# Ongoing XAS collaboration between SCK•CEN and Aalto University

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#### Outline

- Background
- Examples of actinide studies
- Fuel lab introduction
- Collaboration possibilities

# Background

- Personal interest towards synchrotron methods since PhD work on III-V semiconductors
  - Photoluminescence blue shift as a function of In-N bonds in GaInAsN
  - These things can be ACTUALLY measured!
- Visiting scientist at the UW-Madison, USA
  - SSRL and collaboration with S. Conradson (complexity of UO<sub>2+x</sub> ....)
- Arrival to SCK•CEN in 2014
  - Productive PhD students in newly established fuel laboratory
  - Storage full of samples ready for further analysis
  - Collaboration with R. Bes

# Background

- UO<sub>2</sub> and its derivatives are the main fuel used in the light water reactors globally
  - 450 reactors online, 58 under construction (158 ordered or planned)\*)
  - 150 tons of U per reactor per year (→ 65 000 t/year)\*)
- Nuclear fuel complex system
  - Simple binary fluorite consisting of U and O
  - ~100 years of studies simple oxidation of UO<sub>2</sub> still remains a subject of debate
- Spent nuclear fuel the most complex system on Earth
  - Crystal damage from fissions
  - Alternating chemical environment from fission products
  - Extreme radiotoxity

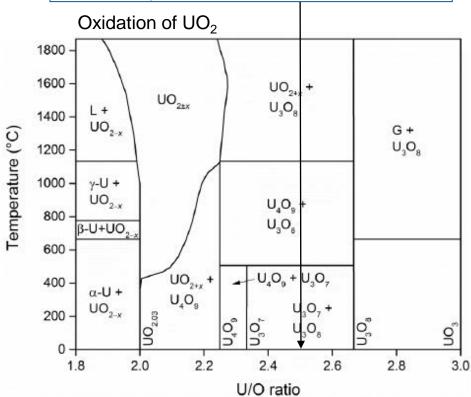
# SCIENTIFIC REPORTS

**OPEN** 

Direct observation of pure pentavalent uranium in U<sub>2</sub>O<sub>5</sub> thin films by high resolution photoemission spectroscopy

Received: 18 January 2018 Accepted: 14 May 2018 Published online: 29 May 2018

T. Gouder, R. Eloirdi @ & R. Caciuffo @



G. Leinders (PhD Thesis, https://lirias.kuleuven.be/handle/123456789/546279)

→ Understanding the behavior of (spent) fuel is a challenge

#### Background

#### Spent fuel inventory

Heavy metal atoms	
U	238.00
Np	237.00
Pu	240.00
Am	242.00
Cm	244.00
Cf	249.00

Rb

Sr

Y

Zr

Nb

Mo

Tc

Ru

Rh

Pd

Pd

Ag

Cd

In

Sn

Sb

Te

Ba

La

Ce

Pr

Nd

Pm

Sm

Eu

Gd

Dy

Tb

Ho

Er

Tm

Yb

Fission produc	ts .	Gaseous and volatile fission product.			
Li	6.94	H	1.00		
Be	9.01	He	4.00		
C	12.01	Br	79.09		
		Kr	83.80		
Zn	65.37	I	126.90		
Ga	69.72	Xe	131.29		
Ge	72.59	Cs	132.91		
As	74.92				
Se	78.96				

85.47

87.62

88.91

91.22

92.91

95.94

97.00

101.07

102.91

106.40

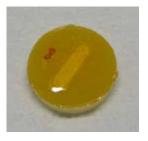
106.40

107.87

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- Challenges (mainly) due to radioactivity
  - Transport of samples
    - Can become costly and time consuming
    - Exempt limit for U and Th = 1000 Bq  $\rightarrow$  sample mass < ~1 g
    - Optimization required (and sometimes possible)
  - Preparation of samples
    - Double sealing required (Kapton)
    - 10 20 mg of UO<sub>2</sub> in BN
    - No powder allowed (pressed pellet)





- Dedicated beamline for radioactive samples
  - MARS at Soleil
  - ROBL / ID26 at ESRF
  - ANKA at KIT
  - ...

Journal of Nuclear Materials 489 (2017) 9-21



Contents lists available at ScienceDirect

#### Journal of Nuclear Materials





Charge compensation mechanisms in  $U_{1-x}Gd_xO_2$  and  $Th_{1-x}Gd_xO_{2-x/2}$  studied by X-ray Absorption Spectroscopy



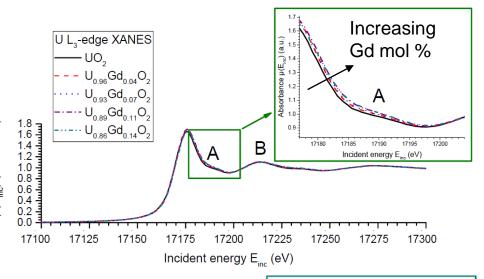
R. Bès <sup>a, \*</sup>, J. Pakarinen <sup>b</sup>, A. Baena <sup>b</sup>, S. Conradson <sup>c</sup>, M. Verwerft <sup>b</sup>, F. Tuomisto <sup>a</sup>

- Practical interest: Gd has large neutron cross-section used as a burnable neutron absorber in the fresh reactor core
- Challenge: Gd<sup>3+</sup> substitution to the position of U<sup>4+</sup> or Th<sup>4+</sup>
- U can be (3+), 4+, 5+, and 6+, Th is always 4+
- Mechanism for charge compensation?
- XANES and EXAFS @ MARS (Soleil, France)

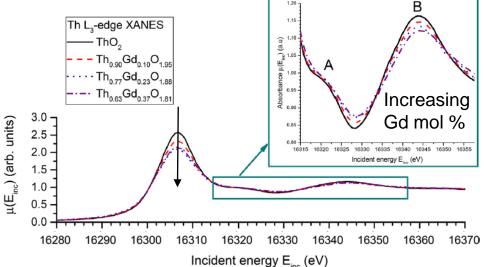
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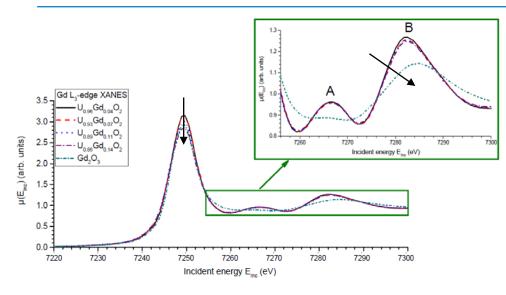
c Synchrotron SOLEIL, Ligne de Lumière MARS, L'Orme des Merisiers, Saint Aubin, BP 48, F-91192 Gif-sur-Yvette Cedex, France



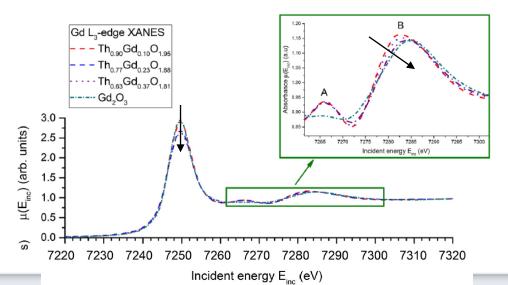
- XANES for U L<sub>3</sub> edge
  - Intensity increases on the highenergy side:
    - Fingerprint for U<sup>5+</sup>formation
    - Linear combination fitting using reference spectra indicated this being the case



- XANES for Th L<sub>3</sub> edge
  - Features remain at fixed positions:
    - Th<sup>4+</sup> is stable (as should be)
    - Uniform reduction is observed, charge is compensated by O vacancies which induce to local lattice disorder and reduced coordination number?



- XANES for Gd L<sub>3</sub> edge in UO<sub>2</sub>
  - White line at fixed position:
    - Gd<sup>3+</sup> is stable
    - Reduction in intensity: disorder
  - Comparison to Gd<sub>2</sub>O<sub>3</sub>
    - Same resonances, A B difference approaches Gd<sub>2</sub>O<sub>3</sub>



- XANES for Gd L<sub>3</sub> edge in ThO<sub>2</sub>
  - White line at fixed position:
    - Gd<sup>3+</sup> is stable
    - Reduction in intensity: disorder
  - Comparison to Gd<sub>2</sub>O<sub>3</sub>
    - Same resonances, A B
      difference approaches Gd<sub>2</sub>O<sub>3</sub>

+ full EXAFS analysis for U L<sub>3</sub>, Th L<sub>3</sub>, and Gd L<sub>3</sub>

#### Conclusion:

- Charge compensation in Gd-doped UO<sub>2</sub> proceeds via disorder and formation of U<sup>5+</sup>
- Charge compensation in Gd-doped ThO<sub>2</sub> proceeds via disorder and formation of O vacancies
- Thus:  $U_{1-x}Gd_xO_2$  vs  $Th_{1-x}Gd_xO_{2-x/2}$

# **Inorganic Chemistry**

Inorganic Chemistry 56 (2017) 6784

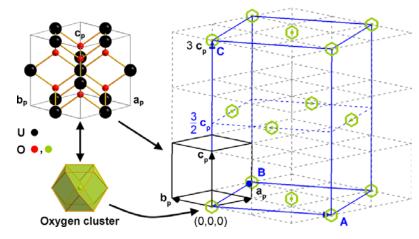
Communication

pubs.acs.org/IC

#### Evolution of the Uranium Chemical State in Mixed-Valence Oxides

Gregory Leinders,\*\*,†© René Bes,‡© Janne Pakarinen,† Kristina Kvashnina,<sup>§,||</sup> and Marc Verwerft†

- Main motivation: breakthrough in solving the crystal structure of U<sub>3</sub>O<sub>7</sub>
- Does chemical state of U match the proposed structure?
- HERFD-XANES @ ID26 (ESRF)



G. Leinders et al., Inorganic Chemistry 55 (2016) 9923

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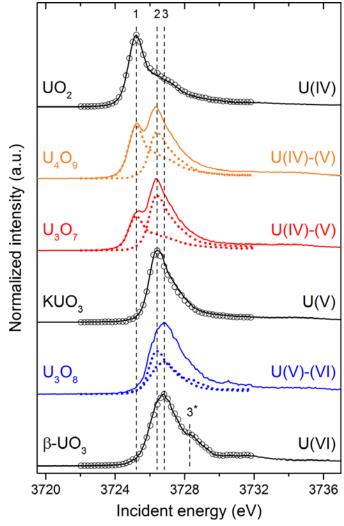
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Helmholtz Zentrum Dresden-Rossendorf (HZDR), Institute of Resource Ecology, P.O. Box 510119, 01314 Dresden, Germany

- "Typical" approach:
  - Well known references and fitting for the unknown
- Excellent agreement with the theory
  - U<sub>3</sub>O<sub>7</sub> matches perfectly
  - world's best U<sub>4</sub>O<sub>9</sub>
- EXAFS was done @ ROBL (ESRF)
  - Analysis ongoing

	relative abundance of valence states (%), ±3%			average U valence, ±0.03	
	U(IV)	U(V)	U(VI)	exptl.	theor.
$U_4O_9$	51	49	0	4.49	4.50
$U_3O_7$	36	64	0	4.64	4.67
$U_3O_8$	0	65	35	5.35	5.33



Inorganic Chemistry 56 (2017) 6784

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#### Journal of Nuclear Materials



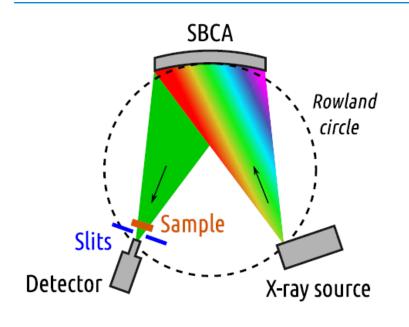


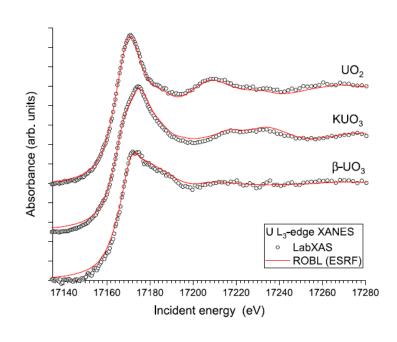
Laboratory-scale X-ray absorption spectroscopy approach for actinide research: Experiment at the uranium L<sub>3</sub>-edge



R. Bès <sup>a, \*</sup>, T. Ahopelto <sup>b</sup>, A.-P. Honkanen <sup>b</sup>, S. Huotari <sup>b</sup>, G. Leinders <sup>c</sup>, J. Pakarinen <sup>c</sup>, K. Kvashnina <sup>d, e</sup>

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- <sup>b</sup> Department of Physics, University of Helsinki, P.O. Box 64, FI-00014, Helsinki, Finland
- <sup>c</sup> Belgian Nuclear Research Centre (SCK CEN), Institute for Nuclear Materials Science, Boeretang 200, B-2400, Mol, Belgium
- d Rossendorf Beamline at ESRF The European Synchrotron, CS40220, 38043, Grenoble Cedex 9, France
- e Helmholtz Zentrum Dresden-Rossendorf (HZDR), Institute of Resource Ecology, P.O. Box 510119, 01314, Dresden, Germany
- Demonstration of lab-XAS @ University of Helsinki
- XANES measurement at U L3 edge comparison to measurement @ ROBL (ESRF)

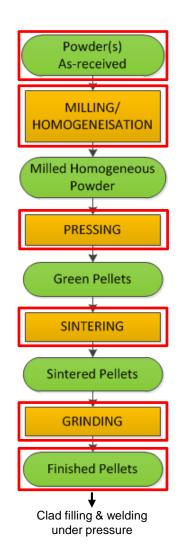




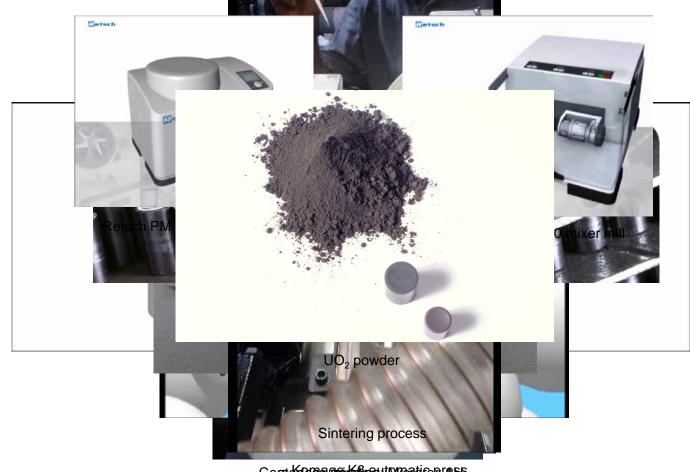
- Lab-XAS: Time consuming compared to synchrotron (24h vs 2h) but all XANES features were reproduced!
- Set-up in the lab offers possibilities: screening of samples before beam time, in-situ development, instrument time availability,...

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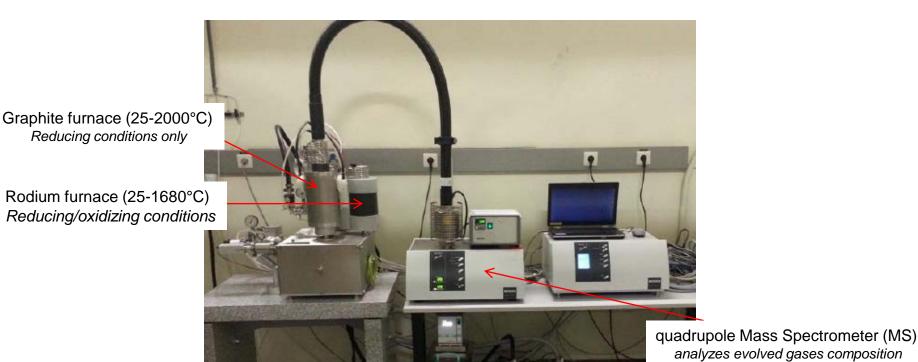
#### Fuel fabrication: from powder to pellet



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#### Fuel lab introduction

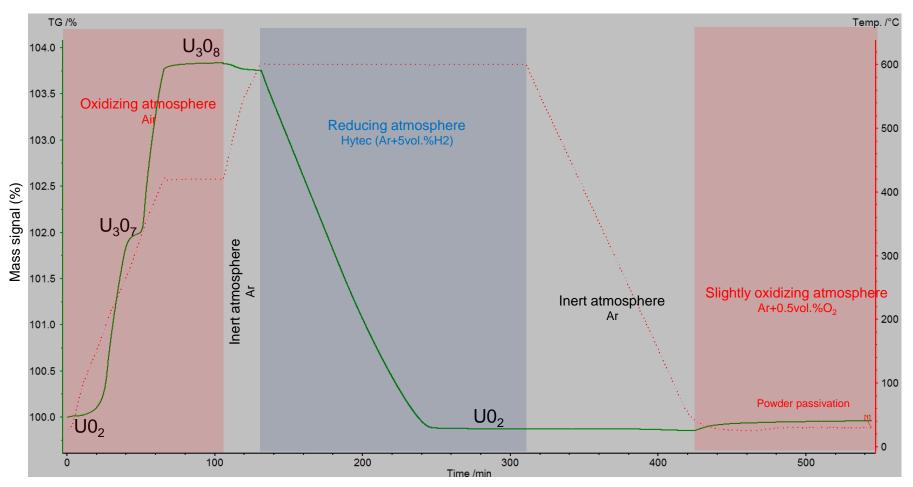
- Silmutaneous Thermal Analyser (STA) Netzsch Jupiter 449 F1
  - → <u>Thermogravimetry</u> (TGA): precise measure of sample mass change in function of time/temp in a controlled atmosphere
  - → <u>Differential Scanning Calorimetry</u> (DSC): precise measure of heat fluxes from sample in function of time/temp in a control, atm.
  - → <u>Mass Spectrometry</u> (MS): analyse the chemical composition of the evolved gases during test



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#### Fuel lab introduction

U02 → U3O8 → UO2 transformation followed by TG



#### Fuel lab introduction

- Complete laboratory for fuel pellet fabrication from powder to pellet (U and Th)
- Coming: new laboratory for MOX fuel processing (everything in glove boxes).
- Complimented by capabilities for full post-processing characterization
  - XAS through collaboration
  - In-house: SEM, FIB, TEM, XRD, Raman (coming),...

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# Collaboration possibilities

- We're always looking for bright students!
- Internships, MSc, and PhD program through SCK•CEN Academy: http://academy.sckcen.be/
- Internships and MSc can be tailored (in case of interest contact directly by email).
- Possibility for a small financial support + affordable housing during your stay.
- For PhDs, there's a yearly competition (deadline for applications ~ end of March).
  - University promotor is always needed
  - Possibility for full SCK•CEN funding

