

ECEN321 : Engineering Statistics

Assignment 4 Submission

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Measurement Error

1. (Navidi 3.1.2)
 - (a) The first thermometer covers the range 16.2 to 16.6 and the second thermometer covers 16.7 to 16.9. Thus the first thermometer shows greater accuracy.
 - (b) The second thermometer as a lower uncertainty, there is the more precise measurement.
2. (Navidi 3.1.8)
 - (a) Uncertainty = $s = \pm 0.6$
 - (b) Bias = 26.18

Linear Combinations of Measurements

3. (Navidi 3.2.2)
$$\sigma_X = \frac{\sigma}{\sqrt{N}}$$
$$N = \left(\frac{\sigma}{\sigma_X}\right)^2 = \left(\frac{1.5}{0.5}\right)^2 = 9$$
4. (Navidi 3.2.6)
$$C = \frac{20.00 - 19.90}{2} = 0.05$$
$$\sigma_C = \sigma_{\frac{h+p}{2}} = \sqrt{\frac{1}{4}0.01^2 + \frac{1}{4}0.02^2} = 0.01118$$

Uncertainties for Functions of One Measurement

5. (Navidi 3.3.4) $T = 300 \pm 0.4$
$$V = 20.04\sqrt{T} = 347.102982$$
$$\frac{dV}{dR} = \frac{10.02}{\sqrt{T}}$$
$$\sigma_V \approx \left|\frac{10.02}{\sqrt{T}}\right| \cdot \sigma_T = \left|\frac{10.02}{\sqrt{300}}\right| \cdot 0.4 = 0.231401987891$$
$$V = 347.1 \pm 0.23 \text{ m/s}$$

Uncertainties for Functions of Several Measurements

6. (Navidi 3.4.2) $V = \frac{\pi}{3}r^2h$, $r = 5.00 \pm 0.02 \text{ cm}$, $h = 6.00 \pm 0.01 \text{ cm}$

(a) $V = \frac{\pi}{3}25 \cdot 6 = 157.079633$

$$\frac{\delta V}{\delta r} = \frac{2\pi}{3}rh, \quad \frac{\delta V}{\delta h} = \frac{\pi}{3}r^2$$

$$\sigma_V = \sqrt{\left(\frac{\pi}{6}rh\right)^2\sigma_r^2 + \left(\frac{\pi}{3}r^2\right)^2\sigma_h^2} = \sigma_V = \sqrt{\left(\frac{\pi}{6}5 \cdot 6\right)^2 0.02^2 + \left(\frac{\pi}{3}25\right)^2 0.01^2} = 1.28361817673$$

$$V = 157 \pm 1.28 \text{ cm}^3$$

(b) $\sqrt{\left(\frac{\pi}{6}5 \cdot 6\right)^2 0.01^2 + \left(\frac{\pi}{3}25\right)^2 0.01^2} = 0.680678408278$

$$\sqrt{\left(\frac{\pi}{6}5 \cdot 6\right)^2 0.02^2 + \left(\frac{\pi}{3}25\right)^2 0.005^2} = 1.26343635931$$

\therefore reducing r to 0.01

7. (Navidi 3.4.14) $R = kl/d^2$, $l = 14.0 \pm 0.1 \text{ cm}$, $d = 4.4 \pm 0.1 \text{ cm}$

(a) $R = 14/4.4^2 k = \frac{16k}{19.36}$

$$\frac{\delta R}{\delta l} = k/d^2 = k/4.4^2 = \frac{k}{19.36}$$

$$\frac{\delta R}{\delta d} = -2kl/d^3 = \frac{-28k}{85.184}$$

$$\sigma_R = \sqrt{\left(\frac{k}{19.36}\right)^2 0.1^2 + \left(\frac{-28k}{85.184}\right)^2 0.1^2} = 0.03327k$$

$$R = \frac{16k}{19.36} \pm 0.033\Omega$$

(b) $\sigma_l = 0.05 \rightarrow \sigma_R = 0.0329k$

$$\sigma_d = 0.05 \rightarrow \sigma_R = 0.0172k$$

\therefore reducing d 's uncertainty