

X857/76/11

Physics Paper 2 — Relationships sheet

WEDNESDAY, 17 MAY 10:15 AM – 12:30 PM





# Relationships required for Physics Higher

$d = \overline{v}t$	W = QV	$V_{rms} = \frac{V_{peak}}{\sqrt{2}}$					
$S = \overline{v}t$	$E = mc^2$	V Z					
v = u + at	$I = \frac{P}{A}$	$I_{rms} = \frac{I_{peak}}{\sqrt{2}}$					
$s = ut + \frac{1}{2}at^2$	A	1					
$v^2 = u^2 + 2as$	$I = \frac{k}{d^2}$	$T = \frac{1}{f}$					
$s = \frac{1}{2}(u+v)t$	$I_1 d_1^2 = I_2 d_2^2$	V = IR					
F = ma	E = hf	$P = IV = I^2 R = \frac{V^2}{R}$					
W = mg	$E_k = hf - hf_0$	$R_T = R_1 + R_2 + \dots$					
$E_w = Fd$ , or $W = Fd$	$v = f\lambda$						
$E_p = mgh$	$E_2 - E_1 = hf$	$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$					
$E_k = \frac{1}{2}mv^2$	$d\sin\theta=m\lambda$	$V_1 = \left(\frac{R_1}{R_1 + R_2}\right) V_S$					
$P = \frac{E}{t}$	$n = \frac{\sin \theta_1}{\sin \theta_2}$	( 1 Z)					
p = mv	2	$\frac{V_1}{V_2} = \frac{R_1}{R_2}$					
Ft = mv - mu	$\frac{\sin \theta_1}{\sin \theta_2} = \frac{\lambda_1}{\lambda_2} = \frac{v_1}{v_2}$	E = V + Ir					
$F = G \frac{m_1 m_2}{r^2}$	$\sin \theta_c = \frac{1}{n}$	$C = \frac{Q}{V}$					
$t' = \frac{t}{}$		Q = It					
$t' = \frac{t}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$		$E = \frac{1}{2}QV = \frac{1}{2}CV^2 = \frac{1}{2}\frac{Q^2}{C}$					
$l' = l \sqrt{1 - \left(\frac{v}{c}\right)^2}$							
$f_o = f_s \left( \frac{v}{v \pm v_s} \right)$ $z = \frac{\lambda_{observed} - \lambda_{rest}}{\lambda_{rest}}$	path difference = $m\lambda$ or $\left(m+\frac{1}{2}\right)\lambda$ where $m=0,1,2$						
	$random uncertainty = \frac{max.value - min.value}{number of values}$						
	or	er of values					
$z = \frac{v}{c}$	$\Delta R = \frac{R_{\text{max}} - R_{\text{min}}}{n}$						
$v = H_0 d$							

## Additional relationships

### Circle

circumference =  $2\pi r$ 

$$area = \pi r^2$$

# Sphere

area = 
$$4\pi r^2$$

volume = 
$$\frac{4}{3}\pi r^3$$

## Trigonometry

$$\sin \theta = \frac{\text{opposite}}{\text{hypotenuse}}$$

$$\cos \theta = \frac{\text{adjacent}}{\text{hypotenuse}}$$

$$\tan\theta = \frac{\text{opposite}}{\text{adjacent}}$$

$$\sin^2\theta + \cos^2\theta = 1$$

# Electron arrangements of elements

		87 <b>Fr</b> 2,8,18,32, 18,8,1 Francium	55 <b>Cs</b> 2,8,18,18, 8,1 Caesium	37 <b>Rb</b> 2,8,18,8,1 Rubidium	2,8,8,1 Potassium	Sodium	Na 11	2,1 Lithium	. <b>.</b> .	Hydrogen	Group 1 (1)
Lanthanides Actinides	88 <b>Ra</b> 2,8,18,32, 18,8,2 Radium	56 <b>Ba</b> 2,8,18,18, 8,2 Barium	38 <b>Sr</b> 2,8,18,8,2 Strontium	20 <b>Ca</b> 2,8,8,2 Calcium	2,8,2 Magnesium	12 <b>Mg</b>	2,2 Beryllium	<b>В</b> 4	(2)	Group 2	
	89 <b>Ac</b> 2,8,18,32, 18,9,2 Actinium	57 <b>La</b> 2,8,18,18, 9,2 Lanthanum	39 <b>Y</b> 2,8,18,9,2 Yttrium	27 <b>Sc</b> 2,8,9,2 Scandium	(3)						
89 <b>Ac</b> 2,8,18,32, 18,9,2 Actinium	57 <b>La</b> 2,8,18, 18,9,2 Lanthanum	104 <b>Rf</b> 2,8,18,32, 32,10,2 Rutherfordium	72 <b>Hf</b> 2,8,18,32, 10,2 Hafnium	40 <b>Zr</b> 2,8,18, 10,2 Zirconium	2,8,10,2 Titanium	(4)				Key	ξ.
90 <b>Th</b> 2,8,18,32, 18,10,2 Thorium	58 <b>Ce</b> 2,8,18, 20,8,2 Cerium	105 <b>Db</b> 2,8,18,32, 32,11,2 Dubnium	73 <b>Ta</b> 2,8,18, 32,11,2 Tantalum	41 <b>Nb</b> 2,8,18, 12,1 Niobium	23 <b>V</b> 2,8,11,2 Vanadium	(5)			[	Ato	
91 <b>Pa</b> 2,8,18,32, 20,9,2 Protactinium	59 <b>Pr</b> 2,8,18,21, 8,2 Praseodymium	106 <b>Sg</b> 2,8,18,32, 32,12,2 Seaborgium	74 <b>W</b> 2,8,18,32, 12,2 Tungsten	42 <b>Mo</b> 2,8,18,13, 1 Molybdenum	24 <b>Cr</b> 2,8,13,1 Chromium	(6)			Name	Atomic number Symbol Flectron arrangement	
92 <b>U</b> 2,8,18,32, 21,9,2 Uranium	60 <b>Nd</b> 2,8,18,22, 8,2 Neodymium	107 <b>Bh</b> 2,8,18,32, 32,13,2 Bohrium	75 <b>Re</b> 2,8,18,32, 13,2 Rhenium	43 <b>Tc</b> 2,8,18,13, 2 Technetium	25 <b>Mn</b> 2,8,13,2 Manganese		Transition elements		(	ber ement	J v
93 <b>Np</b> 2,8,18,32, 22,9,2 Neptunium	61 <b>Pm</b> 2,8,18,23, 8,2 Promethium	108 <b>Hs</b> 2,8,18,32, 32,14,2 Hassium	76 <b>Os</b> 2,8,18,32, 14,2 Osmium	44 <b>Ru</b> 2,8,18,15, 1 Ruthenium	26 <b>Fe</b> 2,8,14,2 Iron	(8)	ı element				
94 <b>Pu</b> 2,8,18,32, 24,8,2 Plutonium	62 Sm 2,8,18,24, 8,2 Samarium	109 <b>Mt</b> 2,8,18,32, 32,15,2 Meitnerium	77 Ir 2,8,18,32, 15,2 Iridium	45 <b>Rh</b> 2,8,18,16, 1 Rhodium	<b>Co</b> 2,8,15,2 Cobalt	(9)	6				
95 <b>Am</b> 2,8,18,32, 25,8,2 Americium	63 <b>Eu</b> 2,8,18,25, 8,2 Europium	110 <b>Ds</b> 2,8,18,32, 32,17,1 Darmstadtium	78 <b>Pt</b> 2,8,18,32, 17,1 Platinum	46 <b>Pd</b> 2,8,18, 18,0 Palladium	28 <b>Ni</b> 2,8,16,2 Nickel	(10)					,
96 <b>Cm</b> 2,8,18,32, 25,9,2 Curium	64 <b>Gd</b> 2,8,18,25, 9,2 Gadolinium	<b>Rg</b> 2,8,18,32, 32,18,1 Roentgenium	79 <b>Au</b> 2,8,18, 32,18,1 Gold	47 <b>Ag</b> 2,8,18, 18,1 Silver	29 <b>Cu</b> 2,8,18,1 Copper	(11)					
97 <b>Bk</b> 2,8,18,32, 27,8,2 Berkelium	65 <b>Tb</b> 2,8,18,27, 8,2 Terbium	110 111 112  Ds Rg Cn 2,8,18,32, 2,8,18,32, 2,8,18,32, 32,17,1 32,18,1 32,18,2  Darmstadtium Roentgenium Copernicium	80 <b>Hg</b> 2,8,18, 32,18,2 Mercury	48 <b>Cd</b> 2,8,18, 18,2 Cadmium	2,8,18,2 Zinc	(12)					
98 <b>Cf</b> 2,8,18,32, 28,8,2 Californium	66 <b>Dy</b> 2,8,18,28, 8,2 Dysprosium		81 <b>T</b> ( 2,8,18, 32,18,3 Thallium	49 <b>In</b> 2,8,18, 18,3 Indium	37 <b>Ga</b> 2,8,18,3 Gallium	Aluminium	<b>A</b> 13	Boron	<b>, —</b> "	(13)	Group 3
99 <b>Es</b> 2,8,18,32, 29,8,2 Einsteinium	67 <b>Ho</b> 2,8,18,29, 8,2 Holmium		82 <b>Pb</b> 2,8,18, 32,18,4 h Lead	50 <b>Sn</b> 2,8,18, 18,4 Tin	32 <b>Ge</b> 3 2,8,18,4 1 Germanium	S:	14 Si	2,4 Carbon		(14)	3 Group 4
100 Fm 2,8,18,32, 30,8,2 Fermium	68 <b>Er</b> 2,8,18,30, 8,2 Erbium		83 <b>Bi</b> 2,8,18, 4 32,18,5 6 Bismuth	51 <b>Sb</b> , 2,8,18, 18,5 Antimony	<b>AS</b> 4 2,8,18,5  A 2,8,18,5	Pho:	<b>7 7</b> 15	Nitrogen	, z v	(15)	4 Group 5
101 <b>Md</b> 2,8,18,32, 31,8,2 Mendelevium	69 <b>Tm</b> 2,8,18,31, 8,2 Thulium		84 <b>Po</b> 2,8,18, 32,18,6	52 <b>Te</b> 2,8,18, 18,6 Tellurium	34 <b>Se</b> 5 2,8,18,6 Selenium	S	3 <b>S</b>	n Oxygen	. 0 ∞	(16)	5 Group 6
102 <b>No</b> 2,8,18,32, 32,8,2 Nobelium	70 <b>Yb</b> 2,8,18,32, 8,2 Ytterbium		85 <b>At</b> 2,8,18, 6 32,18,7 M Astatine	53 	8r 6 2,8,18,7 n Bromine	유 ,		1 Fluorine	, <b>ग</b> ०	(17)	6 Group 7
103 <b>Lr</b> 2,8,18,32, 32,9,2 Lawrencium	71 <b>Lu</b> 2,8,18,32, 9,2 Lutetium					<b>\</b>	18 <b>Ar</b>	e Neon	<b>Z</b> 10	He 2	
103 <b>Lr</b> \$,18,32, \$2,9,2 rencium	71 <b>Lu</b> \$,18,32, 9,2 retium		86 <b>Rn</b> 2,8,18, 32,18,8 Radon	54 <b>Xe</b> 2,8,18, 18,8 Xenon	2,8,18,8 Krypton	Argon	18 <b>Ar</b>	2,8 Neon	2 <b>Z</b> 10	Helium	Group 0 (18)