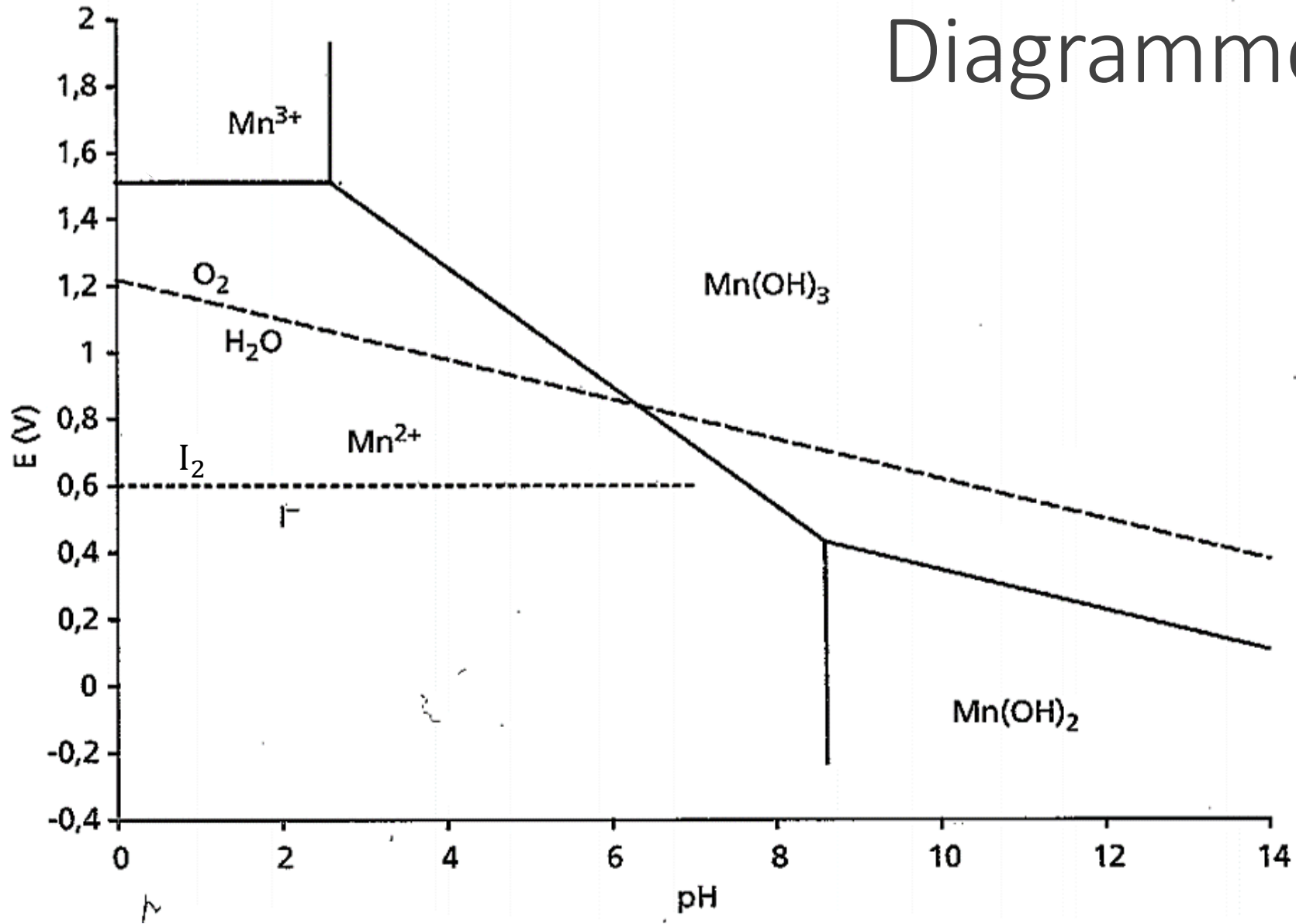


# Diagrammes potentiel- pH (construction exclue)

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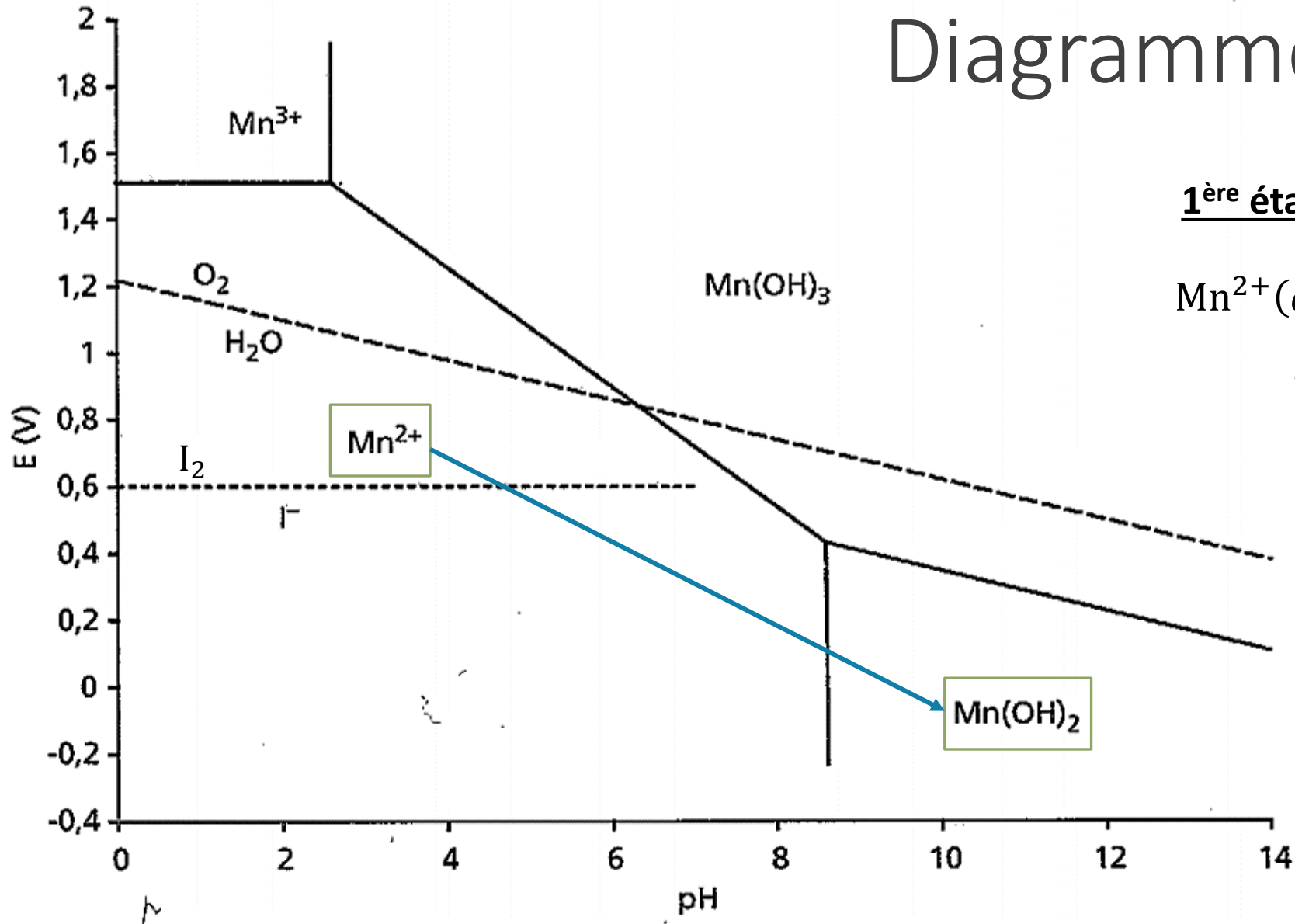
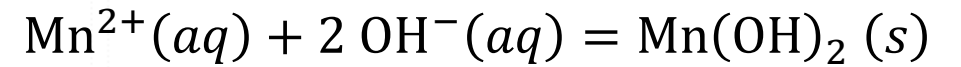
AGREGATION 2019

# Diagramme E-pH simplifié

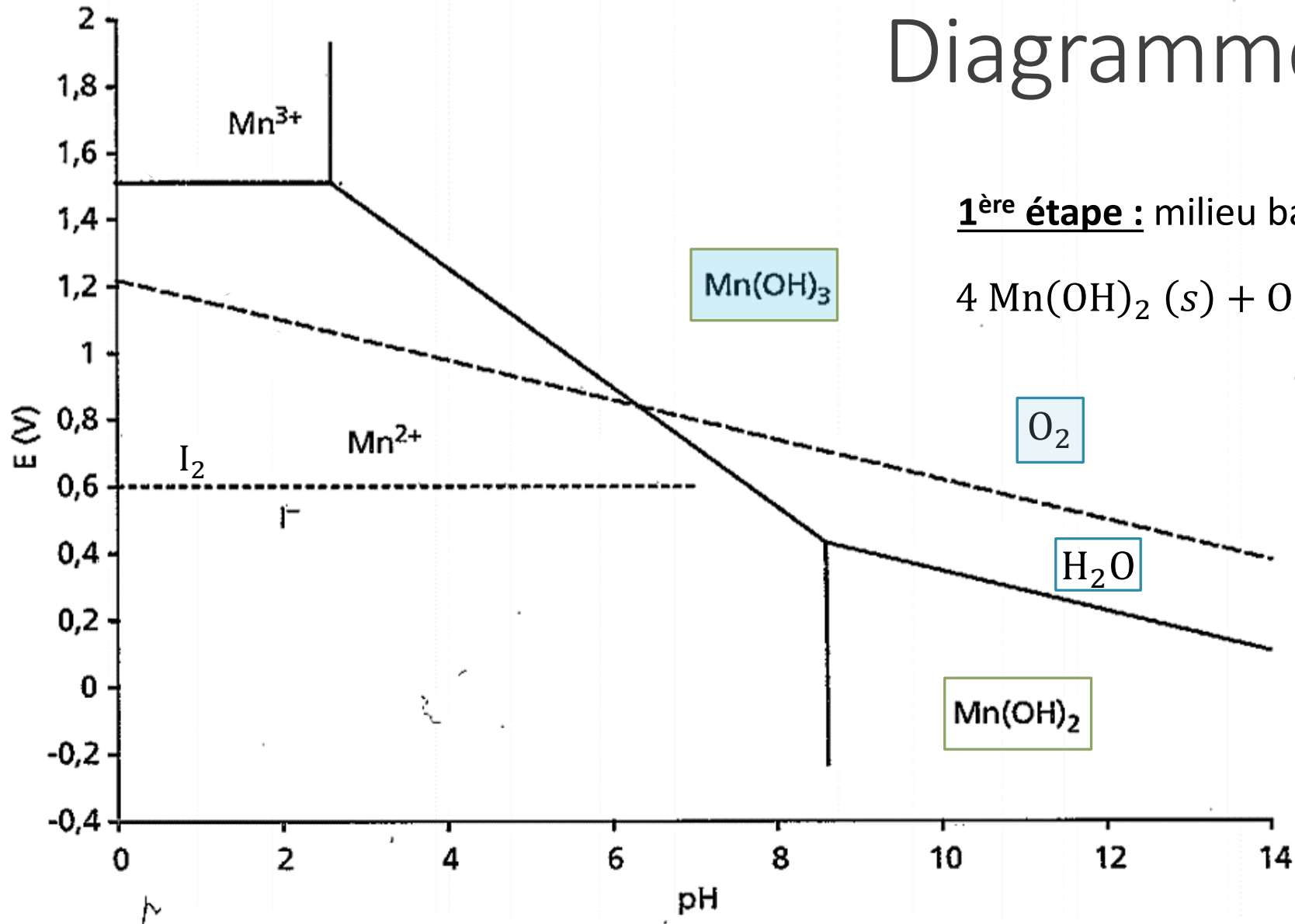


# Diagramme E-pH simplifié

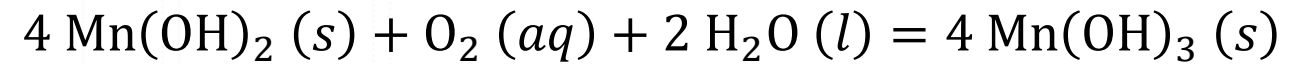
1<sup>ère</sup> étape : milieu basique



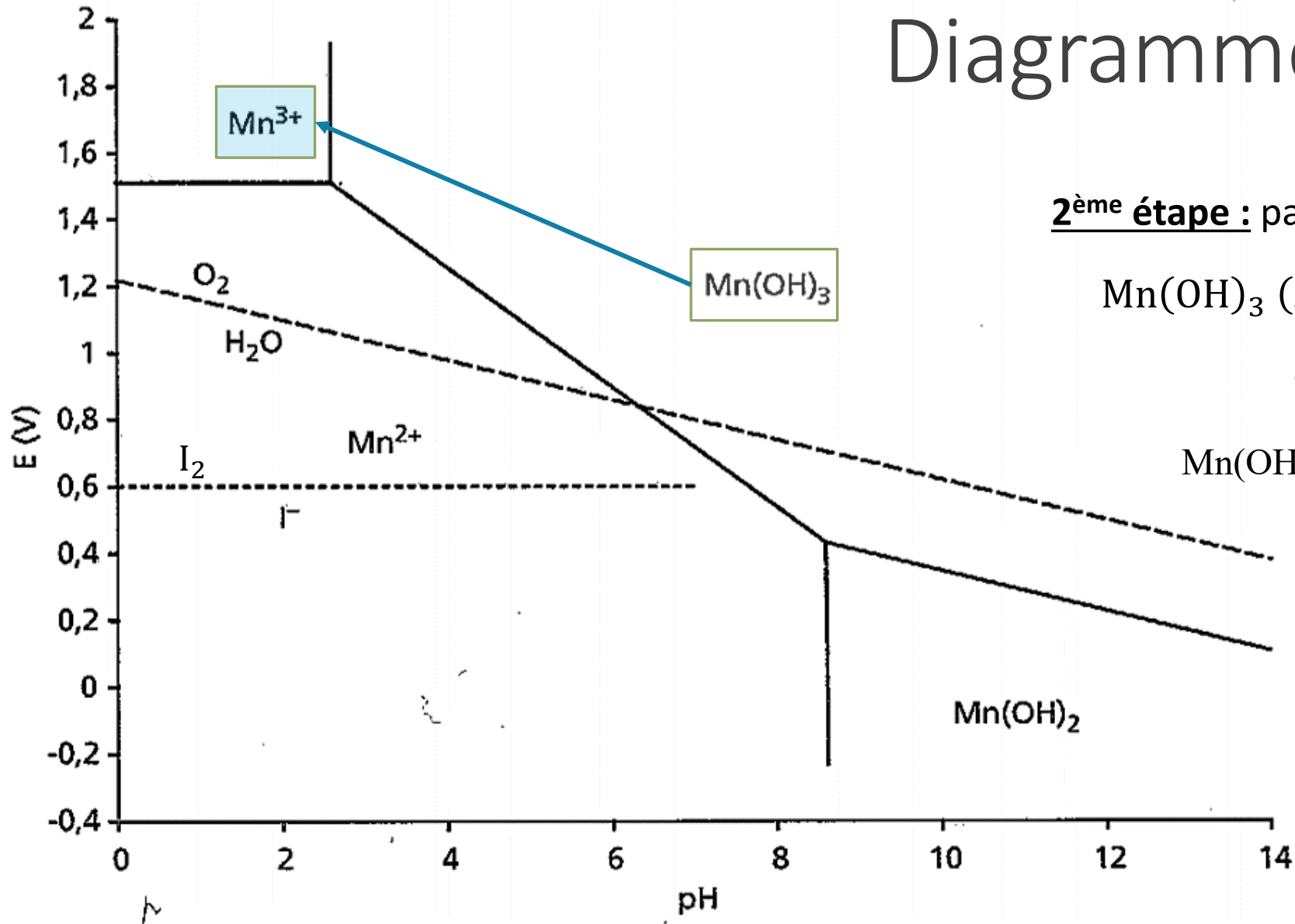
# Diagramme E-pH simplifié



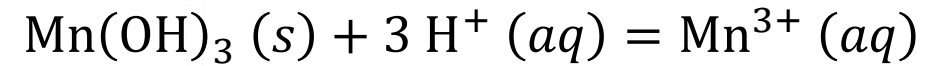
**1<sup>ère</sup> étape** : milieu basique



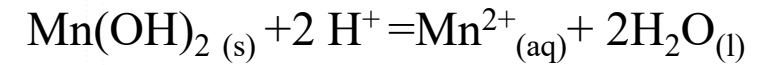
# Diagramme E-pH simplifié



**2<sup>ème</sup> étape** : passage en milieu acide

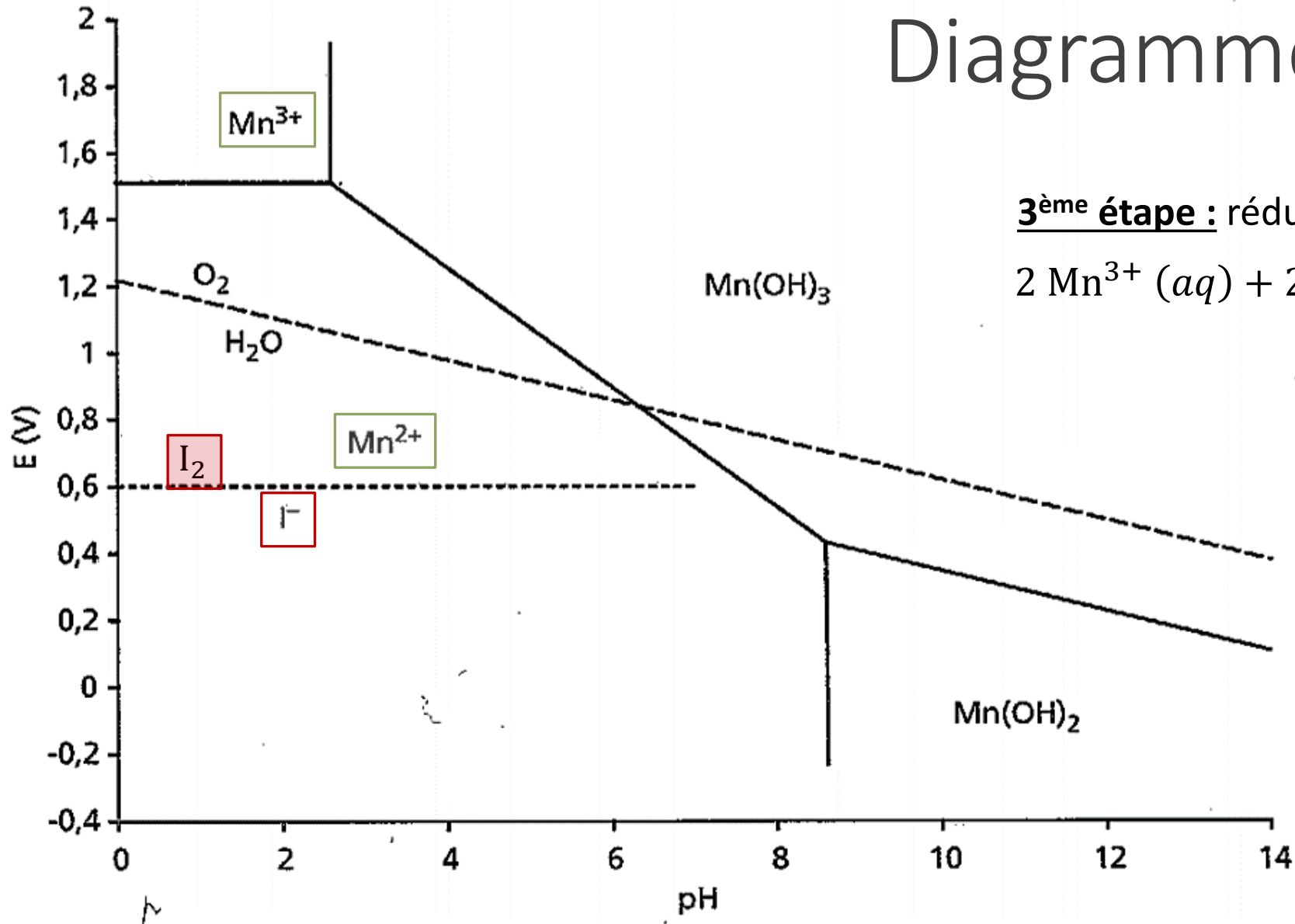
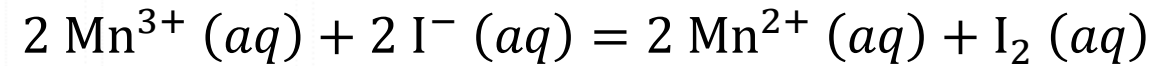


Et



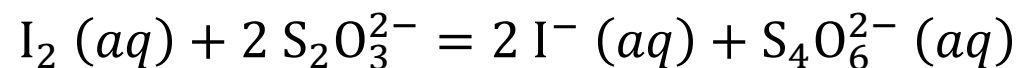
# Diagramme E-pH simplifié

**3<sup>ème</sup> étape** : réduction par les ions iodure



# Dosage iodométrique

**4<sup>ème</sup> étape :** dosage du diiode par les ions thiosulfate



$$C_{\text{thio}} = 1,0 \cdot 10^{-2} \text{ mol. L}^{-1}$$

