Here are the **answers** for all the questions based on the chapter "**Air and Atmosphere**" from your textbook.

# **SECTION A**

- (1) Multiple Choice Questions (MCQs)
  - 1. b) Nitrogen
  - 2. a) 21%
  - 3. b) Oxygen
  - 4. b) Carbon dioxide
  - 5. c) Greenhouse gases
  - 6. b) Carbon dioxide
  - 7. a) Helium
  - 8. b) CO<sub>2</sub>
  - 9. b) Nitrogen
  - 10. c) Nitrogen
- (2) Fill in the Blanks
- 11. gases
- 12. Nitrogen, Oxygen
- 13. **78**%
- 14. Oxygen
- 15. Carbon dioxide, Oxygen
- 16. Carbon dioxide
- 17. Carbon dioxide
- 18. Nitrogen fixation
- 19. Air pollution
- 20. Neon
- (3) True or False
- 21. False
- 22. **True**

- 23. False
- 24. True
- 25. **True**
- 26. **True**
- 27. False
- 28. **True**
- 29. **True**
- 30. False

### SECTION B

- (4) Odd One Out (Give Reason)
- 31. Carbon Dioxide (Others are major components of air, CO<sub>2</sub> is a trace gas)
- 32. Oxygen (Others are noble gases, oxygen is not)
- 33. Oxygen (Others are pollutants, oxygen is essential for life)
- 34. Evaporation (It is a physical change, others are chemical processes)
- 35. Oxygen (Others are greenhouse gases, oxygen is not)
- 36. Combustion (Others are environmental issues, combustion is a process)
- 37. Hydrogen (Others are major components of air, hydrogen is a trace gas)
- 38. Photosynthesis (Others are types of pollution, photosynthesis is a biological process)
- 39. Fossil Fuels (Others are natural processes, fossil fuel burning is man-made)
- 40. Respiration (Others involve nitrogen cycle, respiration does not)
- (5) Matching Questions
- 41. a) Used in photosynthesis
- 42. b) Most abundant gas in the atmosphere
- 43. c) Supports combustion
- 44. d) Used in light bulbs
- 45. e) Used in weather balloons
- 46. f) Greenhouse gas
- 47. g) Used in advertising lights
- 48. h) Causes acid rain
- 49. i) Toxic gas

- 50. j) Protects from UV radiation
- (6) Name the Type of Reaction
- 51. Combustion
- 52. Photosynthesis
- 53. Oxidation (Rusting)
- 54. Acid rain formation
- 55. Combustion
- 56. Oxidation of nitrogen
- 57. Physical change
- 58. Water cycle
- 59. Nitrogen fixation
- 60. Physical change

#### SECTION C

- (7) Short Answer Questions
- 61. Major components of air are nitrogen (78%), oxygen (21%), carbon dioxide (0.03%), argon (0.9%), and other trace gases.
- 62. Nitrogen is essential for plant growth and is a major component of proteins and DNA.
- 63. Oxygen is required for respiration, which provides energy for living organisms.
- 64. The greenhouse effect is the trapping of heat by greenhouse gases, keeping the Earth warm.
- 65. Global warming is caused by excessive greenhouse gases leading to temperature rise.
- 66. Air pollution can cause respiratory diseases, lung infections, and global climate change.
- 67. Acid rain is rainwater that contains acidic pollutants such as sulfuric acid and nitric acid.
- 68. Nitrogen is fixed by bacteria, converting it into ammonia, nitrates, and nitrites for plant use.
- 69. Carbon monoxide binds with hemoglobin, reducing oxygen transport in the body, leading to poisoning.
- 70. Inert gases are non-reactive gases. Examples: Argon, Neon.

## (8) Diagram-Based Questions

(Refer to textbook for labeled diagrams.)

- 71. Labeled diagram of nitrogen cycle
- 72. Illustration of photosynthesis
- 73. Diagram showing greenhouse effect
- 74. Setup showing the presence of oxygen in air
- 75. Flowchart of acid rain formation

### (9) Reasoning-Based Questions

- 76. Nitrogen is required by plants for making proteins and is obtained through nitrogen fixation.
- 77. Carbon dioxide is needed for photosynthesis, where plants use it to produce glucose and oxygen.
- 78. Oxygen is needed for combustion because it supports burning by reacting with fuel.
- 79. Deforestation reduces the absorption of CO<sub>2</sub>, leading to an increase in greenhouse gases.
- 80. Air is a mixture because its components retain their properties and are not chemically combined.
- (10) Chemical Formula & Symbol-Based Questions
- 81. CO<sub>2</sub> (Carbon dioxide)
- 82. O<sub>2</sub> (Oxygen gas)
- 83. 4Fe +  $3O_2$  +  $6H_2O \rightarrow 4Fe(OH)_3$  (Rusting of iron)
- 84.  $6CO_2 + 6H_2O \rightarrow C_6H_{12}O_6 + 6O_2$  (Photosynthesis)
- 85. N<sub>2</sub> (Nitrogen gas)

This completes the answers for all the questions. Let me know if you need any modifications!  $\mathscr{A}$