# Ammonium analysis

## Based on:

Weatherburn, M. W. 1967. Phenol-hypochlorite reaction for determination of ammonia. Analytical Chemistry 39:971-974.

## Sodium salicylate solution

6.8 g sodium salicylate

5 g sodium citrate

5 g sodium tartrate

0.025 g sodium nitroprusside

100 ml ultrapure water

## **Sodium hydroxide solution**

6 g sodium hydroxide

100 ml ultrapure water

## Bleach solution (make fresh each day)

0.2 ml bleach

9.8 ml sodium hydroxide solution

For resin bag extracts, the matrix is 0.1 M HCl/2.0 M NaCl. Extracts from unfertilized plots should follow the protocol for low concentrations and use the low standard curve. Extracts from fertilized plots should be diluted in matrix and run using the high protocol.

For low concentrations (0-5 ppm):

Add the following to each well:

80 µl sample

60 µl salicylate solution (add using multichannel pipet)

60 µl bleach solution (add using multichannel pipet)

For high concentrations (1-10 ppm):

20 µl sample

90 ul salicylate solution (add using multichannel pipet)

90 µl bleach solution (add using multichannel pipet)

Pipet up and down to mix well, incubate 50 min and read plate at 650 nm.

Standard curves: Dilute the 100 ppm stock solution to 10 ppm in a 1.5 ml centrifuge tube (150  $\mu$ l stock:1350  $\mu$ l matrix). Create the following standard curves in 1.5 ml centrifuge tubes.

High			Low		
Concen	μl 10 ppm	μl matrix	Concen	μl 1 ppm	μl matrix
0 ppm	0	1000	0 ppm	0	1000
0.5	50	950	0.05	50	950
1.0	100	900	0.10	100	900
2.0	200	800	0.20	200	800
5.0	500	500	0.50	500	500
10.0	1000	0	1.00	1000	0

Detection limit < 0.05 ppm

Stock ammonium solution: 0.23585 g ammonium sulfate in 500 ml ultrapure DI water

# Nitrate analysis

## Based on:

Doane, T. A., and W. R. Horwath. 2003. Spectrophotometric determination of nitrate with a single reagent. Analytical Letters 36:2713-2722.

## 1 M HCl

Add 500 ml ultrapure water to 1 liter volumetric flask Add 84 ml concentrated HCl to flask and swirl to mix Bring to 1 liter with ultrapure water

#### Saturated vanadium chloride solution

Add 0.35 g vanadium (III) chloride to 50 ml of 1 M HCl; filter if necessary CAUTION: the vanadium chloride powder is very reactive with air! Work quickly.

### 2% sulfanilamide solution

0.2 g sulfanilamide 10 ml of 1 M HCl

### 0.2% NED solution

0.02 g N-(1-naphthyl)-ethylenediamine dihydrochloride 10 ml ultrapure water

### **Reagent solution**

50 ml saturated vanadium chloride solution 3.3 ml 2 % sulfanilamide solution 3.3 ml 0.2 % NED solution 400 ml DI water

Purge 17 ml aliquots with nitrogen or helium and store for up to 3 months frozen

For resin bag extracts, the matrix is 0.1 M HCl/2.0 M NaCl. Extracts from unfertilized plots should follow the protocol for low concentrations and use the low standard curve. Extracts from fertilized plots should be diluted in matrix and run using the high protocol.

For low samples, combine  $100 \mu l$  sample and  $100 \mu l$  reagent. For high samples combine  $10 \mu l$  sample and  $160 \mu l$  reagent. Tap microplate corner to mix well, incubate at least 5 h or overnight, and read plate at 540 nm.

Standard curves: Dilute the 100 ppm stock solution to 10 ppm in a 1.5 ml centrifuge tube (150  $\mu$ l stock:1350  $\mu$ l matrix). Create the following standard curves in 1.5 ml centrifuge tubes.

High			Low		
Concen	μl 10 ppm	μl matrix	Concen	μl 1 ppm	μl matrix
0 ppm	0	1000	0 ppm	0	1000
0.5	50	950	0.05	50	950
1.0	100	900	0.10	100	900
2.0	200	800	0.20	200	800
5.0	500	500	0.50	500	500
10.0	1000	0	1.00	1000	0

Detection limit < 0.05 ppm

Stock nitrate solution: 0.3609 g potassium nitrate in 500 ml ultrapure water

# Inorganic phosphorus analysis

## Based on:

Lajtha, K., C. T. Driscoll, W. M. Jarrell, and E. T. Elliott. 1999. Soil phosphorus: characterization and total element analysis. Pages 115-142 in G. P. Robertson, D. C. Coleman, C. S. Bledsoe, and P. Sollins, editors. Standard Soil Methods for Long-Term Ecological Research. Oxford University Press, New York.

#### **AMP** solution

Add 250 ml ultrapure water to a 500 ml volumetric flask.

Add 53 ml concentrated sulfuric acid.

Dissolve 8.775 g ammonium para-molybdate and bring to volume.

## Malachite green solution

Heat 400 ml ultrapure water to 80°C in an Erlenmeyer flask with stir bar.

Add 1.75 g polyvinyl alcohol and stir to dissolve.

Add 0.175 g malachite green and stir to dissolve.

Cool and bring to 500 ml volume in a volumetric flask.

Add 30  $\mu$ l AMP solution to 150  $\mu$ l sample and tap corner of microplate to mix. After 10 min, add 30  $\mu$ l malachite green solution and pipet up and down mix. Do not place plate on shaker to mix- a nasty precipitate will form. Read microplate at 630 nm after 30 min.

Standard curves: Dilute the 50 ppm stock solution to 1 ppm in a 1.5 ml centrifuge tube (30 µl stock:1470 µl matrix). Create the following standard curve in 1.5 ml centrifuge tubes.

Concen	μl 1 ppm	μl matrix
0 ppm	0	1000
0.02	20	980
0.05	50	950
0.10	100	900
0.20	200	800
0.50	500	500

Detection limit < 0.02 ppm

Stock phosphate solution: 0.2195 g oven-dried KH<sub>2</sub>PO<sub>4</sub> in 1000 ml ultrapure water