## Summary Hartree-Fock Stability of HEG

Evan Curtin

## I. BACKGROUND

The Hartree-Fock procedure has been implemented the quantum many-body problem since its inception and remains the basis for many more advanced techniques even today. In general, the eigenstates of the Hartree-Fock Hamiltonian are solved self-consistently. However, this procedure ensures only that the solution is stationary with respect to the determined orbitals. A method for determining the stability of a Hartree-Fock solution was proposed by Thouless in 1960[1]. The condition for stability of a Hartree-Fock solution is equivalent to the conditions for unstable (complex frequency) many-body oscillations within the Random Phase Approximation (RPA). The condition was rederived into the expression familiar to quantum chemists by Cizek and Paldus in 1967[2]. Furthermore, the stability equations factorize depending on the symmetry of the Hartree-Fock eigenfunctions. To this end, Seeger and Pople outlined a hierarchical approach to systematically evaluate the stability of HF states in the restricted, unrestricted and generalized Hartree-Fock procedures[3]. Recently, the method has been used to aid the in search for the lowest energy Unrestricted Hartree-Fock (UHF) solutions in molecules, as well as the General Hartree-Fock (GHF) solutions in geometrically frustrated hydrogen rings which cannot conform even to the UHF scheme [4][5].

The presence of GHF solutions to the Homogeneous Electron Gas which have lower energy than the RHF solutions was proven in the landmark paper by Overhauser[6]. Later still, the ground state energies of the electron gas were found to great accuracy by Ceperley and Alder [7]. In the past few years, phase diagrams have been determined for the HEG in 2 and 3 dimensions[8][9][10]

## II. REFERENCES

<sup>[1]</sup> D. Thouless, Nuclear Physics **21**, 225 (1960), ISSN 00295582, URL http://linkinghub.elsevier.com/retrieve/pii/0029558260900481.

<sup>[2]</sup> J. Cizek and J. Paldus, The Journal of Chemical Physics 47, 3976 (1967), ISSN 00219606, URL http://scitation.aip.org/content/aip/journal/jcp/47/10/10.1063/1.1701562.

<sup>[3]</sup> R. Seeger and J. A. Pople, The Journal of Chemical Physics 66, 3045 (1977), ISSN 00219606, URL http://scitation.aip.org/content/aip/journal/jcp/66/7/10.1063/1.434318.

<sup>[4]</sup> P. Pulay and Z. Tótha, Preprint 164102, 1 (2016), ISSN 0021-9606, URL http://dx.doi.org/10.1063/1.4964903.

J. J. Goings, F. Ding, M. J. Frisch, and X. Li, The Journal of Chemical Physics 142, 154109 (2015), ISSN 0021-9606,
URL http://dx.doi.org/10.1063/1.4918561http://scitation.aip.org/content/aip/journal/jcp/142/15/10.1063/1.4918561.

<sup>[6]</sup> A. W. Overhauser, Physical Review 128, 1437 (1962), ISSN 0031-899X, URL http://link.aps.org/doi/10.1103/ PhysRev.128.1437.

<sup>[7]</sup> D. M. Ceperley and B. J. Alder, Physical Review Letters 45, 566 (1980), ISSN 0031-9007, URL http://link.aps.org/doi/10.1103/PhysRevLett.45.566.

<sup>[8]</sup> F. Delyon, M. Duneau, B. Bernu, and M. Holzmann, pp. 1–12 (2008), 0807.0770, URL http://arxiv.org/abs/0807.0770.

 <sup>[9]</sup> B. Bernu, F. Delyon, M. Holzmann, and L. Baguet, Physical Review B 84, 115115 (2011), ISSN 1098-0121, URL http://link.aps.org/doi/10.1103/PhysRevB.84.115115.

<sup>[10]</sup> L. Baguet, F. Delyon, B. Bernu, and M. Holzmann, Physical Review Letters 111, 166402 (2013), ISSN 0031-9007, URL http://link.aps.org/doi/10.1103/PhysRevLett.111.166402.