

Table 1. Temperature and Directional Dependent Aerosols added to the POSEIDON Aerosol Database

Chemical Formula (1)	Reference (2)	Polymorph (3)	T (K) (4)	Direction (5)	Notes (6)	(Min, Max) μm (7)
Hibonite						
CaAl ₁₂ O ₁₉	H. Mutschke et al. (2002) ^D	Crystal		Extraordinary ($E \parallel c$)	Natural	(2.00,30)
CaAl ₁₂ O ₁₉	H. Mutschke et al. (2002) ^D	Crystal		Ordinary ($E \perp c$)	Natural	(2.00,30)
Corundum						
Al ₂ O ₃	B. Begemann et al. (1997) ^D	Amorphous			Compact	(7.81, 30)
Al ₂ O ₃	B. Begemann et al. (1997) ^D	Amorphous			Porous	(7.81, 30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	300K	Extraordinary ($E \parallel c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	551K	Extraordinary ($E \parallel c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	738K	Extraordinary ($E \parallel c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	928K	Extraordinary ($E \parallel c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	300K	Ordinary ($E \perp c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	551K	Ordinary ($E \perp c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	738K	Ordinary ($E \perp c$)	Synthetic	(6.67,30)
Al ₂ O ₃	S. Zeidler et al. (2013) ^D	α Crystal	928K	Ordinary ($E \perp c$)	Synthetic	(6.67,30)
Spinel						
MgAl ₂ O ₄	D. Fabian et al. (2001b) ^D	Crystal			Natural	(2, 30)
MgAl ₂ O ₄	D. Fabian et al. (2001b) ^D	Crystal	1223K		Natural, Annealed	(1.67, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	10K		Synthetic	(7.70, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	100K		Synthetic	(7.70, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	300K		Synthetic	(6.71, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	551K		Synthetic	(6.71, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	738K		Synthetic	(6.71, 30)
MgAl ₂ O ₄	S. Zeidler et al. (2013) ^D	Crystal	928K		Synthetic	(6.71, 30)
Fayalite						
Fe ₂ SiO ₄	D. Fabian et al. (2001a) ^D	Crystal		$E \parallel z$ ($E \parallel c^*$)	Synthetic	(2, 30)
Fe ₂ SiO ₄	D. Fabian et al. (2001a) ^D	Crystal		$E \parallel y$ ($E \parallel b^*$)	Synthetic	(2, 30)
Fe ₂ SiO ₄	D. Fabian et al. (2001a) ^D	Crystal		$E \parallel x$ ($E \parallel a^*$)	Synthetic	(2, 30)
Titanium Dioxide						
TiO ₂	T. Posch et al. (2003) ^D	Anatase Crystal		Extraordinary ($E \parallel c$)		(2, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Anatase Crystal		Ordinary ($E \perp c$)		(2, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Rutile Crystal		Extraordinary ($E \parallel c$)		(0.47, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Rutile Crystal		Ordinary ($E \perp c$)		(0.47, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Brookite Crystal		$E \parallel z$ ($E \parallel c^*$)		(2, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Brookite Crystal		$E \parallel y$ ($E \parallel b^*$)		(2, 30)
TiO ₂	S. Zeidler et al. (2011) ^D					
TiO ₂	T. Posch et al. (2003) ^D	Brookite Crystal		$E \parallel x$ ($E \parallel a^*$)		(2, 30)
S. Zeidler et al. (2011) ^D						
Silicon Dioxide						
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	300K	Extraordinary ($E \parallel c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	551K	Extraordinary ($E \parallel c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	738K	Extraordinary ($E \parallel c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	833K	Extraordinary ($E \parallel c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	β Quartz Crystal	928K	Extraordinary ($E \parallel c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	300K	Ordinary ($E \perp c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	551K	Ordinary ($E \perp c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	738K	Ordinary ($E \perp c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	α Quartz Crystal	833K	Ordinary ($E \perp c$)	Natural (Brazil)	(6.26,30)
SiO ₂	S. Zeidler et al. (2013) ^D	β Quartz Crystal	928K	Ordinary ($E \perp c$)	Natural (Brazil)	(6.26,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	295K	A2-symmetry ($E \parallel c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	295K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	346K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	480K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	705K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	α Quartz Crystal	790K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1010K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1125K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1170K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1310K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1394K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1520K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1590K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Quartz Crystal	1646K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)
SiO ₂	D. D. S. Meneses et al. (2014)	β Cristobalite Crystal	1810K	E-symmetry ($E \perp c$)	Cut Crystal	(6.67,30)

Table 1 *continued*

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Chemical Formula (1)	Reference (2)	Polymorph (3)	T (K) (4)	Direction (5)	Notes (6)	(Min, Max) μm (7)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	300K	E y (E a)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	551K	E y (E a)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	738K	E y (E a)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	928K	E y (E a)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	10K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	100K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	200K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	300K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	551K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	738K	E x (E b)	Natural (Burma)	(6.71,30)
Mg _{0.92} Fe _{0.09} SiO ₃	S. Zeidler et al. (2015) ^D	Crystal	928K	E x (E b)	Natural (Burma)	(6.71,30)

NOTE—Aerosol refractive indices, alongside pre-computed Mie properties, added to the aerosol database featured in E. Mullens et al. (2024). Groups of aerosols: Hibonite, Corundum, Spinel, Fayalite, Titanium Dioxide, Silicon Dioxide, Olivine, Orthoenstatite. Chemical formula (1), reference to where refractive index data is from (2), polymorph (3), temperature (if applicable, 4), direction (if applicable, 5), notes (6), and minimum and maximum wavelength of refractive index (where the absolute minimum and maximum of the precomputed aerosol database is 0.2 and 30 μm , 6). ^D refers to refractive indices that can be found on the Database of Optical Constants for Cosmic Dust (DOCCD, <https://www2.astro.uni-jena.de/Laboratory/OCDB/index.html>). Note that special care must be taken for determining which notation correlates to E||a,b,c for (orthorhombic) biaxial crystals. For the Mg₂SiO₄ refractive indices from M. Eckes et al. (2013), it is defined that B1U = E||c, B2U = E||b, B3U = E||a (pers comm., D. D. S. Meneses). For the Mg_{1.72}Fe_{0.21}SiO₄ refractive indices E||x=E||c, E||y=E||b, E||z=E||a (assumed to be the same for Mg_{1.9}Fe_{0.1}SiO₄) and for the Mg_{0.92}Fe_{0.09}SiO₃ refractive indices E||x=E||b, E||y=E||a, E||z=E||c as defined on DOCCD. *We assume that both Fe₂SiO₄ in D. Fabian et al. (2001a) and TiO₂, Brookite in T. Posch et al. (2003) follow the same D_{2h} symmetry group as Mg₂SiO₄ (M. Eckes et al. 2013) and that E||z = B1U = E||c, E||y = B2U = E||b, E||x = B3U = E||a since neither text explicitly assigns their axes. Since in this work we only deal with orthorhombic biaxial crystals, we opt to use crystallographic axis notation (a,b,c) in lieu of optical direction notation (x,y,z). Note that monoclinic and triclinic crystal systems are biaxial and do not have optical directions corresponding to crystallographic axes.

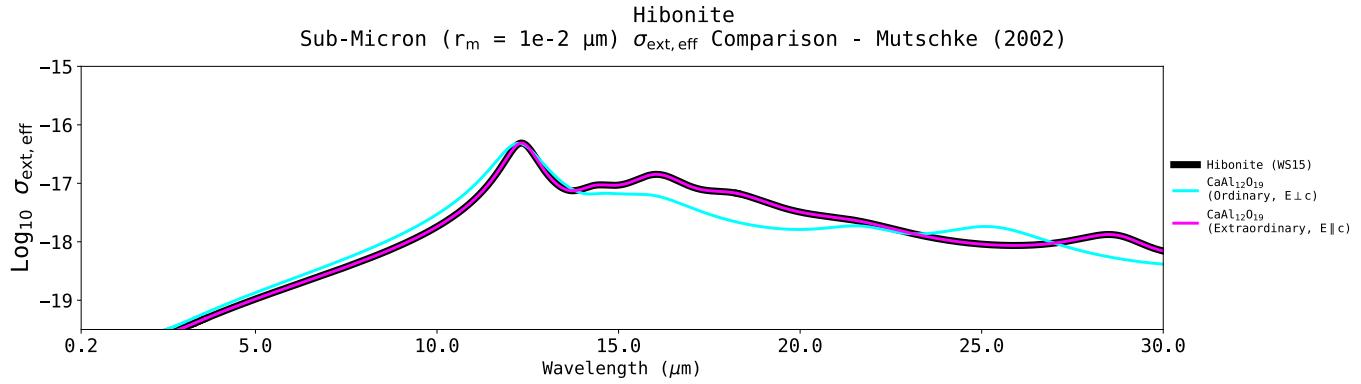


Figure 3. Comparison of effective extinction cross sections ($\sigma_{\text{ext,eff}}$, mean particle radii = 0.01 μm with a lognormal distribution) computed from refractive indices used extensively in exoplanet literature vs those computed from temperature and/or directional-specific refractive indices. In this figure: Hibonite (H. Mutschke et al. 2002).

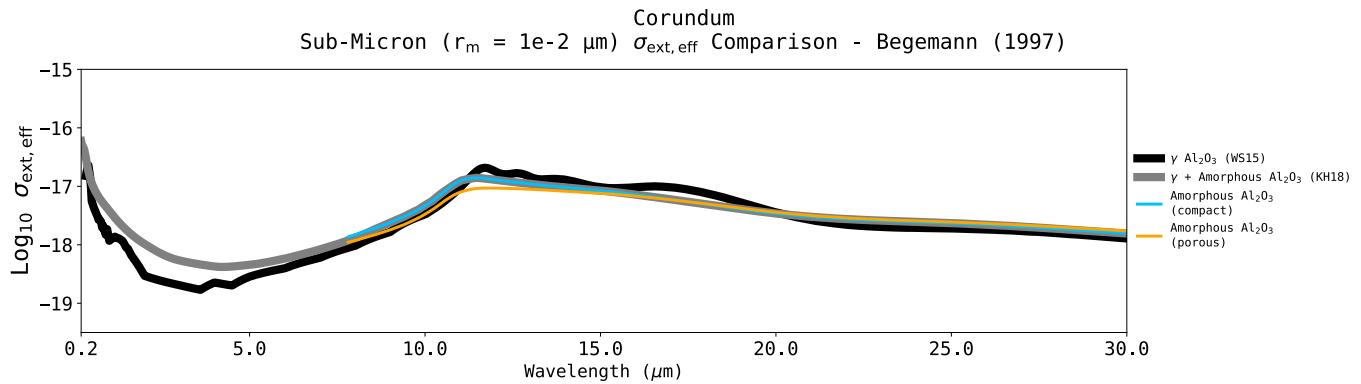


Figure 4. Same as Figure 3, but for Corundum (B. Begemann et al. 1997).

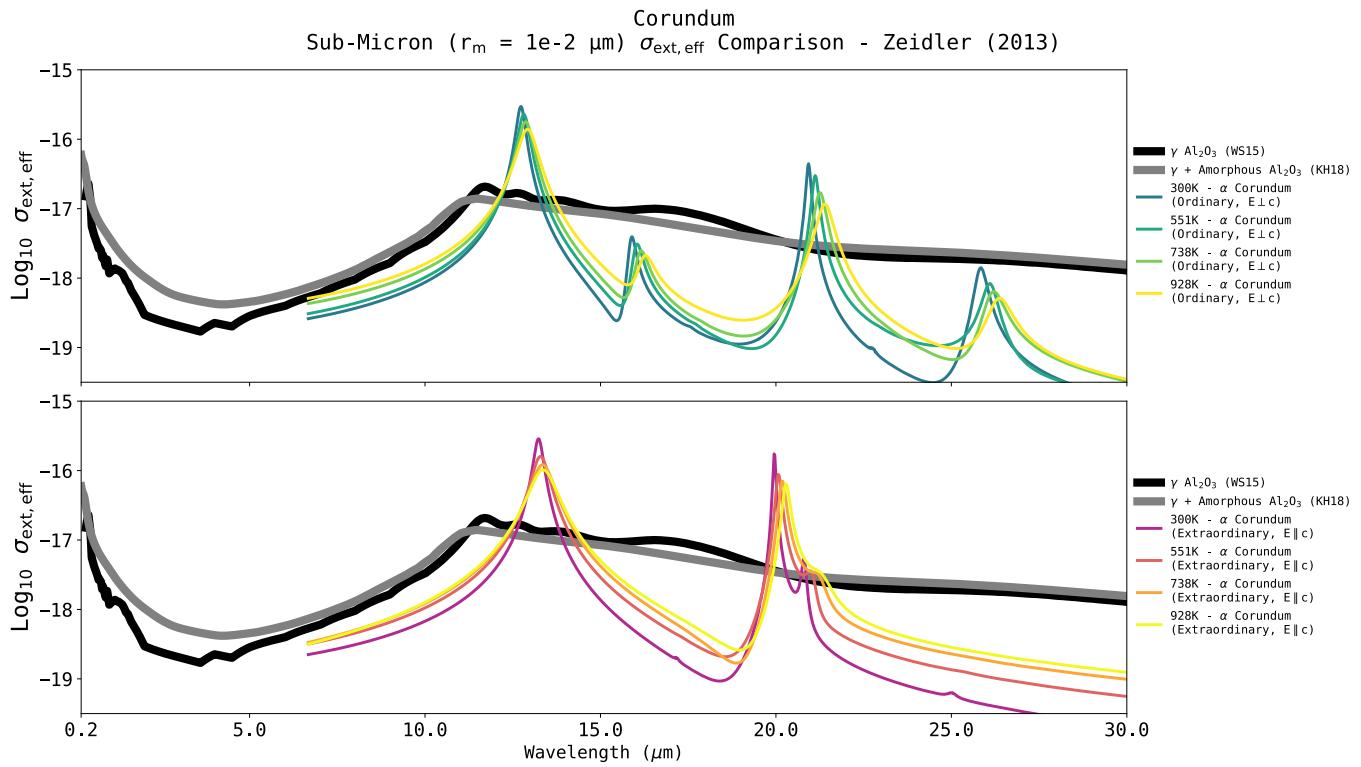


Figure 5. Same as Figure 3, but for Corundum (S. Zeidler et al. 2013).

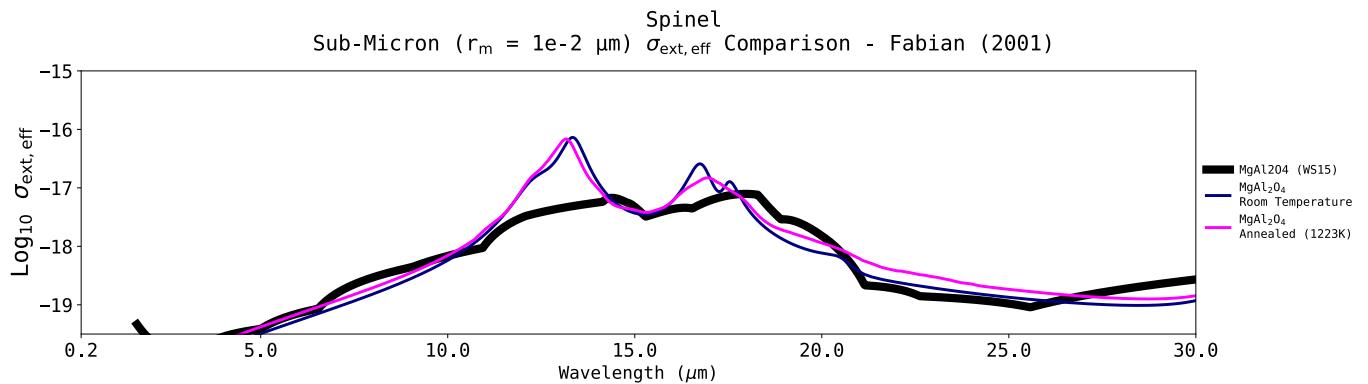


Figure 6. Same as Figure 3, but for Spinel (D. Fabian et al. 2001b).

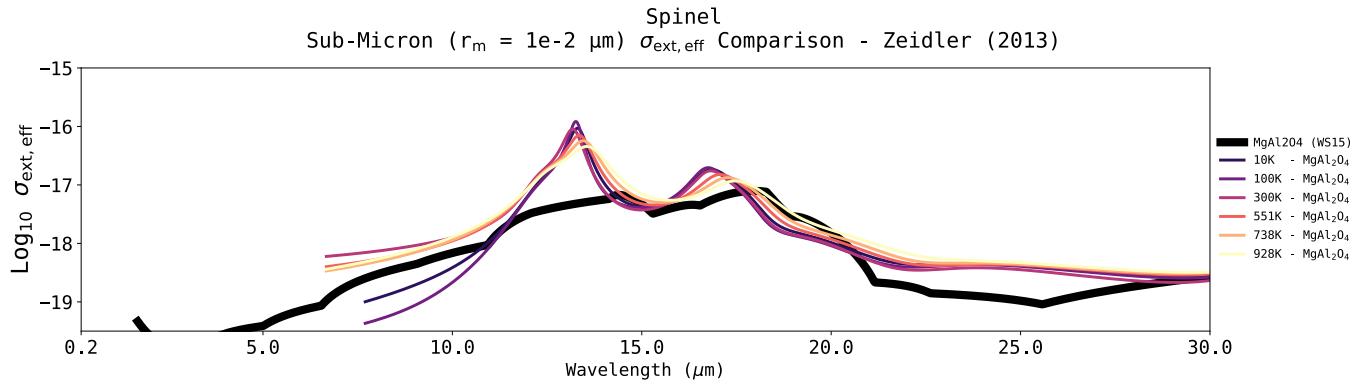


Figure 7. Same as Figure 3, but for Spinel (S. Zeidler et al. 2013).

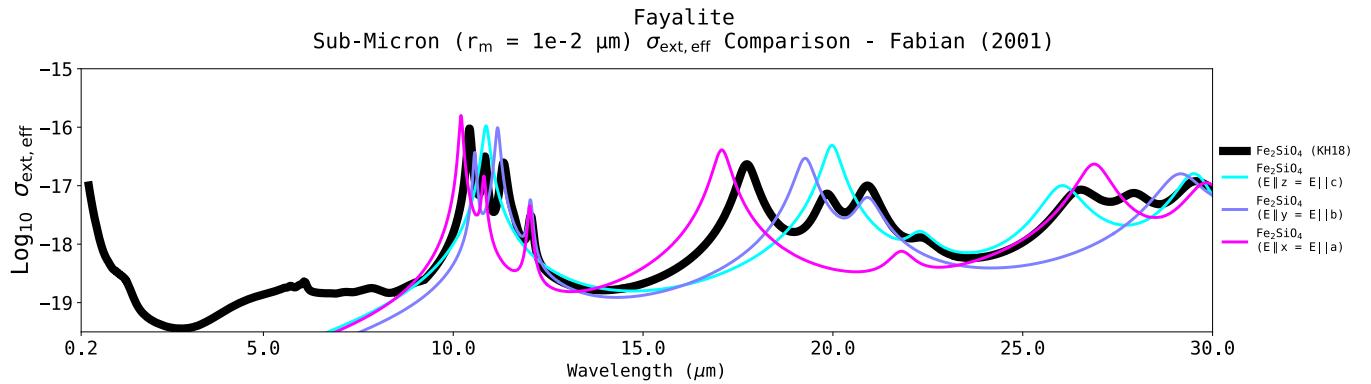


Figure 8. Same as Figure 3, but for Fayalite (D. Fabian et al. 2001a).

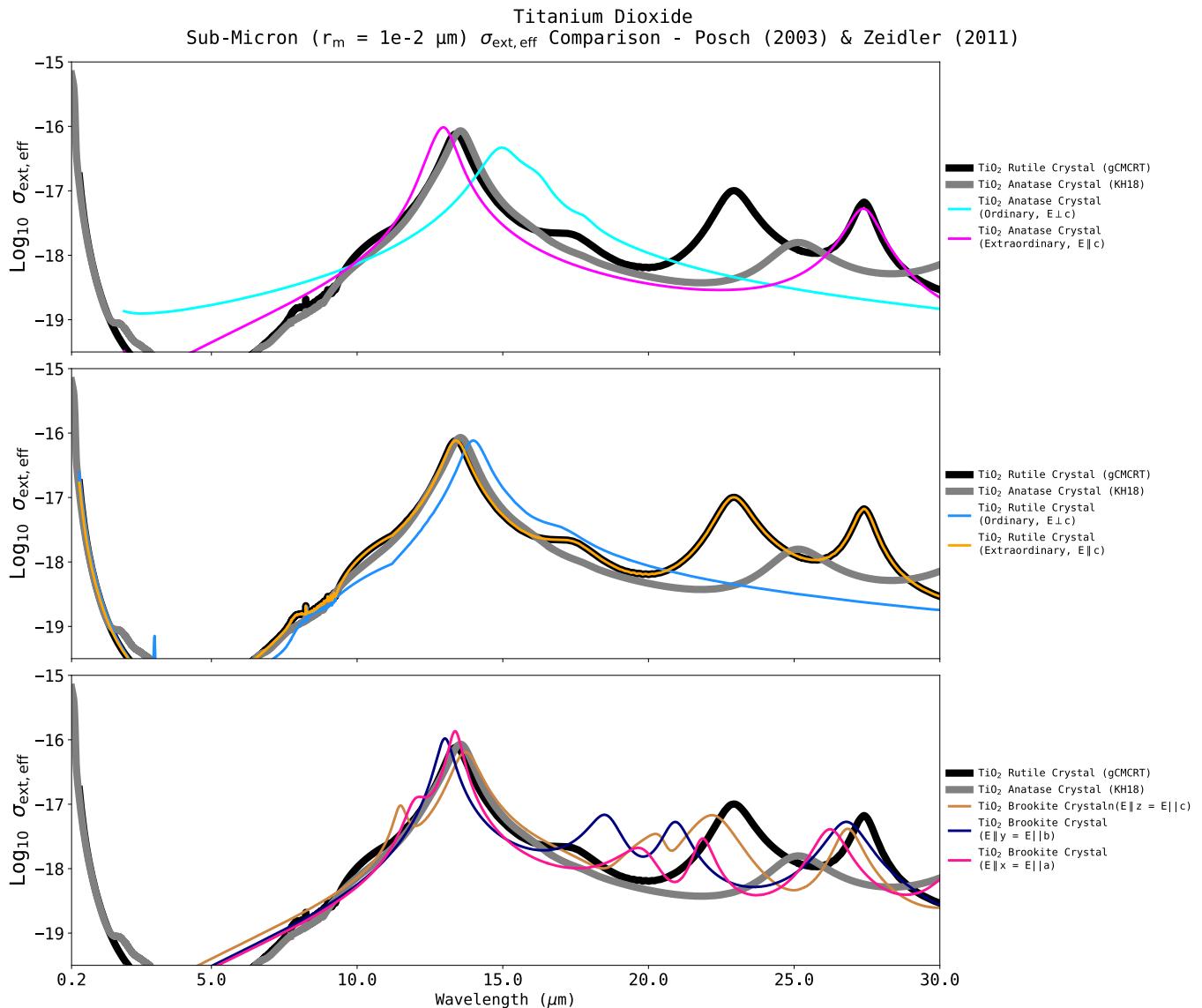


Figure 9. Same as Figure 3, but for Titanium-Dioxide (T. Posch et al. 2003; S. Zeidler et al. 2011).

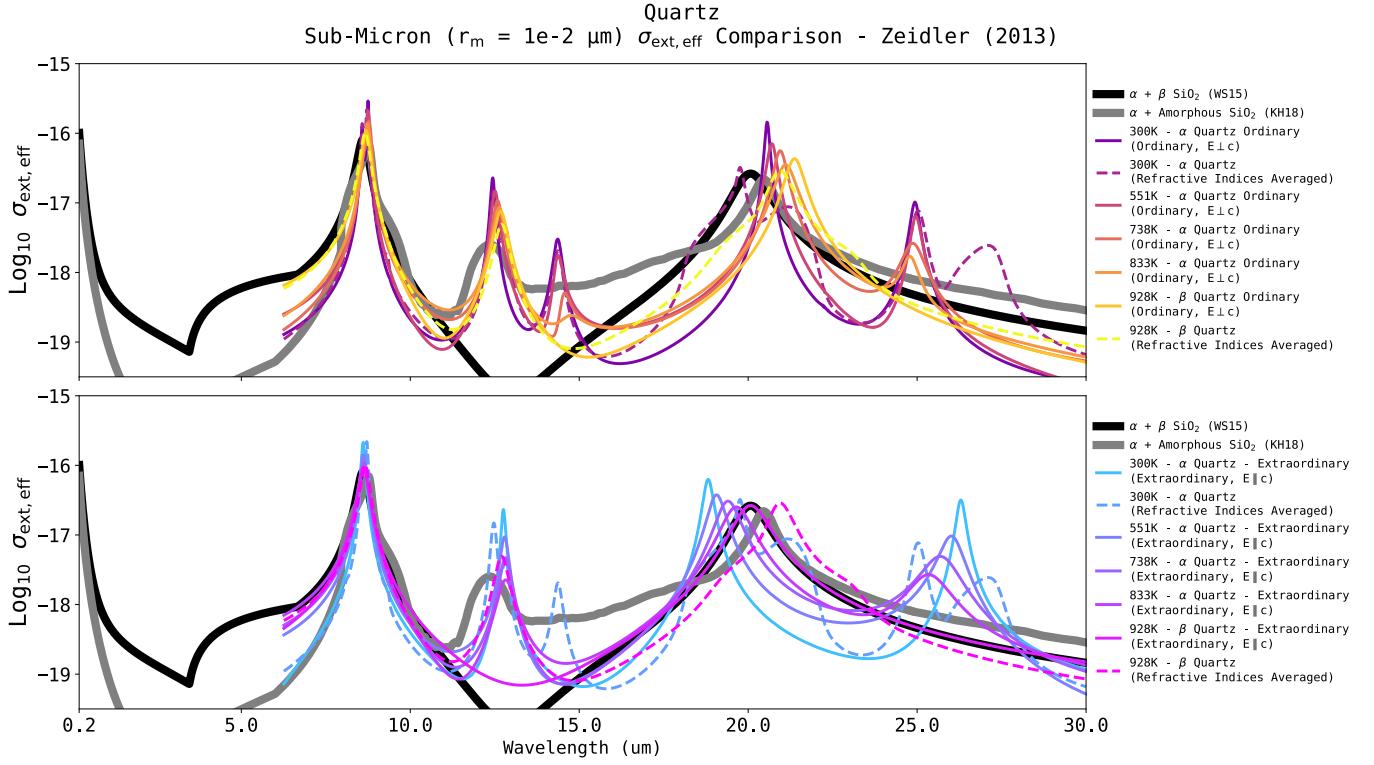


Figure 10. Same as Figure 3, but for Quartz (S. Zeidler et al. 2013).

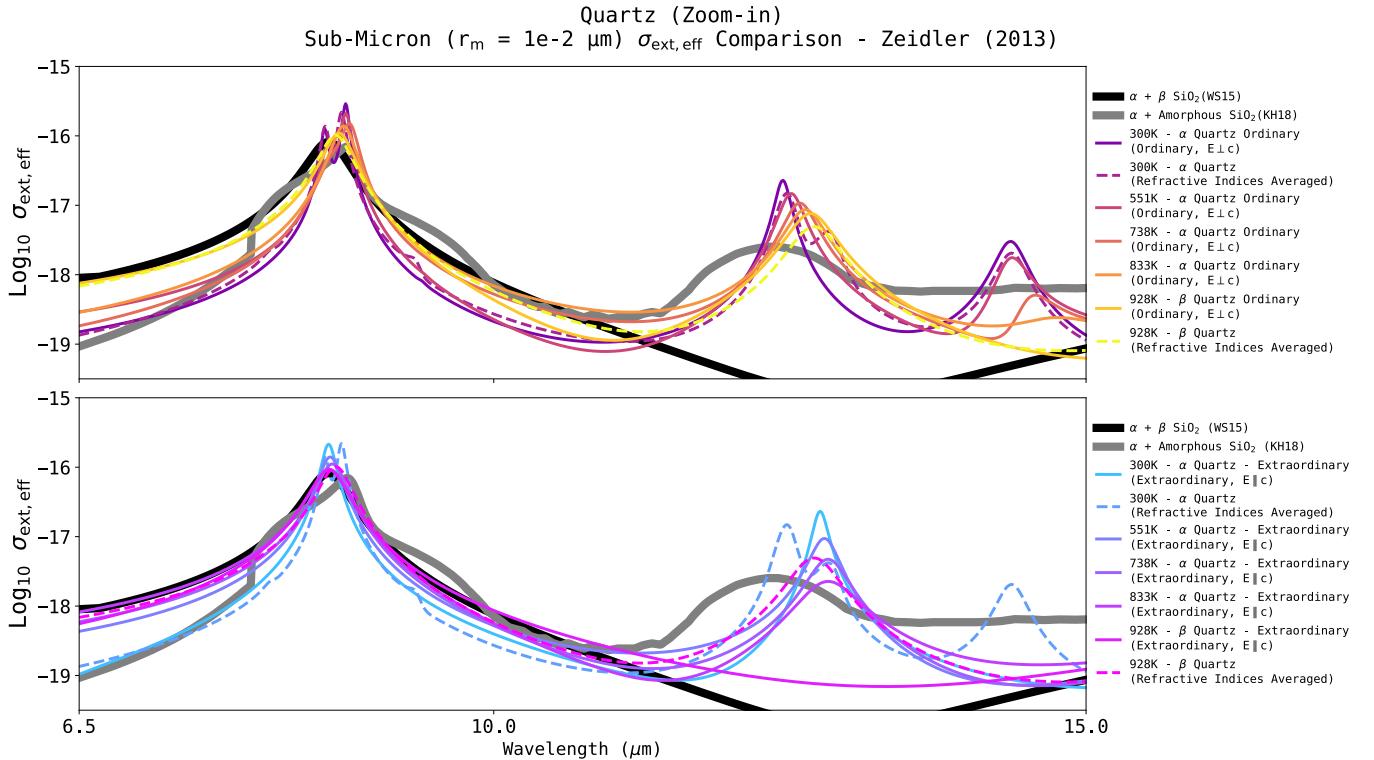


Figure 11. Same as Figure 3, but for Quartz (S. Zeidler et al. 2013), with a zoom-in on JWST MIRI-LRS wavelengths.

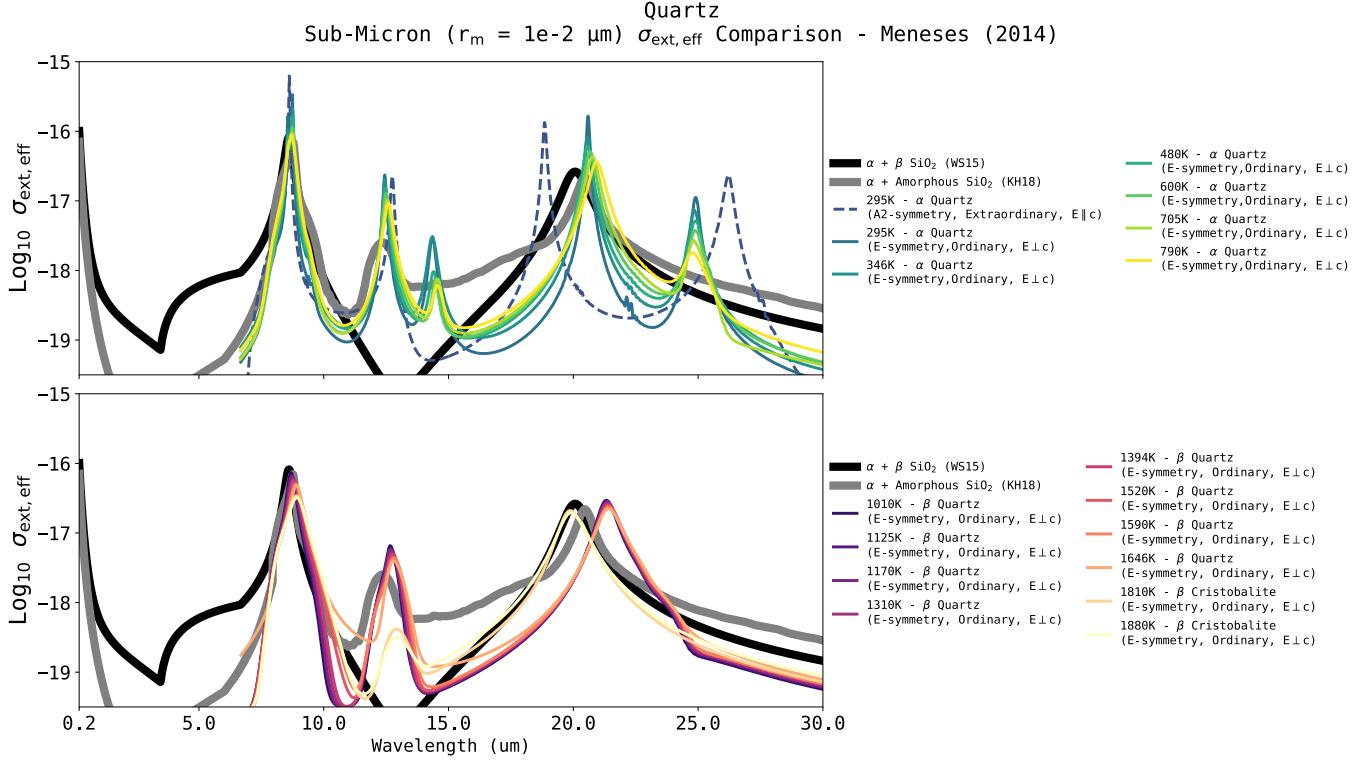


Figure 12. Same as Figure 3, but for Quartz and Cristobalite (D. D. S. Meneses et al. 2014).

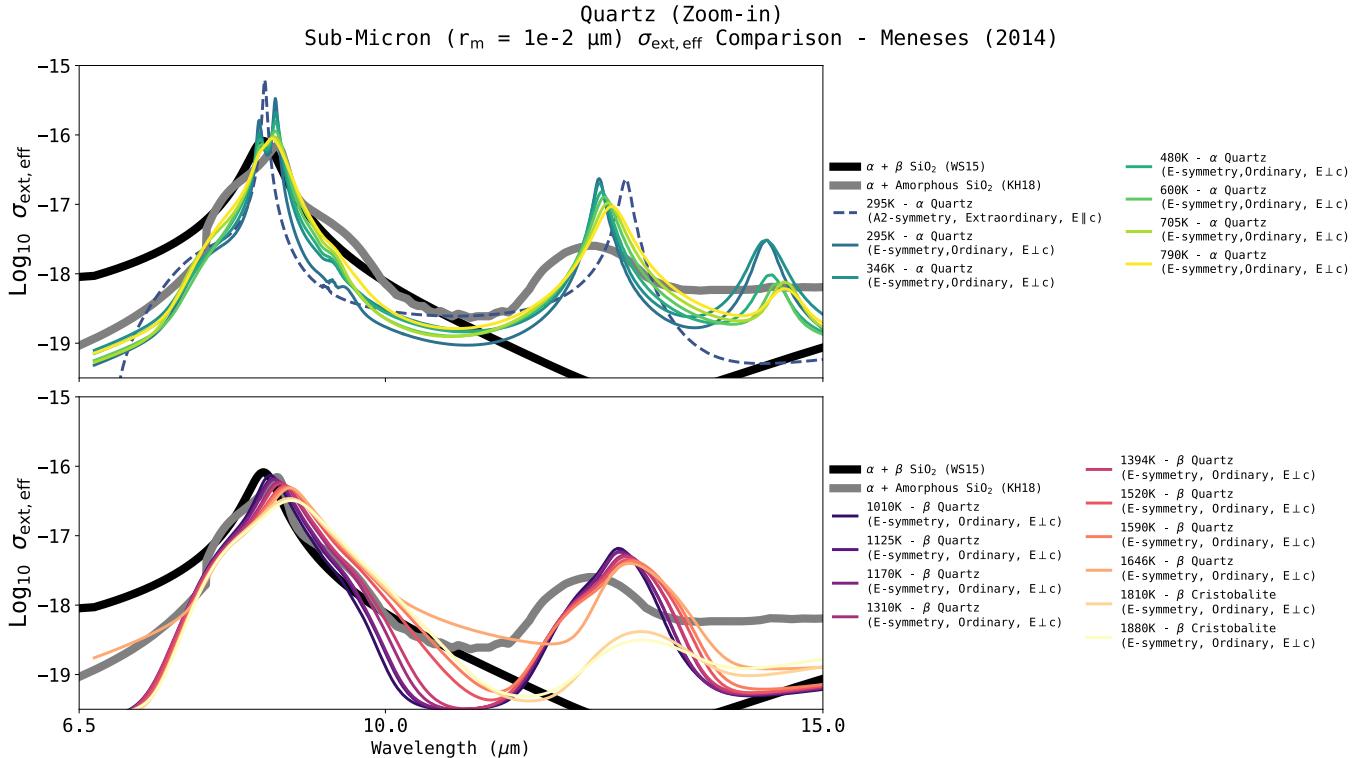


Figure 13. Same as Figure 3, but for Quartz and Cristobalite (D. D. S. Meneses et al. 2014), with a zoom-in on JWST MIRI-LRS wavelengths.

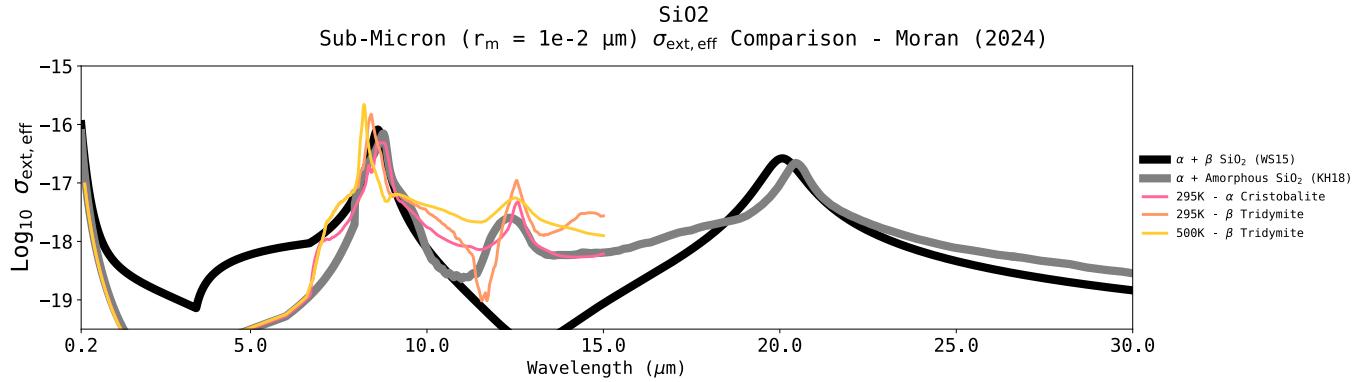


Figure 14. Same as Figure 3, but for Tridymite and Cristobalite (S. E. Moran et al. 2024).

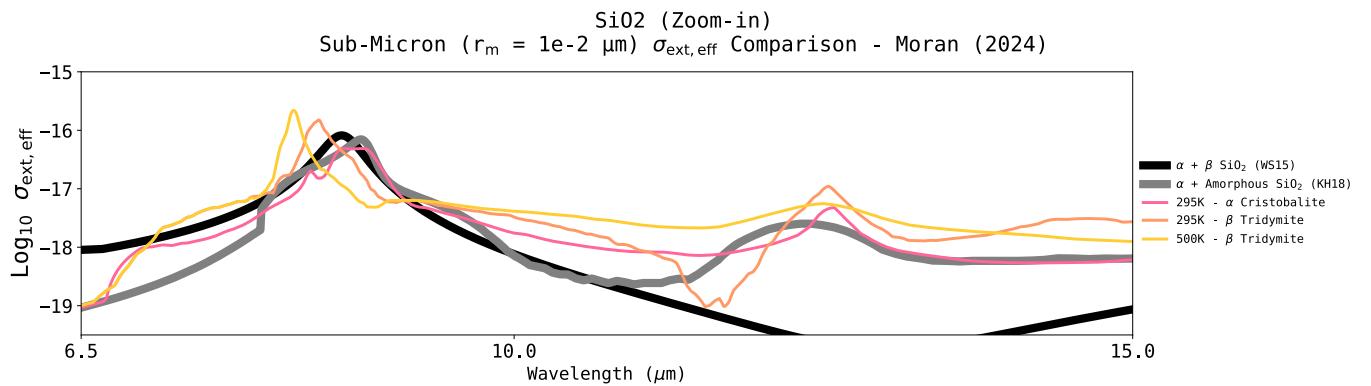


Figure 15. Same as Figure 3, but for Tridymite and Cristobalite (S. E. Moran et al. 2024), with a zoom-in on JWST MIRI-LRS wavelengths.

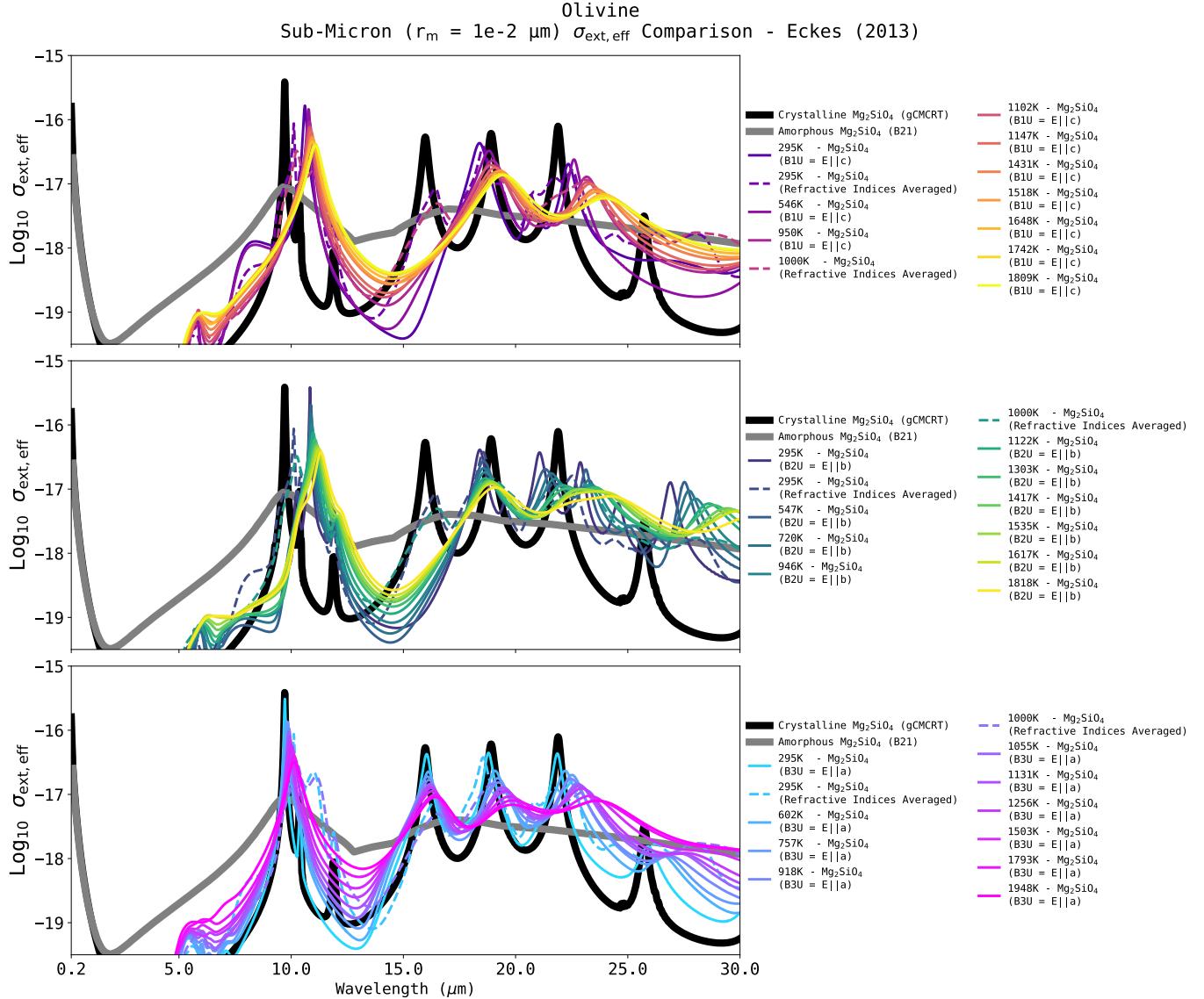


Figure 16. Same as Figure 3, but for Olivine (M. Eckes et al. 2013).

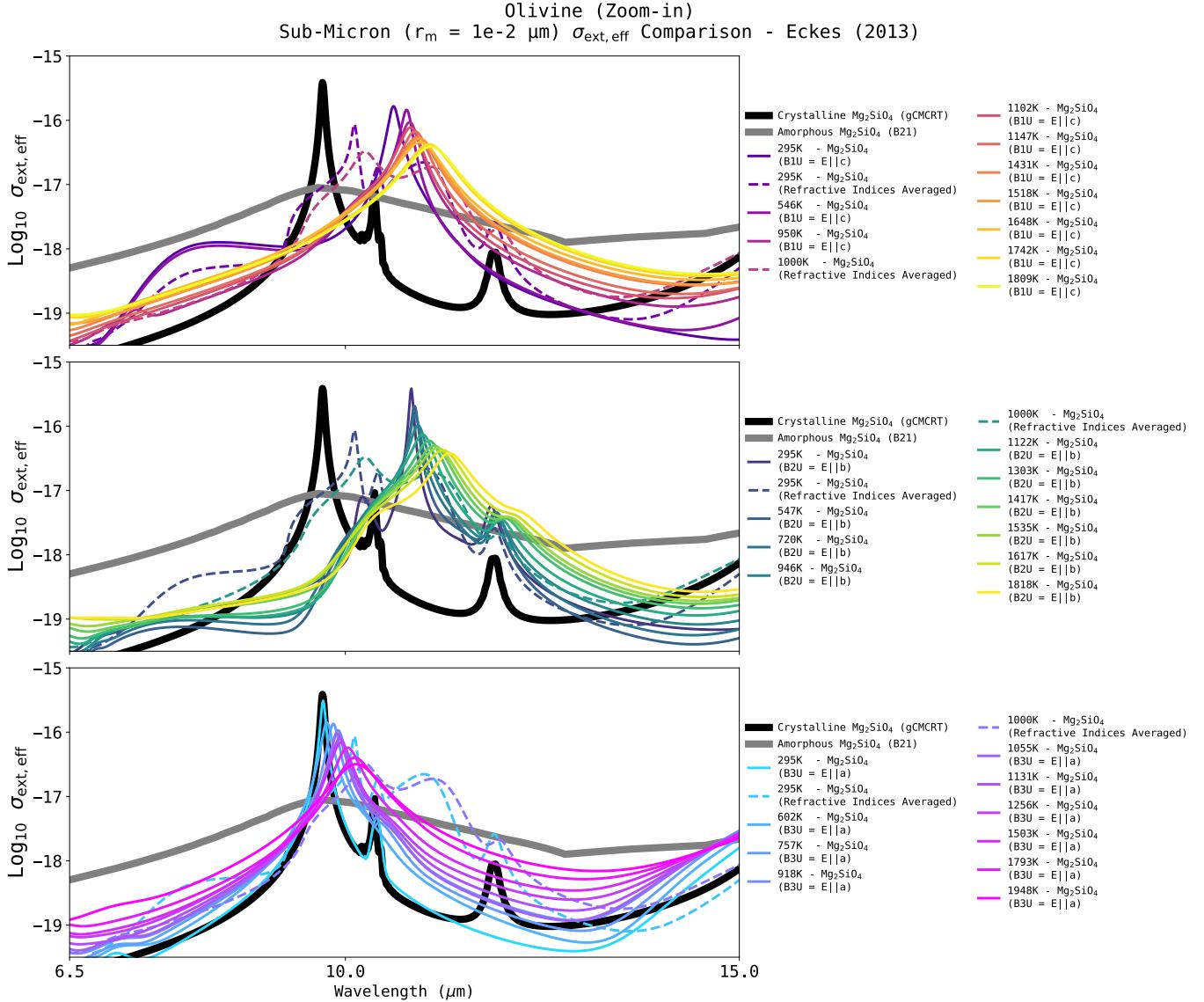


Figure 17. Same as Figure 3, but for Olivine (M. Eckes et al. 2013), with a zoom-in on JWST MIRI-LRS wavelengths.

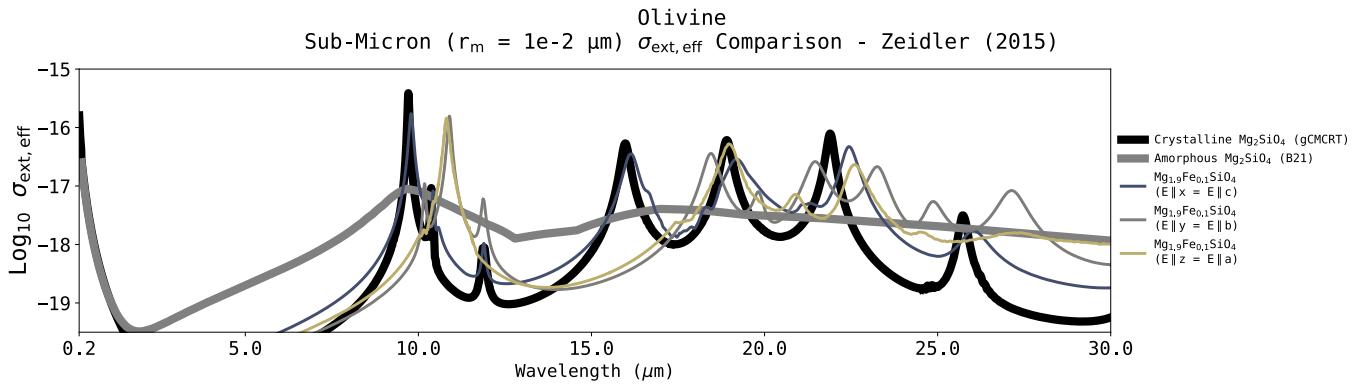


Figure 18. Same as Figure 3, but for Olivine (Strubachtal) (S. Zeidler et al. 2015).

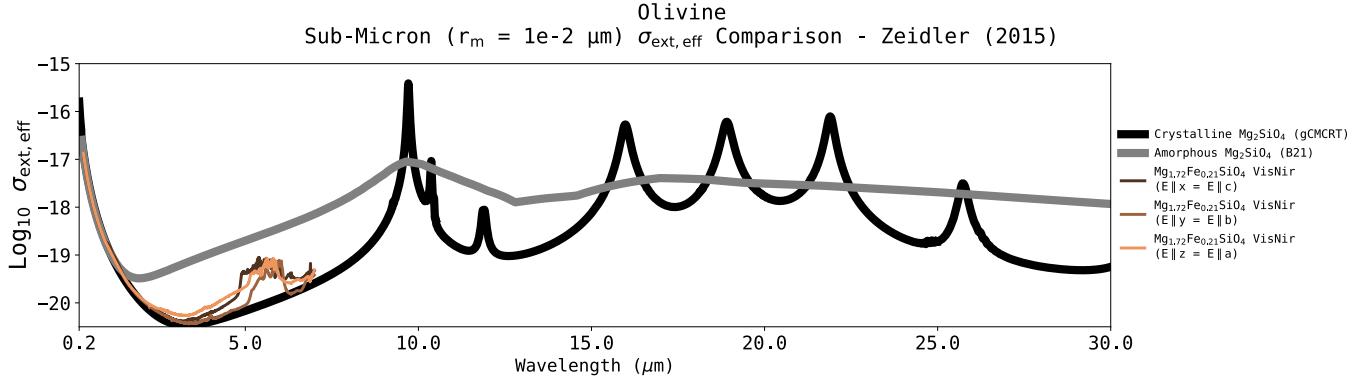


Figure 19. Same as Figure 3, but for Olivine (San Carlos, VisNir) (S. Zeidler et al. 2015).

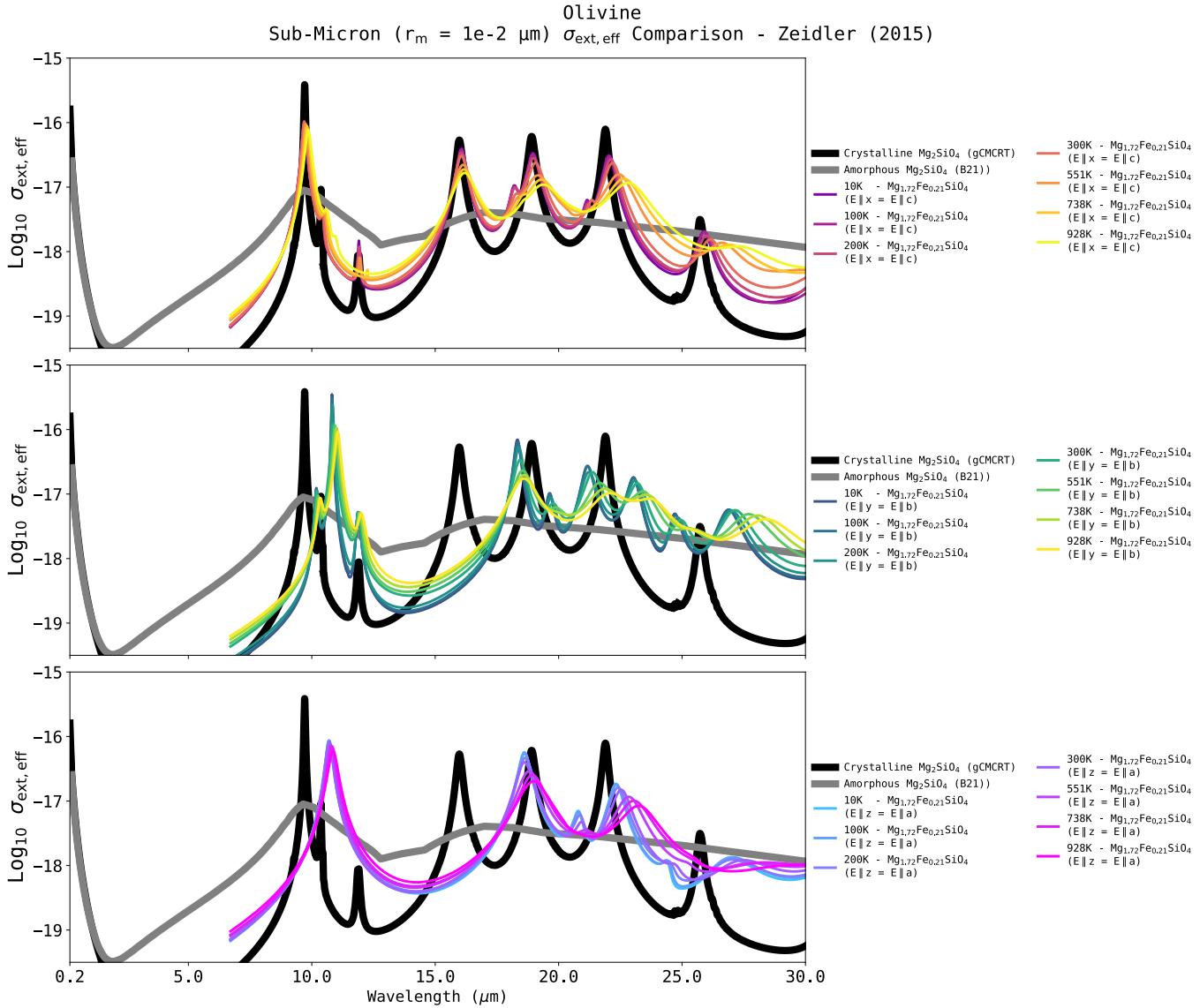


Figure 20. Same as Figure 3, but for Olivine (San Carlos) (S. Zeidler et al. 2015).

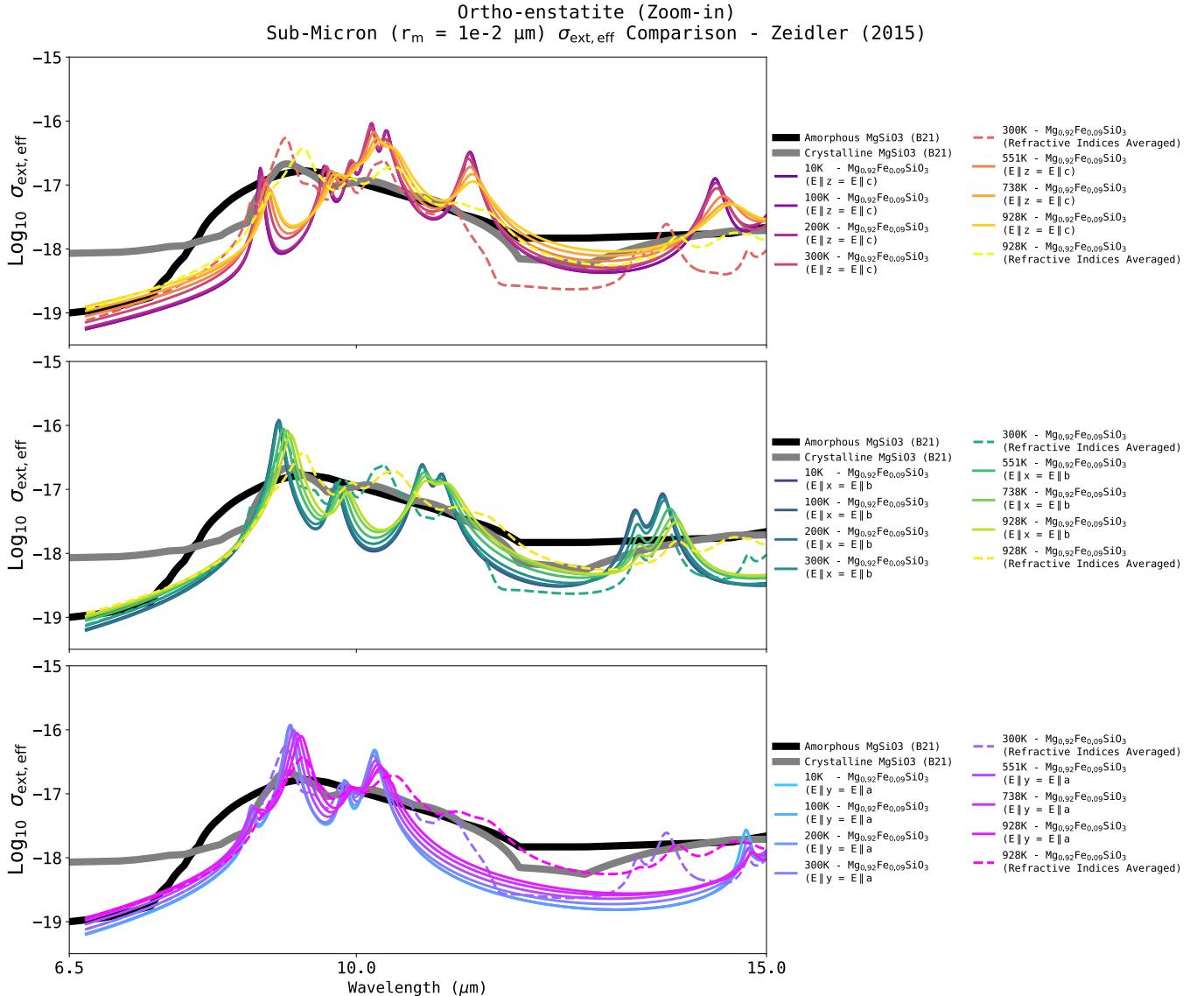


Figure 21. Same as Figure 3, but for Orthoenstatite (S. Zeidler et al. 2015), with a zoom-in on JWST MIRI-LRS wavelengths.

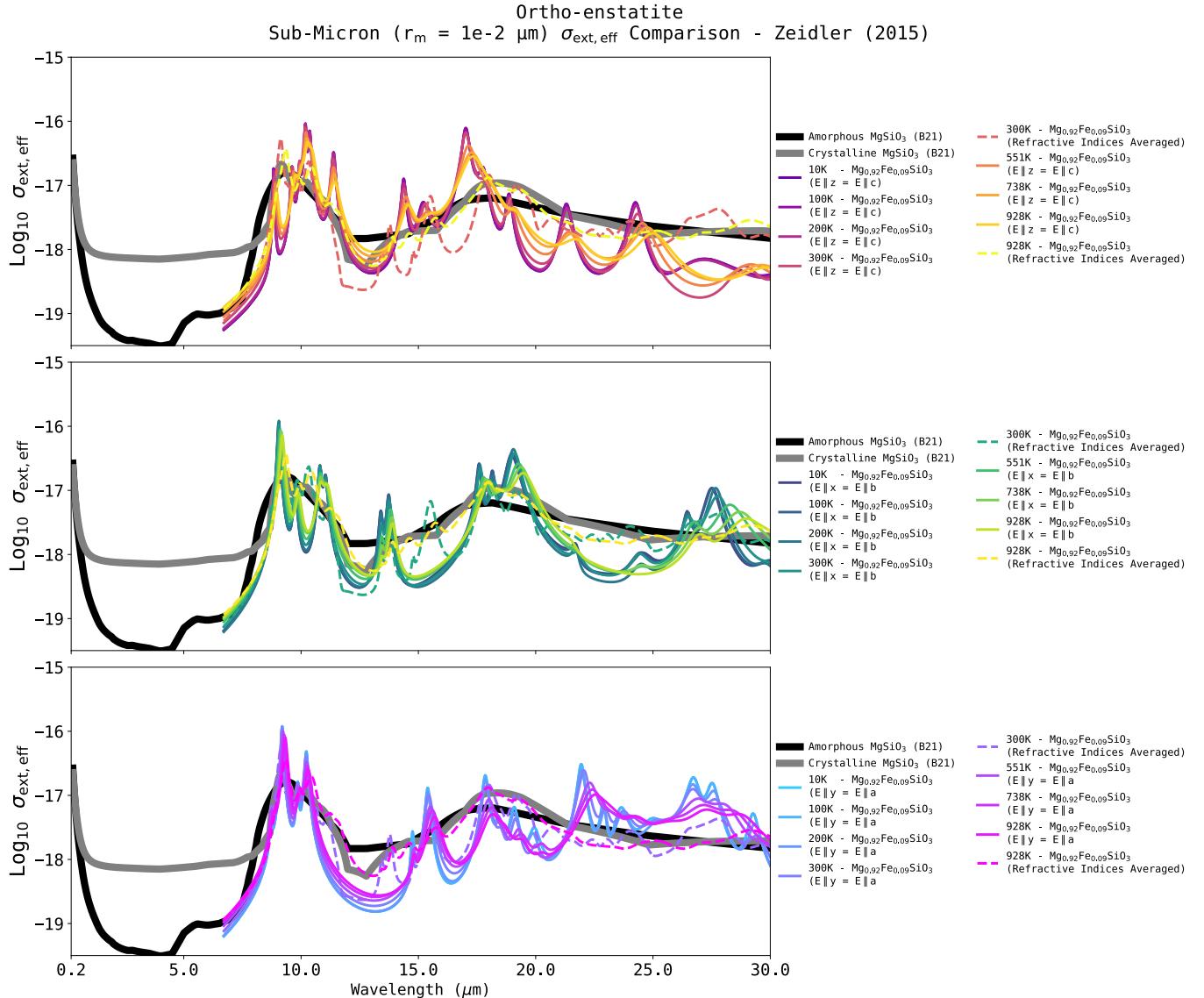


Figure 22. Same as Figure 3, but for Orthoenstatite (S. Zeidler et al. 2015).

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