

# Trajectory Ensemble Analysis with SHARC

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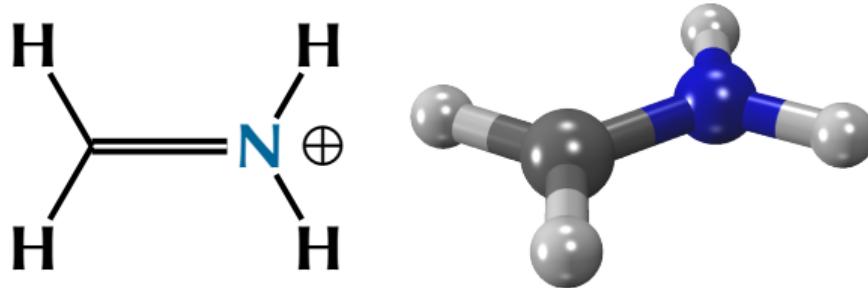
- ① Defining the chemical problem
- ② Choosing the level of theory
- ③ Preparing the initial conditions
- ④ Setting up the trajectories
- ⑤ Running the trajectories
- ⑥ Validating the trajectories
- ⑦ Analyzing individual trajectories
- ⑧ Analyzing the trajectory swarm for statistics

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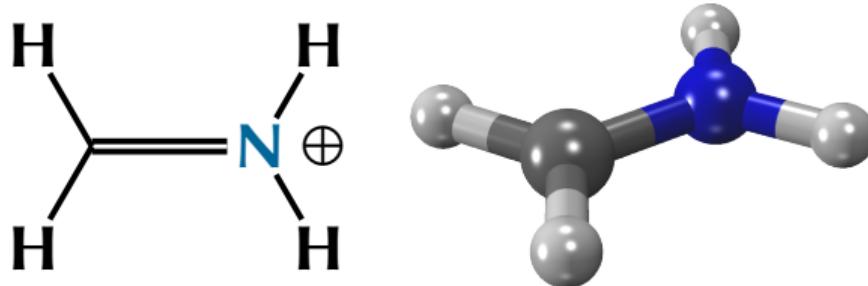
*What was it again that we were interested in about this molecule?*

# The chemical problem



- ▶ Ultrafast dynamics after photo-excitation of **methylene immonium cation**  $\text{CH}_2\text{NH}_2^+$
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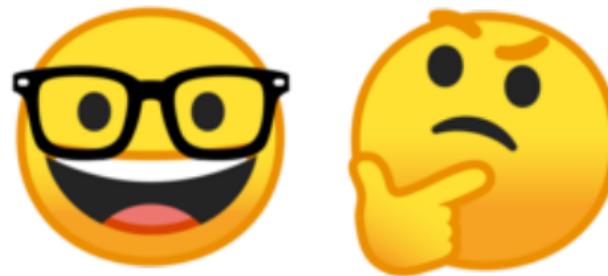
Questions:

- ▶ Involved electronic states?
- ▶ Deactivation processes?
- ▶ Time scales?
- ▶ Photochemical products and yields?



After taking days or weeks to ...

- ▶ choose and validate the level of theory, ...



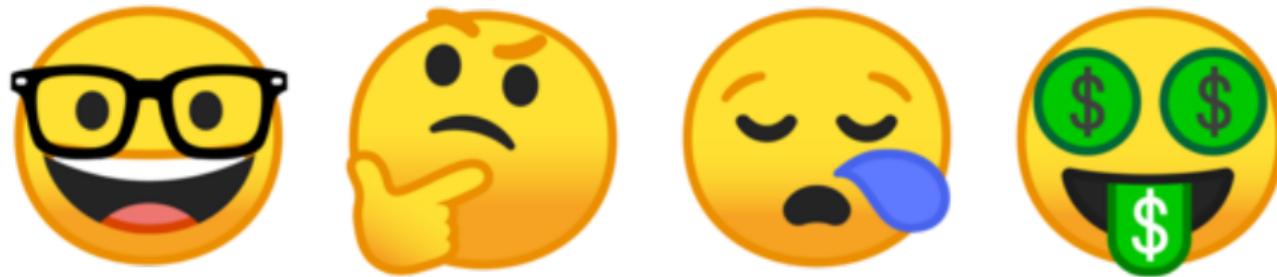
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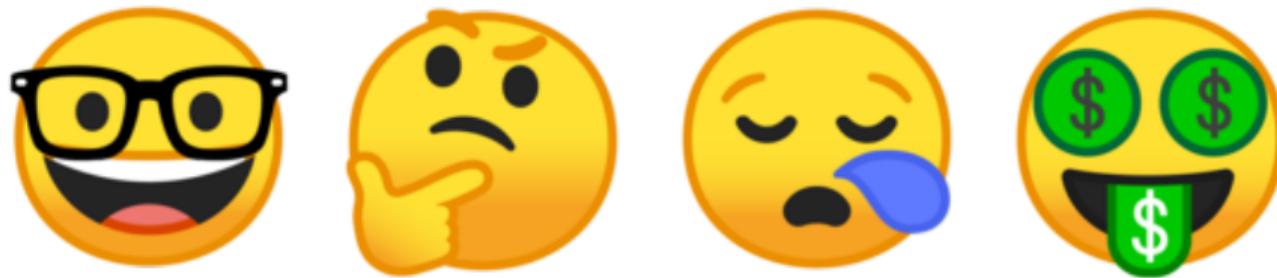
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your simulations are finished and the fun of analyzing them starts!

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*Did anything go wrong?*

# Possible error sources

Computational errors:

- ▶ Network problems
- ▶ I/O errors
- ▶ Job was killed
- ▶ **Recoverable**

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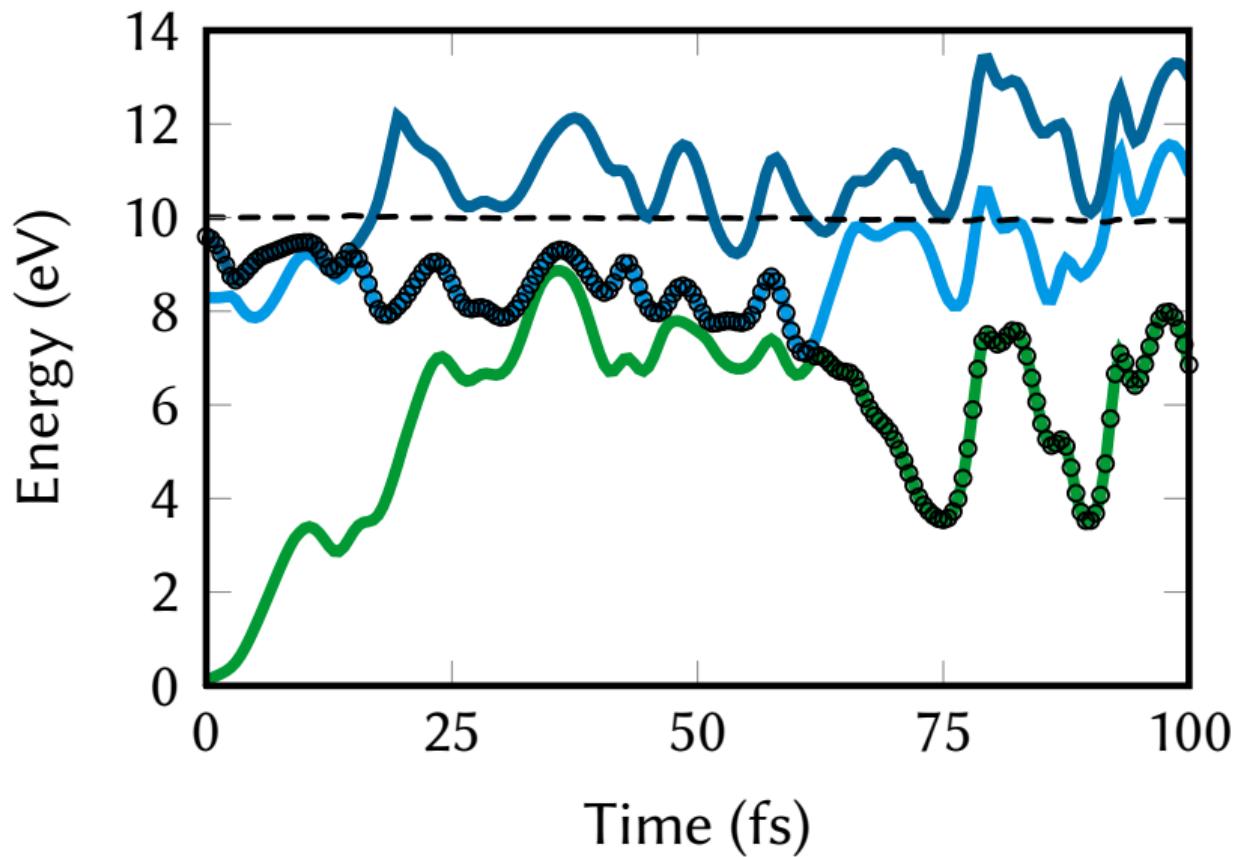
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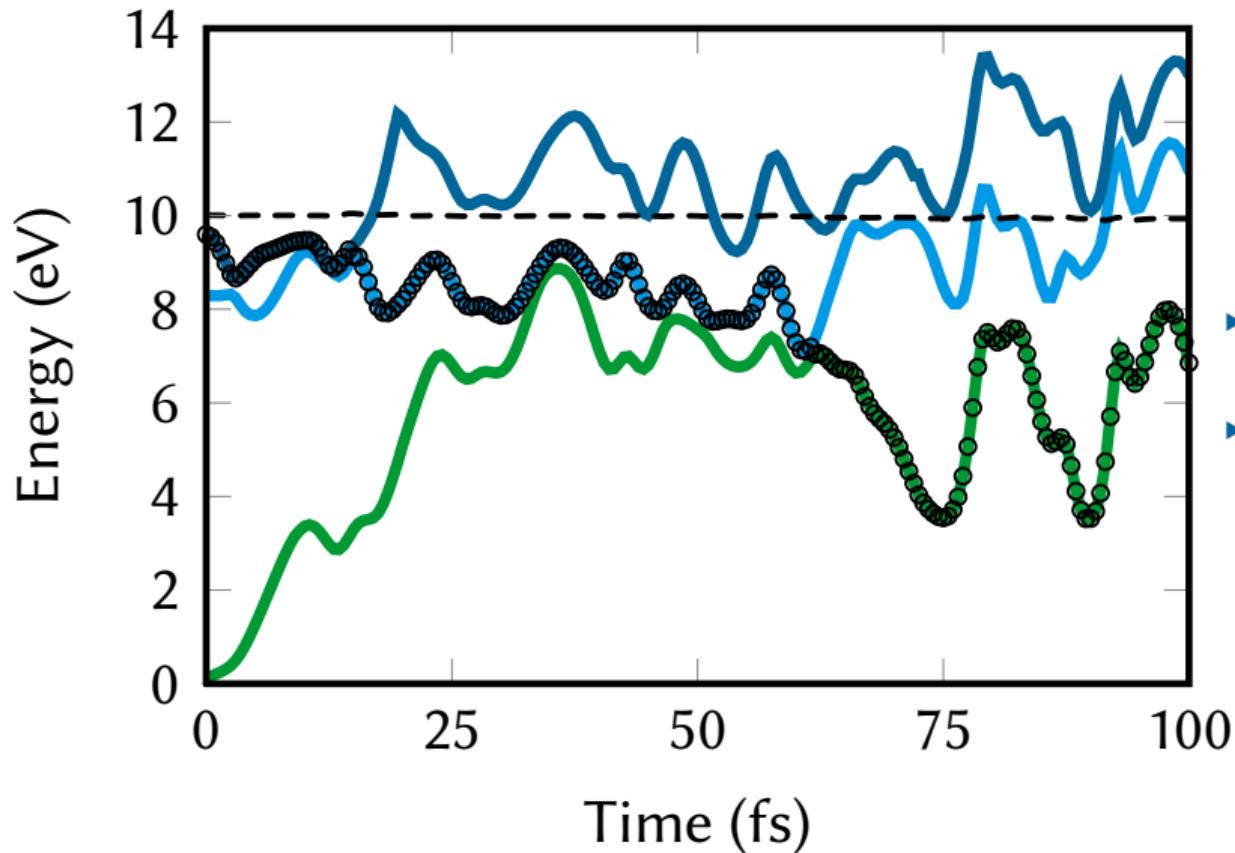
## Numerical artifacts:

- ▶ Total population not conserved
- ▶ Total energy not conserved
- ▶ Non-continuous potential energy surfaces
- ▶ Non-continuous kinetic energy
- ▶ Intruder states
- ▶ Surface hops over large energy gaps
- ▶ ...
- ▶ **Might go unnoticed!**
- ▶ **Checks required**

# Example trajectory I

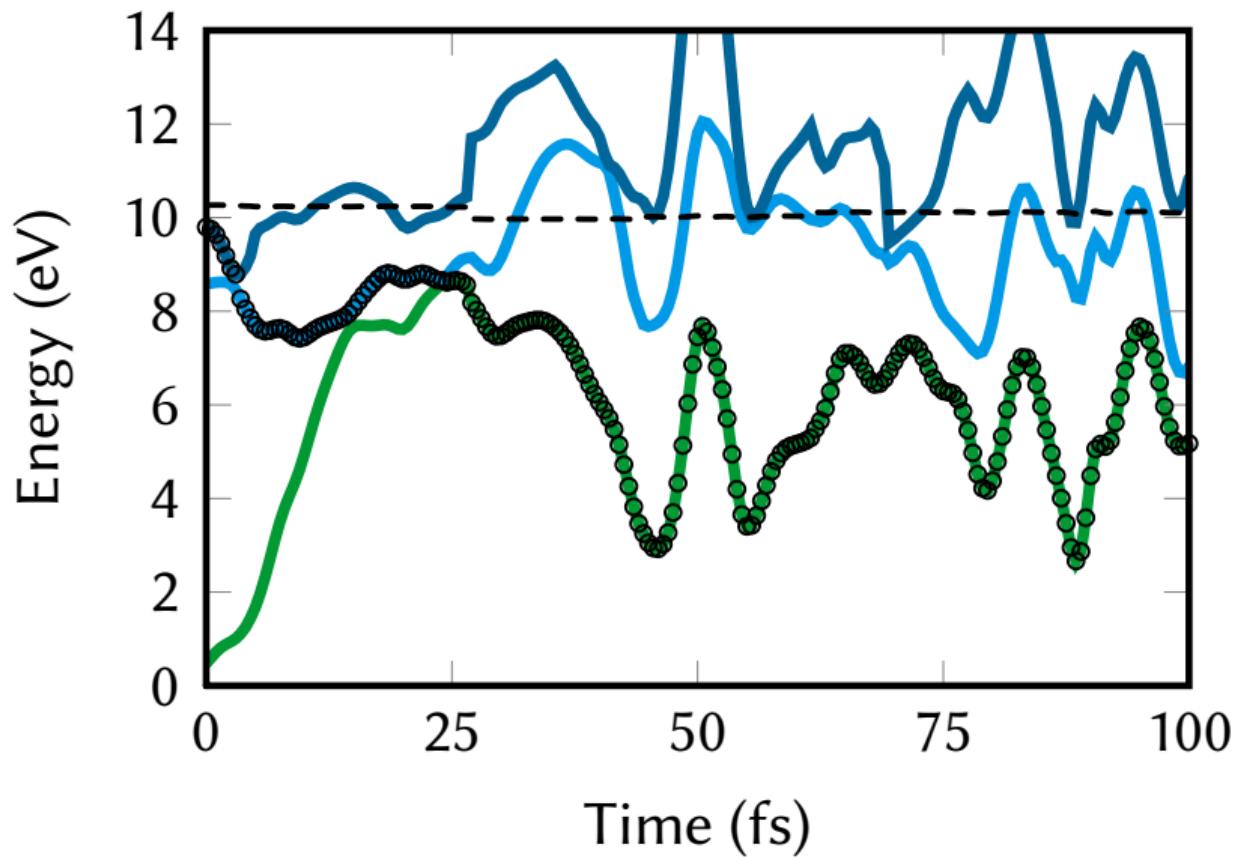


# Example trajectory I

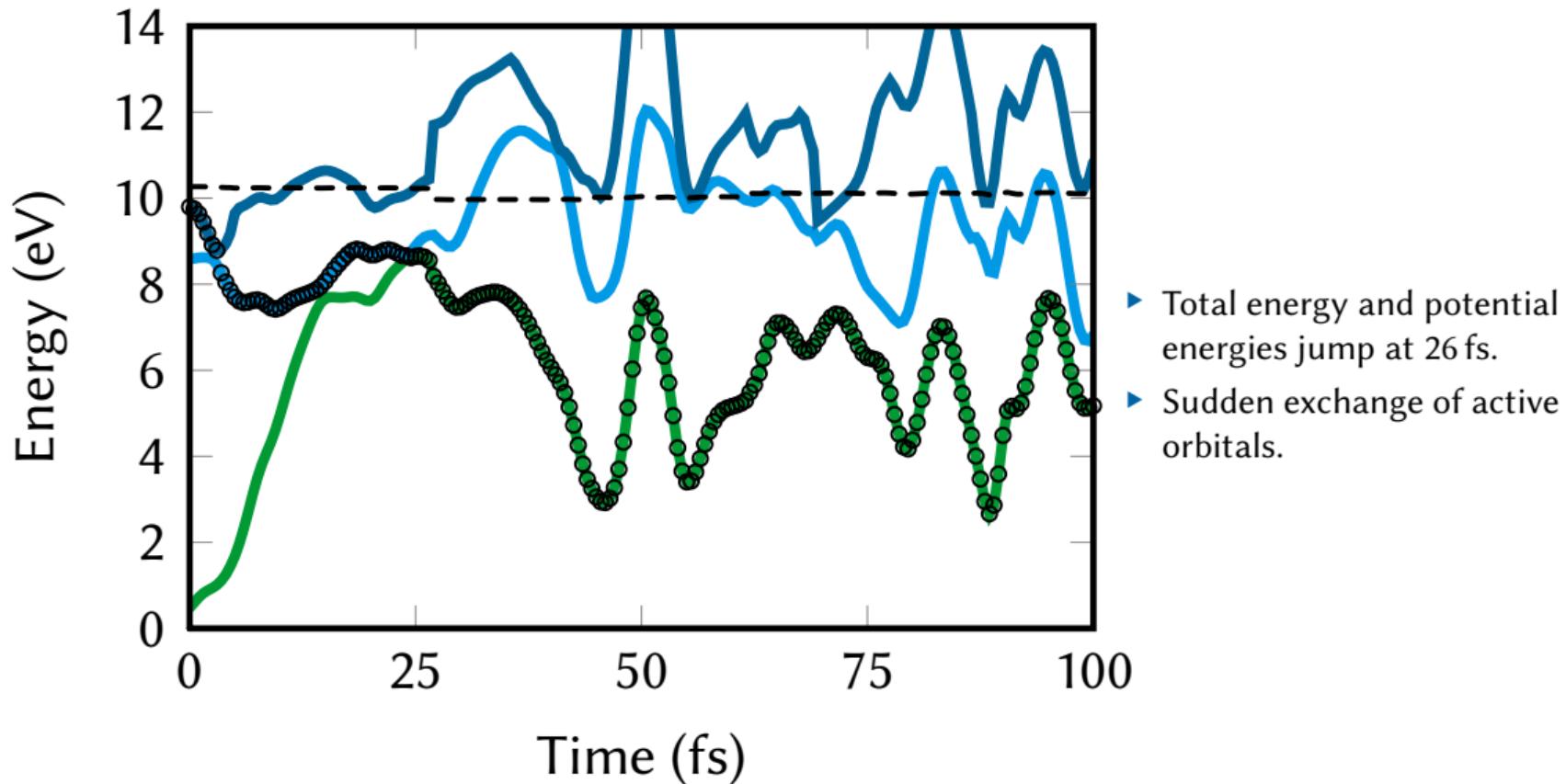


- ▶ Trajectory does not show numerical artifacts.
- ▶ Intruder state at 21 fs (but no problem).

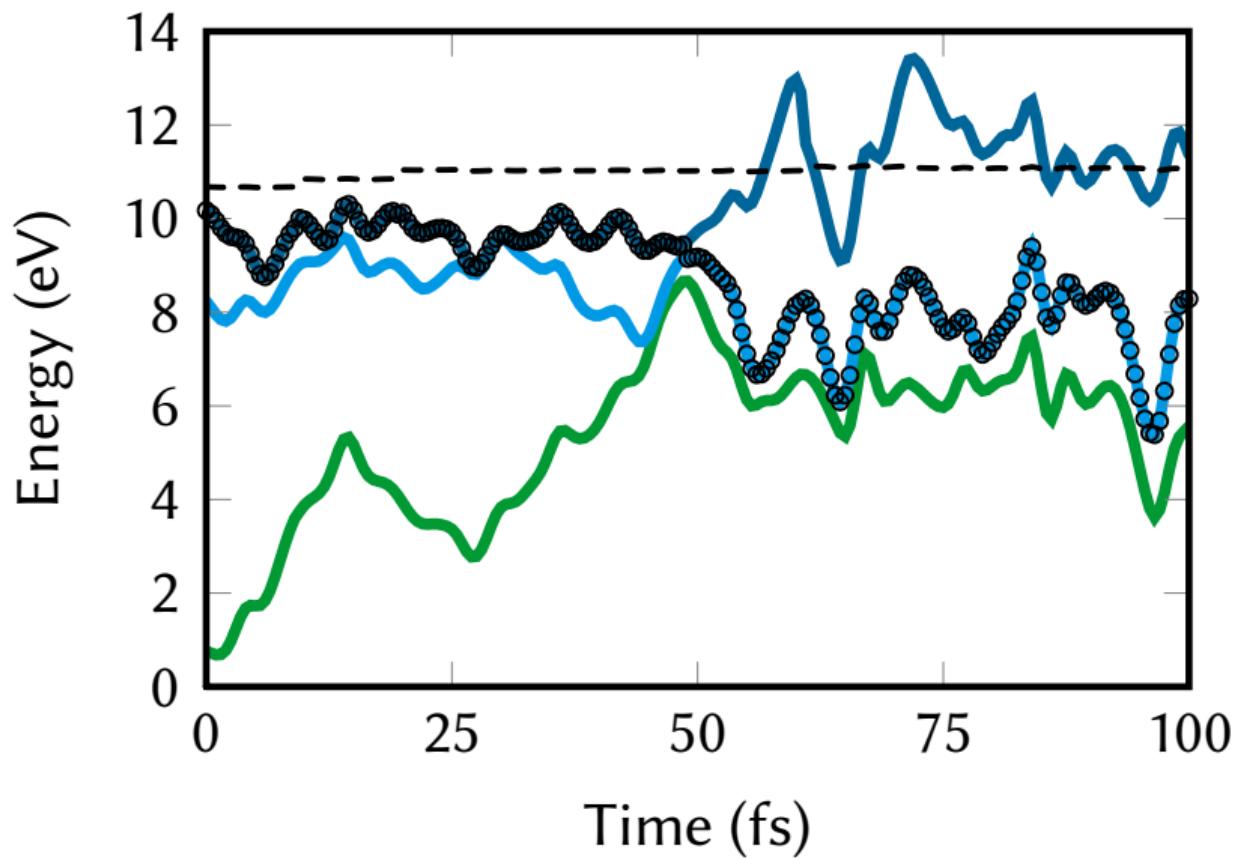
## Example trajectory II



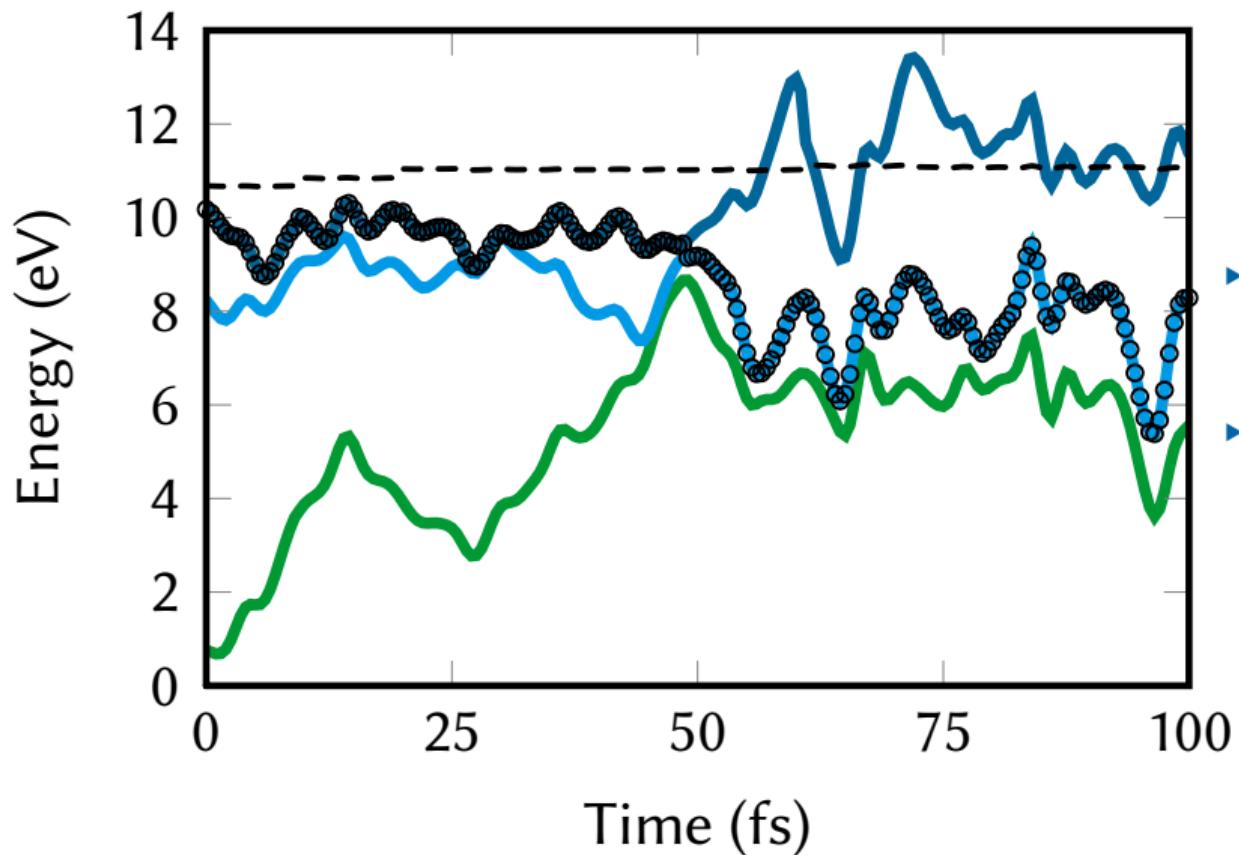
## Example trajectory II



# Example trajectory III

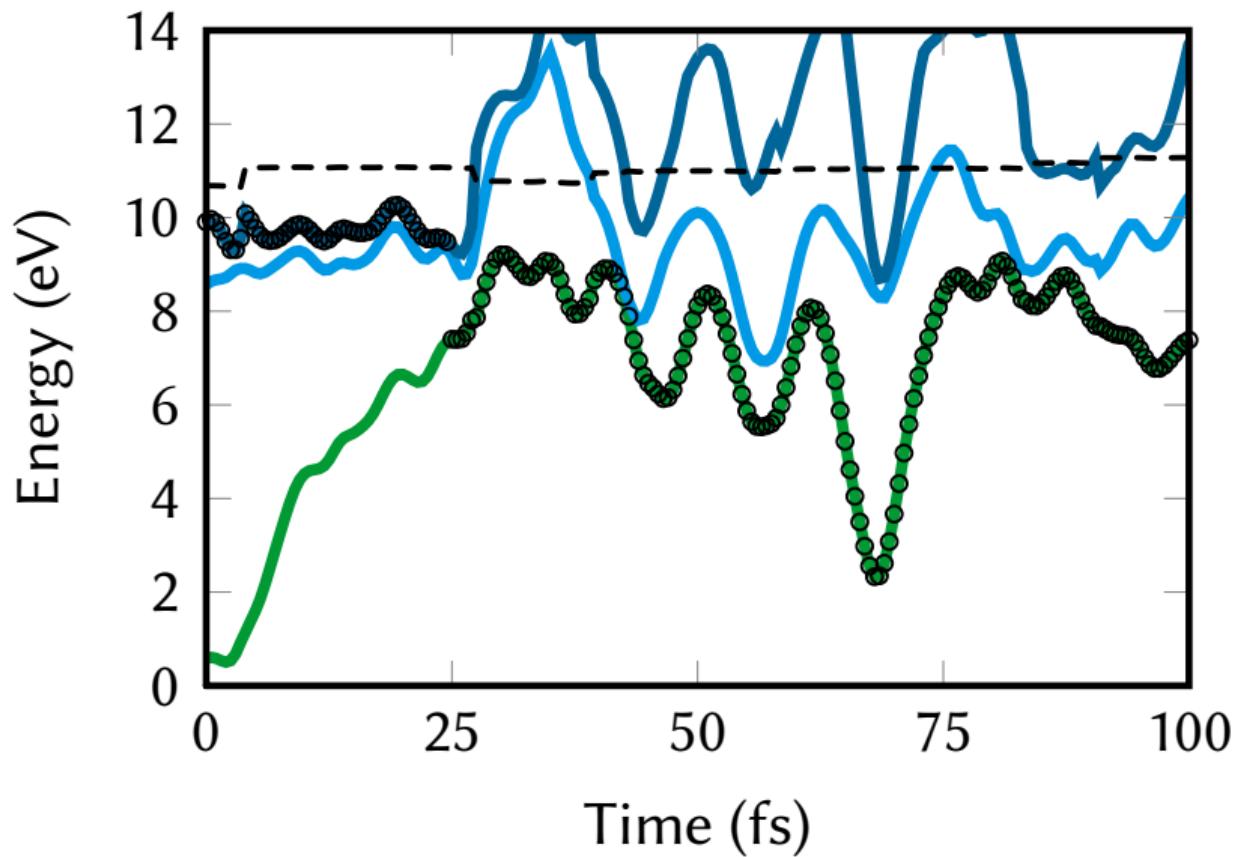


## Example trajectory III

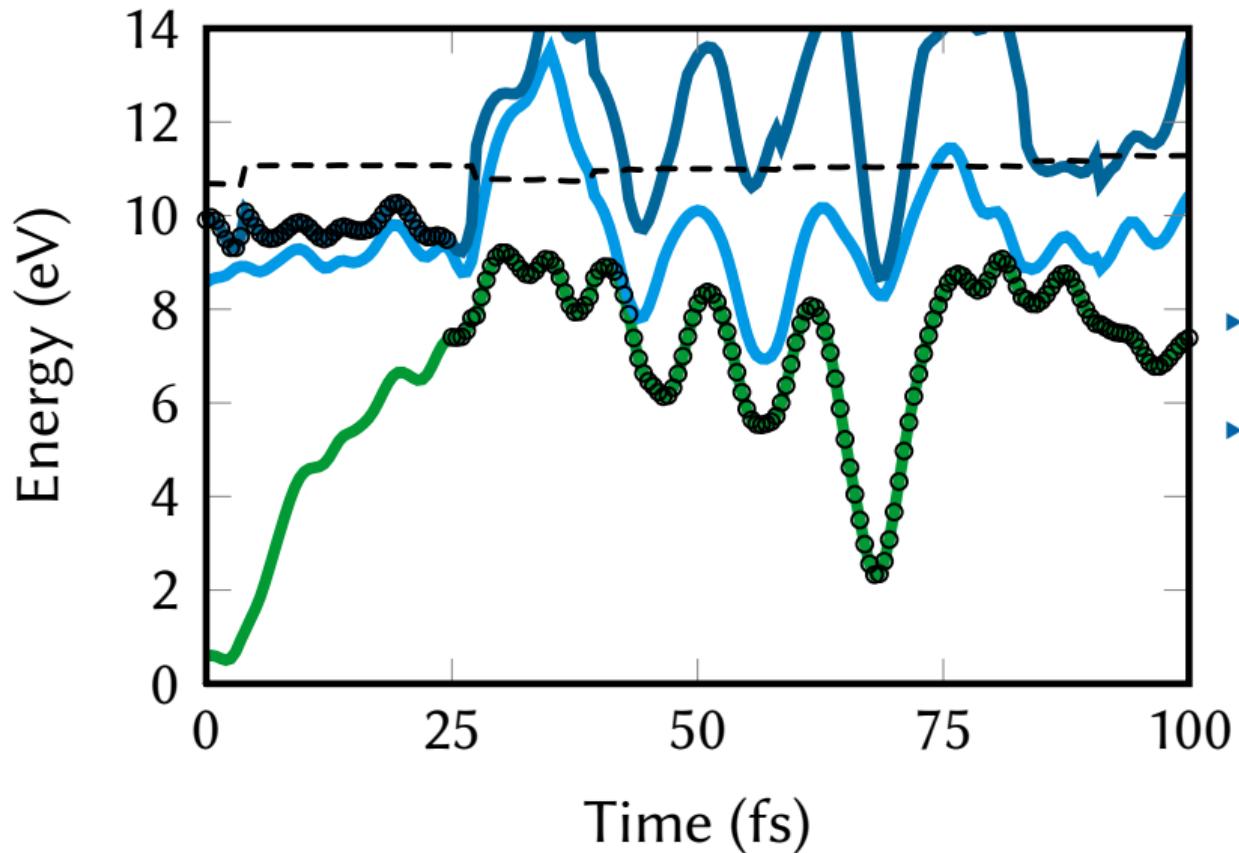


- ▶ Jumps in the total energy while potential energies are smooth.
- ▶ Badly converged gradient computation or too long time steps.

## Example trajectory IV



## Example trajectory IV



- ▶ Surface hop over large energy difference.
- ▶ Different possible reasons, needs closer analysis.

# Trajectory curation

Checking all 210 trajectories shows many problems.

- ▶ Simulations should possibly be repeated with larger active space, like CAS(12,7).
- ▶ Might need different electronic structure settings
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Result: **We keep 90 out of 210 trajectories for analysis.**

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*Isn't this a surprising behaviour?*

# Goals of the individual analysis

Important:

Analyzing the trajectories individually is **not the main way to analyze the results**:

- ▶ It is tedious.
- ▶ It is subjective.
- ▶ It is non-reproducible.
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- ▶ It might suffer from different cognitive biases.

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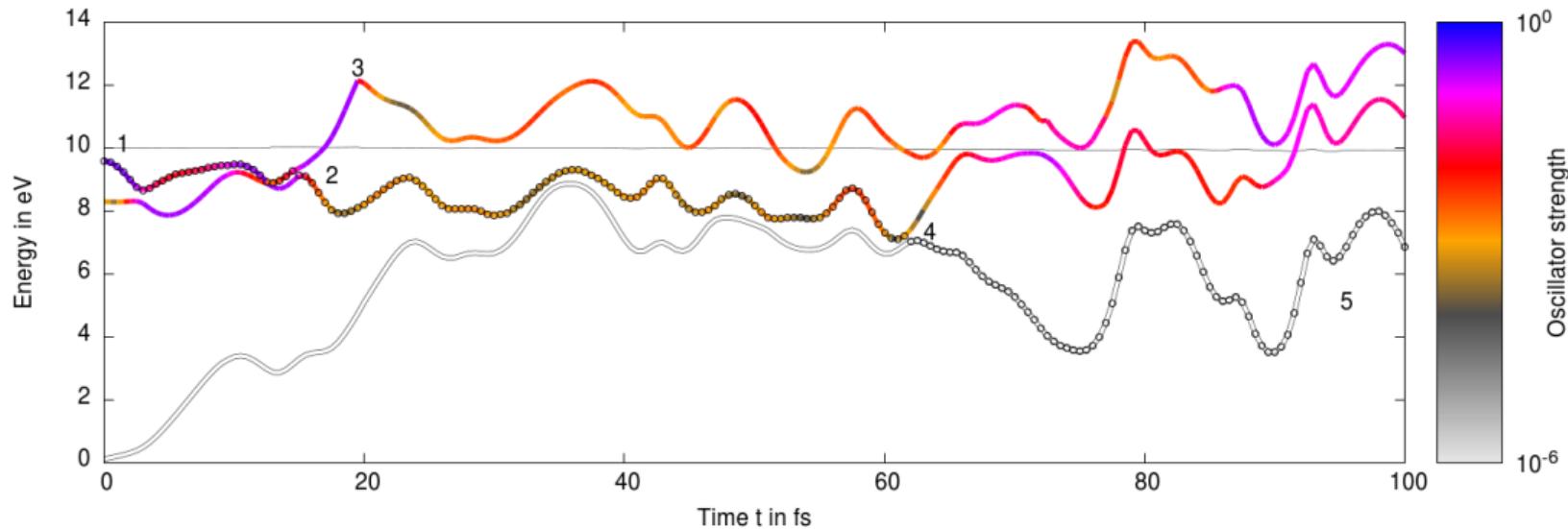
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It can still be useful:

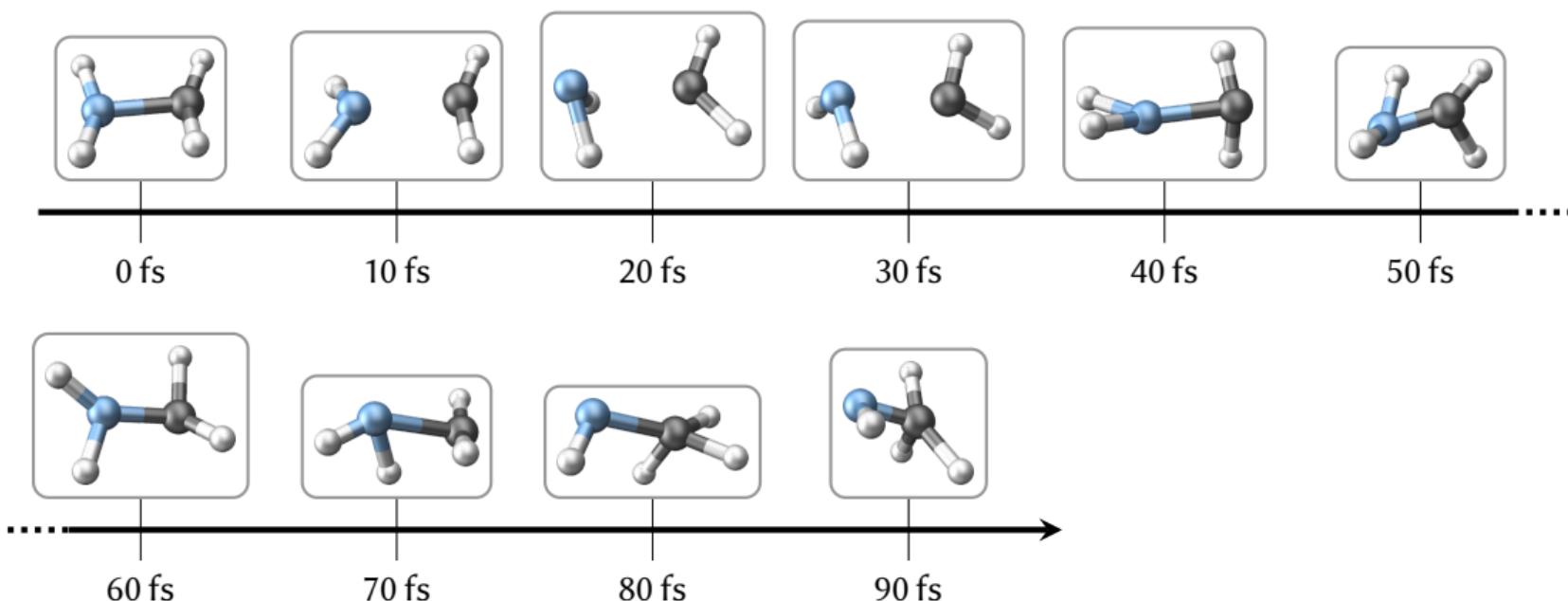
- ▶ Use pattern recognition of human brain to find interesting trends/behaviors/aspects.
- ▶ Formulate hypotheses that can then be tested.

# Example trajectory I: Energies and states



- ▶ Oscillator strength indicates state ( $\pi\pi^*$ ,  $\sigma\pi^*$ , closed shell), see coloring.
- ▶ Two surface hops bring trajectory to ground state, where strong vibrations appear.

# Example trajectory I: Nuclear motion



- ▶ Strong C=N stretch, pyramidalization, torsion, hydrogen migration

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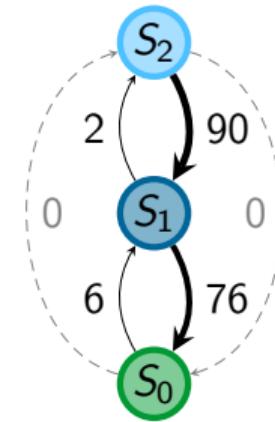
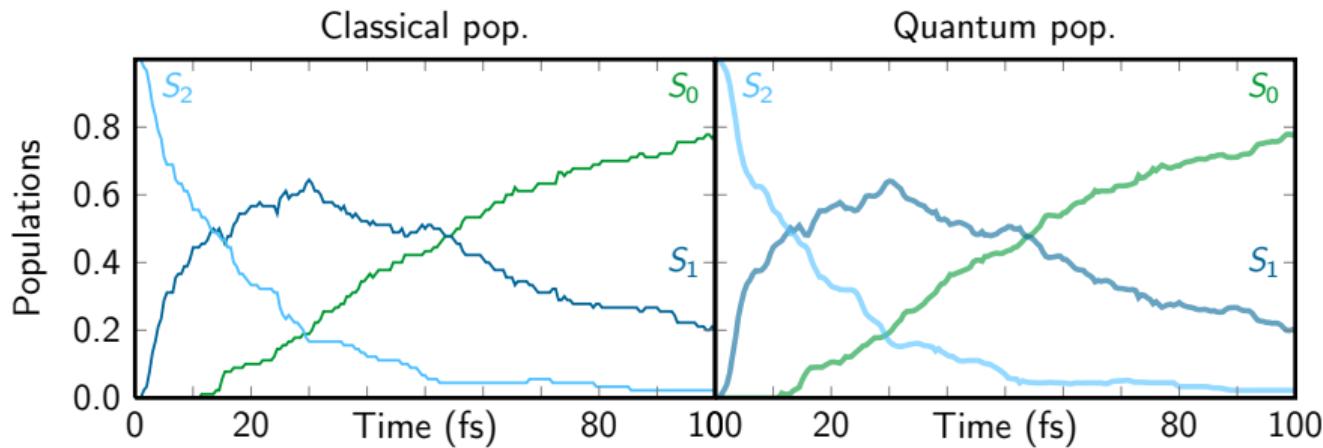
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- ④ **Product ratios:** Most trajectories do not undergo migration, elimination, or dissociation.

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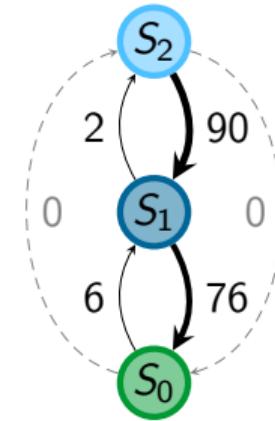
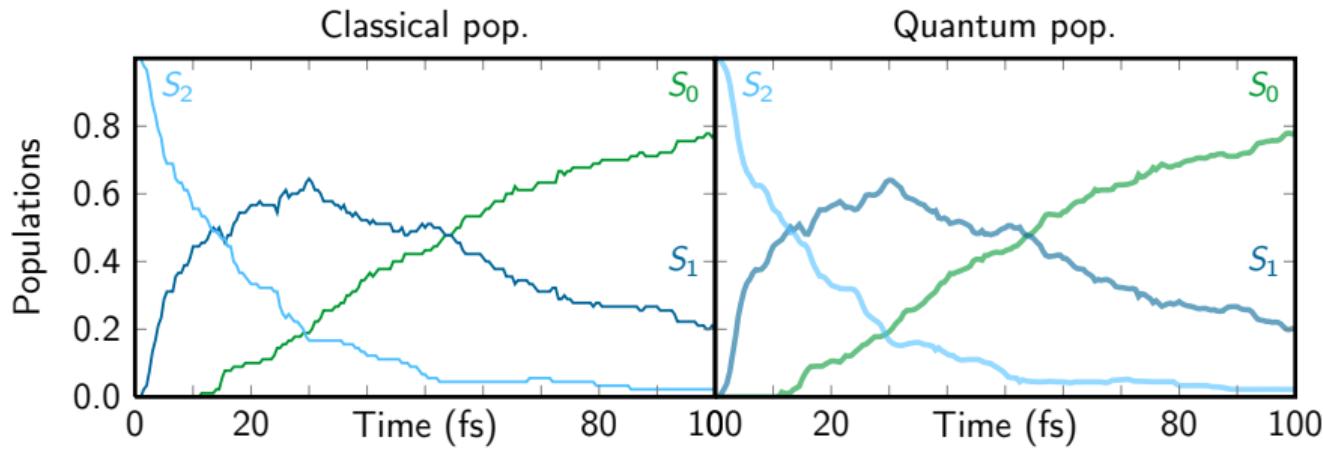


*How can we make sense of all this data?*

# Electronic evolution

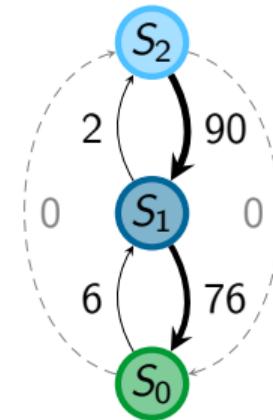
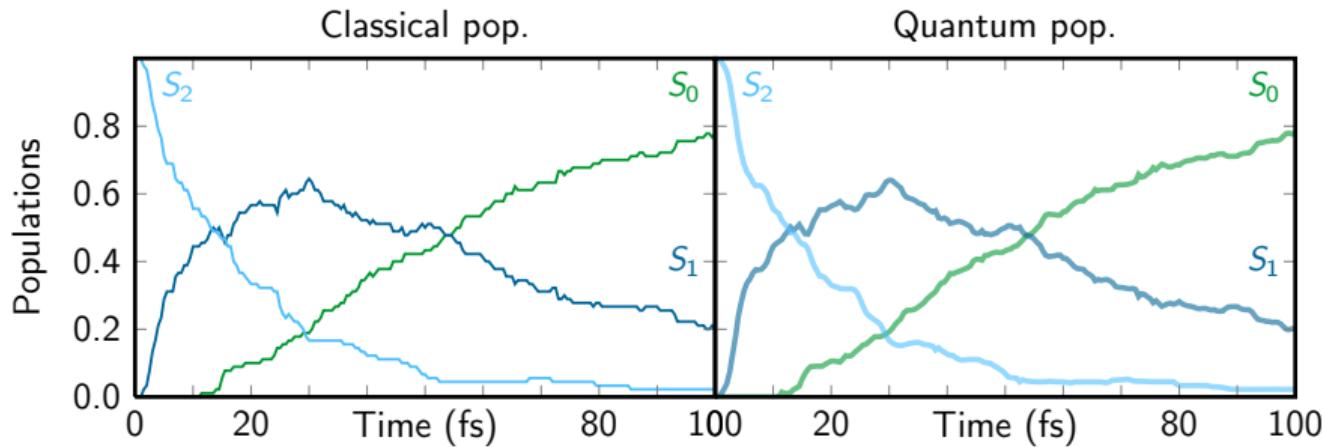


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- ▶ Sequential transfer (verified with hopping analysis)

# Electronic evolution: Kinetic modeling

Sequential, uni-molecular, first-order kinetic model:



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Sequential, uni-molecular, first-order kinetic model:



Corresponding differential equation system:

$$\frac{d}{dt} S_2(t) = -k_{21} S_2(t),$$

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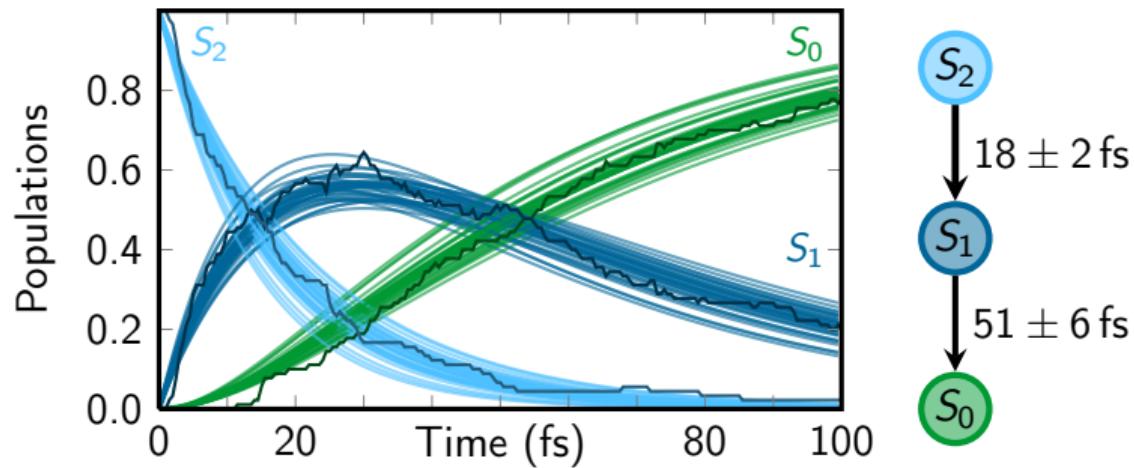
Solutions:

$$\begin{aligned}S_2(t) &= e^{-k_{21}t}, \\ S_1(t) &= -\frac{k_{21}}{k_{21} - k_{10}}e^{-k_{21}t} + \frac{k_{21}}{k_{21} - k_{10}}e^{-k_{10}t}, \\ S_0(t) &= +\frac{k_{10}}{k_{21} - k_{10}}e^{-k_{21}t} - \frac{k_{21}}{k_{21} - k_{10}}e^{-k_{10}t} + 1.\end{aligned}$$

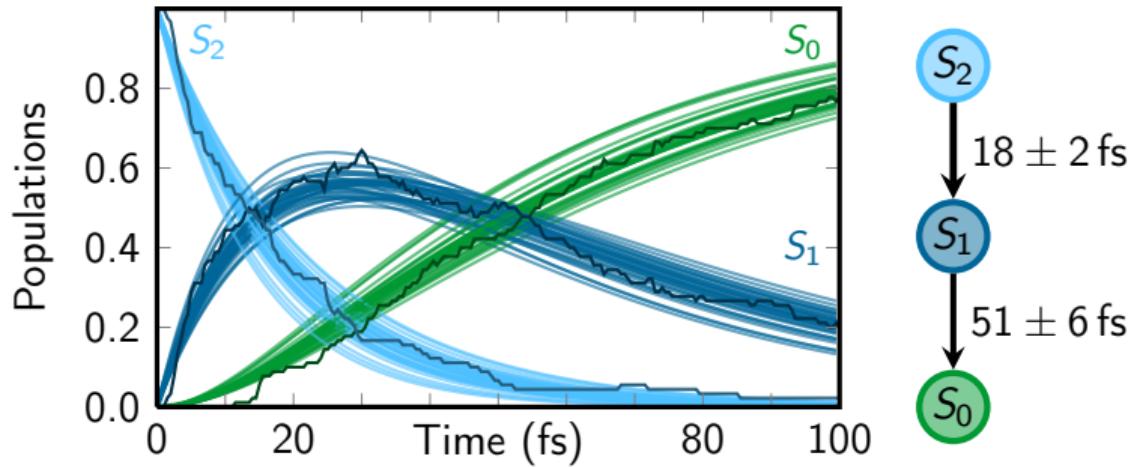
- ▶ Fitting functions
- ▶ Fitting parameters:

$$\tau_{S_2 \rightarrow S_1} = \frac{1}{k_{21}} \text{ and } \tau_{S_1 \rightarrow S_0} = \frac{1}{k_{10}}$$

# Electronic evolution: Time constants

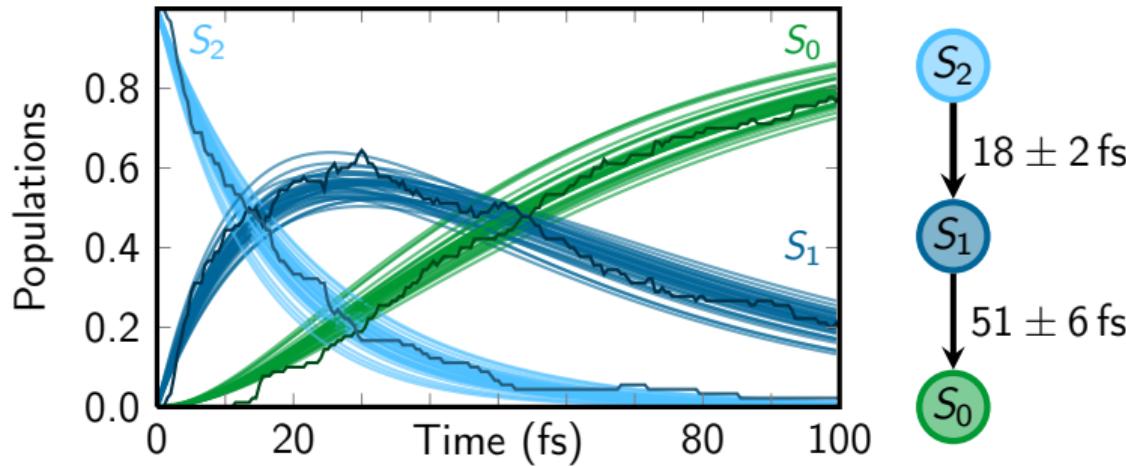


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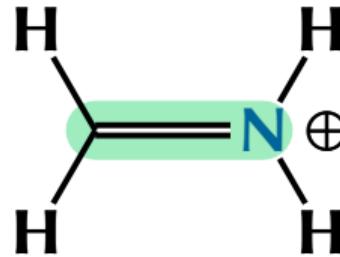
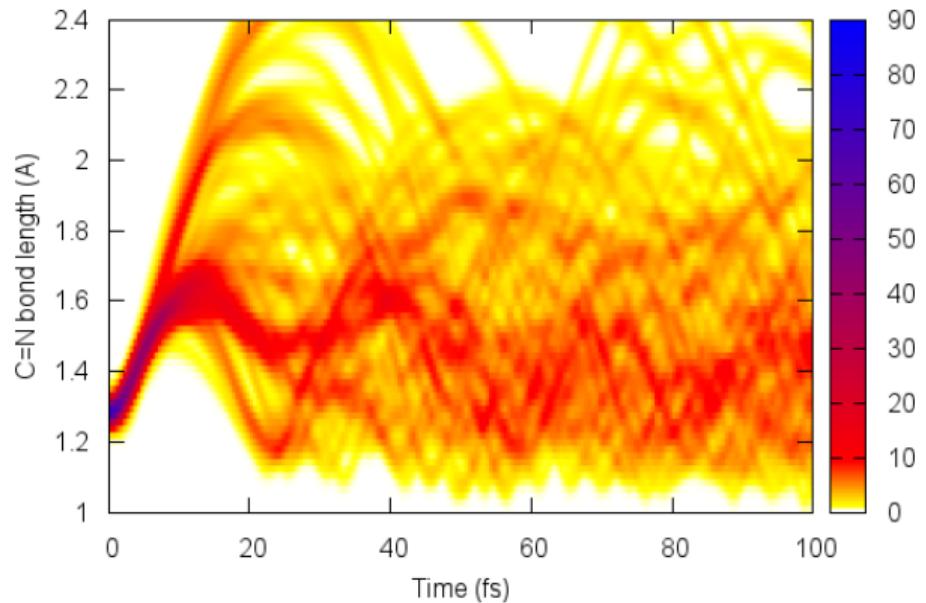
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- ▶ Can compare time scales to experiment

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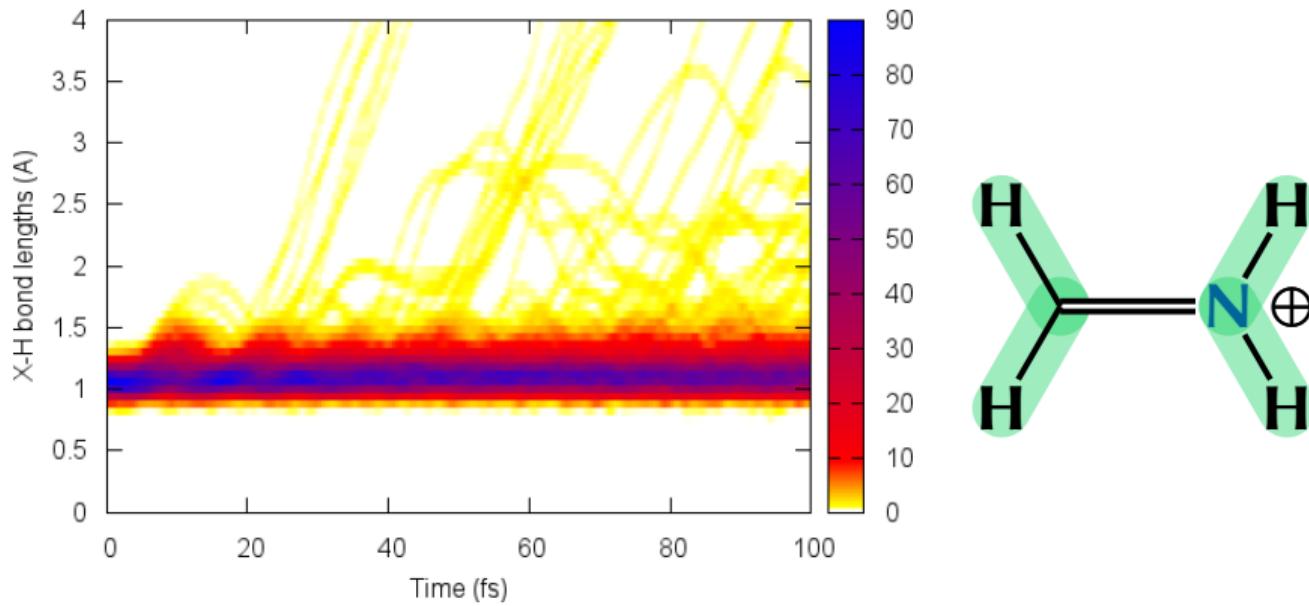
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# Nuclear evolution: C=N bond



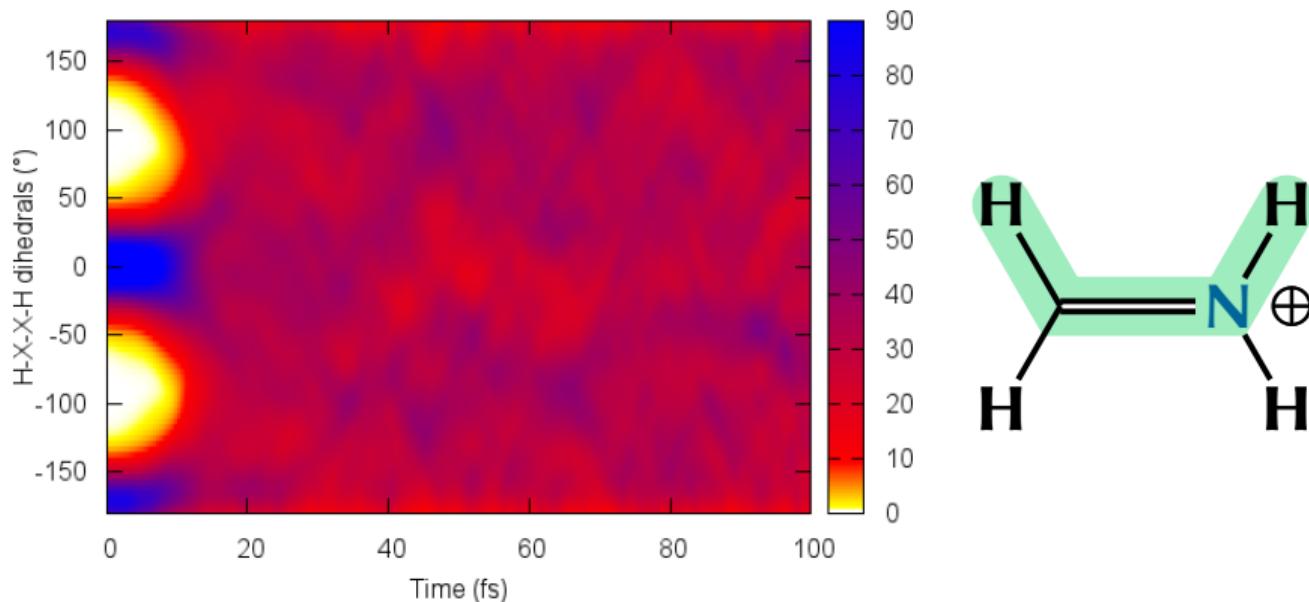
- ▶ Strong increase in bond length
- ▶ Some coherent motion
- ▶ Splitting of trajectory swarm

# Nuclear evolution: C–H and N–H bonds



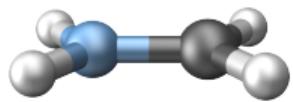
- ▶ Most bonds are stable
- ▶ Some dissociation
- ▶ Some migration

# Nuclear evolution: Dihedral angles

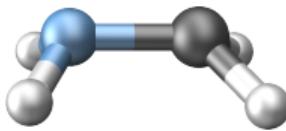


- ▶ Initially planar
- ▶ Double bond is broken in excited state, free rotation around bond

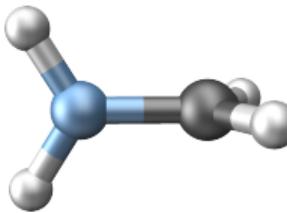
# Nuclear evolution: Hopping geometries and conical intersections



$S_0$  minimum



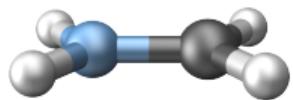
$S_2/S_1$  minimum-energy Coln



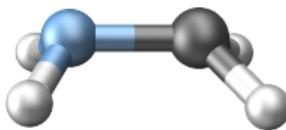
$S_1/S_0$  minimum-energy Coln

- ▶ Extract hopping geometries: find what motion leads to  $S_2/S_1$  and  $S_1/S_0$  crossing points.

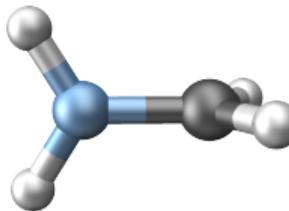
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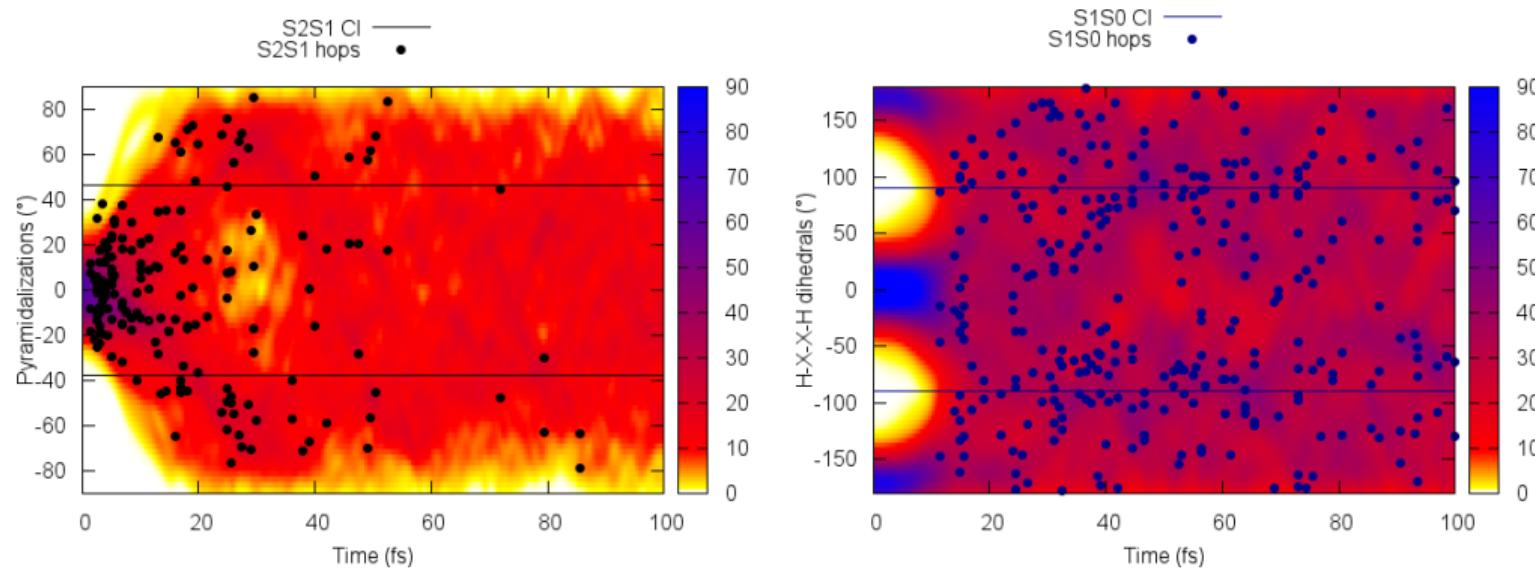
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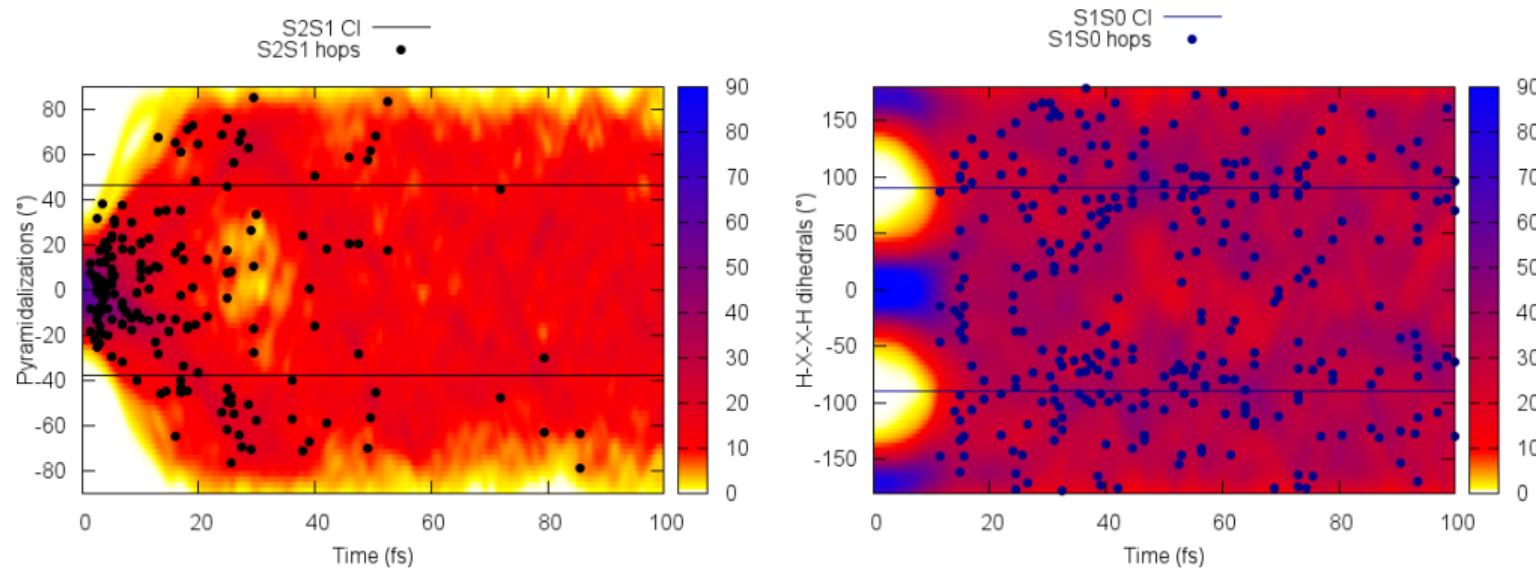
- ▶ Extract hopping geometries: find what motion leads to  $S_2/S_1$  and  $S_1/S_0$  crossing points.
- ▶  $S_2/S_1$  transfer is mediated by pyramidalization and C=N stretch
- ▶  $S_1/S_0$  transfer is mediated by torsion

# Nuclear evolution: Importance of minimum crossing points



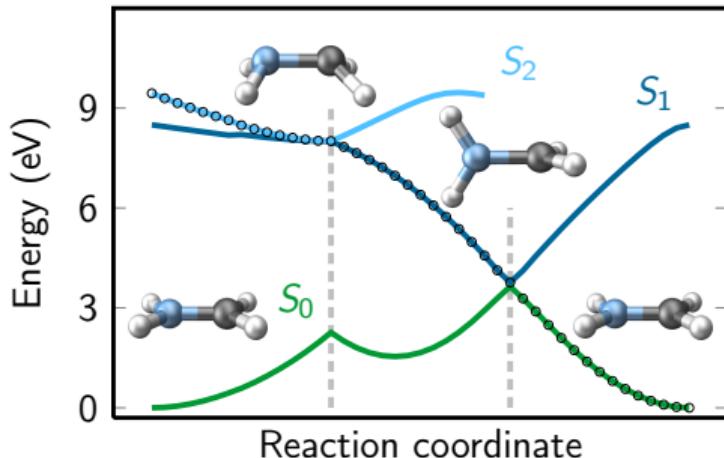
- ▶ In this molecule, there is a 10-dimensional intersection space.
- ▶ Optimized conical intersections are only one point in this space.

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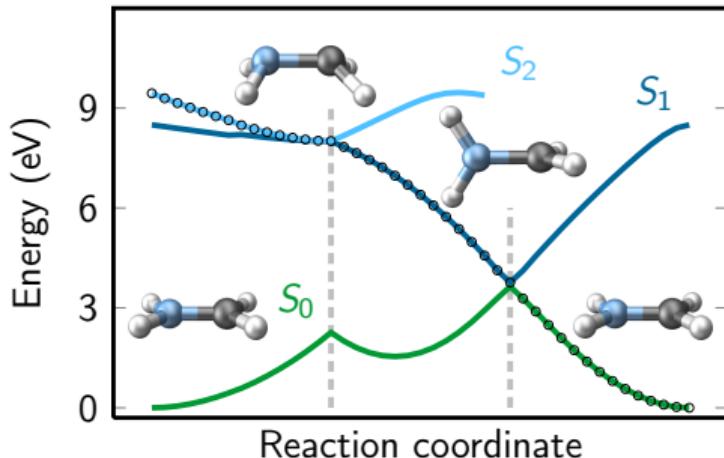
- ▶ In this molecule, there is a 10-dimensional intersection space.
- ▶ Optimized conical intersections are only one point in this space.
- ▶ Trajectories hop at many different geometries distributed around the conical intersection.

# Nuclear evolution: Potential energy surfaces



- ▶ Path from starting point to conical intersections to end point.
- ▶ Path is barrierless.

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② **Relaxation mechanism:** Important motion is a combination of C=N stretch, pyramidalization, and torsional motion. The conical intersections are easy to reach (no barriers). **CONFIRMED**

# Photochemistry products

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Different products:

- ▶ H migration to  $\text{CH}_3\text{-NH}^+$
- ▶ H migration to  $\text{CH-NH}_3^+$
- ▶  $\text{H}_2$  elimination
- ▶ Dissociation to  $\text{CH}_2$  and  $\text{NH}_2$  fragments
- ▶ Any others?

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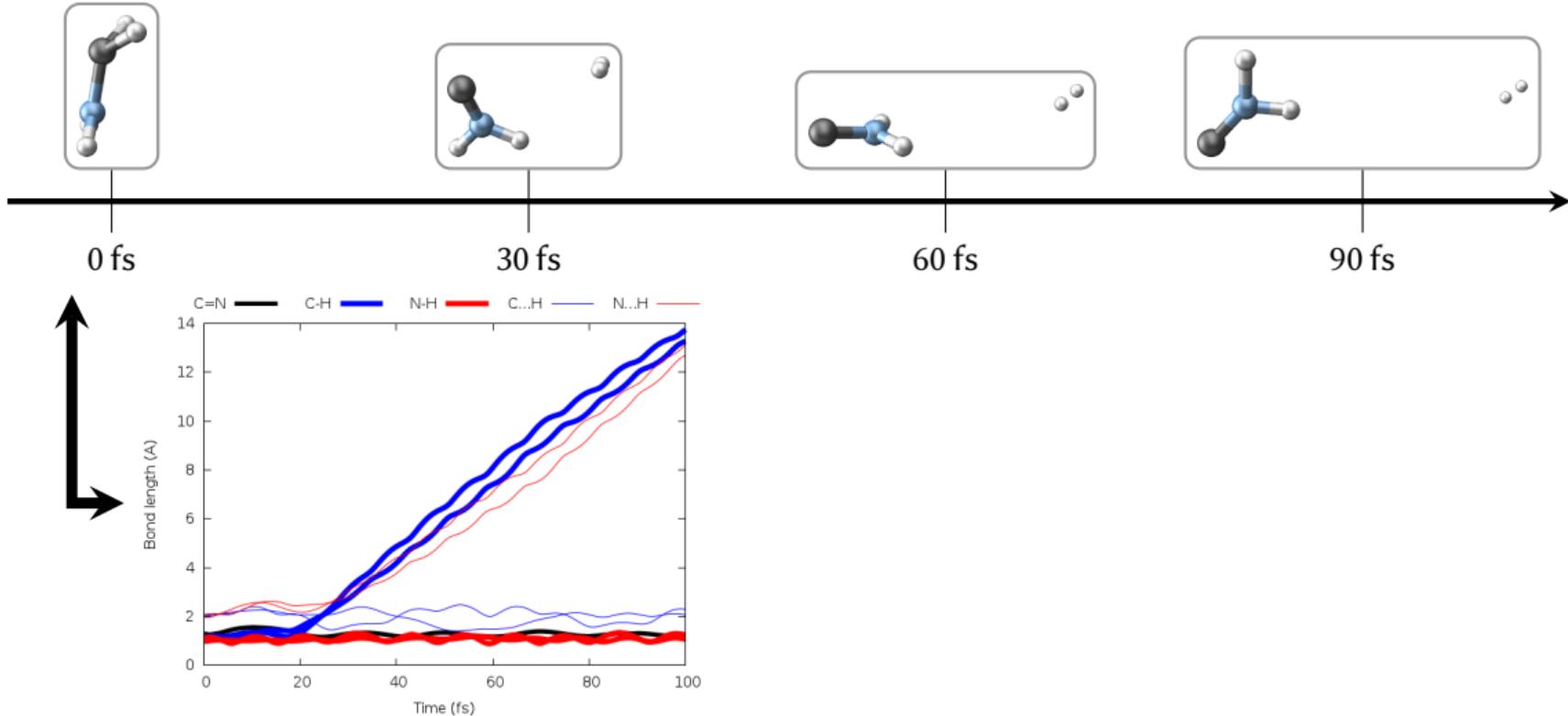
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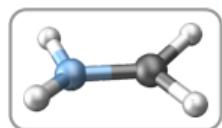
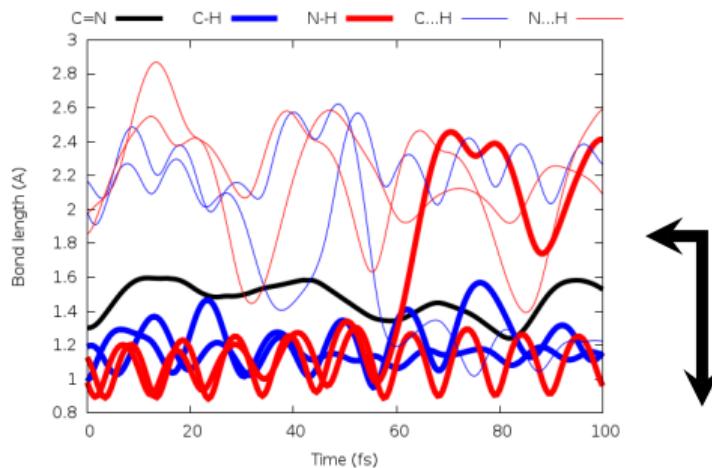
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**Automatically identify products:** through geometry parameters.

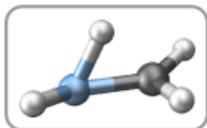
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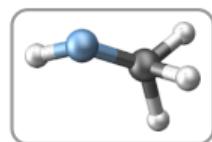
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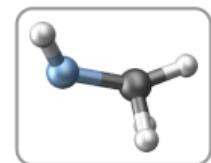
0 fs



30 fs



60 fs



90 fs

# Photochemistry products: Branching ratios

Reaction pathway	Trajectories	Percentage
Unreactive	65	72%
H <sub>2</sub> elimination to C=N <sub>2</sub> <sup>+</sup>	10	11%
H migration to CH <sub>3</sub> -NH <sup>+</sup>	7	8%
H migration to CH-NH <sub>3</sub> <sup>+</sup>	5	6%
C=N dissociation	3	3%

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# Photochemistry products: Branching ratios

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- ③ **Photochemical products:** Possible rearrangements: H migration, H<sub>2</sub> elimination, or C=N dissociation. **CONFIRMED**
- ④ **Product ratios:** Most trajectories do not undergo migration, elimination, or dissociation. **CONFIRMED**

# Simulating time-dependent spectra

Most nonadiabatic dynamics is experimentally measured with time-dependent spectroscopy. Examples:

- ▶ Infrared spectroscopy
- ▶ Transient absorption
- ▶ Photoionization
- ▶ X-ray scattering
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How to compute time-dependent photoluminescence spectrum:

$$S(E, t) = \frac{1}{N_{\text{traj}}} \sum_j^{\text{traj states}} \sum_{\beta} (f_{\text{osc}})_{\alpha \rightarrow \beta}^j(t) \cdot e^{-\frac{4 \ln 2}{\text{FWHM}_E^2} \left( E - \Delta E_{\alpha \rightarrow \beta}^j(t) \right)^2}. \quad (2)$$

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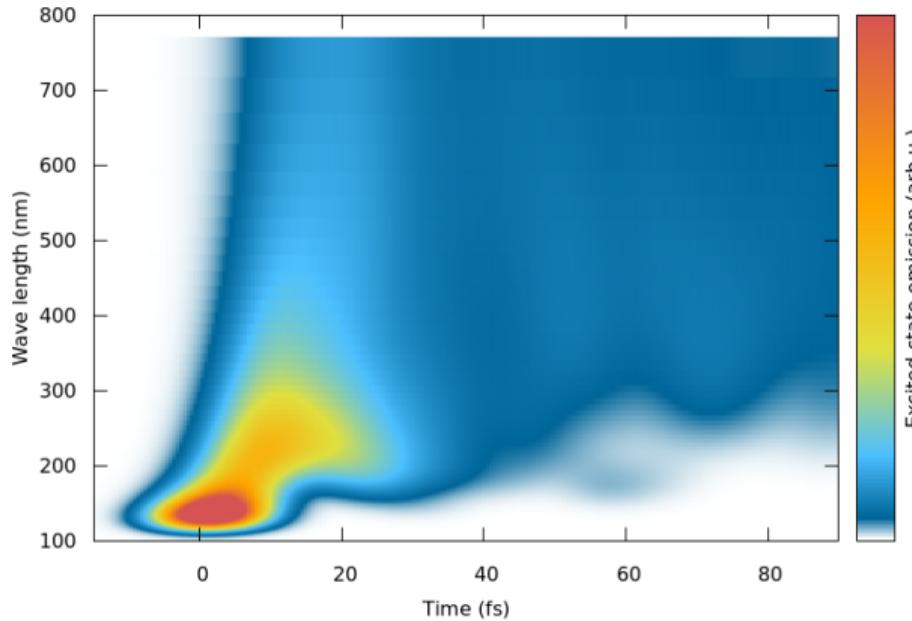
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With broadening through an instrument response function:

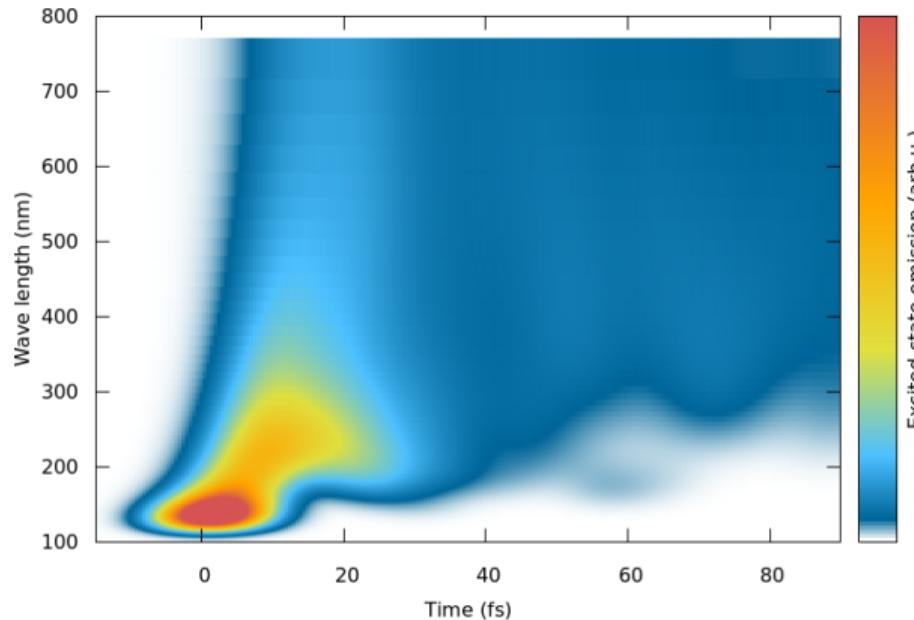
$$S^{\text{conv}}(E, t) = \sum_i^{\text{time steps}} S(E, \Delta t_i) \cdot e^{-\frac{4 \ln 2}{\text{FWHM}_t^2} (t - \Delta t_i)^2}. \quad (3)$$

# Simulating time-dependent spectra: Results



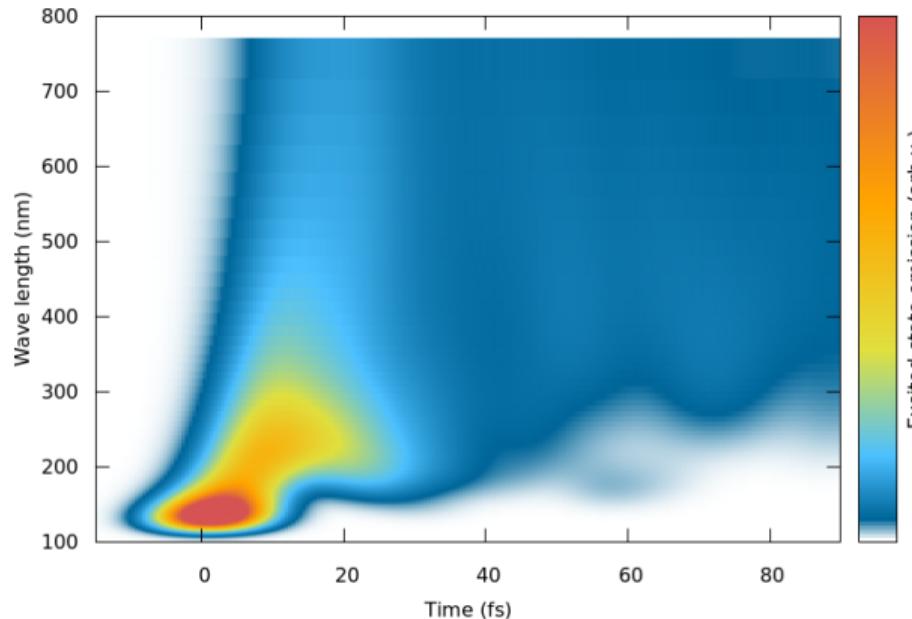
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- ▶ Quick decrease in luminescence energy: decay to ground state
- ▶ Splitting of swarm can be discerned
- ▶ Can be compared to suitable experiment

## Chemical problem

Simulate the photo-induced nonadiabatic dynamics of the methylene immonium cation  $\text{CH}_2\text{NH}_2^+$ .

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Surface hopping coupled to multi-reference configuration interaction (MRCI) with a CAS(6,4) active space.

# Summary I

## Chemical problem

Simulate the photo-induced nonadiabatic dynamics of the methylene immonium cation  $\text{CH}_2\text{NH}_2^+$ .

## Methods

Surface hopping coupled to multi-reference configuration interaction (MRCI) with a CAS(6,4) active space.

## Initial conditions

Wigner distribution around  $S_0$  minimum, excited vertically at 9.29–9.59 eV to the  $S_2$  ( $\pi\pi^*$ ) state.

## Settings

210 trajectories were propagated for 100 fs with a 0.5 fs time step. Typical settings for surface hopping were used (decoherence, kinetic energy rescaling, ...).

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## Execution

A single time step took about 5 min. Therefore, 200 steps took 19 hours and 210 trajectories then cost 4000 CPU hours.

# Summary II

## Settings

210 trajectories were propagated for 100 fs with a 0.5 fs time step. Typical settings for surface hopping were used (decoherence, kinetic energy rescaling, ...).

## Execution

A single time step took about 5 min. Therefore, 200 steps took 19 hours and 210 trajectories then cost 4000 CPU hours.

## Validation

Many trajectories had numerical artifacts so that the swarm was reduced to 90 suitable trajectories.

## Individual analysis

Four hypotheses:

- ① Fast, sequential  $S_2 \rightarrow S_1 \rightarrow S_0$  decay.
- ② Important motion C=N stretch, pyramidalization, and torsion.
- ③ We can have H migration, H<sub>2</sub> elimination, or other processes.
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## Statistical analysis

- ①  $S_2 \rightarrow S_1$  decays with  $18 \pm 2$  fs and  $S_1 \rightarrow S_0$  with  $51 \pm 6$  fs without barriers.
- ② Conical intersections involve C=N stretch, pyramidalization, and torsion.
- ③ We find H<sub>2</sub> elimination (11%), H migration (8%+6%), and C=N dissociation (3%).
- ④ Most trajectories do nothing (72%).

# Thank you for your attention!

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My further thanks goes to:



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FWF