

$$\underbrace{\frac{mv^2}{r}}_{\text{centripetal}} = \underbrace{\frac{e^2}{4\pi\epsilon_0 r^2}}_{\text{coulomb}} \Rightarrow \frac{e^2}{4\pi\epsilon_0 m r}$$

$$E = \frac{1}{2} mv^2 - \frac{e^2}{4\pi\epsilon_0 r} \quad v = \frac{n\hbar}{mr}$$

$$E = \frac{1}{2} m \frac{n^2 \hbar^2}{m^2 r^2} - \frac{e^2}{4\pi\epsilon_0 r}$$

$$E = -\frac{me^4}{2(4\pi\epsilon_0)^2 \hbar^2} \frac{1}{n^2}$$

$$\frac{n^2 \hbar^2}{m^2} = \frac{1}{r} \frac{e^2}{4\pi\epsilon_0 m}$$

$$r = \frac{n^2 \hbar^2 4\pi\epsilon_0}{me^2}$$

Bohr Orbitals

if n is large one has almost continuum levels.

$$\frac{1}{2} m \frac{n^2 \hbar^2}{m^2} \frac{m^2 e^4}{n^4 \hbar^4 (4\pi\epsilon_0)^2}$$

$$= - \frac{e^2}{4\pi\epsilon_0} \frac{me^2}{n^2 \hbar^2 4\pi\epsilon_0}$$

$$\rightarrow - \frac{mc^2}{2} \cdot \underbrace{\frac{e^4}{(4\pi\epsilon_0)^2 c^2 \hbar^2}}_{\text{dimensionless}} \cdot \frac{1}{n^2}$$

$$\rightarrow \alpha \equiv \underbrace{\frac{e^2}{4\pi\epsilon_0 \hbar c}}_{(-13.6 \text{ eV}) \frac{1}{n^2}}$$

make sure
'tis
coulomb

Schrödinger

$$\left(\frac{1}{c^2} \frac{\partial^2}{\partial t^2} - \nabla^2 \right) \psi = 0 \quad \text{say } \psi = e^{i\omega t} \psi$$

$$\left(-\frac{\omega^2}{c^2} - \nabla^2 \right) \psi = 0 \Rightarrow (\nabla^2 + k^2) \psi = 0 \quad k = \frac{2\pi}{\lambda}$$

• Stationary waves
obey this.

$$\left(\nabla^2 + \frac{4\pi^2}{n^2} p^2 \right) \psi = 0 \rightarrow (\hbar^2 \nabla^2 + p^2) \psi = 0$$

+ de Broglie !!!

• then π is the momentum of
the particle"

Definite energy as Monofreq...
"match"

- $E = \frac{p^2}{2m} + V$ (nonrelativistic particle)

$$p^2 = 2mE - 2mV$$

$$(\hbar^2 \nabla^2 + 2mE - 2mV) \psi = 0$$

$$\boxed{\left(-\frac{\hbar^2}{2m} \nabla^2 + V\right) \psi = E \psi}$$

Time independent Schrödinger Equation

matrix · vector = eigenvalue · vector

"Is there a real solution for every E ?"

No! Only certain discrete levels of E are allowed.

- ψ is now called the state function.

Particle in a (1D) box

$$\psi(x=0) = \psi(x=L) = 0$$

$$\left. \begin{aligned} -\frac{\hbar^2}{2m} \frac{d^2 \psi}{dx^2} &= E \psi \end{aligned} \right\} \begin{aligned} \psi &= A \sin(kx) + B \cos(kx) \\ (*) E &= \frac{\hbar^2 k^2}{2m} \end{aligned}$$

Boundary conditions make it non trivial.

When $\psi(x=0)=0$, B is forced to be 0.

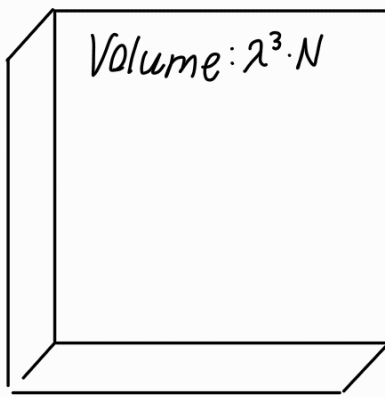
$$\text{When } \psi(x=L)=0, \sin(kL)=0 \Rightarrow k = \frac{n\pi}{L}, \quad k = \frac{2\pi}{\lambda}$$

$$\boxed{\lambda = \frac{2L}{n}}$$

de Broglie wavelengths.

$$(*) E = \frac{\hbar^2 k^2}{2m} = \frac{\hbar^2 n^2 \pi^2}{2mL}$$

"point particles" is no longer in the language of quantum mechanics.



- Lowest E ,
- Largest λ ,
- lowest n . more "quantum"

"Thermalizing a motion"

Thermal wavelength: $\bar{E} = \frac{3}{2} kT = \frac{\overline{p^2}}{2m} = \frac{h^2}{2m\lambda^2} \quad \lambda = \sqrt{\frac{4h^2 m}{3kT}}$

"limiting temperature for quantum interactiveness"