

# Light Refractions

## Snell's Law

$$n_1 \sin \theta_1 = n_2 \sin \theta_2$$

## Critical Angle

$$\theta_c = \sin^{-1} \left( \frac{n_2}{n_1} \right)$$

## Index of Refraction

$$n = \frac{c}{v_{\text{nm}}}$$

# Lenses

## Sign Conventions:

$f$ : positive for convex/positive/converging lenses  
negative for concave/negative/diverging lenses

$s$ : positive on left of lens  
negative on right

$s'$ : positive on right of lens / real image  
negative on left / virtual image

$y$ : positive above optical axis  
negative below

$y'$ : positive above the optical axis  
negative below

## Image/Object/Focal Distance

$$\frac{1}{f} = \frac{1}{s} + \frac{1}{s'}$$

## Object/Image Distance/Height

$$\frac{y'}{y} = -\frac{s'}{s}$$

$$y' = -\frac{s'}{s}y$$

# Mirrors

## Different Sign Conventions

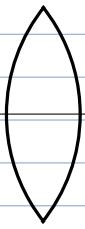
$f$ : positive for concave mirrors  
negative for convex mirrors

$s'$ : positive on left of mirror  
negative on right of mirror

real vs virtual  
from  $s'$  on mirror

Real

Virtual



## Lenses

### Sign Conventions:

f: positive for convex/positive/converging lenses  
negative for concave/negative/diverging lenses

s: positive on left of lens  
negative on right

s': positive on right of lens / real image  
negative on left / virtual image

y: positive above optical axis  
negative below

y': positive above the optical axis  
negative below

## Mirrors

### Different Sign Conventions

f: positive for concave mirrors  
negative for convex mirrors

s': positive on left of mirror  
negative on right of mirror

# Thermal S

## Area Expansion

$$\Delta A = A_0 \alpha \Delta T$$

## Linear Expansion

$$\Delta L = L_0 \alpha \Delta T$$

**Table 20.1** Specific Heats of Some Substances at 25°C and Atmospheric Pressure

Substance	Specific Heat (J/kg · °C)	Substance	Specific Heat (J/kg · °C)
<i>Elemental solids</i>		<i>Other solids</i>	
Aluminum	900	Brass	380
Beryllium	1 830	Glass	837
Cadmium	230	Ice (-5°C)	2 090
Copper	387	Marble	860
Germanium	322	Wood	1 700
Gold	129	<i>Liquids</i>	
Iron	448	Alcohol (ethyl)	2 400
Lead	128	Mercury	140
Silicon	703	Water (15°C)	4 186
Silver	234	<i>Gas</i>	
		Steam (100°C)	2 010

Note: To convert values to units of cal/g · °C, divide by 4 186.

## Volumetric Expansion

$$\Delta V = V_0 \beta \Delta T$$

**Table 19.1** Average Expansion Coefficients for Some Materials Near Room Temperature

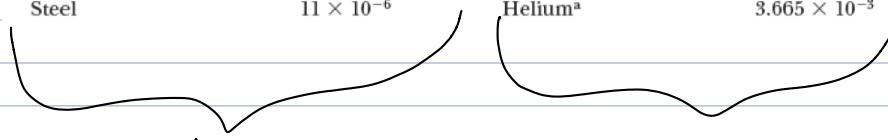
Material (Solids)	Average Linear Expansion Coefficient ( $\alpha$ ) $^{\circ}\text{C}$ $^{-1}$	Material (Liquids and Gases)	Average Volume Expansion Coefficient ( $\beta$ ) $^{\circ}\text{C}$ $^{-1}$
Aluminum	$24 \times 10^{-6}$	Acetone	$1.5 \times 10^{-4}$
Brass and bronze	$19 \times 10^{-6}$	Alcohol, ethyl	$1.12 \times 10^{-4}$
Concrete	$12 \times 10^{-6}$	Benzene	$1.24 \times 10^{-4}$
Copper	$17 \times 10^{-6}$	Gasoline	$9.6 \times 10^{-4}$
Glass (ordinary)	$9 \times 10^{-6}$	Glycerin	$4.85 \times 10^{-4}$
Glass (Pyrex)	$3.2 \times 10^{-6}$	Mercury	$1.82 \times 10^{-4}$
Invar (Ni-Fe alloy)	$0.9 \times 10^{-6}$	Turpentine	$9.0 \times 10^{-4}$
Lead	$29 \times 10^{-6}$	Air <sup>a</sup> at 0°C	$3.67 \times 10^{-3}$
Steel	$11 \times 10^{-6}$	Helium <sup>a</sup>	$3.665 \times 10^{-3}$

For Solids:

$$\gamma = 2\alpha$$

$$\beta = 3\alpha$$

length + Area for  
gases/liquids defined  
by container



# Electrical Charges & Magnetism

## Gravity

$$F_g = G \frac{m_1 m_2}{r^2}$$

## Force from e Charge

$$F_e = k_e \frac{|q_1||q_2|}{r^2}$$

$$k_e = 8.99 \times 10^9 \frac{N \cdot m^2}{C^2}$$

## Elementary Charge:

1 proton:  $1.602 \times 10^{-19} C$

## Voltage & Potential Energy

$$\Delta V = Ed$$

$$V = k \frac{Q}{r}$$

$$\Delta q = -q \Delta V$$

$$U = k \frac{q_1 q_2}{r}$$

$$\Delta q = -q Ed$$

$$\vec{E} = k \frac{\vec{q}_1 \vec{q}_2}{r^2}$$

$$\vec{F} = q \vec{E}$$

## Electric Flux

$$\Phi = EA \quad (\text{when parallel})$$

$$\Phi = E \cdot A \cos(\theta) \quad (\theta \text{ is angle from normal})$$

$$\Phi = \frac{Q_{\text{enc}}}{\epsilon_0}$$

## Oscillations

$$F = |q| V_i B = m \frac{v_i^2}{R}$$

$\epsilon_0$  = Permittivity of free space

$$\epsilon_0 = 8.85 \times 10^{-12} \frac{C^2}{N \cdot m^2}$$

## Angular Frequency

$$\omega = \frac{V}{r}$$

$$= \frac{2\pi}{T} \quad T = \text{period (seconds)}$$

$$= 2\pi f$$

## Magnetism

$$F = |q| v_i B = |q| v B \sin \phi$$

$$T = \frac{m}{q \cdot B}$$

$$A = \frac{C}{s}$$

$$1 \text{ Gauss} = 1 \times 10^{-4} T$$

## Magnetic Field from Long Conductor

$$\vec{B} = \frac{\mu_0 I}{2\pi r}$$

I = Current

$\mu_0$  = Mag Const  
r = dist from cond

## Magnetic & Electric Force:

$$\vec{F} = q(\vec{E} + \vec{v} \times \vec{B})$$

## Magnetic Constant:

$$\mu_0 = 4\pi \times 10^{-7} \frac{N}{A}$$

Positron: mass of  $e^-$   
positive charge of  $e^-$   $(+1.609 \times 10^{-19} C)$

Antiparticle: mass of  $p^+$   
negative charge of  $p^+$   $(-1.609 \times 10^{-19} C)$

## Waves

$$y(x, t) = A \sin [kx - \omega t + \phi]$$

A: Amplitude (usually in m)

k: wave number ( $\text{m}^{-1}$ )

$\omega$ : angular velocity ( $\frac{\text{rad}}{\text{sec}}$ )

t: time (sec)

$\phi$ : phase constant

f: frequency (Hz)

T: period (s)

$\lambda$ : wavelength (m)

V: speed of wave (m/s)

$$f = \frac{1}{T} \quad \text{or} \quad T = \frac{1}{f}$$

$$\omega = 2\pi f = \frac{2\pi}{T}$$

$$k = \frac{2\pi}{\lambda} \quad \text{rad/m}$$

$$V = f \lambda = \frac{\omega}{k}$$

## Doppler Effect

$$f' = f \frac{V \pm V_o}{V \mp V_s}$$

$v_o$  moving towards the source

$v_o$  moving away

$v_s$  moving toward observer

$v_s$  moving away

# Simple Harmonic Motion

$$x(t) = A \cos[\omega t + \phi]$$

$$\omega = 2\pi f = \frac{2\pi}{T}$$

$$v(t) = -A\omega \sin[\omega t + \phi]$$

$f$ : # of times in 1 sec

$T$ : time to do one cycle

$$a(t) = -A\omega^2 \cos[\omega t + \phi]$$

$$\omega = \sqrt{\frac{k}{m}}$$

$$F_s = kx$$

$k$ : spring constant