

Chapter 2 - 2nd Law of Thermodynamics

↳ heat spontaneously flows from high temp to low temp

Einstein Solid

Ideal Gas

↳ Entropy

→ Anacanda python
install language
Monty Python

Combinatorics

one coin
 $P(\text{heads}) = \frac{1}{2}$

multiple coins

$$P(n) = \frac{n}{N}$$

$$P(3 \text{ heads}) = \frac{\Omega(3)}{\Omega(\text{all})}$$
$$= \frac{\Omega(3)}{\sum_{n=0}^n \Omega(n)}$$

← microstate in the macrostate of 3
← all of the microstates

5 coins

microstates { H H T T H — 3 H } macrostates
 T H H H H — 4 H }

how many microstates are in a macrostate?
↳ multiplicity

$$\Omega(n) = \frac{5!}{n!(5-n)!}$$

number of heads
 $\Omega(0) = 1$
 $\Omega(1) = 5$
 $\Omega(2) = 10$
 $\Omega(3) = 10$
 $\Omega(4) = 5$
 $\Omega(5) = 1$


$$\Omega(1) = \frac{5!}{1!(5-1)!} = \frac{5!}{1!4!}$$
$$= \frac{5 \cdot 4 \cdot 3 \cdot 2 \cdot 1}{1 \cdot 4 \cdot 3 \cdot 2 \cdot 1}$$
$$= 5$$
$$\Omega(2) = \frac{5!}{2!3!} = 10$$

$$\Omega(N, n) = \frac{N!}{n!(N-n)!} \leftarrow \text{Notation: } \binom{N}{n}$$

\uparrow
 # of coins

10 atoms each w/ 0 or 1 packets of energy (energy unit)


How many possible ways are there to distribute 4 energy units


 \leftarrow microstate

4 energy packets \leftarrow macrostate

$$\Omega(10, 4) = \frac{10!}{4!6!} = 210$$

What if an atom can have more than one energy packet at a time?

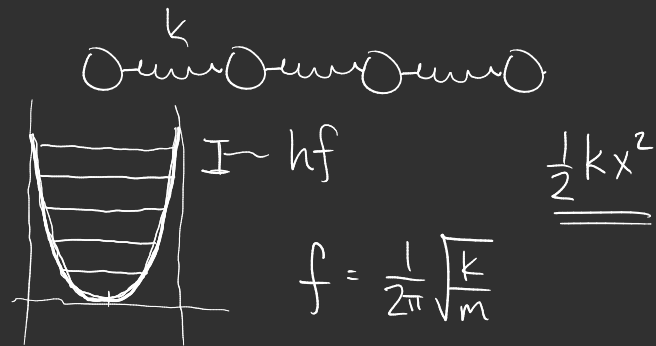

 \leftarrow microstates

4 energy packets \leftarrow macrostate

$$\Omega(N, q) = \frac{(q + N - 1)!}{q!(N-1)!} \quad \binom{q+N-1}{q}$$

\uparrow # of atoms \uparrow # of energy packets
 $q = \text{macrostate}$

This model of a collection of atoms w/ equal size energy quanta distributed among them is the Einstein Solid.



→ Debye Model

• Large Number → addition of small numbers is not important

$$10^{23} + 23 = 10^{23}$$

• Very Large Number

$$10^{10^{23}} \times 10^{23} = 10^{10^{23} + 23} \approx 10^{10^{23}}$$

Stirling's Approximation

$$N! \approx N^N e^{-N} \sqrt{2\pi N} = \frac{N^N}{e^N} \underbrace{\sqrt{2\pi N}}_{\text{very large numbers we leave this off}}$$

↓

$$\ln N! \approx N \ln N - N$$

very large numbers
we leave this off

Two Systems



q_A	mult_A	q_B	mult_B	$\text{mult}_{\text{total}}$
0	1	6	28	28
1	3	5	21	63
2	6	4	15	90
3	10	3	10	100
4	15	2	6	90
5	21	1	3	63
6	28	0	1	28

macrostate w/
the most microstates

total number
of microstates

(462)

Ex. Each system has 3 particles
and 6 total quanta of energy

A



macrostate of 3



$$\Omega(3,1) = 3$$

B



macrostate of 3



$$\Omega(3,5)$$

$$= \frac{7!}{5!2!} = 21$$

Fundamental Assumption of Stat Mech:

all microstates are possible and equally probable

But that does not mean that every microstate will occur.

Not all macrostates are equally probable.

$$P_A(q) = \frac{\Omega_{\text{total}}(q_A)}{\sum_{q_A} \Omega_{\text{total}}(q_A)}$$

We could find the total # of microstates:

$$\Omega(6, 6) = \frac{(6+6-1)!}{6!(6-1)!} = 462$$

$\uparrow \quad \uparrow$
 $N \quad q$

So let's apply Stirling's Approx to Multiplicity $[\ln N! = N \ln N - N]$

$$\Omega(N, q) = \frac{(q+N-1)!}{q!(N-1)!} \approx \frac{(q+N)!}{q!N!}$$

very large number

$$\ln \Omega = \ln(q+N)! - \ln q! - \ln N! \rightarrow N \ln N - N$$

$\underbrace{\ln(q+N)!}_{\ln(q+N)! = (q+N) \ln(q+N) - (q+N)}$ $\underbrace{\ln q!}_{q \ln q - q}$ $\underbrace{\ln N!}_{N \ln N - N}$

$$\ln \Omega = (q+N) \ln(q+N) - \cancel{q} - \cancel{N} - q \ln q + \cancel{q} - N \ln N + \cancel{N}$$

$$\boxed{\ln \Omega = (q+N) \ln(q+N) - q \ln q - N \ln N}$$

high temperature limit $\rightarrow q \gg N$

$$\begin{aligned}
 \ln \Omega &= (q+N) \ln(q+N) - q \ln q - N \ln N \\
 &= \underbrace{q \ln(q+N)}_{\downarrow} + \underbrace{N \ln(q+N)}_{\downarrow} - q \ln q - N \ln N \\
 &= \ln \left[q \cdot \left(1 + \frac{N}{q} \right) \right] \\
 &= \ln q + \ln \left(1 + \frac{N}{q} \right) \\
 &\quad \downarrow \quad \quad \quad \begin{array}{l} \ln(1+x) \approx x \text{ for small } x \\ \rightarrow \frac{N}{q} \end{array} \\
 &= \ln q + \frac{N}{q} \approx \ln(q+N)
 \end{aligned}$$

$$\begin{aligned}
 \ln \Omega &= \cancel{q \ln q} + N + \underbrace{N \ln q}_{\text{small}} + \frac{N^2}{q} - \cancel{q \ln q} - \underbrace{N \ln N}_{\text{small}} \\
 &= N \ln \left(\frac{q}{N} \right) + N + \frac{N^2}{q}
 \end{aligned}$$

$$\ln \Omega(q \gg N) \approx N \ln \left(\frac{q}{N} \right) + N$$

$$\Omega(q \gg N) = e^{N \ln \left(\frac{q}{N} \right) + N} = e^{N \ln \left(\frac{q}{N} \right)} \cdot e^N = \underbrace{\left(e^{\ln \left(\frac{q}{N} \right)} \right)^N} \cdot e^N$$

$$\Omega(q \gg N) = \left(\frac{q}{N} \right)^N e^N = \left| \left(\frac{eq}{N} \right)^N \right|$$

