

# Molecular Pharmacology Notes

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# Chapter 1

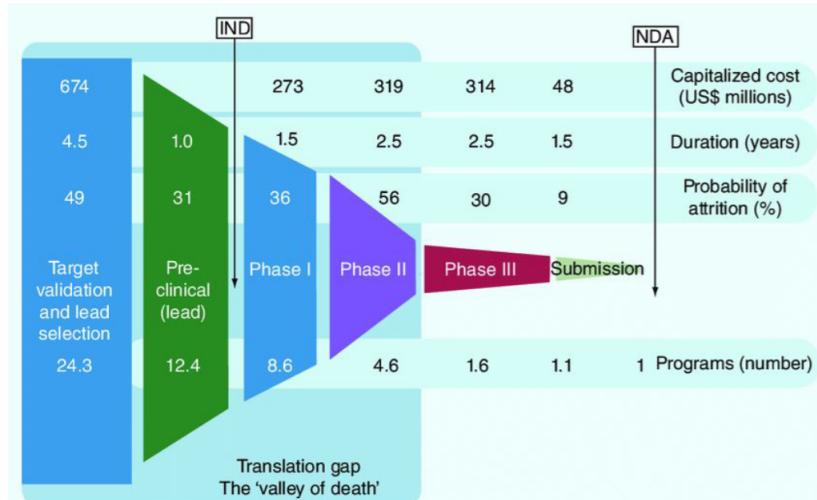
## Pharmacokinetics

### 1.1 What is a drug?

A drug is a chemical that interacts with proteins in the body to affect a physiological function. Once these chemicals are absorbed into the systemic circulation they bind with certain proteins and this changes the functioning of the cell slightly. For example, anticancer drugs bind to proteins on the surface of cancer cells this stimulates the cells to die. In this case cell death is the physiological action of the drug.

No drugs are specific to interacting with just one type of cell or one type of protein and this is what causes side effects. Again using an anticancer drug as an example, the medication works by binding to very rapidly dividing cells, such as cancer cells, however hair cells are also rapidly dividing and that is why one of the side effects of anticancer drugs is hair loss.

Drugs can be generally divided into two main categories: **agonist**, that stimulate a response and **antagonist**, that inhibit a response.



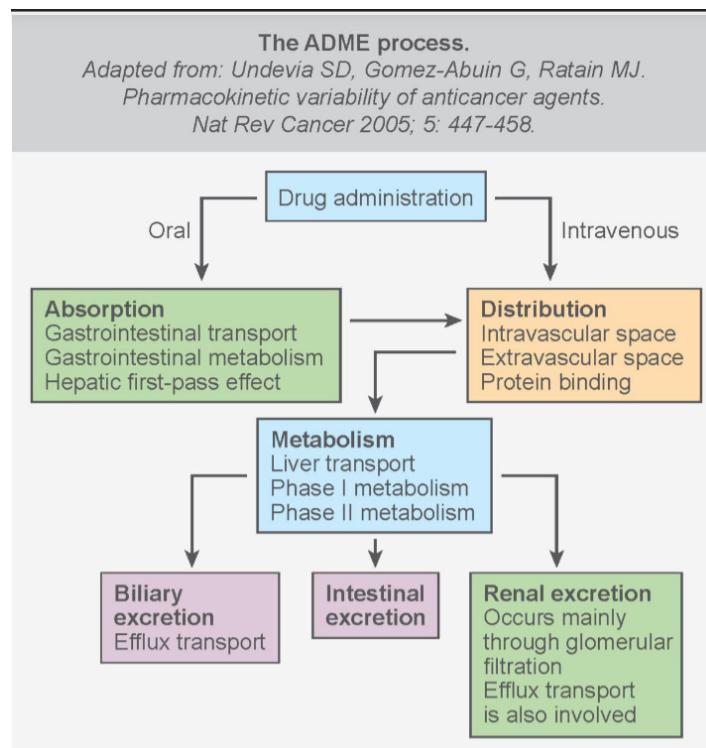
Developing and testing drugs is a very expensive business! Many parameters regarding pharmacokinetics and pharmacodynamics need to be taken studied and fit into strict boundaries for the drug to be accepted.

### 1.2 Pharmacokinetics

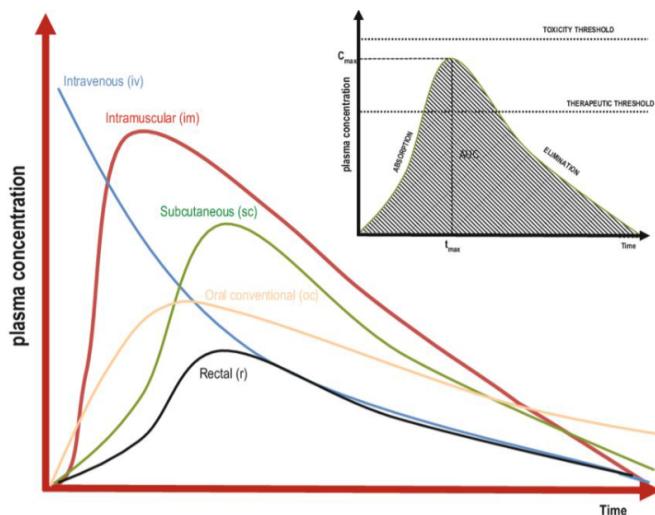
Pharmacokinetics (PK) is the study of how the body interacts with administered substances for the entire duration of exposure. In other words, **what the body does to the drug(s)**!

For a chemical compound to become a marketable drug, it must have favourable properties in addition to **efficacy** (its therapeutic effect) and **safety**. These properties are summarised in the acronym ADME, which refers to **absorption**, **distribution**, **metabolism** and **excretion**.

- Absorption: a compound's ability to pass through barriers such as the intestinal lining, the nasal lining, the lungs or the skin.
- Distribution: how the compound is distributed around the body and its propensity to accumulate in certain tissues and organs.
- Metabolism: how the body breaks down the compound, normally by the liver. The key issues are drug-drug interactions and the effects of the metabolites (the new chemicals created as a result of metabolism).
- Excretion: the rate and process through which the compound exits the body.



Often we refer also to **AADME**, including also **administration**, which also greatly influences the PK properties of drugs and drug concentration in plasma.

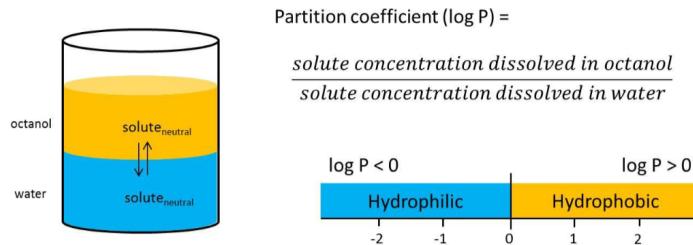


### 1.2.1 Absorption

**Absorption is the process of a drug moving from its site of delivery into the bloodstream.** Absorption is the process of delivering a drug into the bloodstream. Absorption can be accomplished by administering the drug in a variety of different ways (e.g. orally, rectally, intramuscularly, subcutaneously, inhalation, topically, etc.). Note, that if a drug is administered intravenously (placed directly into the bloodstream), the need for absorption is bypassed entirely.

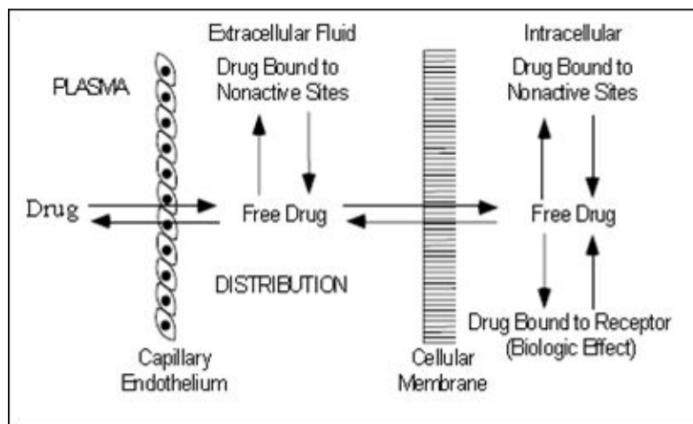
However, the plasmatic membrane cannot be passed freely. Gases ( $CO_2$ ,  $N_2$ ,  $O_2$ , anaesthetic), small uncharged polar molecules (ethanol, urea, water) can pass easily. Inversely, large uncharged polar molecules (sugar), ions and charged polar molecules (amino acids, ATP, proteins, nucleic acids) cannot.

The need for drug molecules to cross lipid bilayers in passing from one body compartment to another requires that the structure has properties that impart solubility in both a hydrophobic medium and water. Each part of a drug molecule contributes hydrophobic or hydrophilic properties to the molecule as a whole and that contribution is known as the **Hansch partition coefficient** of the group.



### 1.2.2 Distribution

**Drug distribution is the process of delivering a drug from the bloodstream to the body.** The process of transferring a drug from the bloodstream to tissues is referred to as distribution. The same principles that govern drug absorption (e.g. ionization of a drug, lipophilicity of a drug, size of a drug, pH of the environment, etc.) also govern the rate and extent that a drug will distribute to various tissues in the body. In addition to that, there are additional factors at play, particularly non-specific binding to proteins.



#### Volume of distribution

The concept of “apparent volume of distribution” is a concept that seeks to predict how extensively a drug is distributed throughout the body. The apparent volume of distribution,  $Vd$ , is mathematically calculated by dividing the dose that is administered ( $mg$ ) by the plasma concentration  $C$  ( $mg/L$ ).

$$Vd = \frac{Dose}{C}$$

Based on the above equation:

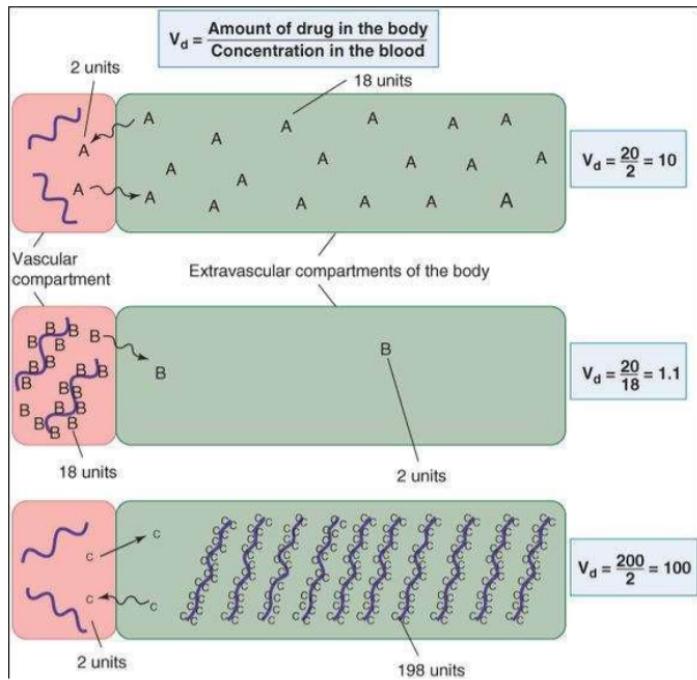


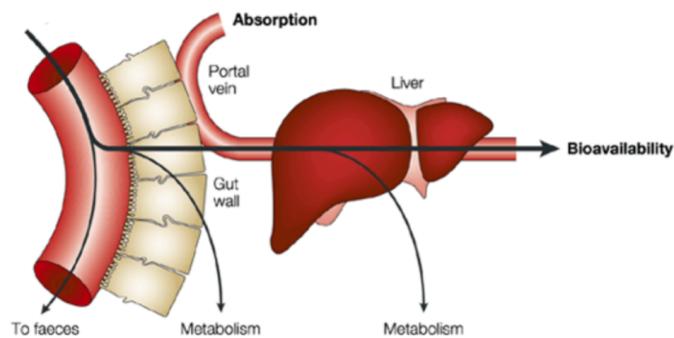
Figure 1.1: Drug A diffuses freely between the two compartments. Drug B binds avidly to plasma proteins and is retained in the plasma compartment (low  $V_d$ ). Drug C binds avidly to molecules in peripheral tissues. Low concentration in blood, thus high  $V_d$  and higher total dose required to achieve measurable plasma centration.

- A drug with a high  $V_d$  has a propensity to leave the plasma and enter the extravascular compartments of the body, meaning that a higher dose of a drug is required to achieve a given plasma concentration. (High  $V_d \rightarrow$  More distribution to other tissue)
- Conversely, a drug with a low  $V_d$  has a propensity to remain in the plasma meaning a lower dose of a drug is required to achieve a given plasma concentration. (Low  $V_d \rightarrow$  Less distribution to other tissue)

Another way to think about  $V_d$  is that  $V_d$  is equal to the amount of space that a drug must fill up such that a given dose of a drug will achieve a specific plasma concentration. There is an assumption here; that is, calculation of the apparent  $V_d$  presumes that the drug concentration is the same everywhere throughout the body. We know though that this is not true since most drugs are not uniformly distributed. More on this on the NIH website[2].

**Binding to plasma proteins** highly influences the volume of distribution and in general, drug efficacy, distribution, and disposition. Serum albumin displays an extraordinary ligand binding capacity, and  $\alpha$ -1-Acid glycoprotein is the main carrier for basic and neutral drugs. High- and low-density lipoproteins play a limited role in drug binding.

Drug distribution is also influenced by physio-anatomical features. For example, brain capillaries are different from general capillaries in the body, forming the **blood-brain-barrier** (BBB). The BBB is composed of specialized endothelial cells, pericytes, a capillary basement membrane, and astrocyte end-feet, all working together to protect the brain from harmful substances while allowing essential nutrients and molecules to enter. The BBB's selectivity is crucial for maintaining a stable and optimal environment for brain function. The placenta is also a barrier (placental barrier) to be taken into account.



**Figure 1.2: First pass effect-** The large reduction in concentration of an orally absorbed drug due to passage of the drug through the liver. Some drug will be lost during absorption from the intestine and more from metabolism in the liver. Remember, the liver sees many things as potential toxins and tries to detoxify them. Drugs are included in that list of things that the liver tries to remove. **Bioavailability**– the fraction of the drug that escapes the liver and reaches the systemic circulation due to the first pass effect. If a drug is given intravenously, it is considered to have 100% bioavailability since it is all in the blood stream. The same drug given orally will have less bioavailability since some is never absorbed (stays in the gut and ends up in the feces) or is modified by the liver before it hits the vascular system.

### 1.2.3 Metabolism

The primary objective of drug metabolism is to facilitate a drug's excretion by increasing its water solubility (hydrophilicity). The involved chemical modifications incidentally decrease or increase a drug's pharmacological activity and/or half-life. The principal organs of drug metabolism are the liver and (for orally taken drugs) the small intestine. Drugs completely inactivated during the first-pass through these organs must be given parenterally, similarly to poorly absorbed drugs.

#### Cytochromes P450 and transporters

CYPs are heme proteins used extensively by the liver to metabolize drugs. The types and amounts of CYPs vary by species, sex and genetic makeup. The main families of CYP450 enzymes involved in drug metabolism are the monooxygenases of the CYP1, CYP2 and CYP3 families. Changes in CYPs result in different levels of drug effectiveness and toxicity. Differences in cyp450 is responsible for the unique responses of horses and cats to certain drugs.

Drugs can interact with CYPs in 3 different ways:

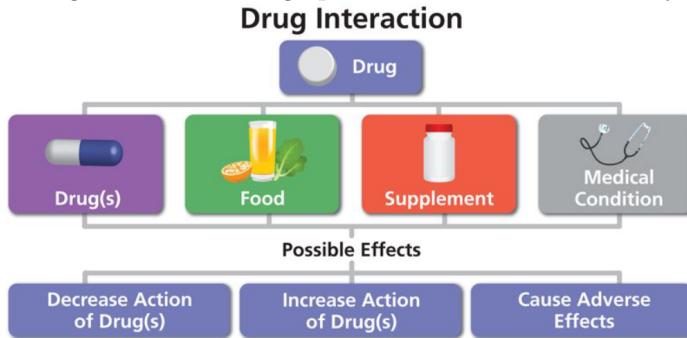
- Drugs may just be metabolized by the CYPs. Generally this makes drugs less available to the animal as it leads to faster excretion.
- Drugs may inhibit CYP activity. Erythromycin and omeprazole inhibit CYP activity. This prevents the CYPs from metabolizing other drugs being administered to the animal, resulting in abnormally high levels of the unmetabolized drugs.
- Drugs may induce CYP activity. Phenytoin, phenobarbital, griseofulvin and rifampin are CYP inducers. When they are present, the CYPs metabolize other drugs at a much higher rate, resulting in low working levels of those drugs.

However, drug metabolism does not happen only in the liver! **P-glycoprotein 1** is important protein of the cell membrane that pumps many foreign substances out of cells. P-gp is extensively distributed and expressed in the intestinal epithelium where it pumps xenobiotics back into the intestinal lumen, in liver cells where it pumps them into bile ducts, in the cells of the proximal tubule of the kidney where it pumps them into urinary filtrate (in the proximal tubule), and in the capillary endothelial cells composing the blood-brain barrier and blood-testis barrier, where it pumps them back into the capillaries.

Another important family of cell transporters are the **ABC-type ATPases**. Of this family we also find **P-glycoprotein (Pgp)**, also known as multidrug resistance protein 1 (MDR1) or ATP-binding cassette sub-family B member 1 (ABCB1), which is a transmembrane protein that acts as an efflux pump, transporting various substances out of cells. It plays a role in drug resistance in cancer and also has physiological functions in various tissues like the intestines, liver, and brain.

### Other factors influencing drug metabolism

There's a plethora of factors that contribute in differences in drug metabolism, e.e. sex, age, polymorphism, diseases.... Another important aspect to consider is drug interactions. The interactions can happen with other drugs, supplements or medical conditions. For example, ingesting grapefruit blocks CYP3A4 enzyme in enterocytes and in the liver, which metabolizes the drug felodipine. Usually CYP3A4 metabolizes the drug to the 15%, but grapefruit increases bioavailability to 45%



#### 1.2.4 Excretion

**Excretion is the removal of drugs and their metabolites from the body.** The kidney is the principal drug-excreting organ. All renally excreted drugs reach urine via glomerular filtration. Many drugs additionally are secreted into the proximal tubule through the opportunistic (molecular similarity-based) use of organic cation and anion transporters. Lipophilic drugs are additionally secreted into the proximal tubule via passive diffusion, as they can easily cross the membranes of nephron cells.

Excretion can also happen by the bile (and thereby with stool), sweat, exhaled air, saliva, and breast milk, but they play a much smaller role compared to urine. An exception is the excretion of volatile anesthetics with exhaled air.

#### Drug Clearance

Drug clearance (body clearance, total body clearance, or  $Cl_T$ ) is a pharmacokinetic term for describing drug elimination from the body without identifying the mechanism of the process. Clearance is calculated based on plasma drug concentration data.

$$Cl_T = \frac{\text{elimination rate}}{\text{plasma concentration}(C_p)}$$

$$Cl_T = \frac{dD_E/dt}{C_p} = \frac{\mu\text{g}/\text{min}}{\mu\text{g}/\text{ml}} = \text{ml}/\text{min}$$

Where  $D_E$  is the amount of drug eliminated and  $dD_E/dt$  is the rate of elimination. Clearance relates to volume of distribution through the following equation:

$$Cl_T = \frac{kC_pV_D}{C_p} = kV_D$$

Clearance is the product of  $V_D$  and  $k$ , both of which are constant. As the plasma drug concentration decreases during elimination, the rate of drug elimination,  $dD_E/dt$ , decreases accordingly, but clearance remains constant. Clearance is constant as long as the rate of drug elimination is a first-order process.

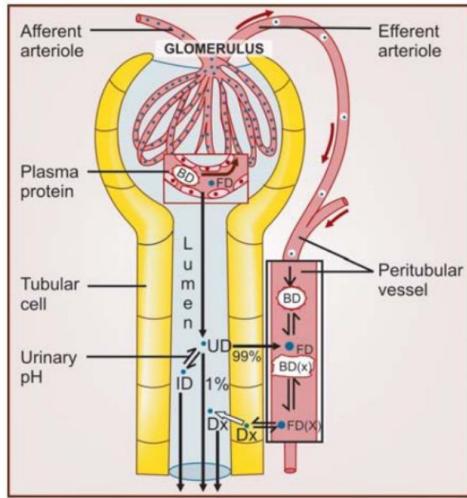


Figure 1.3: Schematic depiction of glomerular filtration, tubular reabsorption and tubular secretion of drugs. FD - free drugs; BD - bound drug; UD - unionized drug; ID - ionized drug; Dx - actively secreted organic acid (or base) drug.

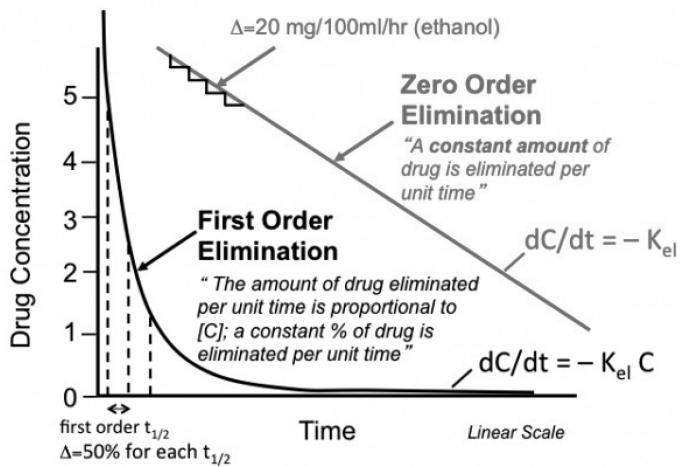


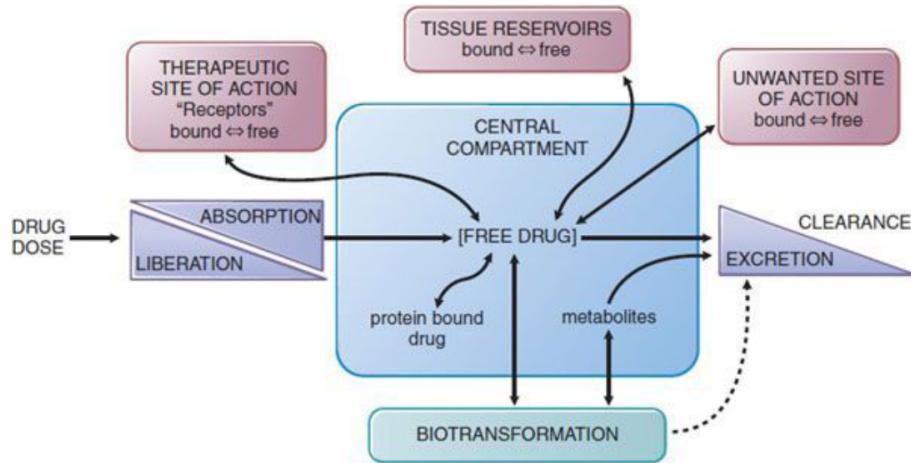
Figure 1.4: In medical science, the term half-life typically refers to the elimination half-life. The elimination half-life is defined as the time required for the concentration of a specific substance, typically a drug, to decrease to half of its initial amount in the body. Although different drugs have varying half-lives, they all share a common principle—after one half-life, 50% of the initial drug amount is removed from the body. Most clinically relevant drugs follow first-order pharmacokinetics, meaning their drug-elimination rates are directly proportional to plasma concentrations.[1] In contrast, a few drugs follow zero-order elimination, in which the drug amount decreases by a constant amount over time, regardless of initial concentration, for example, ethanol. For more info check [1].

### Free Drug is the Active Fraction

A basic tenet of clinical pharmacology is that only unbound (free) drug molecules can interact with target receptors that are present on cell membrane or with enzymes that are located inside the cell, and therefore the intensity and duration of drug action are mediated via the time course of unbound drug concentrations at the site of action. With few exceptions, most drug target receptors or enzymes are located outside of the blood circulation in the target tissues. Although unbound drug concentrations in blood can be readily measured, direct assessment of unbound drug concentration at the action site in target tissues is seldom possible due to inaccessibility of the action sites.

For this reason, the unbound drug concentration in blood (plasma) is often used to establish pharmacokinetics-pharmacodynamics (PK-PD) relationship by applying the so-called **free drug hypothesis**. The hypothesis assumes that the unbound drug concentration in blood is the same as that in the site of action at steady state. From literature, the free drug hypothesis appears valid for many drugs. The unbound concentrations of drugs in peripheral tissues and brain are quantitatively similar to those in plasma. However, the hypothesis is not applicable for some drugs. For examples, the unbound concentrations of morphine, gabapentin, and atenolol in the brain are significantly lower than those in plasma.

Over the years, with the progress of molecular biology, it has become evident that efflux drug transporters (such as P-gp, BCRP, and MRPs) play an important role not only in drug excretion but also in tissue distribution (drug transport), particularly for brain uptake. For drugs that are substrates of efflux transporters, at steady state, the unbound drug concentrations in tissues are expected to be lower than unbound drug concentrations in plasma, when efflux transporters are involved in tissue distribution. Based on the involvement of efflux transporters, drugs can generally be categorized into two classes: drugs that are not substrates of efflux transporters (Class I) and drugs that are substrates of efflux transporters (Class II). It is expected that the free drug hypothesis will not be valid for drugs that are substrates of efflux transporters. Conceivably, the free drug hypothesis is also not applicable for drugs that are substrates of influx transporters. Unbound drug concentration in tissues is expected to be higher for influx transporter drugs than that in plasma.

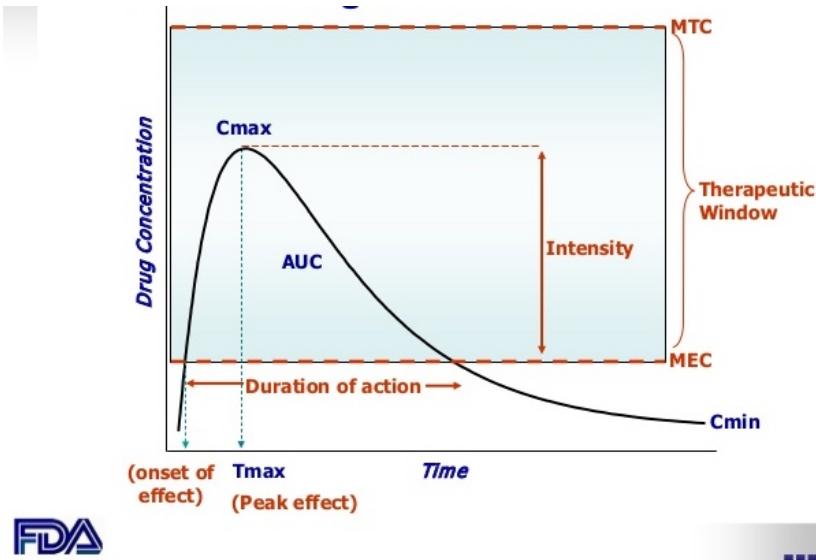


### 1.2.5 General PK considerations

Other basic PK concepts are :

- MEC: minimum effective concentration
- MTC: minimum toxic concentration

Everything under MEC is considered "sub therapeutic level", everything above MTC is considered at "toxic level". The first intersection of drug concentration with MEC is called "onset time", and the second intersection "termination of action".



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# Chapter 2

## Pharmacodynamics

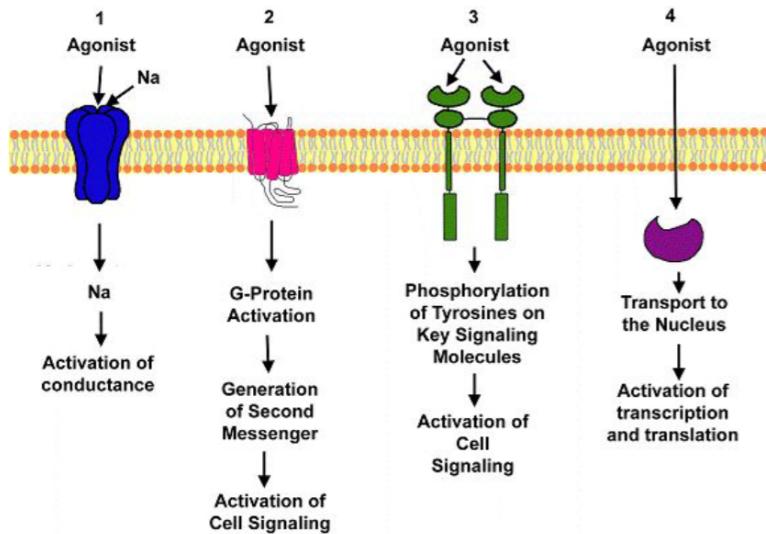
pharmacodynamics is the other branch, together with pharmacokinetics, of pharmacology.

In particular, pharmacodynamics is the study of **how a drug affects an organism**, whereas pharmacokinetics is the study of how the organism affects the drug. Both together influence dosing, benefit, and adverse effects.

Pharmacodynamics places particular emphasis on dose-response relationships, that is, the relationships between drug concentration and effect. One dominant example is drug-receptor interactions as modeled by  $L + R \rightleftharpoons LR$ , where  $L$ ,  $R$ , and  $LR$  represent ligand (drug), receptor, and ligand-receptor complex concentrations, respectively.

### 2.1 Drug Targets

Drug targets can be any structure in the cell and often they are cell receptors.



### 2.2 Dose Response Curve

The pharmacological effect of a drug depends on its concentration at the site of action, which in turn is determined by the dose of the drug administered. Such a relationship is called the **D-R**.

The idea is that the extent to which the desired response alters as the dose changes. The dose is plotted on the horizontal axis and the response on the vertical axis.

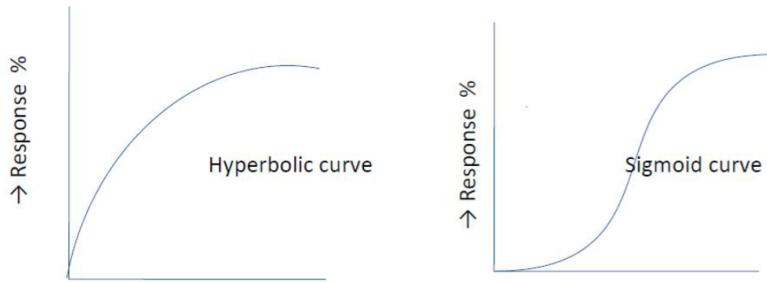


Figure 2.1: Difference in plotting between dose and logDose

### 2.2.1 Graded dose-response graphs

Graded dose-response curves are graphical representations of the relationship between the dose of the drug and the effect it achieves. More specifically, the concentration of the drug is used, rather than the actual dose.

They are also usually plotted on a logarithmic scale of concentration which stretches the lower end of the dose range, to inflate the useful (20%-80%) response range where dose-effect relationship is linear, and to compress the clinically uninteresting high and low dose ranges.

Graded dose-response curves provide insights to these properties:

- Affinity ( $K_d$ ): the tendency of a drug to bind to the receptor, or the extent to which a drug can produce a response when all available receptors or binding sites are occupied.
- Efficacy: the capacity of a drug-receptor complex to elicit response.
- Potency: the quantity of a drug required to produce the desired response (more potent drugs produce effects at lower concentrations). Potency refers to the concentration ( $EC_{50}$ ) or dose ( $ED_{50}$ ) of a drug. The potency of a drug depends on the affinity ( $K_d$ ) and in part on the efficacy with which drug-receptor interaction is coupled to response.

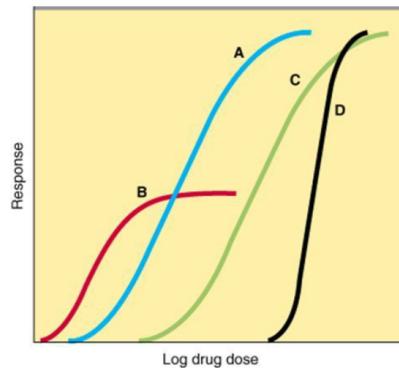


Figure 2.2: Graded dose-response curves for drugs, illustrating different pharmacological potencies and different maximal efficacies. Potency:  $B > A > C > D$ , Efficacy: all same except for B (lowest).

### 2.2.2 Therapeutic Index (Therapeutic Window)

*Solely the dose determines that a thing is not a poison*

Vitamin A, essential for sustaining life, can be lethal if taken in very doses, and botulin toxin can not cause damages (and even be beneficial) at very low doses.

The therapeutic index is a ratio that compares the blood concentration at which a drug becomes toxic and the concentration at which the drug is effective. The larger the therapeutic index (TI), the safer the drug is. If the TI is small (the difference between the two concentrations is very small), the drug

must be dosed carefully and the person receiving the drug should be monitored closely for any signs of drug toxicity.

$$TI = \frac{TD_{50}}{ED_{50}}$$

However, nowadays the term "therapeutic window" or "therapeutic range" is preferred, which is determined after all the studies on the drug of interest. Importantly, this window is not fixed, rather it depends on the age and other physiological characteristics.

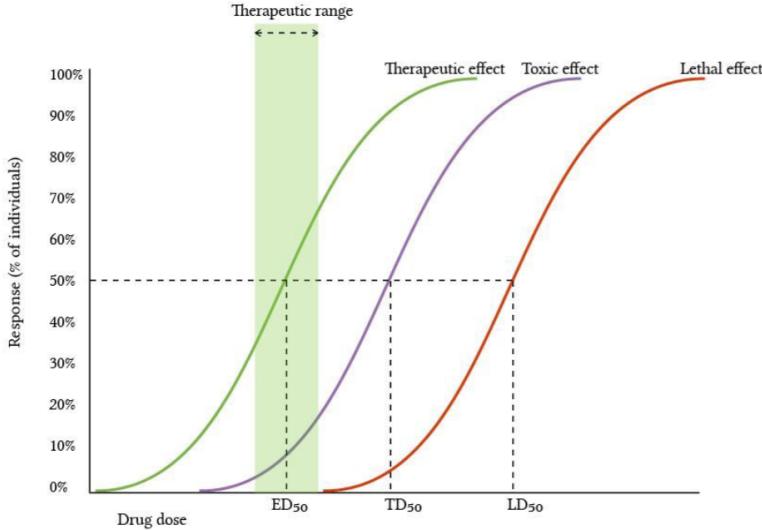
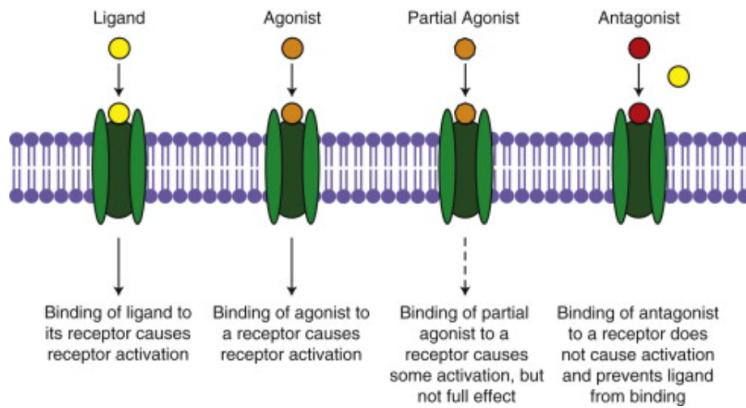


Figure 2.3:

## 2.3 Agonists and Antagonists

Basically any receptor can be subjected to agonist and antagonist drug.

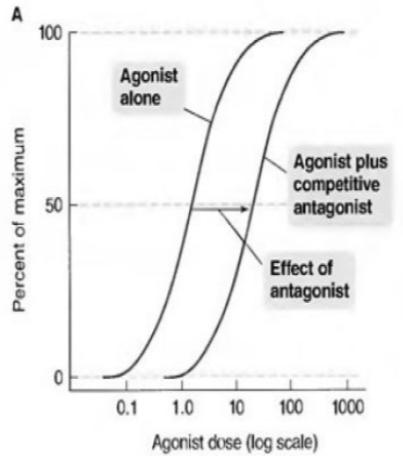
- Agonists: drugs that occupy receptors and activate them.
- Antagonists: drugs that occupy receptors but do not activate them. They also can lessen the activation by agonists.



### 2.3.1 Agonists

Agonism describes the process of binding of a drug to its specific receptor, activate it and produce some molecular and cellular response.

In other words, agonists are the chemical substances, or drugs, that can interact (combine) with a receptor (affinity) and thereby initiate a cellular reaction (efficacy). E.g. adrenaline in cardiac arrest stimulates  $\beta_1$  receptor - increasing the force of contraction and heart rate. Their efficacy is 1 (or 100%).



A **partial agonist** binds to its receptor but produces a smaller effect than a full agonist. Their efficacy is between 0 – 1.

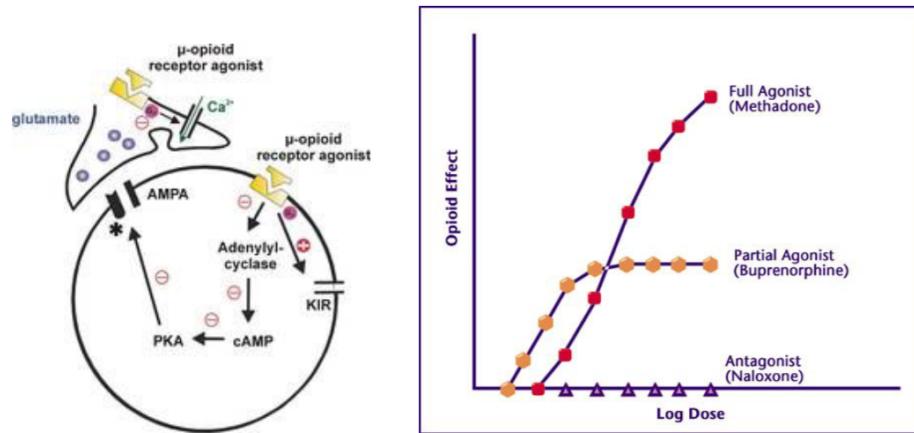


Figure 2.4: Pain killers can be full agonists, partial agonists or antagonists. Naloxone is an opioid antagonist medication used to reverse or reduce the effects of opioids, particularly in cases of overdose. It works by blocking the effects of opioids on the body, restoring breathing if it has been depressed by opioids.

### Inverse Agonist

Inverse agonists bind to a receptor and produce the opposite effect. In this case, the efficacy is  $-1$ . Examples are benzodiazepines, that produce CNS depression, while  $\beta$  carbolines produce excitation, anxiety and convulsion.

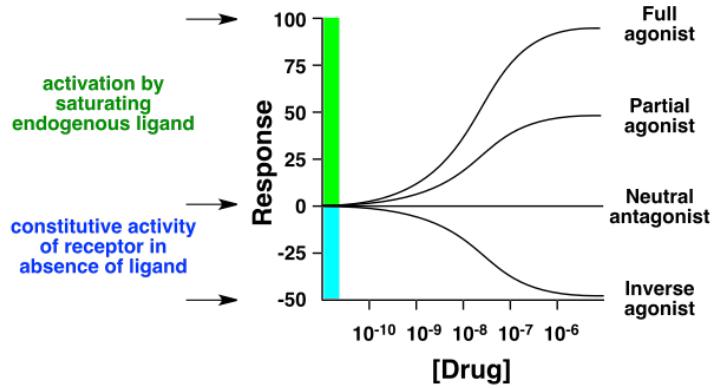


Figure 2.5: An agonist is a chemical that activates a receptor to produce a biological response. In contrast, an antagonist blocks the action of the agonist, while an inverse agonist causes an action opposite to that of the agonist.

### 2.3.2 A little about GABA<sub>A</sub>

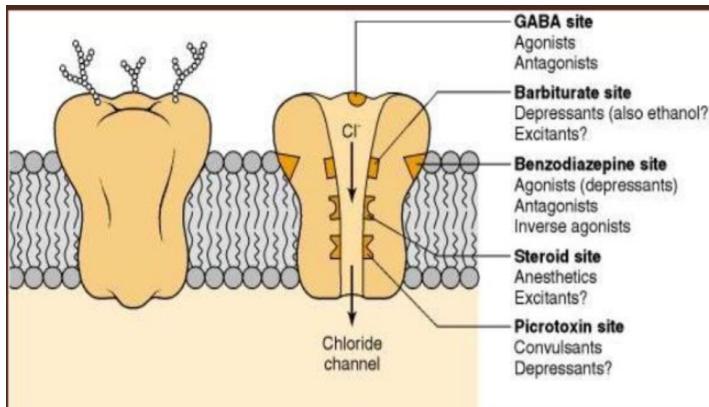


Figure 2.6: GABA<sub>A</sub> and its site of actions.

The GABA<sub>A</sub> receptor is an ionotropic receptor and ligand-gated ion channel. Its endogenous ligand is  $\gamma$ -aminobutyric acid (*GABA*), the major inhibitory neurotransmitter in the central nervous system. Accurate regulation of GABAergic transmission through appropriate developmental processes, specificity to neural cell types, and responsiveness to activity is crucial for the proper functioning of nearly all aspects of the central nervous system (CNS). Upon opening, the GABA<sub>A</sub> receptor on the postsynaptic cell is selectively permeable to chloride ions ( $Cl^-$ ).

This receptor is the target of many drugs and active compounds:

- **GABA site**

- Agonists: Mimic GABA and activate the receptor (e.g., muscimol<sup>1</sup>).
- Antagonists: Block GABA binding, inhibiting the receptor (e.g., bicuculline<sup>2</sup>).

- **Barbiturate site**

<sup>1</sup>Muscimol (also known as agarin or pantherine) is the principal psychoactive constituent of *Amanita muscaria* and *Amanita pantherina*. Muscimol is a potent and selective orthosteric agonist for the GABA<sub>A</sub> receptor and displays sedative-hypnotic, depressant and hallucinogenic psychoactivity.

<sup>2</sup>Bicuculline is a phthalide-isooquinoline compound that is a light-sensitive competitive antagonist of GABA<sub>A</sub> receptors. It was originally identified in 1932 in plant alkaloid extracts. Since it blocks the inhibitory action of GABA receptors, the action of bicuculline mimics epilepsy; it also causes convulsions. This property is utilized in laboratories around the world in the in vitro study of epilepsy, generally in hippocampal or cortical neurons in prepared brain slices from rodents. [4]

- Barbiturates<sup>3</sup> (e.g., phenobarbital): Bind here and enhance GABA action by prolonging the duration of chloride channel opening.
- May also be affected by ethanol.
- High doses can directly activate the receptor even in the absence of GABA.

- Benzodiazepine site

- Agonists (e.g., diazepam, lorazepam): Potentiate GABA by increasing the frequency of channel opening.
- Antagonists (e.g., flumazenil): Block the site, reversing benzodiazepine effects.
- Inverse Agonists: Reduce GABAergic activity (may cause anxiety or seizures).

### 2.3.3 Antagonists

Drugs that bind to its receptor and does not activate it, will prevent the action of an agonist. The affinity is higher than the agonist and the efficacy is 0.

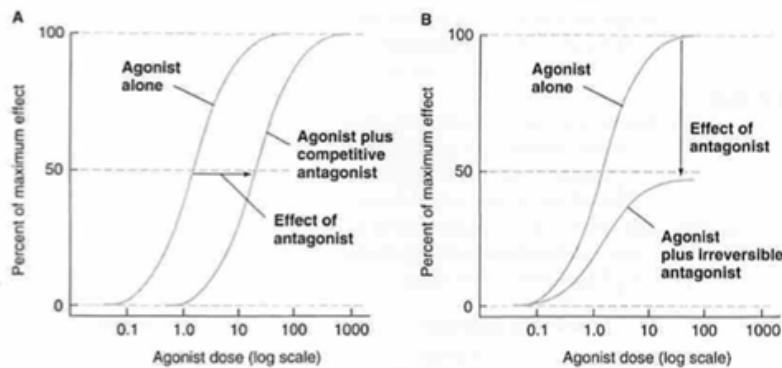


Figure 2.7: Characteristics of antagonists: **a)** **competitive**: shift agonist curve to the right. Resembles substrate [S], blocks active site but can be overcome by ↑ [S], ↓ potency, ↑  $K_m$  and  $V_{max}$  is unchanged. **b)** **non-competitive**: shift agonist curve downward. USUALLY Does not resemble [S], dose not block active site, therefore no ↑ [S] can overcome the effect, ↓ efficacy and potency, ↓  $V_{max}$ , and  $K_m$  is unchanged. ↑ EC50 of agonist (half maximally effective concentration for producing a given effect. [3])

Antagonism can be:

- Competitive

- The antagonism is reversible, agonist and antagonist compete for the same receptor.
- Antagonism can be overcome by increased concentration of agonist.
- When we plot the dose-response curve we see a parallel shifting.

- Non-Competitive

- Irreversible binding of receptor.
- Agonist cannot compete with antagonist.
- There will be resynthesis of the receptor and action will be regained.

<sup>3</sup>**Barbiturates** are a class of depressant drugs that are chemically derived from barbituric acid. They are effective when used medically as anxiolytics, hypnotics, and anticonvulsants, but have physical and psychological addiction potential as well as overdose potential among other possible adverse effects. Barbiturates have largely been replaced by benzodiazepines and nonbenzodiazepines ("Z-drugs") in routine medical practice, particularly in the treatment of anxiety disorders and insomnia, because of the significantly lower risk of overdose, and the lack of an antidote for barbiturate overdose. Despite this, barbiturates are still in use for various purposes: in general anesthesia, epilepsy, treatment of acute migraines or cluster headaches, acute tension headaches, euthanasia, capital punishment, and assisted suicide.

**Competitive** antagonists are sub-classified as reversible (**surmountable**) or irreversible (**insurmountable**) competitive antagonists, depending on how they interact with their receptor protein targets. Reversible antagonists, which bind via noncovalent intermolecular forces, will eventually dissociate from the receptor, freeing the receptor to be bound again. Irreversible antagonists bind via covalent intermolecular forces. Because there is not enough free energy to break covalent bonds in the local environment, the bond is essentially "permanent", meaning the receptor-antagonist complex will never dissociate. The receptor will thereby remain permanently antagonized until it is ubiquitinated and thus destroyed.

**Non-competitive** antagonist is a type of insurmountable antagonist that may act in one of two ways: **by binding to an allosteric site of the receptor, or by irreversibly binding to the active site of the receptor**. While the mechanism of antagonism is different in both of these phenomena, they are both called "non-competitive" because the end-results of each are functionally very similar. Unlike competitive antagonists, which affect the amount of agonist necessary to achieve a maximal response but do not affect the magnitude of that maximal response, non-competitive antagonists reduce the magnitude of the maximum response that can be attained by any amount of agonist.

### 2.3.4 Allosteric Modulators

Allosteric modulators affect the interaction of the receptor and the ligand by binding to separate sites on the receptor. These effects are transmitted through changes in the receptor protein.

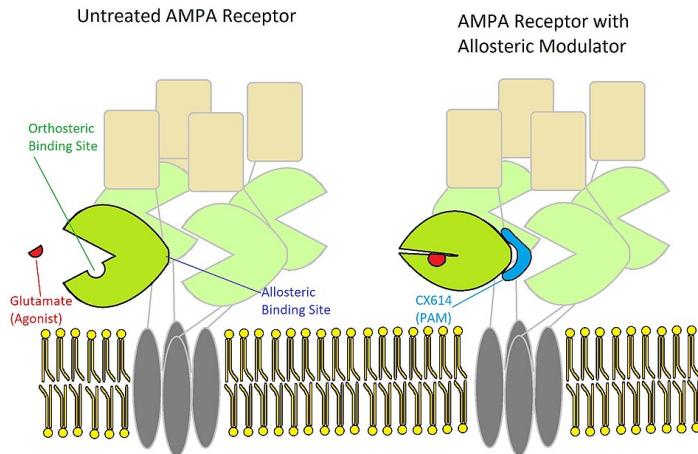


Figure 2.8: Some modulators act to stabilize conformational changes associated with the agonist-bound state. This increases the probability that the receptor will be in the active conformation, but does not prevent the receptor from switching back to the inactive state. With a higher probability of remaining in the active state, the receptor will bind agonist for longer. AMPA receptors modulated by aniracetam and CX614 will deactivate slower, and facilitate more overall cation transport. This is likely accomplished by aniracetam or CX614 binding to the back of the "clam shell" that contains the binding site for glutamate, stabilizing the closed conformation associated with activation of the AMPA receptor.

Allosteric modulators can be:

- Positive allosteric modulators (PAMs) → enhance receptor activity.
- Negative allosteric modulators (NAMs) → reduce receptor activity (non-competitive inhibition).
- Silent allosteric modulators (SAMs) → bind but don't modulate alone.

Important: All NAMs are non-competitive antagonists, but not all non-competitive antagonists are allosteric — some bind irreversibly to the orthosteric site (the main agonist site).

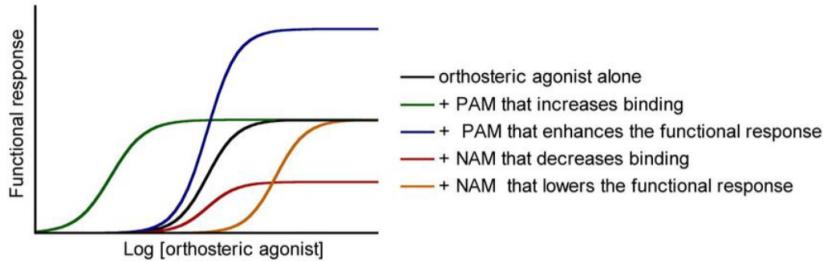


Figure 2.9: PAMs shift initial agonist response curve to lower agonist concentrations by increasing affinity and/or increase maximum response by increasing efficacy. NAMs shift curves to higher concentrations by decreasing affinities and/or lower maximum responses by decreasing efficacies. If compared to PAMs, the effects of NAMs are inverse.

## 2.4 Take Home Message

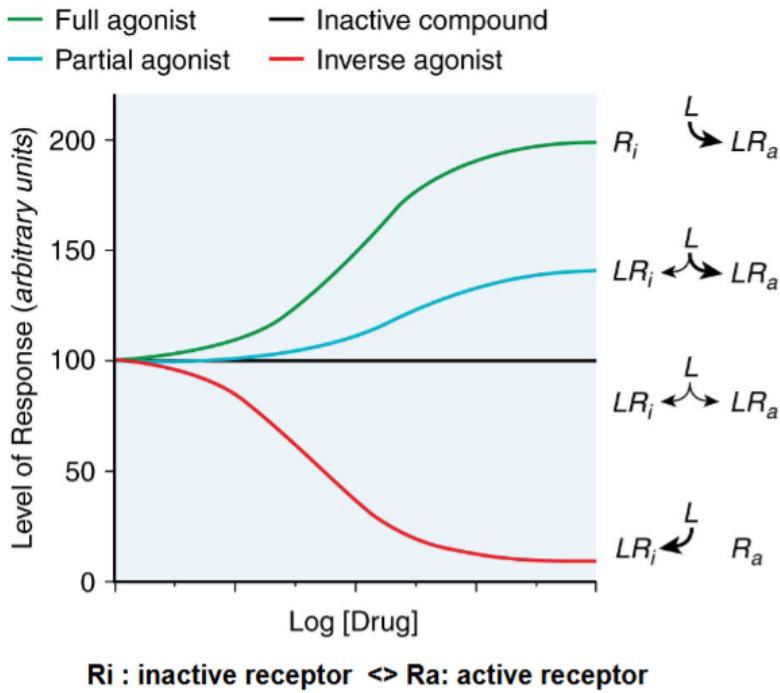


Figure 2.10:

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