

$$P_{1,2} A_1$$

$$P_{\text{gauge}} = P_{\text{abs}} - P_{\text{atm}}$$

use P_{gauge}

$$\int d\vec{r} = \int \vec{r} dt$$

$$\int d\vec{v} = \int \vec{a} dt$$

$$\vec{r}(t) - \vec{r}(t_0) = \int_{t_0}^t \vec{a}(t) dt$$

$$\vec{r} = \frac{\sum \vec{r}_i m_i}{\sum m_i}$$

$$= \int \vec{r} \rho dV$$

$$I = \int \rho r^2 dV, K_E = \frac{1}{2} I \omega^2$$

$$E_{\text{sp}} = \frac{1}{2} k (x - x_0)^2$$

$$E_{\text{cap}} = \frac{1}{2} C V^2, E_{\text{ind}} = \frac{1}{2} L I^2$$

$$x_a x_b = x_0$$

$$m \rightarrow$$

$$W_{\text{sp}} = \frac{1}{2} k [(x_b - x_0)^2 - (x_a - x_0)^2]$$

$$W = \int \vec{F} \cdot d\vec{s}, \dot{W} = \vec{F} \cdot \vec{v}$$

$$\dot{W}_{\text{flow}} = \dot{m}(p v) \rightarrow \text{transport term}$$

$$\text{Fourier's Law Conduction: } \dot{Q} = -KA \frac{dT}{dx}$$

$$\text{Newton's Cooling (convection): } \dot{Q} = hA(T_{\text{surf}} - T_{\infty})$$

$$\text{Stefan Boltzmann (radiat.): } \dot{Q} = \epsilon \sigma A_s (T_{\text{surf}}^4 - T_{\text{amb}}^4)$$

$$\sigma = 5.67 \cdot 10^{-8} \frac{\text{W}}{\text{m}^2 \text{K}^4}$$

$$\text{Power cycle thermal efficiency: } \eta = \frac{W_{\text{net}}}{Q_{\text{in}}} = \frac{W_{\text{net}}}{Q_{\text{in}}} \leq 1$$

$$\text{refridgeration: } \text{COP} = \frac{Q_{\text{out}}}{W_{\text{in}} - W_{\text{out}}} \geq 1$$

$$\text{heat pump: } \text{COP} = \frac{Q_{\text{out}}}{W_{\text{in}}}$$

$$\dot{S}_{\text{gen}} \rightarrow 0 = \text{impossible}$$

$$= 0 \rightarrow \text{ideal}$$

$$> 0 \rightarrow \text{possible}$$

$$Tds = du + pdv$$

$$Tds = dh - vdp$$

$$1 \text{ slug} = 32.2 \text{ lb}_m$$

$$1 \text{ lb}_f = 32.2 \frac{\text{lb}_m \cdot \text{ft}}{\text{s}^2} \rightarrow 1 = 32.2 \frac{\text{lb}_m \cdot \text{ft}}{\text{lb}_f \cdot \text{s}^2}$$

$$1 \text{ lb}_f = 1 \frac{\text{slug} \cdot \text{ft}}{\text{s}^2} \rightarrow 1 = 1 \frac{\text{slug} \cdot \text{ft}}{\text{lb}_f \cdot \text{s}^2}$$

$$6.02 \cdot 10^{23} \text{ particles} \cdot \frac{1 \text{ gram-mole}}{\text{gram-mole}} \cdot \frac{\text{lb}_m}{\text{lb}_m \cdot \text{mole}} \cdot \frac{1000 \text{ g}}{\text{kg}} \cdot \frac{1 \text{ kg}}{2.2 \text{ lb}_m} = 2.4 \cdot 10^{26} \text{ particles} \cdot \frac{1 \text{ lb}_m \cdot \text{mole}}{\text{lb}_m \cdot \text{mole}}$$

$$\text{sp. vol.: } v = \frac{1}{\rho} = \frac{V}{m}$$

$$\text{sp. weight: } \gamma = \rho g$$

$$\text{sp. grav.: } SG = \frac{\rho}{\rho_{\text{ref, water}}}$$

$$\dot{m}_{\text{out}} = \int \rho(\vec{v}_{\text{rel}} \cdot \hat{n}) dA \rightarrow \vec{P}_{\text{transport}} = \dot{m} \vec{v}_{\text{abs}} = \int \rho(\vec{v}_{\text{rel}} \cdot \hat{n}) \vec{v}_{\text{abs}} dA = \rho V_{\text{rel}} v_{\text{abs}} A$$

assumptions: (ρ) incomp. fluid, const. ρ | uniform velocity | flow is \perp to boundary | $\vec{v}_{\text{rel}} = \vec{v}_{\text{abs}}$

$$\text{control: } \frac{d}{dt}(\vec{P}_{\text{sys}}) = \sum \vec{F}_{\text{ext}} + \sum_{\text{in}} \dot{m} \vec{v} - \sum_{\text{out}} \dot{m} \vec{v}$$

$$\rightarrow \text{storage: } \dot{m} \frac{d\vec{v}}{dt} + \vec{v} \frac{dm}{dt}$$

$$\text{finite: } \vec{P}_{\text{system}} - \vec{P}_{\text{sys, init}} = \int \vec{F}_{\text{ext}} dt + \dots$$

$$\text{Force} = \text{impulse}$$

$$\frac{d}{dt}(E_{\text{sys}}) = \dot{W}_{\text{in, net}} + \dot{Q}_{\text{in, net}} + \sum_{\text{net}} \dot{m} (h + \frac{v^2}{2} + gz)$$

$$h = \text{sp. enthalpy} \quad \dot{Q} = ke \quad gz = pe$$

$$h = u + pv \rightarrow u = \text{sp. int. energy}$$

$$pv = \text{sp. flow energy} = P(\frac{1}{\rho})$$

$$C_v = \frac{du}{dT} \quad C_p = \frac{dh}{dT} \quad P = \text{const.}$$

$$\text{incompressible: } u_2 - u_1 = C_v(T_2 - T_1) \quad h_2 - h_1 = u_2 - u_1 + v(P_2 - P_1) \quad (p = C_1)$$

$$\text{ideal gases: } u_2 - u_1 = C_v(T_2 - T_1) \quad h_2 - h_1 = C_p(T_2 - T_1) \quad C_p = C_v + R_{\text{sp}}$$

$$\text{Kelvin-Planck: } \frac{Q_{\text{in}}}{Q_{\text{out}}} \rightarrow \text{have to have } Q_{\text{out}}$$

$$\text{Clausius: } T_C \rightarrow T_H \rightarrow \text{impossible w/o } W_{\text{in}}$$

$$\text{Power cycle: } \eta = 1 - \frac{Q_{\text{out}}}{Q_{\text{in}}} = 1 - \left(\frac{T_C}{T_H} + \frac{T_C \dot{S}_{\text{gen}}}{\dot{Q}_{\text{in}}} \right) = 1 - \frac{T_C}{T_H} \text{ if } \dot{S}_{\text{gen}} = 0$$

$$\text{Refridgeration: } \text{COP} = \frac{1}{\frac{Q_{\text{out}}}{\dot{Q}_{\text{in}}} - 1} = \frac{1}{T_H/T_C + \frac{T_H \dot{S}_{\text{gen}}}{\dot{Q}_{\text{in}}} - 1} = \frac{T_C}{T_H - T_C} \text{ if } \dot{S}_{\text{gen}} = 0$$

$$\text{Heat Pump: } \text{COP} = \frac{T_H}{T_H - T_C} \text{ for heat pump}$$

$$\text{ideal gas: } S_2 - S_1 = C_v \ln \left(\frac{T_2}{T_1} \right) + R \ln \left(\frac{V_2}{V_1} \right)$$

$$S_2 - S_1 = C_p \ln \left(\frac{T_2}{T_1} \right) - R \ln \left(\frac{P_2}{P_1} \right) \quad (S_2 - S_1 = 0) \text{ or } P_1 V_1^K = P_2 V_2^K, \text{ where } K = \frac{C_p}{C_v} \quad K_{\text{air}} \approx 1.4$$

$$\text{incompressible: } S_2 - S_1 = C_v \ln \left(\frac{T_2}{T_1} \right)$$

$$T_2 = T_1 \text{ (incompressible)}$$

$$\text{isentropic: adiabatic/reversible: } 0 = \dot{m}(S_{\text{in}} - S_{\text{out}})$$

$$\text{Universal: } pV = n R_u T$$

$$\text{(for moles)}$$

$$R_u = 8.314 \frac{\text{J}}{\text{mol} \cdot \text{K}}$$

$$= 1545 \frac{\text{ft} \cdot \text{lb}_f}{\text{lbmol} \cdot ^\circ \text{R}}$$

$$\text{Specific: } pV = m R_{\text{sp}} T$$

$$R_{\text{sp}} = \frac{R_u}{M}$$

$$M = \frac{m}{n}$$

$$QR = 0F + 454.67$$

$$\vec{v}_{\text{fluid/grd}} = \vec{v}_{\text{fluid/bdy}} + \vec{v}_{\text{bdy/grd}}$$

$$\vec{v}_{\text{abs}} = \vec{v}_{\text{rel}} + \vec{v}_{\text{boundary}}$$

$$\vec{v} = \vec{r} \times \vec{\omega}$$

$$\vec{L} = \int (\vec{r} \times \vec{v}) \rho dV = \omega \int r^2 \rho dV$$

$$W = - \int p dV$$

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$$\text{who does work?}$$

$$\text{surf. on gas}$$

$$\text{gas on surrounding}$$

$$W = - \int p dV$$

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