

# Refrigeration and Air Conditioning

## INTRODUCTION To RAC

### Refrigeration & Air Conditioning

What is Refrigeration:

Process of maintaining lower temperature as compared to surroundings.

Difference b/w Cooling & Refrigeration?

Taking example of the cup of tea. At room temperature the cup of hot tea will lose its heat is cooling.

But if it is cooled below the atmospheric temperature then it is Refrigeration.

⇒ How do we maintain lower temperature than surrounding?

⇒ The substance used for this purpose is known as refrigerant.

### Unit of Refrigeration

TON of Refrigeration  $\rightarrow$  US  $\rightarrow$  TR = 2000 lbs  
 $\rightarrow$  = 907.7 kg short ton

TONNES = 1000 kgs metric ton

$$1 \text{ TON} = 907.7 \text{ kg.}$$

TR

Amount of heat required to remove from 1 ton of water at  $0^\circ\text{C}$  in order to convert it into ice at  $0^\circ\text{C}$  in one day (24 hrs).

$$\dot{Q}_{TR} = 210 \text{ KJ/min} = \frac{210 \text{ KJ}}{60 \text{ s}} = 3.5 \text{ KW}$$

if capacity is  $\dot{Q}_{TR}$  means 3.5 KW  
 $2\dot{Q}_{TR}$  means  $2(3.5) \text{ KW}$

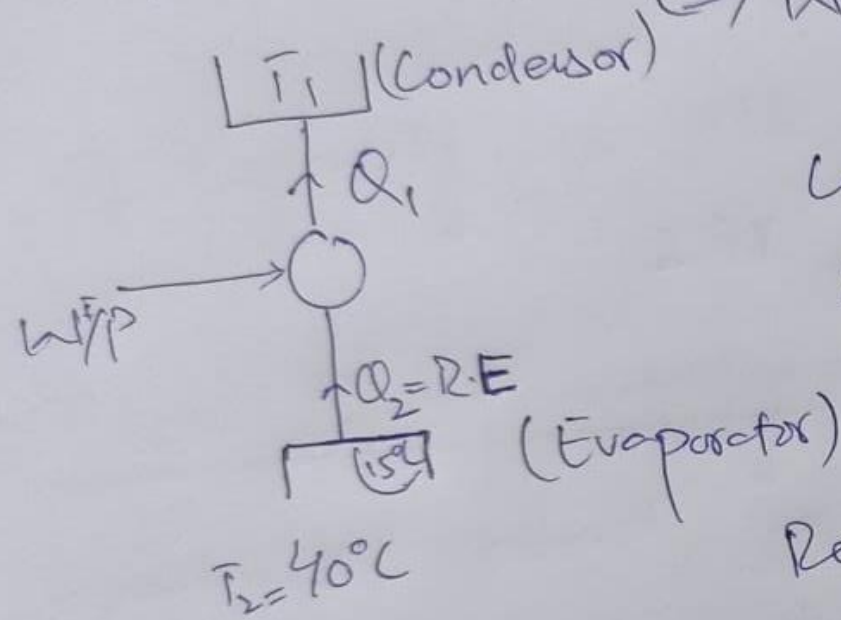
2<sup>ND</sup> Law of Thermodynamics:- Clausius's Statement

Impossible to construct a device which transfers heat from lower temperature to higher temperature w/o any external input.



Refrigerator

Evaporator  $\Rightarrow$  Refrigerator boils in evaporator  
 $\Rightarrow$  where the refrigerant picks up heat

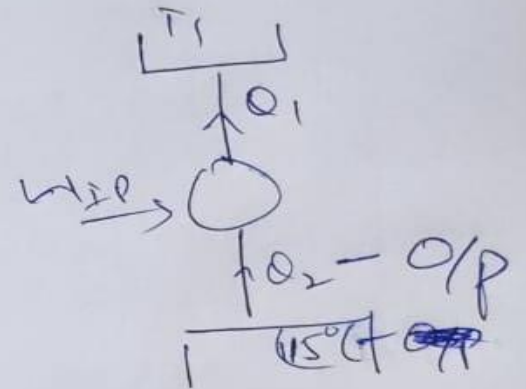


Condensor [condensation takes place]  
 Evaporator [Boiling takes place.]

Remember in Refrigeration  
 Condenser o/p has the higher temperature  
 Evaporator o/p has the lower temperature.

COP [Coefficient of Performance]

$$\frac{R.E}{W I/p} = \frac{\text{Desired Effect}}{W I/p} = \frac{O/p}{I/p}$$



Refrigeration Capacity = mass flow rate  $\times$  R.E  $T_2$

$$= \frac{\text{kg}}{\text{s}} \times \frac{\text{kJ}}{\text{kg}}$$

R.C = kW

$$P I/p = m \times W I/p$$

$$= \text{kW units}$$

$$COP = \frac{O/P}{I/P} = \frac{R.C}{P I/P}$$

So what COP tells us?

⇒ The more the mass flow rate of the refrigerant the higher will be the cooling effect.

⇒ Means higher will be the Refrigeration Capacity.

~~⇒ Higher will be the COP.~~

⇒ Capacity of fridges / Refrigerators are known.

For 1-2 person

3-4 "

Family

it is 50 - 200 liters, 1.7 cuft - 7 cuft

250 - 300 liters, 9 cuft - 14 cuft

400 - 500 liters, 14 - 18 cuft

$$COP = \frac{R.C \rightarrow \text{Fixed}/\text{unit}}{P.I/p}$$

COP ↑

means

$P.I/p \downarrow$

COP ↓

means

$P.I/p \uparrow$

COP signifies the running cost of the refrigerator

⇒ The higher the COP lesser will be the power input & the lesser will be the running cost

⇒ Star rating of refrigerators

$\star\star$   
 $P.I/p \neq 2\text{-star} > \star\star\star 3\text{-star} > \star\star\star\star\star 5\text{-star}$



$$\text{COP}_{AC} > \text{COP}_R$$

(As we know COP defines running cost. so it means

Running cost AC < Running cost R X does AC have lower cost?  
No. Why?

Why!

⇒ Refrigerated space for AC is large (A whole Room)

⇒ Continuous removal heat is required.

In a room full of appliances (fans, tube lights), people generate energy. AC works to reject that heat

⇒ Room is not insulated. So heat leakages take place. That also ~~heat~~ is to be removed by A.C

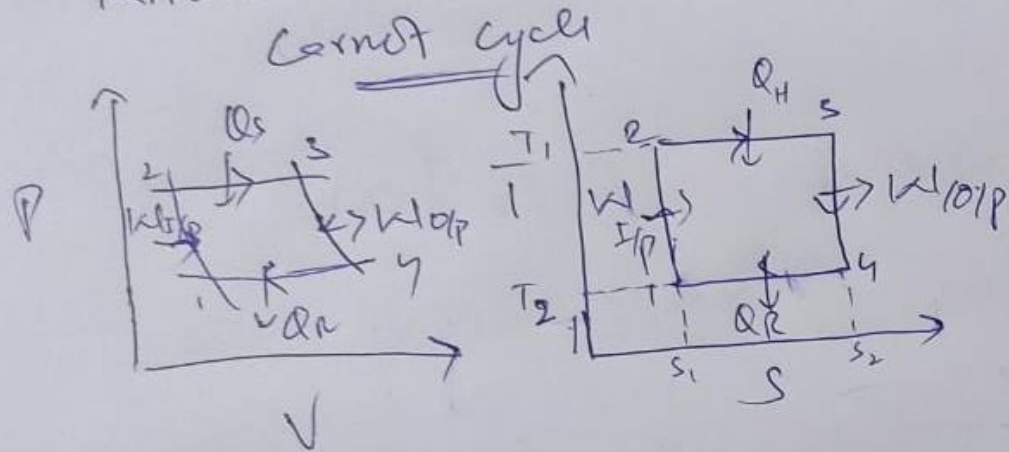
⇒ Comparatively Refrigerator have less refrigerated space  
⇒ Well insulated.

So, COP of AC is higher but it doesn't mean it will have a lower running cost.

\* Running ~~AE~~ cost  $_{AC} >$  Running cost Refrigerator

# Ideal Refrigeration Cycle.

Known as Reversed Carnot cycle.



Now Reverse it

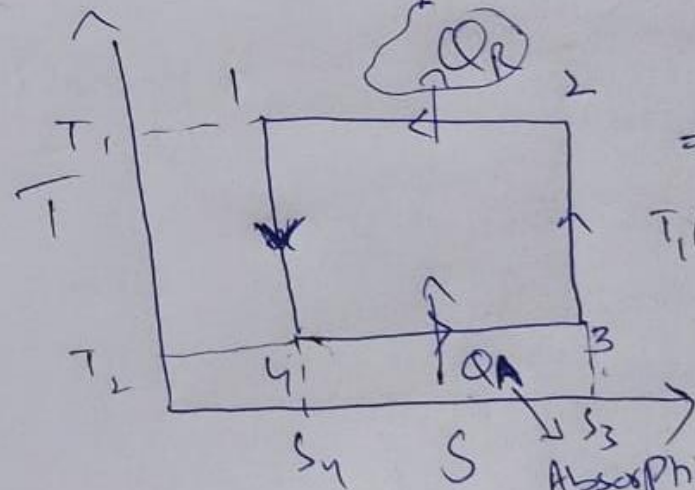
When reversing the H.E we get H.P

2- Isothermal

2- <sup>Rev.</sup> Adiabatic

$$\eta = \frac{O/P}{I/P} = 1 - \frac{Q_R}{Q_S}$$

$$\eta = 1 - \frac{T_2(S_2 - S_1)}{T_1(S_2 - S_1)} \quad \eta_{H.E} = 1 - \frac{T_2}{T_1}$$



$$\begin{aligned} \text{COP} &= \frac{Q_L}{Q_H - Q_L} \\ &= \frac{T_2(S_4 - S_3)}{T_1(S_4 - S_3) - T_2(S_4 - S_3)} \\ &= \frac{T_2(S_4 - S_3)}{T_1 - T_2(S_4 - S_3)} \\ \text{Absorption COP} &= \frac{T_2}{T_1 - T_2} \end{aligned}$$



$$\text{COP} = \frac{T_2}{T_1 - T_2} = \frac{T_L}{T_H - T_L} \quad (\text{IDEAL Refrigeration cycle})$$

Not possible practically?

Because it has two Isothermal process  
& two Adiabatic Process.

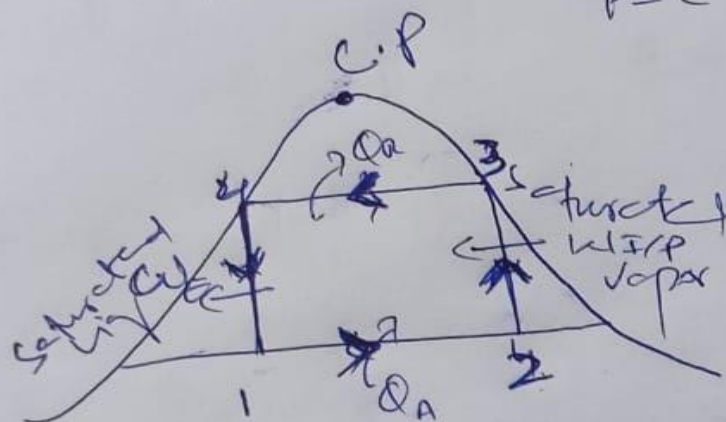
- Isothermal Process is the slowest Process
- Adiabatic Process is the fastest Process
- This combination is not possible together.



→ During the Phase change Pressure and Temperature (Thermodynamic properties remain constant)

L-V

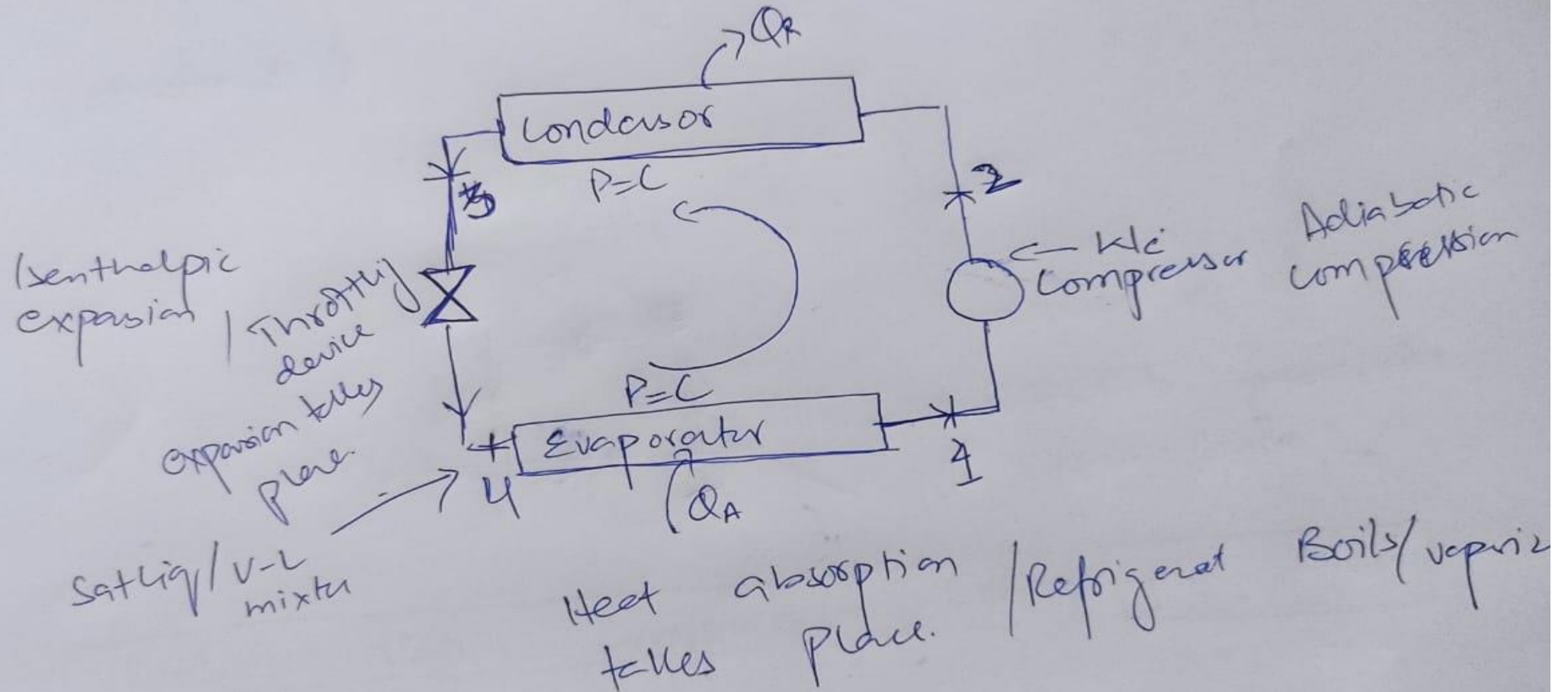
V-L  $(T = C)$   
 $P = C$



idea of  
VCRS generates  
from here

VCRS : Vapor Compression Refrigeration cycle

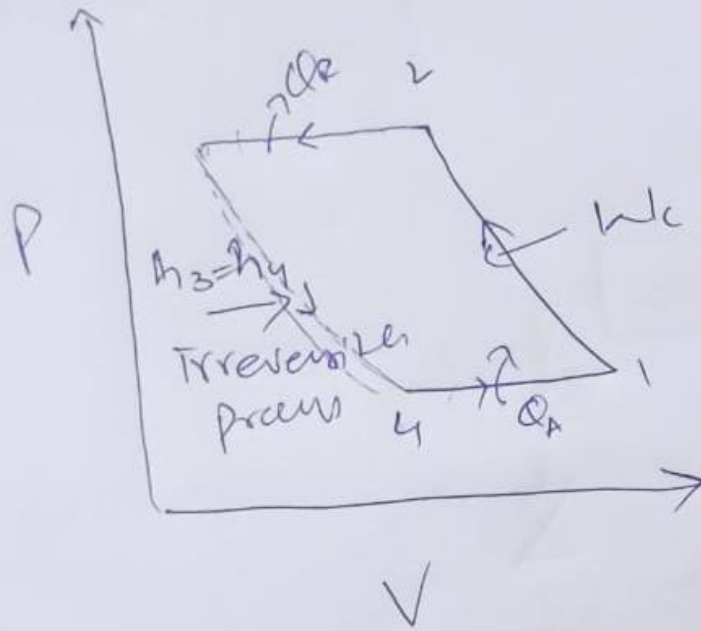
# VCRS: Vapor Compression Refrigeration cycle



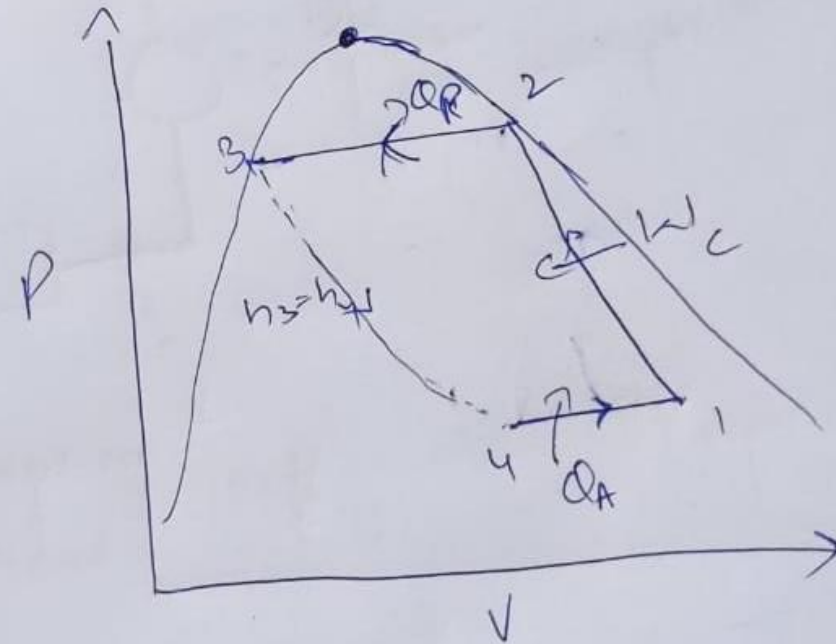
$h_3 = h_4$  so  $T_3 = T_4$  ? NO, Why

$$h = u + P v \uparrow \uparrow$$

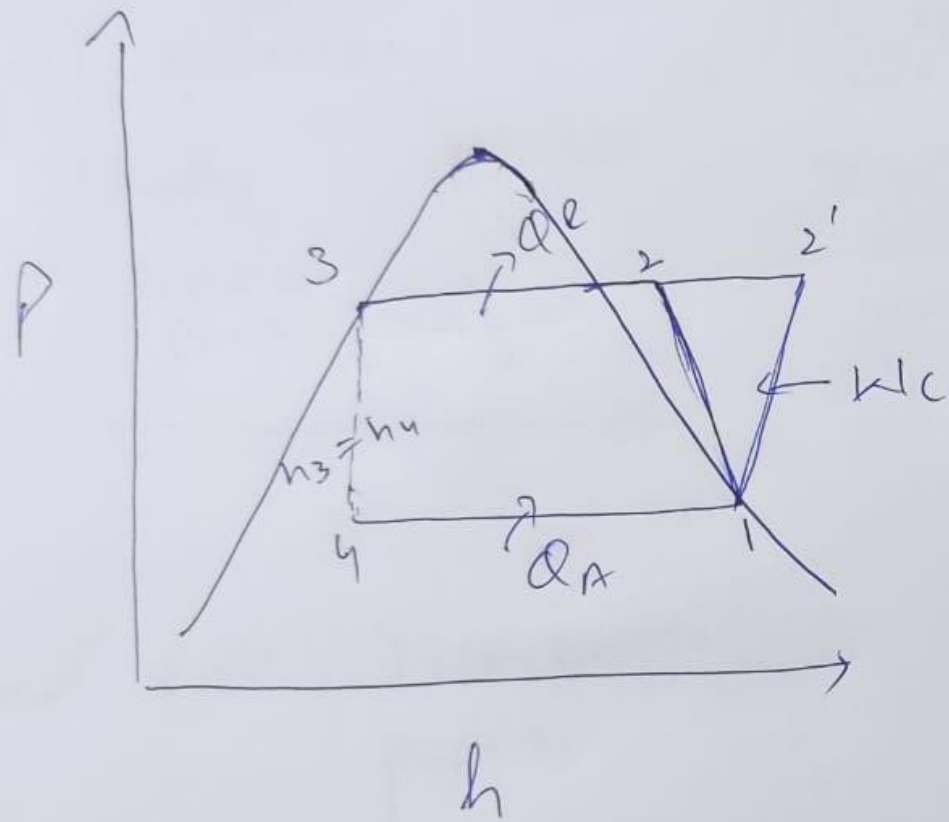
Temperature reduces  
so does  
internal energy



Dotted line: Irreversible process



# P-h Diagram

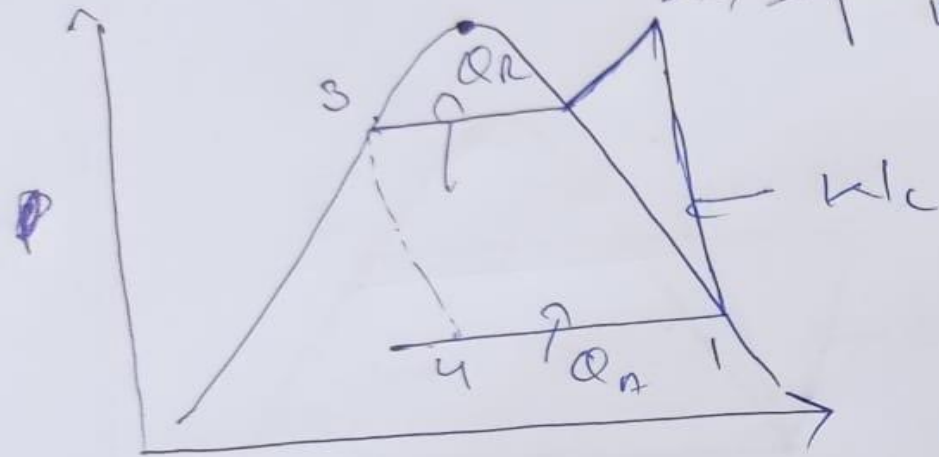


Process 1-2 False way?  
 $\uparrow h = \uparrow u + \uparrow Pv$  in compression

So in compression  
 the enthalpy  
 process  
 is increasing



# T-S Diagram



Process 1-2  
Isentropic Compression

Process 2-3  
Constant Pressure heat rejection

Process 3-4  
Isentropic Expansion

Process 4-1  
Constant Pressure heat Absorption.

Process 1-2 :- Rev. ~~Constant~~ Adiabatic Compression

SPEED  $h_1 + \frac{C_1^2}{2} + g z_1 + q_v = h_2 + \frac{C_2^2}{2} + g z_2 + w_{cv}$

Adiabatic

$$h_1 + g z_1 = h_2 + w_{cv}$$

$$h_1 - h_2 = w_{cv} \quad \text{etc. } \text{---} h_1 - h_2$$

$$w_c = -w_{cv} = h_2 - h_1$$

$$w_c = h_2 - h_1$$

$$\text{Power of compressor} = \dot{m} \times w_c = \dot{m} (h_2 - h_1)$$

Process 2-3 ∴ Constant Pressure heat reject.

$$h_2 + \frac{c_2^2}{2} + g z_2 + \cancel{h_{2v}} = h_3 + \frac{c_3^2}{2} + g z_2 + h_{2cv}$$

$$h_2 + q = h_3$$

$$q = h_3 - h_2$$

$$\boxed{q_R = -q = h_2 - h_3}$$

$$\boxed{q_R = h_2 - h_3}$$

$$\text{Total } q_R = m \cdot x (h_2 - h_3) = q_c$$

condenser.

Process 3-4

Throttling  
 $h_3 = h_4$

Process 4-1

constant Pressure

Evaporator  
heat

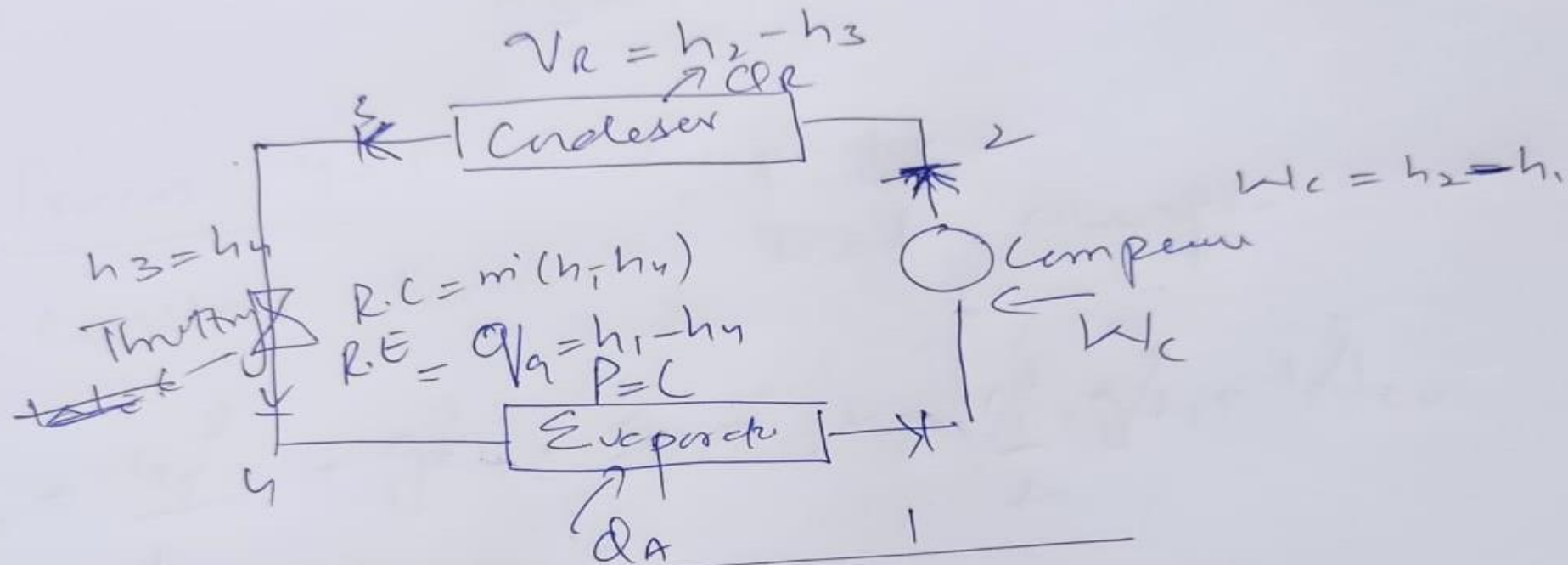
absorption

$$h_4 + \frac{c_4^2}{2} + g z_4 + q = h_1 + \frac{c_1^2}{2} + g z_1 + \cancel{h_{1cv}}$$

$$q = h_1 - h_4$$

$$\boxed{q_{41} = q = h_1 - h_4}$$





$$COP = \frac{Q_L}{Q_H - Q_L} = \frac{R.E}{W.I/P}$$

$$COP = \frac{h_1 - h_4}{h_2 - h_1} = \frac{h_1 - h_3}{h_2 - h_1}$$

$$\therefore h_3 = h_4$$

$$\text{COP} = \frac{m(h_1 - h_3)}{m(h_2 - h_1)} = \frac{R.C}{P.I/P}$$

These are the basic parts  
of vcl's.

~~Compressor.~~