosol composition



[IPCC, 2013]

Sulfate Nitrate Ammonium Organics (OA)



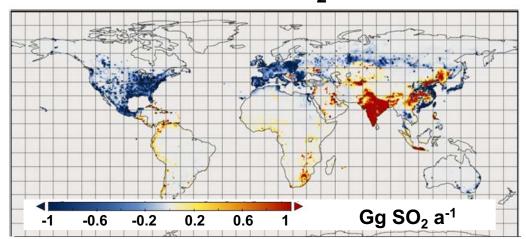
SOA: secondary organic aerosol

POA: primary organic aerosol

Organic Aerosol Fraction is Changing

Changes in sulfate alter the contribution of OA to PM_{2.5}

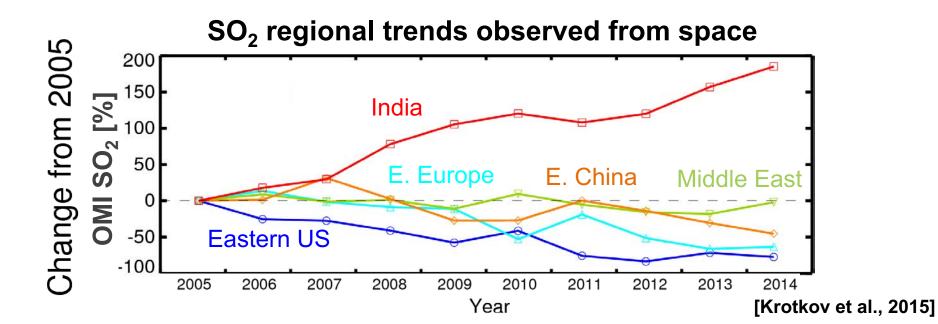
2010 minus 2005 SO₂ emissions



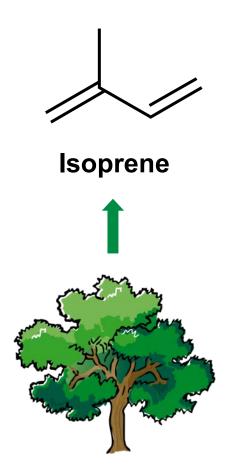
Global bottom-up emission inventory trends (left)

Corroborated by surface and satellite observations (below)

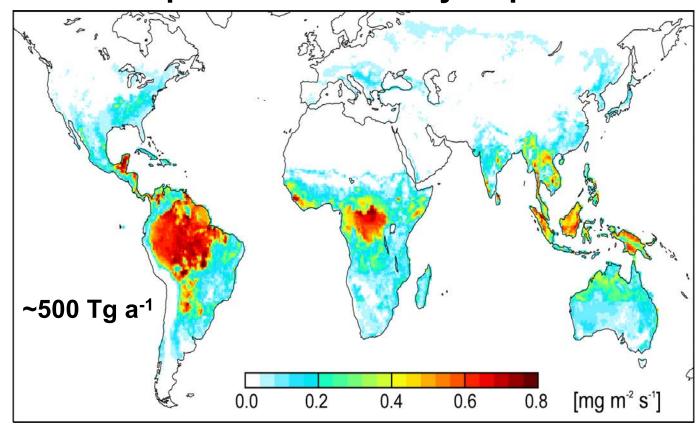
[Klimont et al., 2013]



Isoprene as a precursor of secondary organic aerosol



Most isoprene is emitted by tropical trees



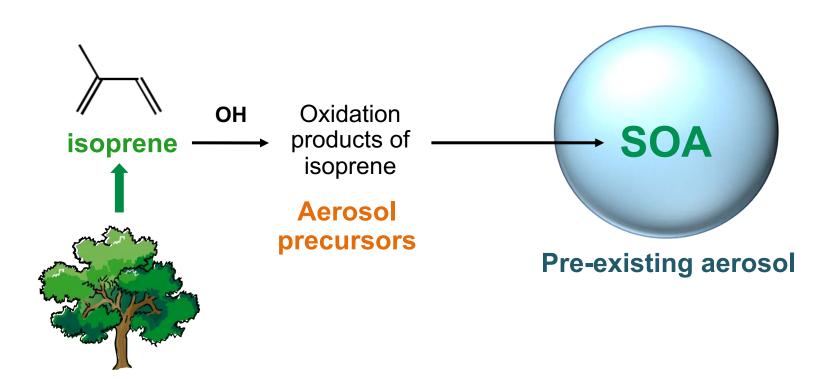
[Guenther et al., 2012]

Factors that affect emissions:

plant type, temperature, light, soil moisture, CO₂, plant physiology

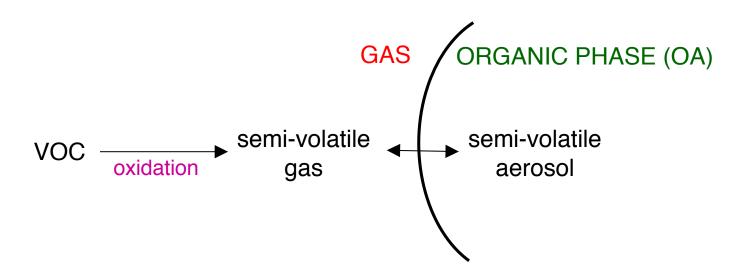
Isoprene as a precursor of secondary organic aerosol

Isoprene is oxidized to form compounds that then condense to pre-existing aerosol to form secondary organic aerosol (**SOA**)

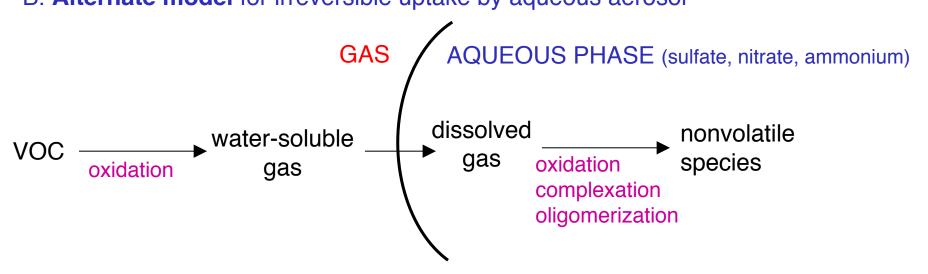


Two approaches to represent secondary organic aerosol

A. Classical model for reversible uptake by pre-existing organic aerosol



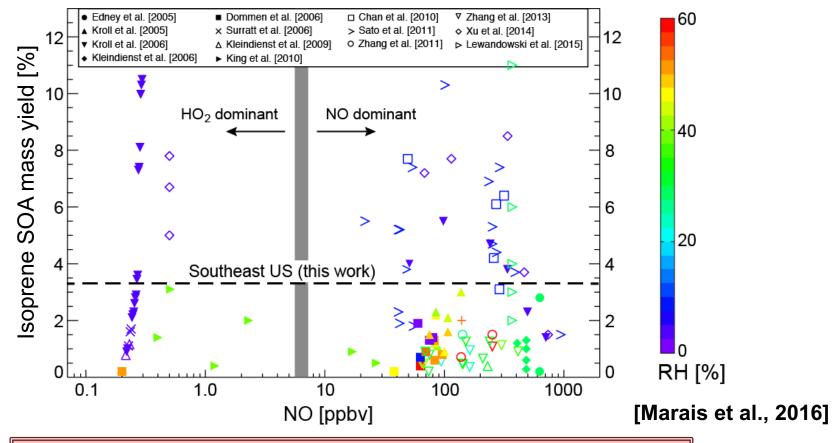
B. Alternate model for irreversible uptake by aqueous aerosol



Model Parameterizations of Isoprene Organic Aerosol

Traditional model based on chamber studies conducted at conditions far from ambient atmosphere

Compilation of chamber study isoprene + OH SOA mass yields

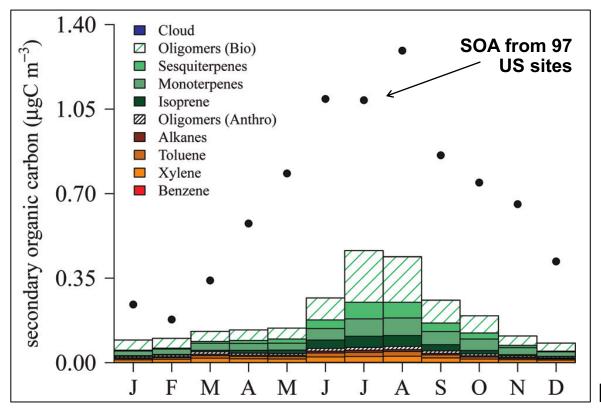


Southeast US Boundary- Layer Summer Conditions

RH = $72 \pm 17 \%$ NO = $0.053 \pm 0.140 \text{ ppbv}$ isoprene = $0.78 \pm 0.85 \text{ ppbv}$

The classical model routinely misrepresents SOA

Measured vs modeled SOA across the US



[Carlton et al., 2010]

Model captures the seasonality (peaks in summer is due to biogenic SOA), but not the magnitude

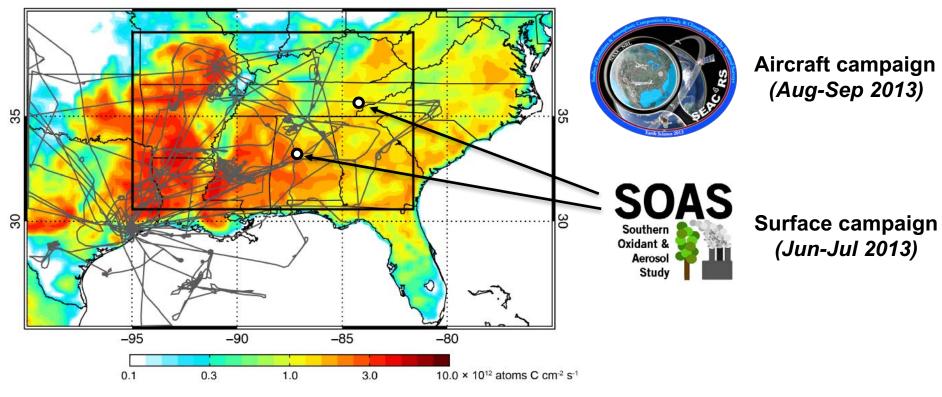
Limits ability to determine impact of isoprene SOA on climate and human health

Increasing evidence that SOA formation is instead by irreversible uptake to aqueous aerosol

The Southeast United States

Multiple summer 2013 campaigns to understand biogenic-anthropogenic interactions

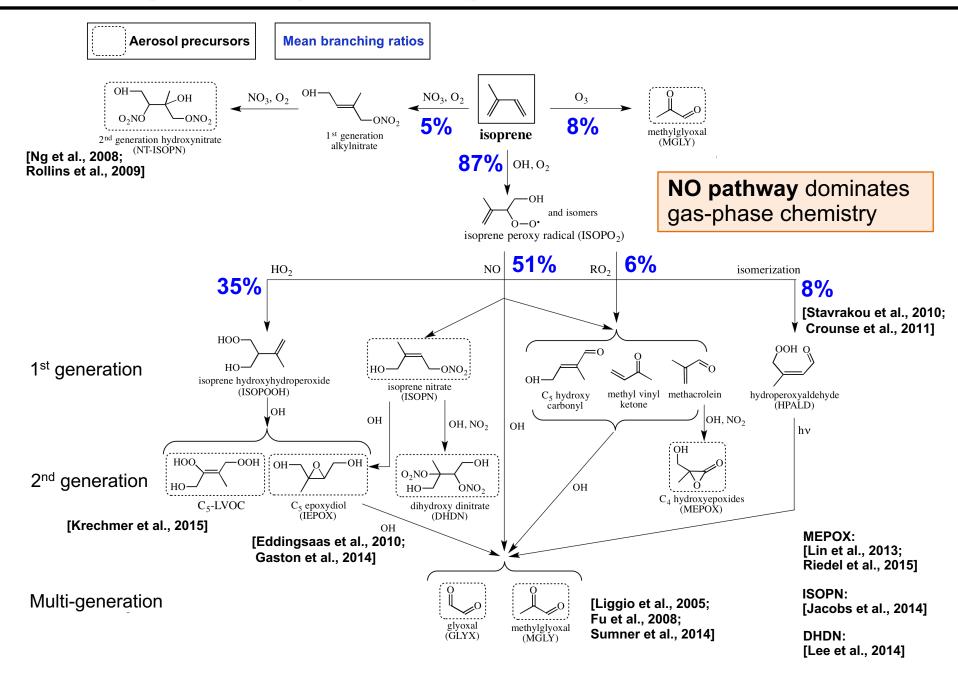
MEGAN isoprene emissions, SEAC⁴RS flight tracks, and SOAS monitoring sites



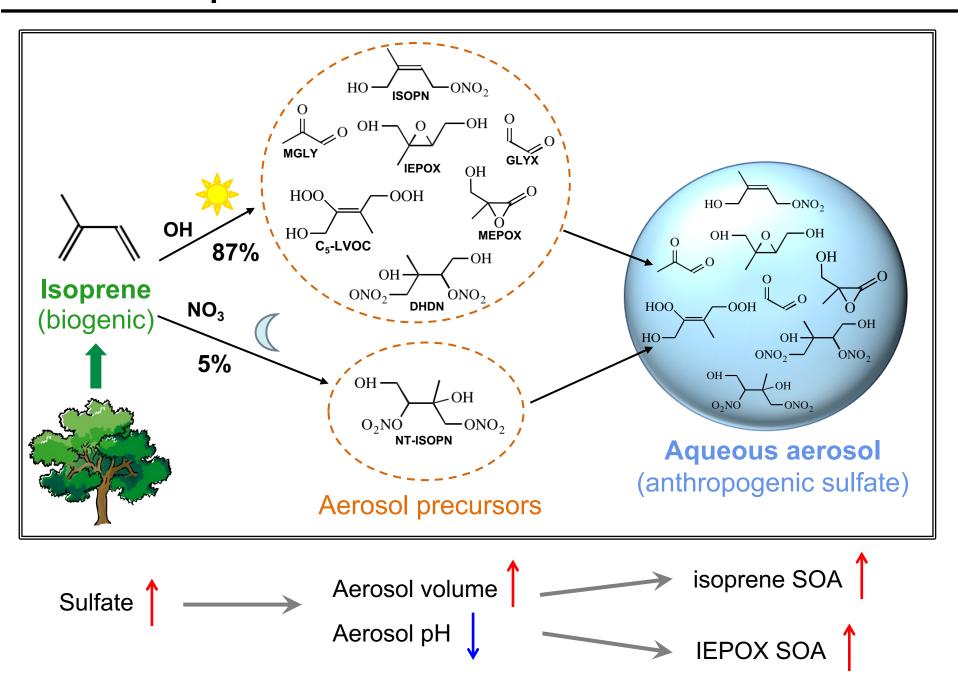
[Kim, P. et al., 2015]

In summer the Southeast US is a large source of **biogenic isoprene** (high temperatures) and **anthropogenic sulfate** (from oxidation of SO₂)

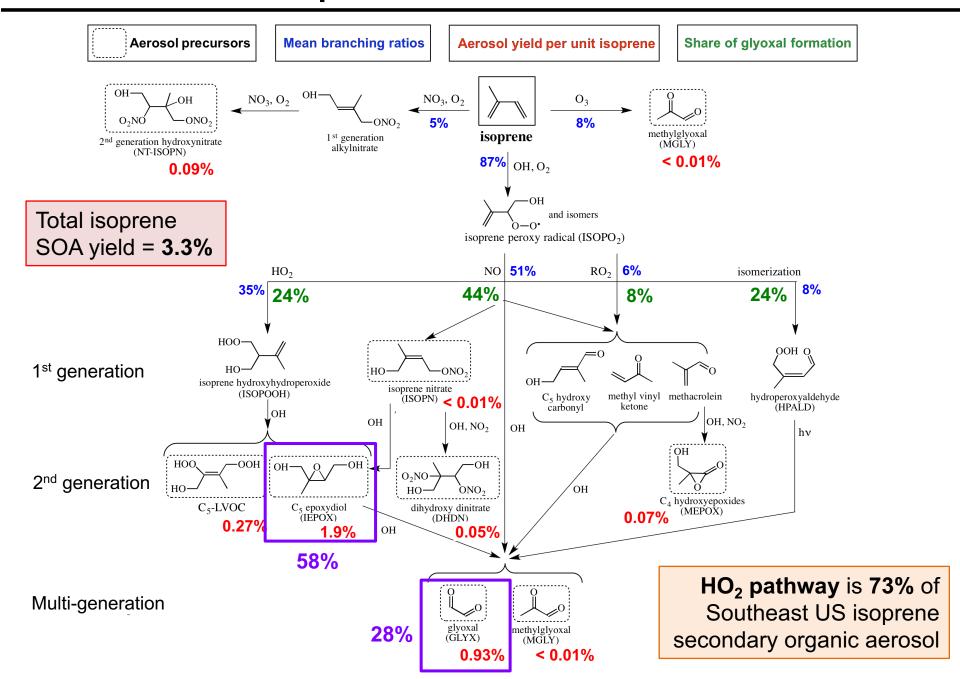
Gas-phase Isoprene SOA precursors in GEOS-Chem



Aqueous-Phase Mechanism Framework



GEOS-Chem Isoprene SOA Yields in the Southeast US

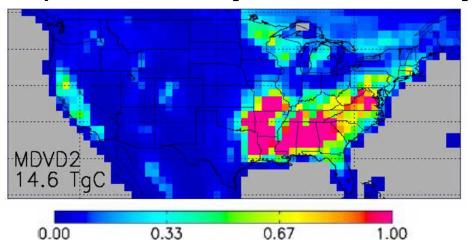


OA-HCHO Relationship

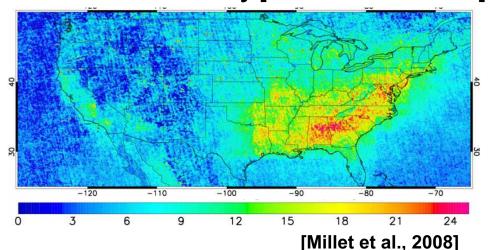
Relationship should be sensitive to isoprene SOA yields

Isoprene is the largest source of HCHO

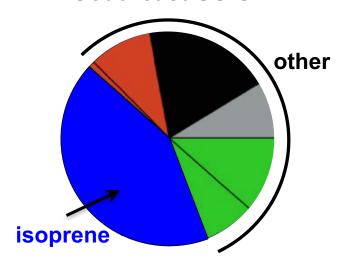
Isoprene Emissions [10¹³ atoms C cm⁻² s⁻¹]



HCHO Column Density [10¹⁶ molecules cm⁻²]



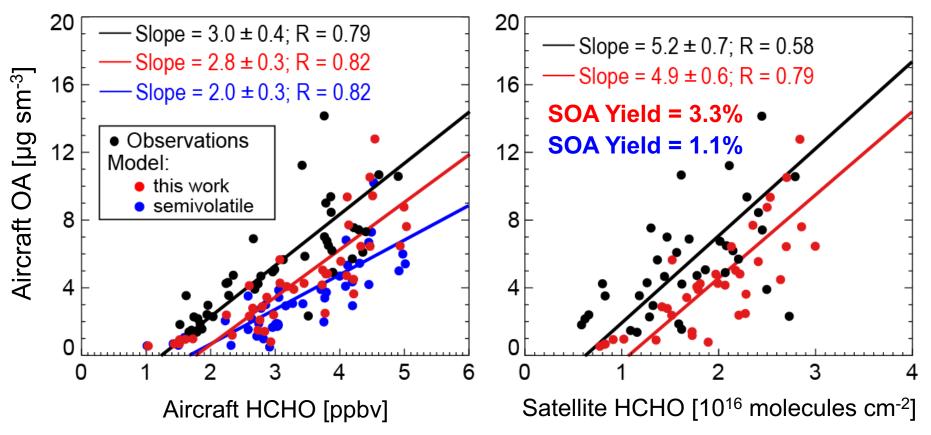
Isoprene SOA is 40% of OA Southeast US OA



Adapted from Kim et al. [2015]

OA-HCHO Relationship Constrains Isoprene SOA Yields

OA-HCHO Relationships during SEAC⁴RS (Aug-Sep 2013)



Traditional scheme underestimates the slope

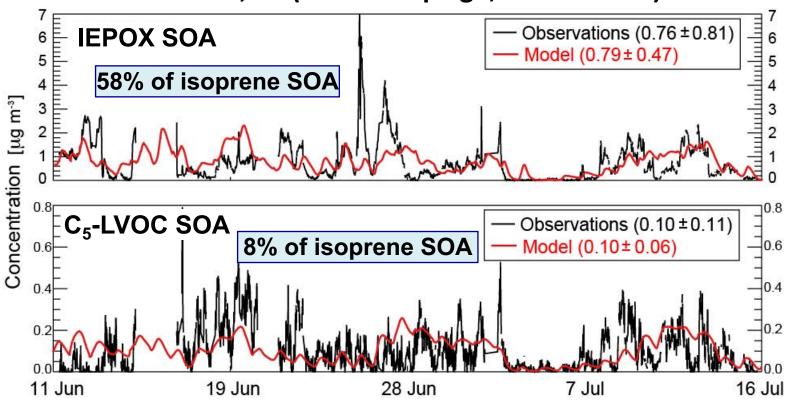
Irreversible uptake scheme reproduces observations: 3.3% isoprene SOA yield.

[Data from T. Hanisco, G. Wolfe, H. Arkinson, L. Zhu, J. L. Jimenez, P. Campuzano-Jost]

Observational Constraints on Isoprene SOA Components

ISOP
$$\xrightarrow{OH, O_2}$$
 ISOPO₂ $\xrightarrow{HO_2}$ ISOPOOH \xrightarrow{OH} $\xrightarrow{75\%}$ IEPOX C_5 -LVOC

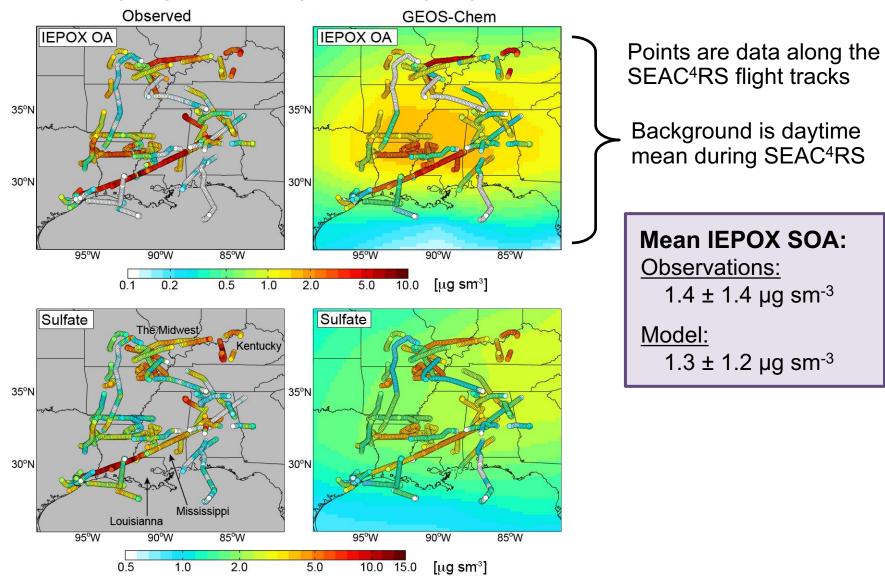
Secondary organic aerosol from IEPOX and C₅-LVOC at Centreville, AL (SOAS campaign; Jun-Jul 2013)



[Data from D. A. Day, W. Hu, J. Krechmer, J. L. Jimenez]

Spatial Distribution of IEPOX SOA

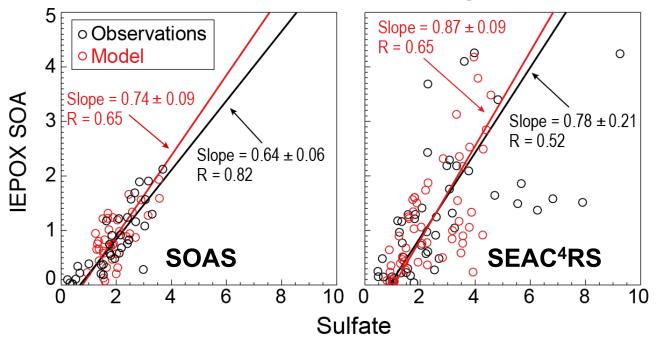
SEAC⁴RS (Aug-Sep 2013) boundary-layer IEPOX SOA and sulfate



[Data from P. Campuzano-Jost, J. L. Jimenez]

What modulates IEPOX OA in the Southeast US?

IEPOX SOA and Sulfate correlation during SOAS and SEAC4RS

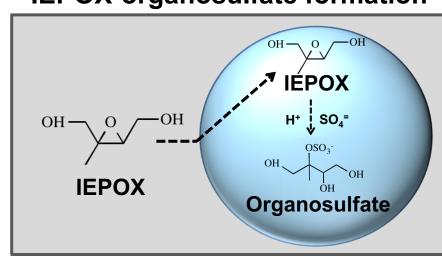


Similar relationship between sulfate and IEPOX OA in the observations and model

Correlation identified throughout the **Southeast US**:

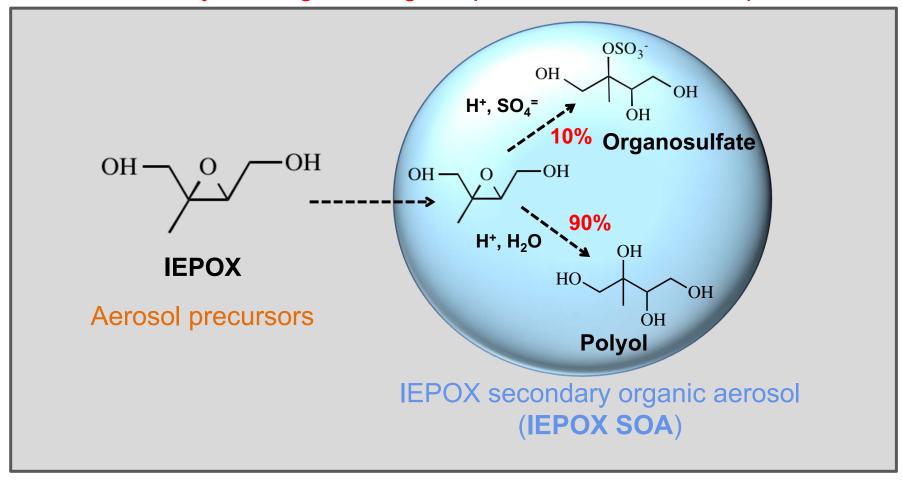
Budisulistiorini et al. [2013, 2015]; Xu et al., [2015a, 2015b]; Hu et al. [2015]

IEPOX-organosulfate formation



Sulfate correlation not due to nucleophilic addition

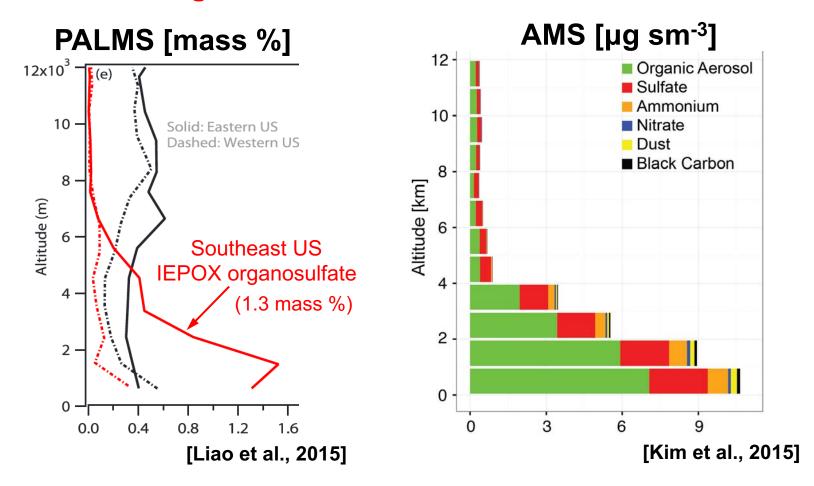
Acid-catalyzed ring cleavage to produce non-volatile species



In our mechanism acid-catalyzed sulfate addition is 10% and acid-catalyzed H₂O addition is 90% of the fate of IEPOX

Aircraft observations constrain organosulfate formation

Additional support for limited role of sulfate channel from SEAC⁴RS PALMS IEPOX-organosulfate observations:

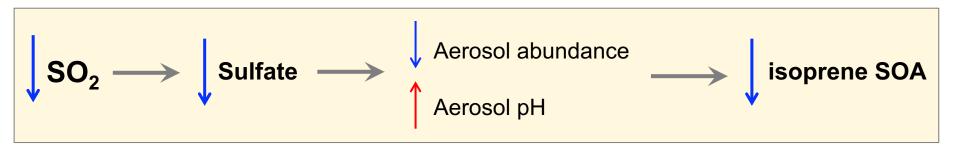


Boundary-layer IEPOX-organosulfate: **0.14 μg sm**-³ (10% of IEPOX SOA)

IEPOX-organosulfates long-lived, so remain intact throughout the aerosol lifetime

Sulfate impacts aerosol acidity and volume

Anthropogenic sulfate influences isoprene SOA formation



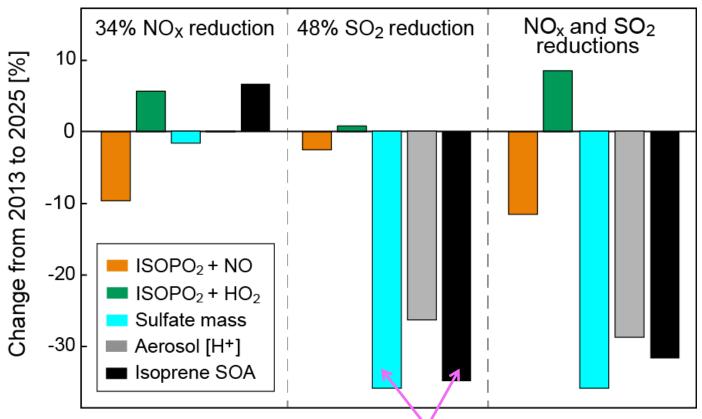
Aqueous aerosol abundance impacts all isoprene SOA precursors

Aerosol acidity impacts IEPOX

Effect of Anthropogenic Emission Reductions

Test the effect of future SO₂ and NO_x emission controls on isoprene SOA

Changes in sulfate, aerosol pH, and isoprene SOA

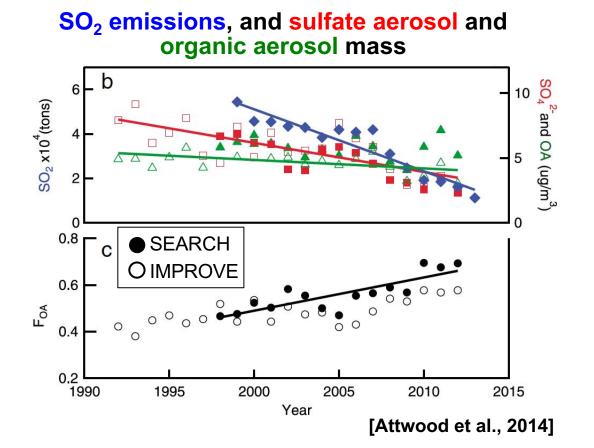


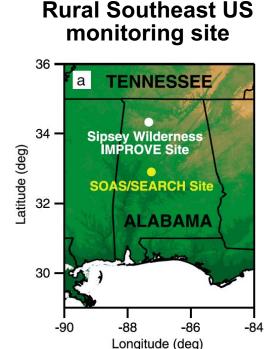
Near-equivalent decrease in sulfate and isoprene SOA

Policy implication: Dual benefit from targeting SO₂ sources

Organic Aerosol Fraction is Increasing – Southeast US

The increasing contribution of organic aerosol is apparent at a rural monitoring site in the Southeast US



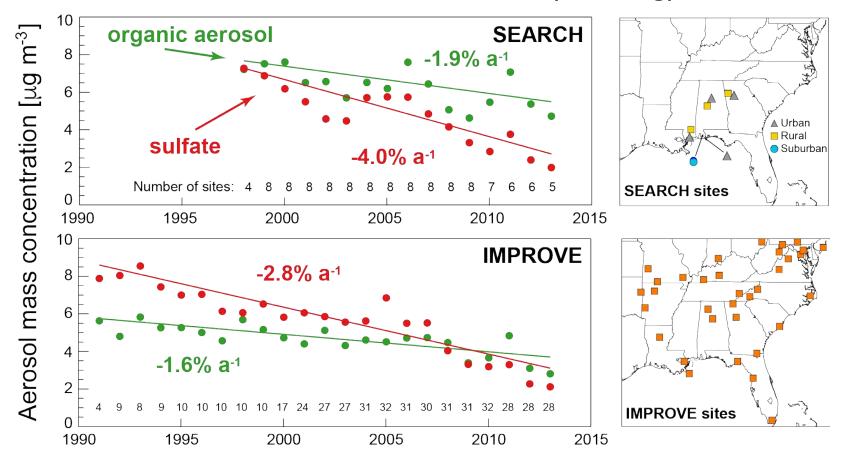


Site impacted by urban, industrial, **biogenic**, and agricultural emissions.

F_{OA} (fraction of organic aerosol) increased from 40% (1992) to 60% (2012).

Observed decline in sulfate and OA in the Southeast US

Observed 1991-2013 trends in summertime (Jun-Aug) sulfate and OA

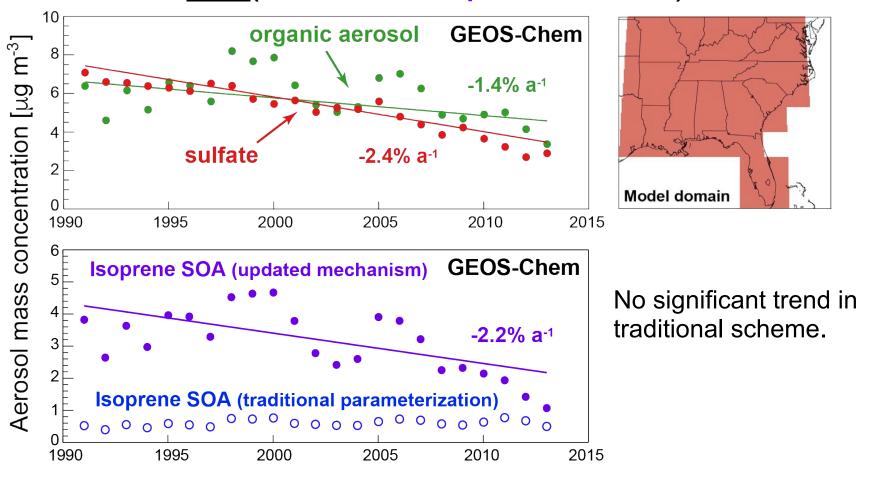


Steeper decline in sulfate at SEARCH than IMPROVE sites – greater urban influence. Similar OA trends supports biogenic SOA driving the trend

OA instead of sulfate is now the dominant PM_{2.5} component in the Southeast US

Modelled OA decreases due to decline in isoprene SOA

Model 1991-2013 trends in summertime sulfate and OA, and <u>isoprene</u> SOA (traditional and updated schemes)

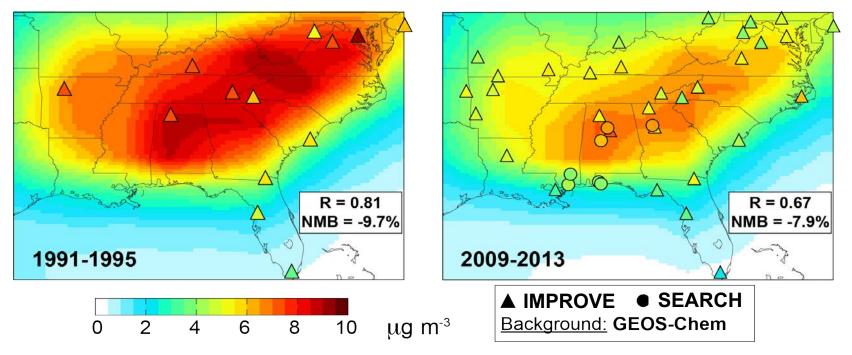


Model includes annual trends in anthropogenic SO_2 , NO_x , and VOCs emissions. Large OA interannual variability due to isoprene emissions.

Majority of decline in modelled OA is due to isoprene SOA

Spatial distribution of organic aerosol trends

Spatial distribution of five-year mean summertime OA from the model and observations at the start and end of the record

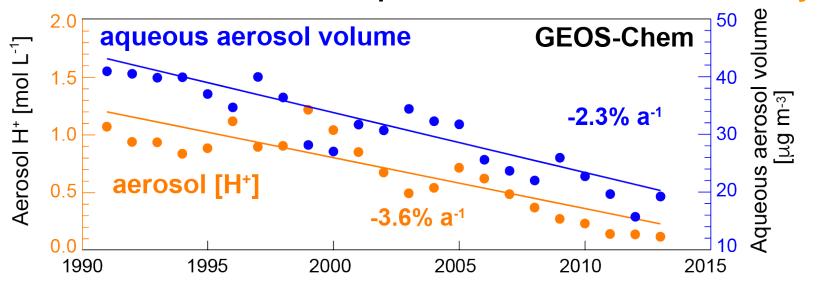


No significant change in OA spatial distribution in the observations or model supports biogenic SOA driving the OA trend.

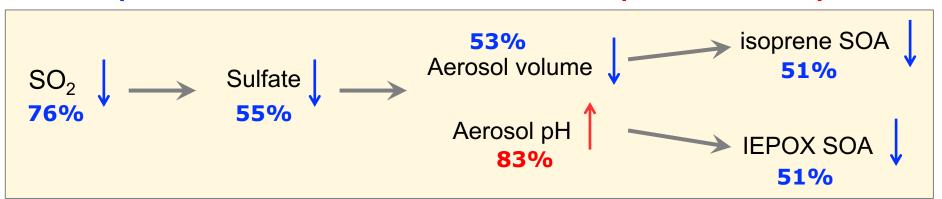
Small model normalized mean bias (NMB) and similar change in OA in the model and observations

Modelled isoprene SOA decreases due to decline in sulfate

Model trends in summertime aqueous aerosol volume and acidity

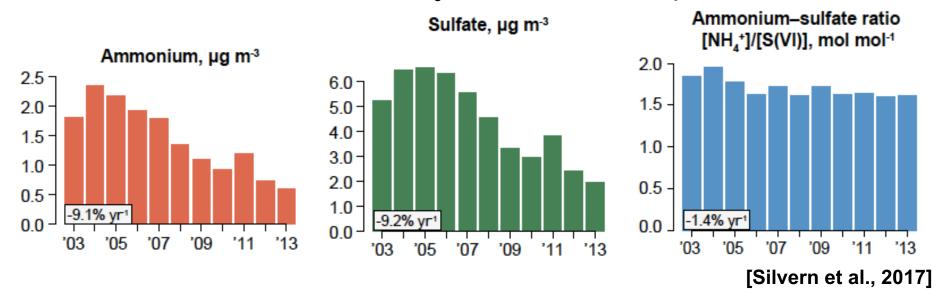


Decline in sulfate (dominant aqueous aerosol component) decreases aqueous aerosol volume and increases aqueous aerosol pH

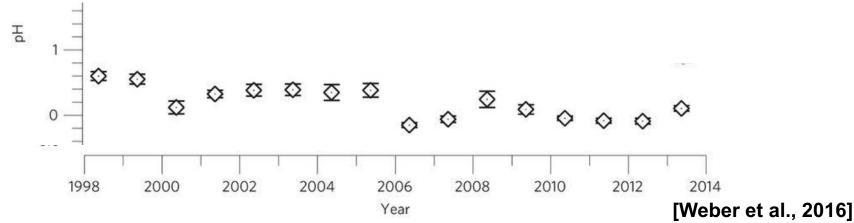


Aerosol Acidity Trend is Controversial

Observations (2003-2013) show decrease in ammonium and sulfate even though emissions of NH₃ have remained steady:

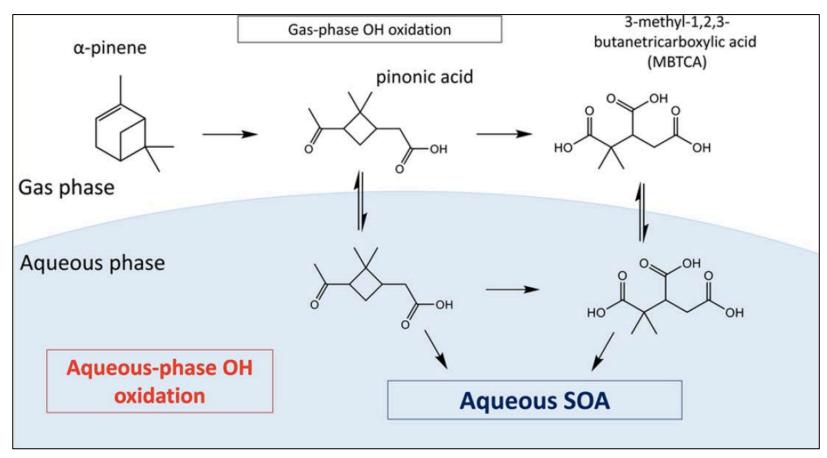


Aerosol acidity may instead be increasing, according to a thermodynamic model driven with surface observations



What about monoterpenes?

First chamber study showing aqueous-phase monoterpene SOA formation:



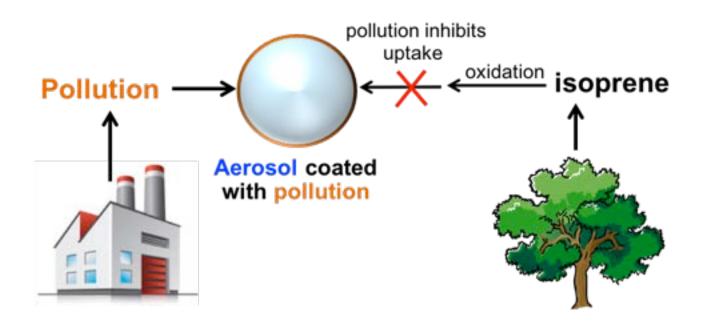
[Aljawhary et al., 2016]

Implies relationship between aqueous aerosol abundance (sulfate) and monoterpene SOA.

Concluding Remarks

- Coupled explicit isoprene SOA formation mechanism to detailed gasphase chemistry in the GEOS-Chem model
- Extensively evaluated isoprene SOA composition with surface and aricraft observations.
- Linear relationship between sulfate and isoprene SOA that in the model is due to dependence of isoprene SOA on aqueous aerosol abundance and acidity.
- This represents a dual air quality benefit of SO₂ emission controls by concurrently decreasing sulfate and isoprene SOA.
- We find support for long-term (1991-2013) decline in OA in the Southeast US in summer as due to decline in isoprene SOA driven by decline in sulfate.
- The implication is that countries in the isoprene-rich tropics will experience devastating air quality degradation if SO₂ emissions increase there.

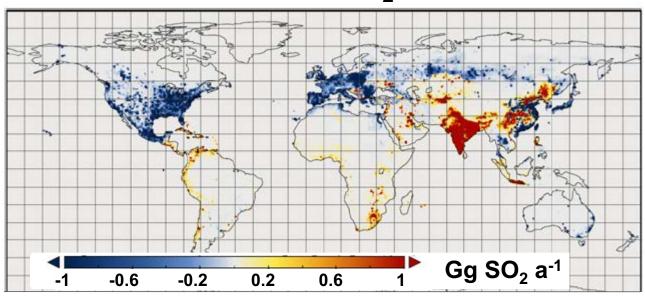
What is the global impact of complex biogenic-anthropogenic interactions constrained with process-based information from chamber experiments?



Collaborators: Jason Surratt (UNC Gillings School of Public Health, Chapel Hill) Also submitted a NERC DTP studentship application.

What is the impact of future development in the isoprene-rich tropics on local air quality?

2010 minus 2005 SO₂ emissions

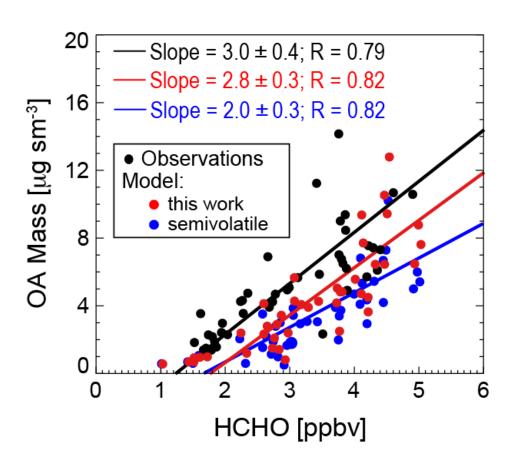


[Klimont et al., 2013]

Investigate how biogenic SOA responds to changes in SO₂ emissions

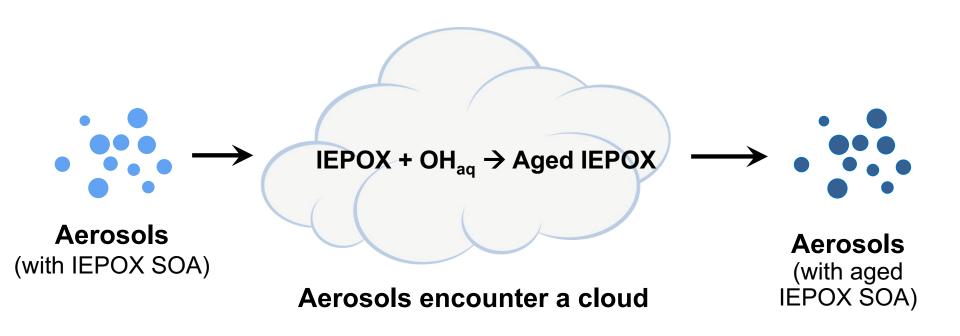
Collaborators: Paul Palmer (U. Edinburgh), Ben Langford, Eiko Nemitz (CEH)

What is the contribution of OA sources to the relationship between OA and formaldehyde?



<u>Led by:</u> NASA (Jin Liao, Thomas Hanisco)

Apply new laboratory and field measurements to the model to investigate interactions between clouds and biogenic SOA.



<u>Led by:</u> José Jimenez, Pedro Campuzano-Jost (UC Boulder).