

# Physics Collectanea

A Personal Compendium of Physics Notes

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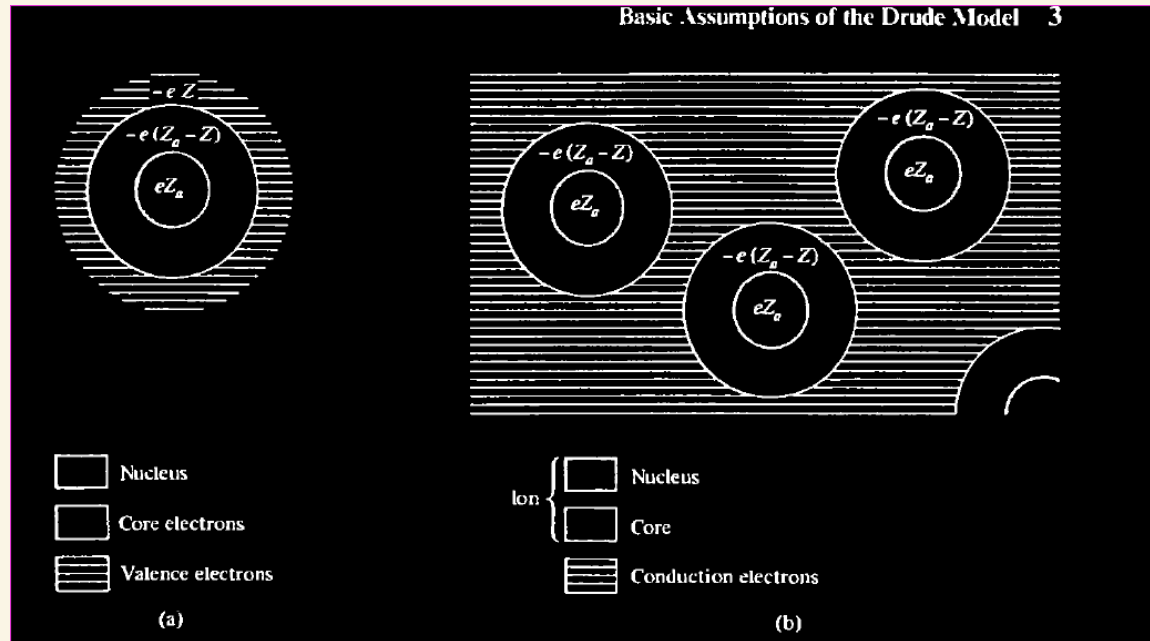
Part I

Condensed Matter

# Chapter 1

## Basic Models of Metals

### 1.1 The Drude Model



The Drude model assumes that we have immovable nucleus with most external electrons free. A nucleus of some element, has  $Z$  protons, and subsequently  $Z$  electrons with total charge  $-eZ$ .

Let's try to derive some things. We know that electrical force  $\vec{F} = -e\vec{E}$  and  $\vec{F} = m\vec{a}$ . Thus we have:

$$\vec{a} = \frac{-eE}{m} \quad (1.1)$$

After a collision, the electron accelerates for a time  $t$ , gaining velocity:

$$\Delta \vec{v} = \vec{a} \cdot t = \frac{-eEt}{m} \quad (1.2)$$

So the final velocity is:

$$\vec{v}(t) = \vec{v}_0(t) + \Delta \vec{v} \quad (1.3)$$

Well, we assume that the collision randomize  $\vec{v}_0$  (it's isotropic), its **average is zero**:

$$\langle \vec{v}_0 \rangle = 0 \rightarrow \langle \vec{v}(t) \rangle = \frac{-eE\langle t \rangle}{m}$$

We know that  $\mathbf{j} = -nev$ , so we have:

$$\mathbf{j} = -nev_{avg} = -ne \left( \frac{-eE\tau}{m} \right) = \left( \frac{ne^2\tau}{m} \right) E \quad (1.4)$$

Let's define the **conductivity** ( $\sigma$ ) that tell us how easily a material allows electric charges to move when you apply an electric field (that's created when you make a differential in potential). Then:

$$\mathbf{j} = \sigma \mathbf{E} \quad (1.5)$$

Notice that conductivity is the inverse of resistivity:

$$\sigma = \frac{1}{\rho} \quad (1.6)$$

We can determine the relaxation time  $\tau$  usgin the resistivity:

$$\tau = \frac{m}{\rho ne^2} \quad (1.7)$$