

Problema 1: $m = \frac{D H}{T} \rightarrow H = \frac{m T}{D}$; $C_m = c : de$

• Combinando el primer y segundo principio:

$$T dS = du - H dm = c dT - \frac{m T}{D} dm$$

$$\rightarrow dS = c \frac{dT}{T} - \frac{1}{D} m dm$$

$$\therefore \int_{S_0}^S dS = c \int_{T_0}^T \frac{dT'}{T'} - \frac{1}{D} \int_{m_0}^m m dm$$

$$S - S_0 = c \ln\left(\frac{T}{T_0}\right) - \frac{1}{2D} (m^2 - m_0^2)$$

$$\therefore S = S_0 + c \ln\left(\frac{T}{T_0}\right) - \frac{(M^2 - M_0^2)}{2mD}$$

$$S = m S ; c = m c ; M = m m$$

• Entalpía: $H = U + H M = c(T - T_0) + \frac{T M^2}{mD}$

$$h = c(T - T_0) + \frac{T m^2}{D} = c(T - T_0) + \frac{D}{T} H^2$$

• Energía libre de Helmholtz: $A = U - TS$

$$A = c(T - T_0) - T \left\{ S_0 + c \ln\left(\frac{T}{T_0}\right) - \frac{(M^2 - M_0^2)}{2mD} \right\}$$

$$A = cT - cT_0 - TS_0 - cT \ln\left(\frac{T}{T_0}\right) + \frac{T}{2mD} M^2 - \frac{T}{2mD} M_0^2$$

$$A = cT \left[1 - \ln\left(\frac{T}{T_0}\right) \right] + \frac{T}{2mD} M^2 - T \left(S_0 + \frac{M_0^2}{2mD} \right) - cT_0$$

• Energía libre de Gibbs $G = A - H M$

$$G = cT \left[1 - \ln\left(\frac{T}{T_0}\right) \right] - \frac{T}{2mD} M^2 - T \left(S_0 + \frac{M_0^2}{2mD} \right) - cT_0$$

Problema 3: El estado crítico es un estado donde P, T y V satisfacen simultáneamente 3 ecuaciones: Ecuación de estado

$$\left(\frac{\partial P}{\partial V}\right)_T = 0 \quad \vee \quad \left(\frac{\partial^2 P}{\partial V^2}\right)_T = 0. \text{ Así, para}$$

$$P = \frac{mRT}{V-mb} \exp\left[-\frac{am}{RTV}\right],$$

tenemos

$$i) \quad \left(\frac{\partial P}{\partial V}\right)_T = \frac{mRT}{V-mb} \cdot \left(\frac{am}{RTV^2}\right) \exp\left[-\frac{am}{RTV}\right] - \frac{mRT}{(V-mb)^2} \exp\left[-\frac{am}{RTV}\right]$$

$$\left(\frac{\partial P}{\partial V}\right)_T = \frac{mRT}{V-mb} \exp\left[-\frac{am}{RTV}\right] \left\{ \frac{ma}{RTV^2} - \frac{1}{V-mb} \right\}$$

$$\rightarrow \left(\frac{\partial P}{\partial V}\right)_T = P \left\{ \frac{ma}{RTV^2} - \frac{1}{V-mb} \right\} \quad (i)$$

$$ii) \quad \left(\frac{\partial^2 P}{\partial V^2}\right)_T = \left(\frac{\partial P}{\partial V}\right)_T \left\{ \frac{ma}{RTV^2} - \frac{1}{V-mb} \right\} + P \left\{ -\frac{2ma}{RTV^3} + \frac{1}{(V-mb)^2} \right\}$$

$$\left(\frac{\partial^2 P}{\partial V^2}\right)_T = P \left\{ \frac{ma}{RTV^2} - \frac{1}{V-mb} \right\}^2 + P \left\{ -\frac{2ma}{RTV^3} + \frac{1}{(V-mb)^2} \right\} \quad (ii)$$

$$\text{De (i): } \left(\frac{\partial P}{\partial V}\right)_T = 0 \Rightarrow \frac{V^2}{V-mb} = \frac{ma}{RT} \quad (iii)$$

$$\text{De (ii): } \left(\frac{\partial^2 P}{\partial V^2}\right)_T = 0 \Rightarrow \frac{V^3}{(V-mb)^2} = \frac{2ma}{RT} \quad (iv)$$

Dividiendo (iv) en (iii):

$$\frac{V_c}{V_c - mb} = 2 \Rightarrow V_c = 2mb$$

Reemplazando en (iii):

$$\frac{4m^2 b^2}{mb} = \frac{ma}{RT_c} \rightarrow T_c = \frac{a}{4Rb}$$

Reemplazando en la ec. de estado:

$$P_c = \frac{mRT_c}{V_c - mb} \exp\left[-\frac{am}{RT_c V_c}\right] = \frac{ma/4Rb}{mb} \exp\left[-\frac{am}{4Rb \cdot 2mb}\right] = \frac{a}{4b^2} e^{-2}$$

$$P_c = \frac{a}{4e^2 b^2}$$

Para obtener la expresión de los "estados correspondientes" usamos las variables reducidas, $N^* = V/V_c$, $\gamma = T/T_c$, $\beta = P/P_c$

$$\rightarrow P = P_c \beta = \frac{m R T_c \gamma}{V_c N^* - m b} \exp \left[- \frac{a m}{R T_c \gamma V_c N^*} \right]$$

$$\# \frac{m R T_c \gamma}{P_c (V_c N^* - m b)} = \frac{m R T_c}{V_c P_c} \cdot \frac{\gamma}{N^* - \frac{m b}{V_c}} = \frac{\cancel{m} \cancel{R} / 4b}{2mb \cdot \frac{\cancel{a}}{4e^2 \cancel{b}^2}} \cdot \frac{\gamma}{N^* - \frac{mb}{2mb}} = \frac{e^2}{2} \frac{\gamma}{N^* - \frac{1}{2}}$$

$$\# \frac{a m}{R T_c V_c} \times \frac{1}{\gamma N^*} = \frac{\cancel{a} \cancel{m}}{\frac{\cancel{a}}{4b} \times 2mb} \times \frac{1}{\gamma N^*} = \frac{2}{\gamma N^*}$$

$$\therefore \beta = \frac{\gamma e^2}{2N^* - 1} \cdot e^{-\frac{2}{\gamma N^*}} = \frac{\gamma}{2N^* - 1} e^{-2(\frac{1}{\gamma N^*} - 1)}$$

$$\Rightarrow \boxed{\beta = \frac{\gamma}{2N^* - 1} e^{-2(\frac{1}{\gamma N^*} - 1)}}$$

Problema 3: Sólido: $\ln P = \ln P_1 - \frac{\ln(P_2/P_1)T_0^2}{T^2}$ (1)

(a) Líquido: $\ln P = \ln P_2 - \frac{2\ln(P_2/P_1)T_0}{T}$ (2)

En el punto triple se igualan las presiones de vapor:

$$\ln P_2 - \frac{2\ln(P_2/P_1)T_0}{T_{Pt}} = \ln P_1 - \frac{\ln(P_2/P_1)T_0^2}{T_{Pt}^2}$$

$$\Rightarrow \ln(P_2/P_1) - 2\ln(P_2/P_1)\frac{T_0}{T_{Pt}} + \ln(P_2/P_1)\frac{T_0^2}{T_{Pt}^2} = 0$$

$$\ln(P_2/P_1) \left[1 - \frac{2T_0}{T_{Pt}} + \frac{T_0^2}{T_{Pt}^2} \right] = 0$$

$$\Rightarrow T_{Pt}^2 - 2T_0T_{Pt} + T_0^2 = 0$$

$$T_{Pt} = T_0 = 250 \text{ K}$$

En la expresión para la presión de vapor del sólido:

$$\ln P_{Pt} = \ln P_1 - \ln P_2 + \ln P_1 = 2\ln P_1 - \ln P_2$$

$$\ln P_{Pt} = \ln(P_1^2/P_2) \Rightarrow P_{Pt} = \frac{P_1^2}{P_2} = 0,5 \text{ atm}$$

(b) A partir de (2):

$$\frac{dP}{P} = 2\ln(P_2/P_1) \times \frac{dT}{T^2} \Rightarrow \frac{dP}{dT} = 2\ln\left(\frac{P_2}{P_1}\right) \times \frac{P}{T^2}$$

La ecuación de Clapeyron:

$$\frac{dP}{dT} = \frac{h_v}{T(v''' - v'')} \approx \frac{h_v}{Tv'''} \quad \leftarrow \text{suponiendo que el volumen molar de vapor es mucho mayor que el volumen molar de líquido.}$$

Así, comparando

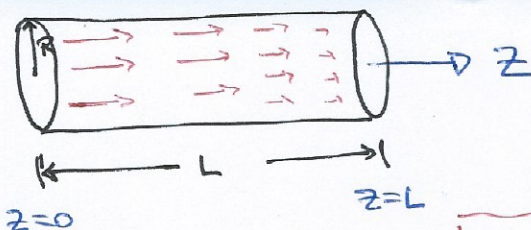
$$2 \ln\left(\frac{P_2}{P_1}\right) \times \frac{P}{T^2} \approx \frac{l_v}{RT^3} \Rightarrow l_v \approx 2 \ln\left(\frac{P_2}{P_1}\right) \times \frac{PT^3}{T}$$

Podemos suponer que el vapor se comporta como un gas ideal, de manera que $PT^3 \approx RT$, y así:

$$l_v \approx 2R \ln\left(\frac{P_2}{P_1}\right) \approx 2 \times 0,69 \times R$$

$$l_v \approx 1,39R \approx 2,75 \left(\frac{\text{cal}}{\text{mol} \cdot \text{K}}\right)$$

Problema 4:



$$T(z) = T_0 \left[1 + \cos\left(\frac{\pi z}{L}\right) \right]$$

(a)

Ley de Fourier: $\vec{q} = -K \vec{\nabla} T \rightarrow \vec{q} = -K \frac{dT}{dz} \hat{k} \equiv q_z \hat{k}$

$$q_z = -K \frac{dT}{dz} = -K \times T_0 \times \left(-\frac{\pi}{2L}\right) \times \sin\left(\frac{\pi z}{L}\right) \Rightarrow q_z = \frac{K\pi T_0}{2L} \sin\left(\frac{\pi z}{L}\right)$$

$$\Rightarrow \vec{q} = \frac{\pi K T_0}{2L} \sin\left(\frac{\pi z}{L}\right) \hat{k}$$

$$q_z(z=L) = \frac{K\pi T_0}{2L}$$

$$q_z(z=\frac{L}{2}) = \frac{K\pi T_0}{2L} \cdot \sin\left(\frac{\pi}{4}\right)$$

$$q_z(z=\frac{L}{2}) = \frac{K\pi T_0}{2L} \cdot \frac{\sqrt{2}}{2}$$

\Rightarrow La razón entre la densidad de flujo en $z=L$ y $z=L/2$ es

$$\eta = \frac{q_z(L/2)}{q_z(L)} = \frac{(K\pi T_0/2L) \cdot \sqrt{2}/2}{(K\pi T_0/2L)}$$

$$\Rightarrow \eta = \frac{\sqrt{2}}{2} \approx 0,7$$

(b) $q_z(z=L/3) = \frac{K\pi T_0}{2L} \cdot \sin\left(\frac{\pi}{6}\right) = \frac{K\pi T_0}{2L} \times \frac{1}{2}$

$$A = \pi r^2 = \pi \left(\frac{R}{2}\right)^2 = \frac{\pi}{4} R^2$$

$$\Rightarrow \dot{Q}(z=L/3) = q_z(L/3) \cdot A = \frac{K\pi T_0}{2L} \times \frac{1}{2} \times \frac{\pi}{4} R^2 = \frac{\pi^2 R^2 K T_0}{16L}$$

$$\Rightarrow \dot{Q}(z=L/3) = \frac{\pi^2}{16} \times \frac{R^2 K T_0}{L}$$