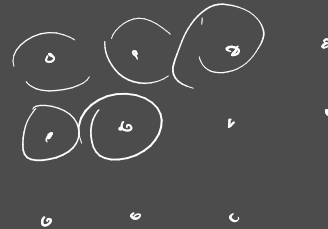
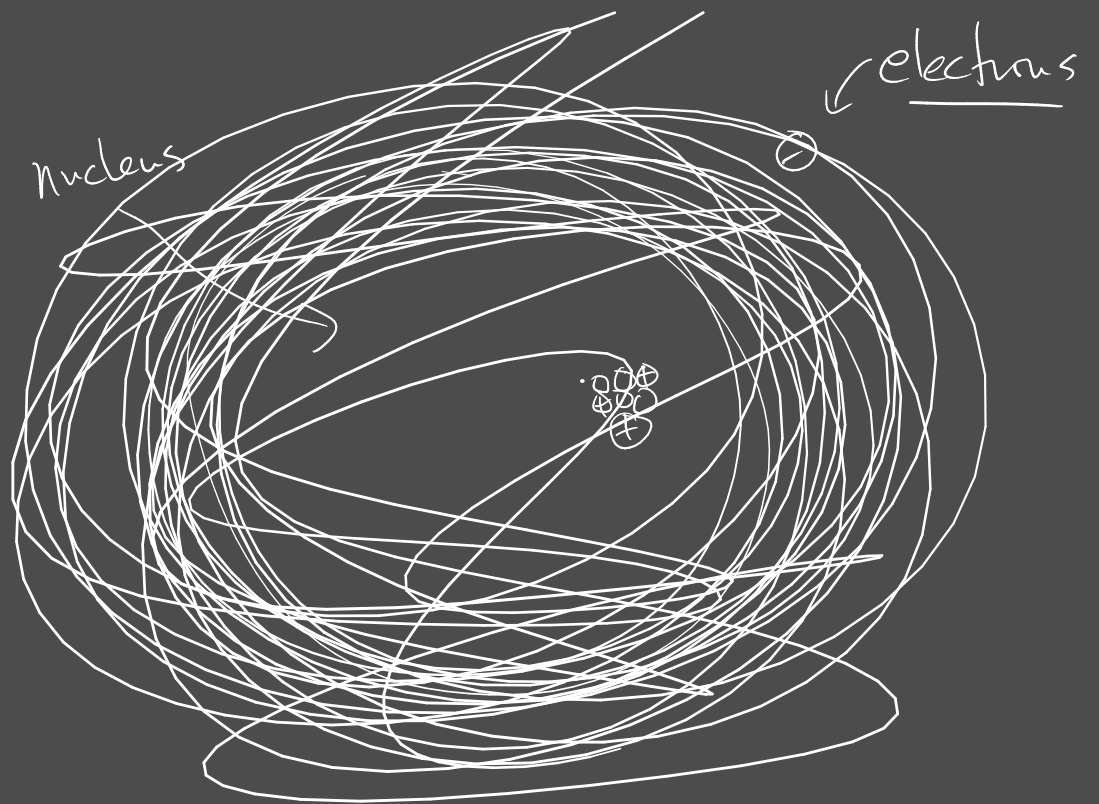


# Chapter 12

## Atom

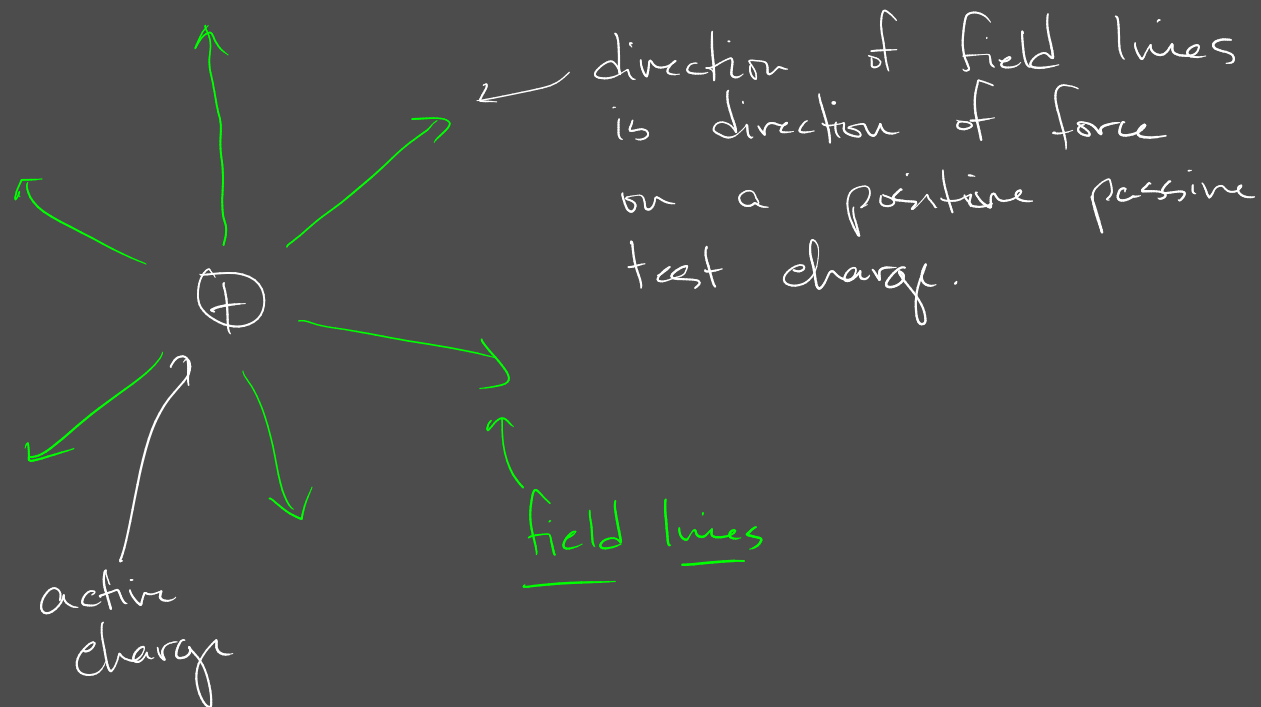
nucleus

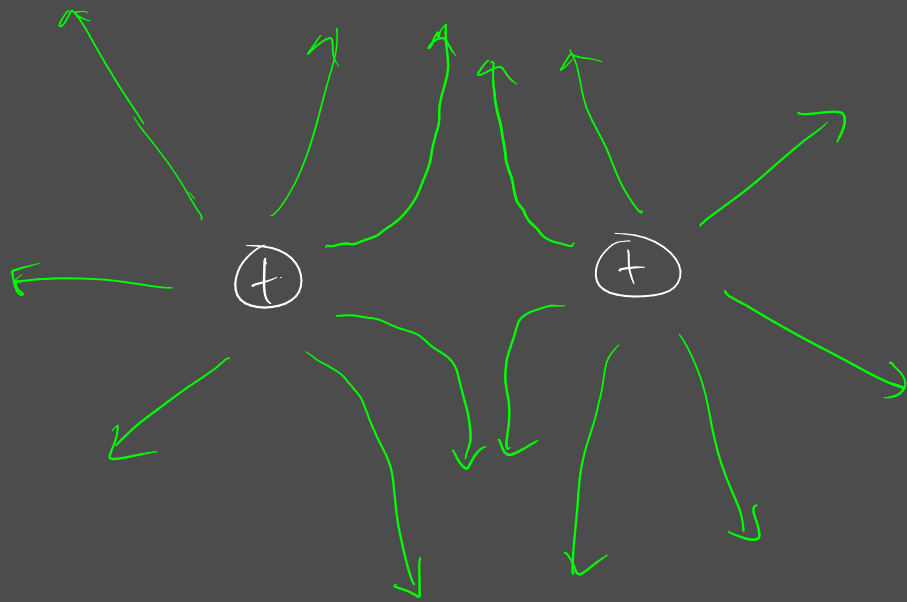
- protons - positive
- electrons - negative
- neutrons - no charge



Electric Field  $\rightarrow$  force field that results from charge

Electric Force =  $\underbrace{\text{charge in a field}}_{\text{"passive" charge being acted on by the field.}} \times \underbrace{\text{Field caused by another charge}}_{\substack{\text{"active" charge} \\ \downarrow \\ \text{producing an electric field}}}$

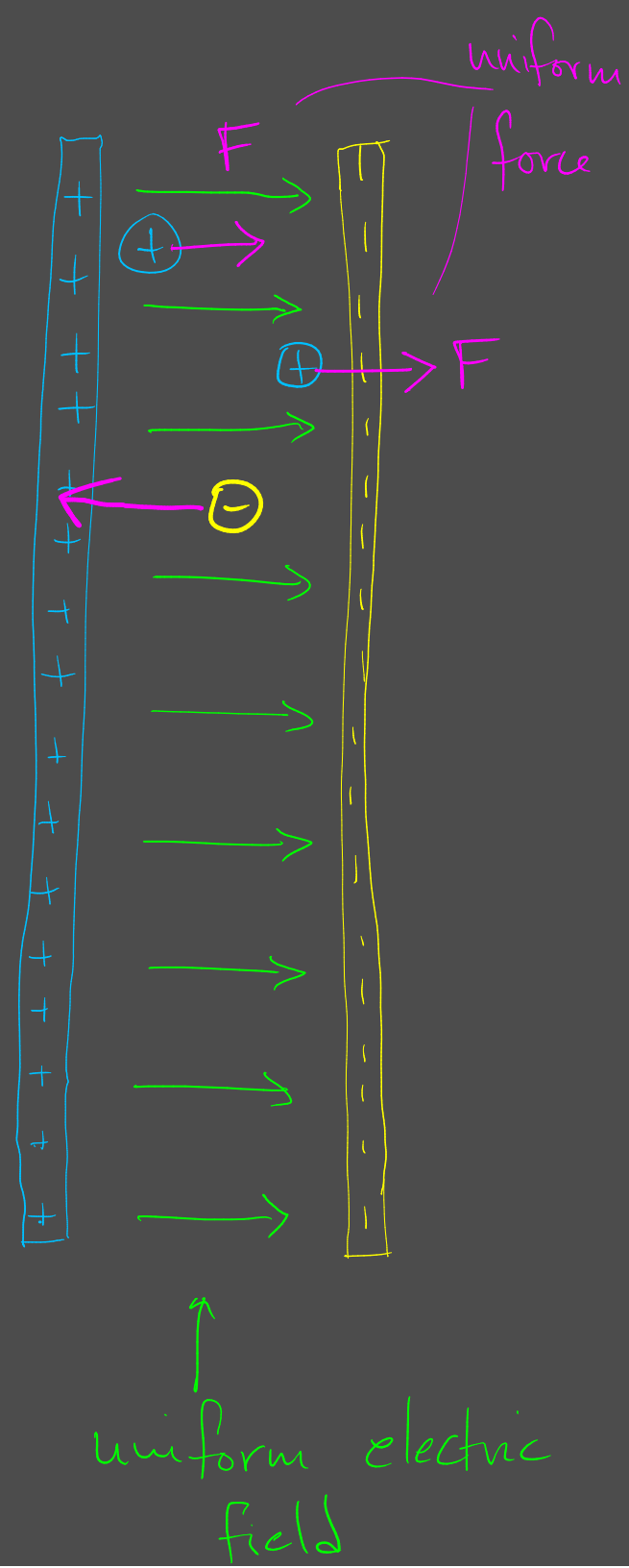




charge  $\rightarrow$  1 unit of charge  
 $\rightarrow$  1 Coulomb

1 elementary charge  $\rightarrow$  charge of one electron or proton  
 $\downarrow$   
 fundamental

$\hookrightarrow 6.24 \times 10^{18}$  electrons = 1 Coulomb



↓ What is the charge of 1 electron in terms of Coulombs?

$$1 \text{ elementary charge} = \frac{1 \text{ C}}{6.24 \cdot 10^{18} e} = 1.6 \cdot 10^{-19} \text{ C / elementary charge}$$

always true | Electric Field can be complicated

$$\text{Force} = \overset{\text{passive}}{\text{charge}} \times \text{Field}$$

$q_2$

$$q_1 = 10 \text{ C}$$

⊕

$$q_2 = 2 \text{ C}$$

⊕

10 m

Coulomb's Law

$$F = \frac{k \cdot q_1 \cdot q_2}{r^2}$$

only true for two point charges

$$k = 9 \cdot 10^9 \frac{\text{N m}^2}{\text{C}^2}$$

↳ Coulomb constant

$$\text{Electric Field}$$

point charge

$$E = \frac{k q_1}{r^2}$$

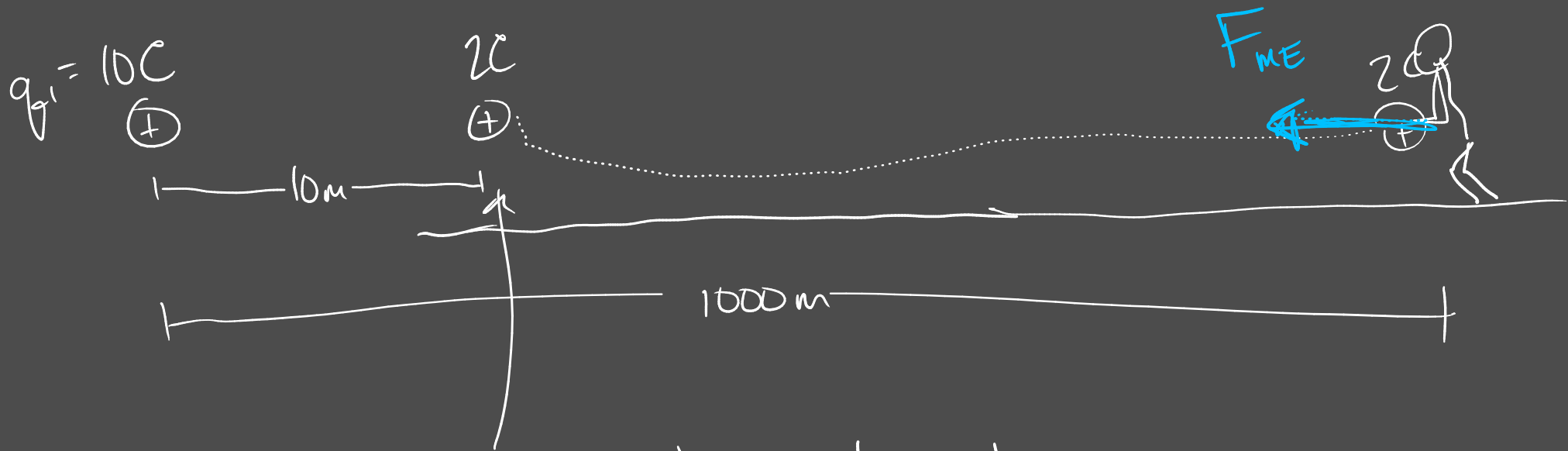
$$\left| \begin{array}{l} 2 \mu\text{C} = 2 \cdot 10^{-6} \text{C} \\ 0.6 \mu\text{C} = 0.6 \cdot 10^{-6} \text{C} \\ 10 \text{mC} = 10 \cdot 10^{-3} \text{C} \end{array} \right.$$

Electric Field

$$E \quad \left[ \frac{\text{N}}{\text{C}} \right]$$

$$F = q \cdot E$$

$$\begin{array}{ccc} \uparrow & \uparrow & \uparrow \\ [N] & = [C] \cdot & \left[ \frac{N}{C} \right] \end{array}$$

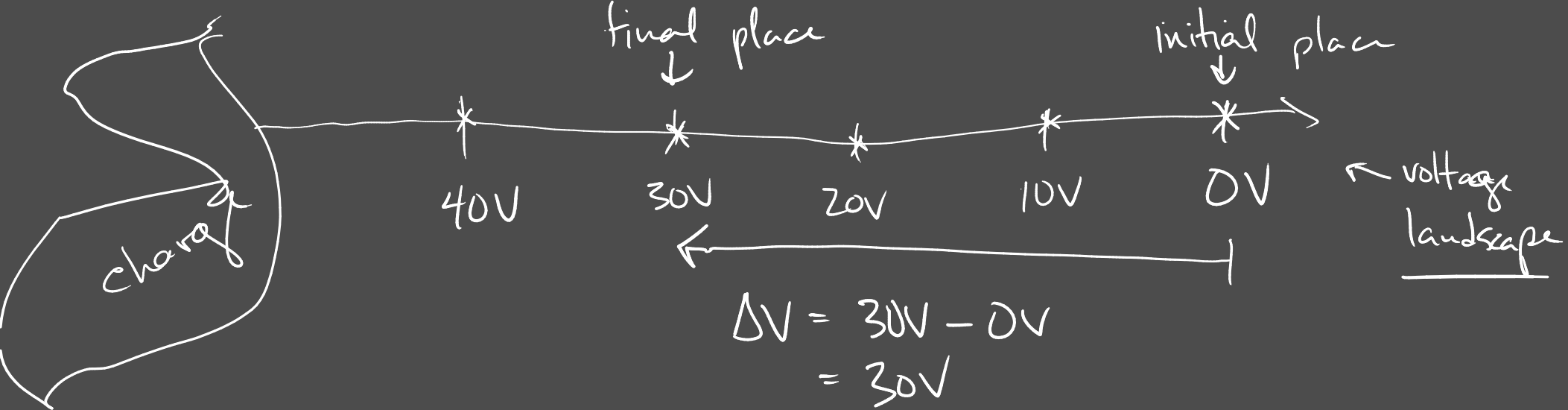


Work has been done to  
get the charge to here.

The charge now has potential energy

Potential Energy = Voltage (Electric Potential)  
unit of charge

$$\frac{[\text{Joule}]}{[\text{Coulomb}]} = [\text{Volts}]$$



important thing is change in voltage  
 always need a reference point.

~~$$V = \frac{U_E}{q}$$~~

$$\Delta V = \frac{\Delta U_E}{q} \quad \left[ \frac{\text{Joule}}{\text{Coulomb}} \right] = [\text{Volt}]$$

$$\Delta U_E = \Delta V \cdot q$$

↑        ↑        ↑  
 Joules   Volts   Coulomb

$$\#1 \mid q_e = 1.6 \cdot 10^{-19} \text{ C} \leftarrow \text{fundamental charge}$$

$$\frac{10.4 \cdot 10^{-5} \text{ C}}{1.6 \cdot 10^{-19} \text{ C}} = 6.5 \cdot 10^{14} \text{ charges}$$

$$\#2 \mid F = 24 \text{ N}$$

$$d_2 = \frac{1}{3} d_1$$

$$F = \frac{k q_1 q_2}{d^2} = 24 \text{ N}$$

$$F = \frac{k q_1 q_2}{\left(\frac{d}{3}\right)^2} = \frac{k q_1 q_2}{\frac{d^2}{9}} = k q_1 q_2 \cdot \frac{9}{d^2} \\ = 9 \cdot \underbrace{\frac{k q_1 q_2}{d^2}}_{24 \text{ N}}$$



#4]  $q_1 = 10 \cdot 10^{-6} \text{ C}$

$d = 12 \text{ cm}$

$\uparrow$   
 $.12 \text{ m}$

$$F = \frac{k q_1 q_2}{d^2}$$

$$F = \frac{9 \cdot 10^9 \cdot 10 \cdot 10^{-6} \cdot 10 \cdot 10^{-6}}{(.12)^2} = 62.5 \text{ N}$$

#7]  $E = 30 \frac{\text{N}}{\text{C}}$  upward

$q = -6 \text{ C}$

#8]  $q = 12 \cdot 10^{-6} \text{ C}$

$F = 41 \text{ N down}$

$E = ?$

Electric Force = charge · Field

$\frac{30 \text{ N}}{\cancel{\text{C}}} \cdot \cancel{6 \text{ C}} = \underline{180 \text{ N}}$

$F = q \cdot E$

$\frac{41 \text{ N}}{12 \cdot 10^{-6} \text{ C}} = \frac{12 \cdot 10^{-6} \text{ C} \cdot E}{12 \cdot 10^{-6} \text{ C}}$

$E =$

$$5\text{ C}$$

$$12\text{ V} \rightarrow 0\text{ V}$$

$$\Delta U_E = q_f \cdot \Delta V$$

$$= q_f \cdot (V_f - V_i)$$

$$= 5\text{ C} \cdot (0\text{ V} - 12\text{ V})$$

$$= 5\text{ C} \cdot -12\text{ V}$$

$$\Delta U_E = -60\text{ J}$$