

RECAP

$$de = dq - dw = dq - Pdv$$

↑ BY FLUID

$$\text{IDEAL GAS: } de = C_v(T) dT$$

$$C_v = C_v(\text{SPECIES}, T)$$

$$\text{IDEAL GAS LAW: } P = \rho R T$$

e IS WELL-SUITED FOR NO-THROUGH-FLow SYSTEMS

FOR SYSTEMS W/ MOVING BOUNDARIES

E.G. ICE ENGINE

OR THROUGH-FLow

E.G. TURBINE / COMPRESSOR

A BETTER SUITED VARIABLE IS ENTHALPY

$$h \equiv e + Pv = e + \frac{P}{\rho}$$

$$dh = de + v dP + P dv$$

$$\downarrow$$
$$dq - \cancel{Pdv} + v dP + \cancel{Pdv}$$

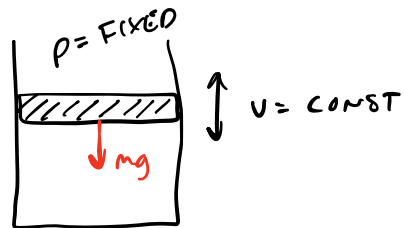
$$\rightarrow dh = dq + v dP$$

PROCESS OF INTEREST: CONSTANT PRESSURE

$$\rightarrow dP = 0$$

$$\rightarrow dh = dq$$

EX.



$$\left(\frac{\partial h}{\partial T} \right)_P = \left(\frac{\partial a}{\partial T} \right)_P$$

$$\underbrace{\hspace{1cm}}_{\equiv C_p}$$

FOR IDEAL GAS:

$$h = e + Pv = e + RT$$

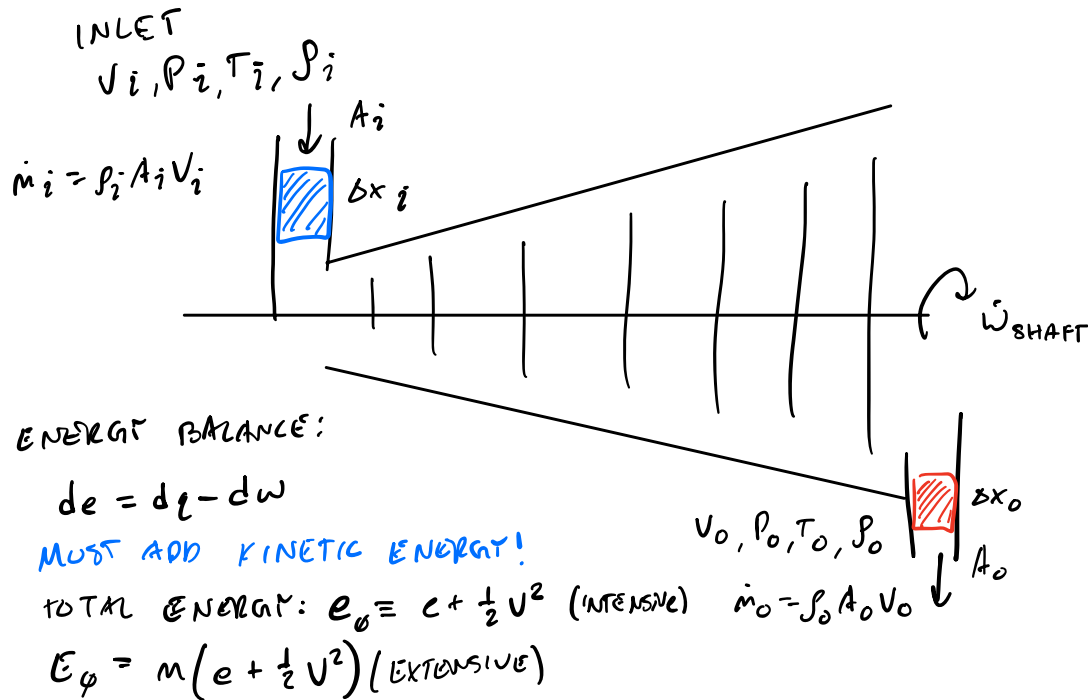
$$dh = de + RdT = C_v dT + R dT = (C_v + R) dT$$

$$\rightarrow \left(\frac{\partial h}{\partial T} \right)_P = C_v + R$$

$$C_p = C_v + R$$

OR

$$\bar{C}_p = \bar{C}_v + \bar{R}$$



1st LAW REWRITES:

$$de_o = dq - dw \quad \text{OR} \quad \Delta E_o = \Delta Q - \Delta W$$

$$\Delta \equiv \text{OUT} - \text{IN}$$

$$\text{IF ADIABATIC: } \Delta \dot{E}_o = \Delta \dot{Q} - \Delta \dot{W}$$

FOR TURBINE, OVER TIME Δt

$$\text{MASS IN} = \text{MASS OUT} = \dot{m} \Delta t$$

STEADY STATE

$$\Delta W = \dot{W} \Delta t$$

work power

MUST ACCOUNT FOR WORK PERFORMED:

- BY FLUID AT ENTRANCE
- ON FLUID AT EXIT
- BY FLUID ON BLADES

POWER $\dot{W} = \dot{W}_{\text{INLET}} + \dot{W}_{\text{OUTLET}} + \dot{W}_{\text{SHAFT}}$

WORK $W = (\quad \quad \quad) \Delta t$

WITH: $\dot{W}_{\text{INLET}} \Delta t = \underbrace{p_i A_i}_{\text{FORCE}} \underbrace{(-\Delta x_i)}_{\text{DISPLACEMENT}} \leftarrow \text{NEGATIVE SINCE VOLUME DECREASES}$

$\dot{W}_{\text{OUTLET}} \Delta t = p_o A_o \Delta x_o$

→ REASSEMBLE:

$\dot{m} \Delta t \left(e_o + \frac{1}{2} V_o^2 - e_i - \frac{1}{2} V_i^2 \right) = -(-p_i A_i \Delta x_i + p_o A_o \Delta x_o + \dot{W}_{\text{SHAFT}} \Delta t)$

÷ THROUGH BY Δt :

$\dot{m} \left(e_o + \frac{1}{2} V_o^2 - e_i - \frac{1}{2} V_i^2 \right) = - \left(\underbrace{-p_i A_i \frac{\Delta x_i}{\Delta t}}_{\dot{V}_i = \frac{\dot{m}}{\rho_i}} + \underbrace{p_o A_o \frac{\Delta x_o}{\Delta t}}_{\dot{V}_o = \frac{\dot{m}}{\rho_o}} + \dot{W}_{\text{SHAFT}} \right)$

→ $\dot{m} \left[\underbrace{\left(e_o + \frac{p_o}{\rho_o} \right)}_{h_o} - \underbrace{\left(e_i + \frac{p_i}{\rho_i} \right)}_{h_i} + \frac{1}{2} V_o^2 + \frac{1}{2} V_i^2 \right] = - \dot{W}_{\text{SHAFT}}$

$\dot{m} \left[\Delta h + \frac{1}{2} \Delta V^2 \right] = - \dot{W}_{\text{SHAFT}}$

$\Delta V^2 = \Delta(V^2)$

IF HEAT TRANSFER $\neq 0$

$\dot{m} \left(\Delta h + \frac{1}{2} \Delta V^2 \right) = \dot{Q} - \dot{W}_{\text{SHAFT}}$

→ $h_o \equiv h + \frac{1}{2} V^2$

ENTROPY

ANOTHER VARIABLE OF STATE

- MEASURES AMOUNT OF DISORDER WITHIN SYSTEM
- PREDICTS TOWARDS WHICH STATE A SYSTEM SPONTANEOUSLY TENDS

FOR A REVERSIBLE PROCESS $ds \equiv \frac{dq}{T} \geq 0$

2ND LAW OF THERMO

REVERSIBLE PROCESS

- INFINITELY SLOW
- FRICTIONLESS
- DUE TO SMALL DISTURBANCES



→ INCREASE FORCE ON PISTON, 1 CM IN AT A TIME