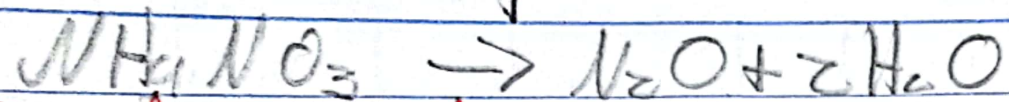
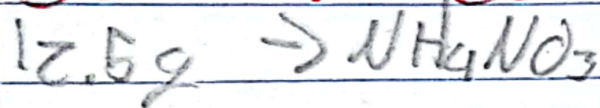


Tarea Estequiometria

30/9/22



~~Balanceada~~



$$12.5\text{g} \left( \frac{1\text{mol NH}_4\text{NO}_3}{80.06\text{g}} \right) \left( \frac{2\text{mol H}_2\text{O}}{1\text{mol NH}_4\text{NO}_3} \right) \left( \frac{18.02\text{g H}_2\text{O}}{1\text{mol H}_2\text{O}} \right) = 5.63\text{g H}_2\text{O} \downarrow$$