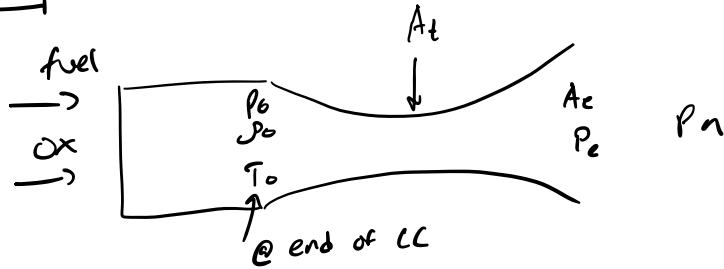


Recap



$$T_0 = T_0(\bar{m}, Q_R, p_0) \quad \begin{array}{l} \uparrow \quad \uparrow \\ \text{molar mass} \quad \text{Heat released} \\ \quad \quad \quad \text{unit mass} \end{array}$$

Express T in terms of $\left\{ \begin{array}{l} p_0, p_0, T_0 \\ \bar{m}, Q_R, \sigma \\ \text{Geometry} \end{array} \right.$

Remarks

1) H_2/O_2 rockets w/ very low \bar{m}

→ very high I_{sp} ($\sim 450s$)

u_e ($\sim 4500 \text{ m/s}$)

rockets w/ higher \bar{m} , lower I_{sp}/u_e

2) Low \bar{m} , high $\frac{M_{\text{tank}}}{M_p}$

b/c:

i) p_{H_2} is low, fixed mass H_2 needs large (heavy) tank

ii) H_2 liquid @ 77 K, need massive insulation

Alternative: solid rocket, no need for insulation or turbopumps
much lighter

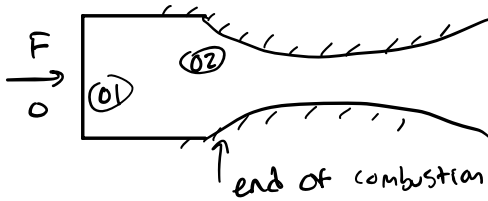
We'll study combustion in chapter 6

For now, link T to T_0 (end of CC)

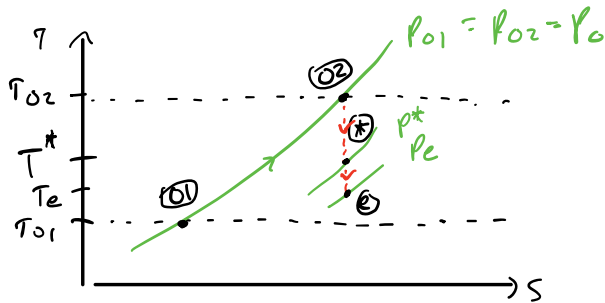
1. Performance

simplifications:

- combustion products = perfect gas, const. composition
- combustion = const. pressure process
- expansion: steady, 1D, isentropic (NO shock)



Book uses ① & ②, prefer ① & ② to indicate stagnation,



From energy w/m CC

$$\frac{d}{dt} \int_{CV} \rho \left(e + \frac{u^2}{2} \right) dV + \int_{CS} \rho \left(h + \frac{u^2}{2} \right) \underline{u}_{rel} \cdot d\underline{A} = \dot{Q} - \dot{W}_{shft} - \dot{W}_{shft} - \int_{CS} p u_b \cdot d\underline{A}$$

$$\dot{m}(h_{02} - h_{01}) = \dot{Q} = \dot{Q}_R \quad \text{Always true (e-bal)}$$

\uparrow
 heat released
 unit mass

$$Q_R = C_p (\bar{T}_{O2} - T_{O1}) \quad \text{only true if } C_p = \text{const} \quad (\text{Not a good assumption for } \Delta T \sim 2000 \text{ K})$$

From $\dot{Q} = Q_p \dot{m}$

$$\dot{Q}_3 = \dot{Q}_n \rightarrow \frac{\dot{Q}_n}{c_p} = T_{02} - T_{01} \rightarrow T_{02} = T_{01} + \frac{\dot{Q}_n}{c_p} \quad (A)$$

In process $\textcircled{O_2} \rightarrow \textcircled{e}$, $\dot{Q} = 0 \Rightarrow h_{O_2} = h_{O_e} = h_e + \frac{u_e^2}{2}$

$$\frac{u_e^2}{2} = h_{O_2} - h_e \quad (\text{energy balance})$$

$$\frac{u_e^2}{2} = c_p(T_{O_2} - T_e) \quad (\text{only if } c_p = \text{const})$$

$$u_e = \sqrt{2c_p(T_{O_2} - T_e)} = \sqrt{2c_p T_{O_2} \left(1 - \frac{T_e}{T_{O_2}}\right)} \quad \begin{array}{l} \text{depends on } M \\ \hookrightarrow \text{depends on } A \end{array}$$

If $\textcircled{O_2}$ to \textcircled{e} is also isentropic

$$\frac{T_e}{T_{O_2}} = \left(\frac{p_e}{p_{O_2}}\right)^{\frac{\gamma-1}{\gamma}}$$

$$\Rightarrow u_e = \sqrt{2c_p T_{O_2} \left(1 - \left(\frac{p_e}{p_{O_2}}\right)^{\frac{\gamma-1}{\gamma}}\right)} \quad (B)$$

$$\text{Recall } \begin{cases} c_p = \frac{\gamma}{\gamma-1} R \\ R = \frac{\bar{R}}{\bar{M}} \end{cases}$$

$$u_e = \sqrt{2 \frac{\gamma \bar{R}}{(\gamma-1) \bar{M}} T_{O_2} \left[1 - \left(\frac{p_e}{p_{O_2}}\right)^{\frac{\gamma-1}{\gamma}}\right]} \quad (C)$$

$$\text{For fixed } \begin{cases} \gamma \\ T_{O_2} \\ p_e \\ p_{O_2} \end{cases} \quad u_e \uparrow (\text{Exp} \uparrow) \text{ as } \bar{M} \downarrow$$

About $\dot{Q}_R \rightarrow$ plug (A) into (B)

$$u_e = \sqrt{2c_p \left(T_{O_1} + \frac{\dot{Q}_R}{c_p}\right) \left[1 - \left(\frac{p_e}{p_{O_2}}\right)^{\frac{\gamma-1}{\gamma}}\right]}$$

Let's note $T_{O_1} \ll \dot{Q}_R / c_p$

$$\rightarrow u_e \approx \sqrt{2 \dot{Q}_R \left[1 - \left(\frac{p_e}{p_{O_2}}\right)^{\frac{\gamma-1}{\gamma}}\right]}$$

$$\text{introduce } \bar{Q}_R \equiv \dot{Q}_R \cdot \bar{M}$$

\uparrow heat released per mole propellant

$$\bar{Q}_R = \frac{\bar{Q}_R}{\bar{m}} \Rightarrow u_e \approx \sqrt{2 \frac{\bar{Q}_R}{\bar{m}} \left[1 - \left(\frac{P_e}{P_{02}} \right)^{\frac{\gamma-1}{\gamma}} \right]}$$

↑ ↑
propellant nozzle property
properties

For high u_e , look for high \bar{Q}_R & low \bar{m}

Ratio $\frac{\bar{Q}_R}{\bar{m}}$ depends upon $\frac{\text{Fuel}}{\text{oxidizer}}$ ratio (lean or rich compared to stoichiometric)

Using a high fuel: oxidizer ratio (rich)

→ $\begin{cases} \text{low } \bar{m} \\ \text{low } \bar{Q}_R \end{cases}$ b/c not all fuel burns

→ Gain in low \bar{m} offsets loss in low \bar{Q}_R