

Application of SFEE.

Nozzle

A device which increases velocity of any fluid.

⇒ By reducing pressure Pressure Energy is converted to K.E

& Diffuser

Increases pressure
Decreases velocity

Mach Number:-

$$= \frac{\text{velocity of fluid}}{\text{Sonic velocity}}$$

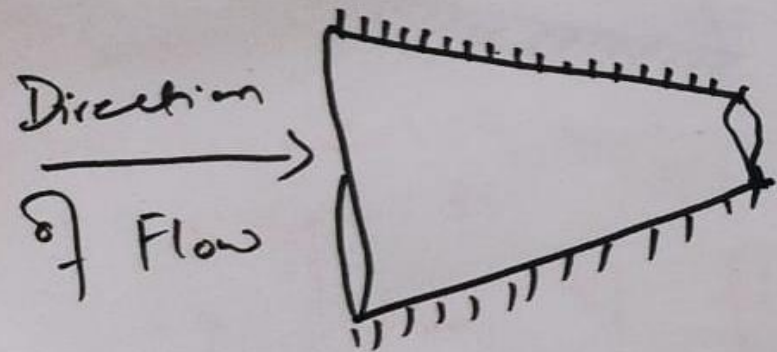
In an ideal gas

$$\text{sonic velocity} = \sqrt{\gamma RT}$$

(1)

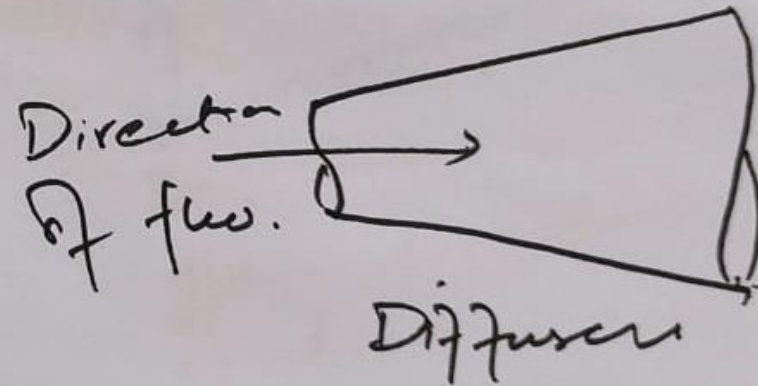
$M > 1$	(Supersonic Flow)	Velocity of Fluid $>$ Sonic velocity
$M < 1$	(Subsonic Flow)	Velocity of fluid $<$ Sonic Velocity
$M = 1$	(Sonic Flow)	Velocity of fluid $=$ Sonic Velocity

Subsonic Flow $M < 1$



Nozzle

Cross section area decreases along the direction of flow.

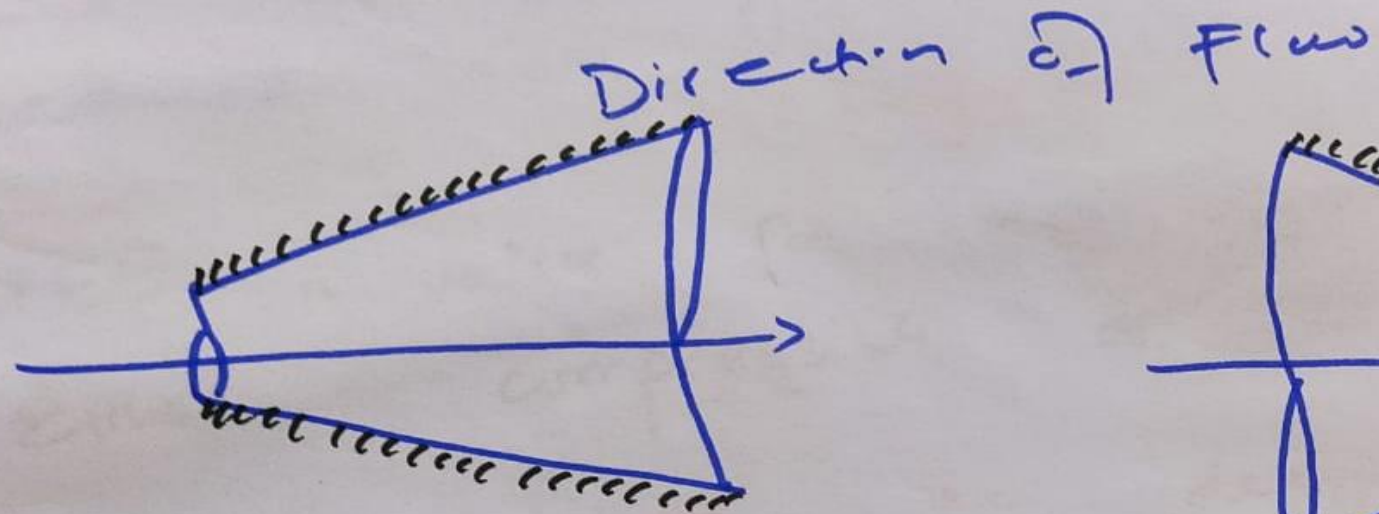


Diffuser

Cross section area increases along the direction of flow. (2)

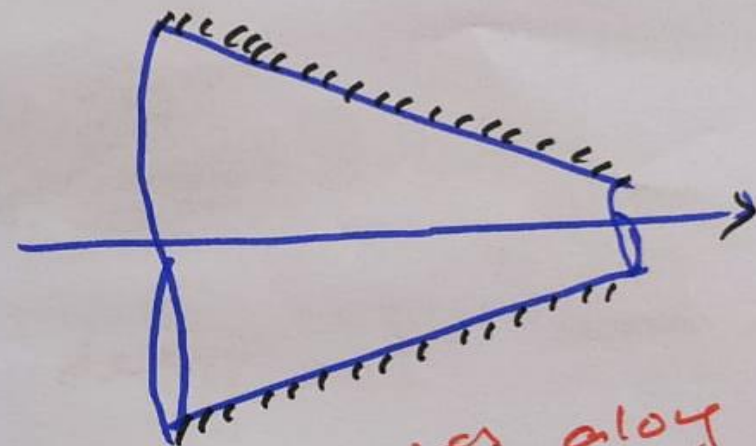
Supersonic Flow

$$M > 1$$



Increases
along the
direction of flow

Nozzle



Decreases along
the direction
of flow.

Diffuser

13

= For compressible Fluid Flow

→ When the velocity is high, its volume is reduced.

(Importance)

→ SFEE is valid for any type of fluid.
compressible or incompressible.

Either

When would be the design for ~~sub~~ sonic flow

$$M=1?$$

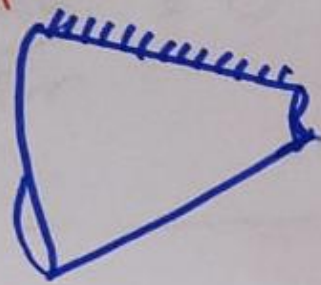
For a fluid to have

$$M=1$$

feasible
of time

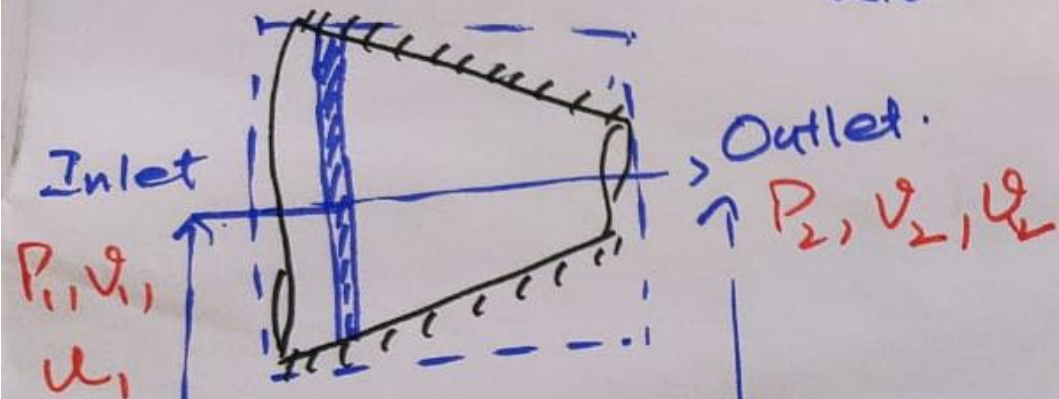
is not
layer period

(4)



Nozzle

kinetic transfer = 0



Well Insulated
Perfect Insulation
 $z_1 \rightarrow z_2$

$$\frac{dQ}{dm} = 0$$

$$1) \frac{dd}{dm} = 0$$

$$2) \frac{dw}{dm} = 0$$

$$(iii) z_1 = z_2 \quad \Delta PE = 0$$

$$4) \Delta KE \neq 0$$

Inlet
Sp. Enthalpy $h_1 = u_1 + P_1 v_1$

Outlet
Sp. Enthalpy $h_2 = u_2 + P_2 v_2$ (5)

SFEE

$$h_1 + \frac{v_1^2}{2} + \cancel{gz_1} + \cancel{\frac{d\phi}{dm}} \stackrel{?}{=} 0$$
$$= h_2 + \frac{v_2^2}{2} + \cancel{gz_2} + \cancel{\frac{d\phi}{dm}} \stackrel{?}{=} 0$$

$$h_1 + \frac{v_1^2}{2} = h_2 + \frac{v_2^2}{2}$$

$$h_1 - h_2 = \frac{v_2^2}{2} - \frac{v_1^2}{2}$$

if $v_2 \gg v_1$

(b)

$V_1^2 = \text{neglected}$

$$\boxed{\frac{V_2^2}{2} = h_1 - h_2}$$

$$V_2^2 = 2(h_1 - h_2)$$

$$V_2 = \sqrt{2(h_1 - h_2)}$$

unity
 $\frac{J}{1g}$

SFEE:

$$\frac{J}{1g} \rightarrow \boxed{h_1 + \frac{V_1^2}{2} + g z_1 + \frac{dQ}{dm} = h_2 + \frac{V_2^2}{2} + g z_2 + \frac{dW}{dm}}$$

(7)

$$K.E = \frac{1}{2} m v^2 \rightarrow \text{Nm or Joule}$$

$$\bar{E} = U + K.E + P.E$$

"Dividing by mass"

$$e = u + \frac{v^2}{2} + g^2$$

$\rightarrow \frac{J}{kg}$

$\downarrow m^2/s^2$ (Dimension of v)

$$v^3/2$$

$$\approx \frac{J}{kg}$$

$$J = Nm$$

$$N = kg m s^{-2}$$

$$\frac{kg m s^{-2} m}{kg}$$

$$= m^2/s^2$$

(e)

So

$$V_2 = \sqrt{2(h_1 - h_2)} \Rightarrow J/1g$$

$\leftarrow J/1g$

SFEE:

$$h_1 + \frac{V_1^2}{2000} + \frac{gz_1}{1000} + \frac{\cancel{dQ}}{dm} = h_2 + \frac{V_2^2}{2000} + \frac{gz_2}{1000} + \frac{\cancel{dW}}{dm}$$

$\leftarrow J/1g$

$\leftarrow J/1g$

$\leftarrow J/1g$

$\leftarrow J/1g$

$\leftarrow J/1g$

$$V_2 = \sqrt{2000(h_1 - h_2)}$$

$$\boxed{V_2 = 44.72 \sqrt{\Delta h}}$$

(9)

Four Declared cars

$$1) \frac{dQ}{dm} = 0$$

$$H_T = 0$$

$$H_T = 0$$

} → Almost always

$$2) \frac{dH}{dm} = 0$$

$$3) \Delta PE = 0$$

$$z_1 = z_2$$

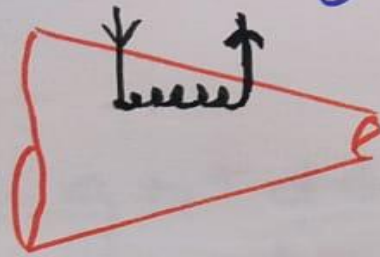
$$4) \Delta KE \neq 0$$

$$U_2 = \sqrt{2gh}$$
$$U_2 = \sqrt{2(h_1 - h_2)}$$

(10)

Special ~~eq~~ con

This can
happen



Electric Work = Power

Heat $\frac{dW}{dt}$

$$= UI$$
$$= I^2 R T \rightarrow dQ$$

But Δ work transfer
is always zero.

In case of Diffuser "same approach"
will be used

Possibility Different " h_1 & h_2 value", (11)

Turbine & Compressor

ΔFEE

$$h_1 + \frac{v_1^2}{2} + g z_1 + \frac{dQ}{dm} = h_2 + \frac{v_2^2}{2} + g z_2 + \frac{dW}{dm}$$

External work $\begin{cases} \rightarrow \text{Shaft work} \\ \rightarrow \text{Electrical work} \end{cases}$

\downarrow
External work.

Turbine:-

Rotodynamic Machine

Provides us shaft work.

(11)

Turbine...

and

Consumes thermal Energy
Enthalpy in case of steam
produces work (shaft
work)

Turbine: [Thermodynamic System]

work from system [+ve]

Compressor :-

Reciprocating & Centrifugal
[work done on the system (-ve)]

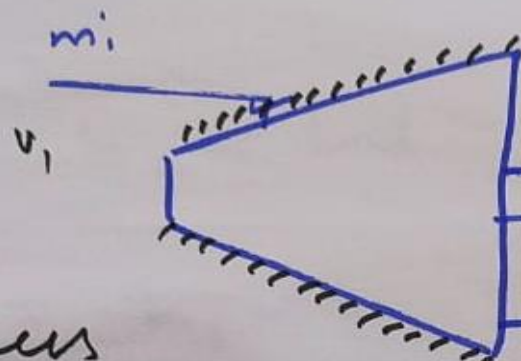
(13)

SFEE for turbine:

Assumption:

No Friction

Loss / Friction Loss



$$m_1 = m_2 = m$$

$$\frac{dW}{dm} \neq 0 \quad \frac{dm}{dt}$$

1) assumption :- Heat lost = 0

2) Isentropic (Frictionless)
 $dS = 0$ (Reversible) + Adiabatic

3) $v_1 = v_2$
 Inlet outlet

$$\Delta K.E = 0$$

4) $z_1 = z_2 \quad \Delta P.E$

(14)

$$h_1 = h_2 + \frac{dw}{dm}$$

SFEE

\Rightarrow

$$h_1 = h_2 + w$$

$$\frac{dw}{dm} \approx w \cdot \frac{1}{g}$$

$\frac{J}{kg}$

or

$12 \frac{J}{kg}$

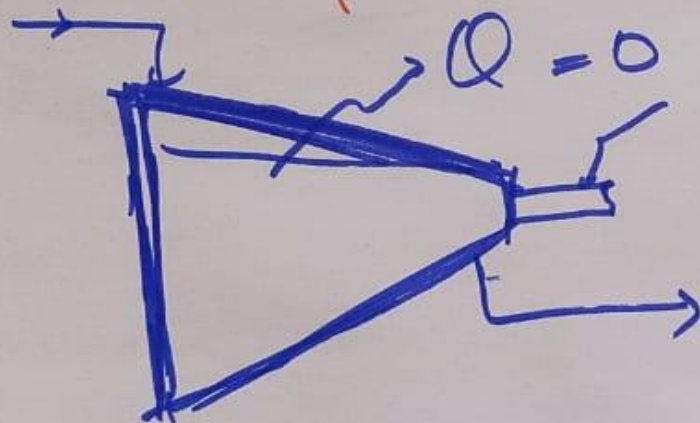
$$w = h_1 - h_2$$

$$h_1 > h_2$$

work

transfer for turbine.

Compressor ::



w (-ve)

$$1) \Delta KE = 0$$

$$2) \Delta PE = 0$$

$$3) \Delta Q = 0$$

(15)

$$h_1 = h_2 + |w|$$

$$h_1 = h_2 + (-w)$$

$$|w| = h_2 - h_1$$

$$KJ/kg \text{ or } J/kg$$

SFEE

for

compressor

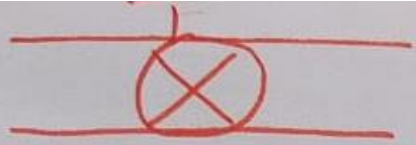
Throttling

Process:-

Irreversible
Process

Rapid reduction of pressure.

(16)



valves / tubes

① Examples of throttling process

→ Capillary tube

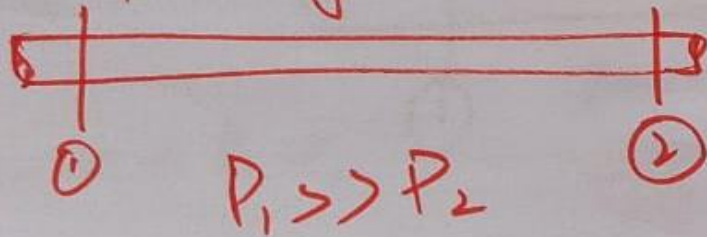
is example of throttling process.

⇒

→ Partially open valve.

→ Adjustable valve

→ Capillary tube ②



By virtue of friction pressure is reduced

③



Porous plugs

$P_1 \gg P_2$
 $P_2 \ll P_1$

(17)

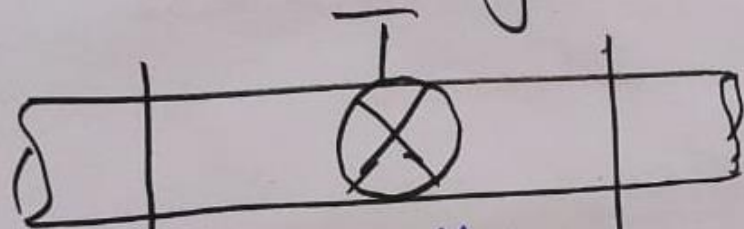
Throttling process

⇒ Isentropic process

$$h_1 = h_2$$

" But it doesn't mean all Isentropic processes are throttling processes "

Reel gas :-



$$h_1 = u_1 + P_1 v_1$$

(1)

partially opened

(2)

$$h_2 = u_2 + P_2 v_2$$

(18)

if

$$P_1 v_1 \gg P_2 v_2$$

$$u_1 - u_2 = P_2 v_2 - P_1 v_1$$

Temp will increase along flow

$$u_1 - u_2 = -ve$$

$$u_2 > u_1$$

$$T_2 > T_1$$

if $P_2 v_2 \gg P_1 v_1$

$$T_1 > T_2$$

Temp will ~~increase~~ decrease
along the direction
of flow)

So it doesn't mean that temperature
will always reduce in case of throttling
because of reduction in pressure.

It can also increase in direction
of flow, meanwhile pressure decreases

(19)

For Ideal gas

$$h = f(T)$$

$$h_1 = h_2$$

$$T_1 = T_2$$

Throttling
case
for ideal
gases

temperature remain
constant.

In case of real gases we are

getting both heating
($T_1 > T_2$)

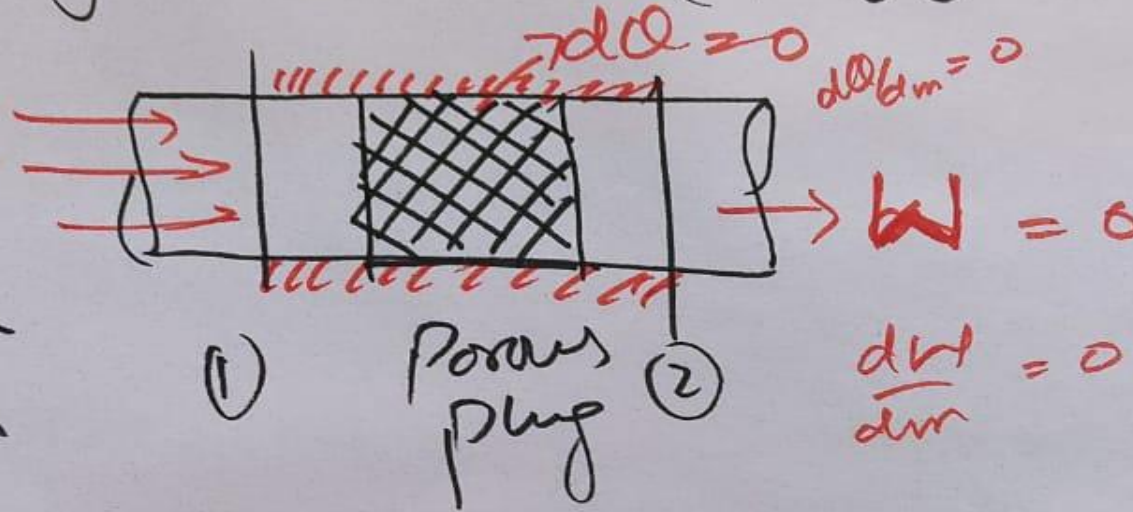
$$T_2 > T_1$$

Joule's Thomson
coefficient of expansion

(26)

Throttling Process (SFEE)

Not time
enough
plus Not
enough
to transfer
heat



Joule Thompson
Process

$$\begin{aligned} z_1 &= z_2 \\ v_1 &= v_2 \end{aligned} \quad \left. \begin{aligned} \Delta P.E. &= 0 \\ \Delta KE &= 0 \end{aligned} \right\}$$

SFEE

$$h_1 = h_2$$

(2)