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Hidden alternative structures of proline isomerase essential for catalysis

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A long-standing challenge is to understand at the atomic level how protein dynamics contribute to enzyme catalysis. X-ray crystallography can provide snapshots of conformational substates sampled during enzymatic reactions1, while NMR relaxation methods reveal the rates of interconversion between substates and the corresponding relative populations^{1,2}. However, these current methods cannot simultaneously reveal the detailed atomic structures of the rare states and rationalize the finding that intrinsic motions in the free enzyme occur on a timescale similar to the catalytic turnover rate. Here we introduce dual strategies of ambient-temperature X-ray crystallographic data collection and automated electron-density sampling to structurally unravel interconverting substates of the human proline isomerase, cyclophilin A (CYPA, also known as PPIA). A conservative mutation outside the active site was designed to stabilize features of the previously hidden minor conformation. This mutation not only inverts the equilibrium between the substates, but also causes large, parallel reductions in the conformational interconversion rates and the catalytic rate. These studies introduce crystallographic approaches to define functional minor protein conformations and, in combination with NMR analysis of the enzyme dynamics in solution, show how collective motions directly contribute to the catalytic power of an enzyme.

It has become widely accepted that not only the chemical steps of an enzymatic reaction, but also protein conformational rearrangements, contribute to the rate acceleration of enzymes^{1,3–6}. For human CYPA, a peptidyl-prolyl cis/trans isomerase, NMR relaxation experiments revealed that millisecond motions during catalysis occur in a 'dynamic network' that could be described as a two-state interconversion between enzyme conformations bound to the cis- and trans-Pro substrates. Further, the free enzyme samples the same two conformations on a similar timescale, but with the equilibrium shifted far towards one substate4. Although these 'major' and 'minor' conformations are sampled during turnover, studies aimed at modelling the catalytic mechanism of CYPA⁷⁻¹⁰ have focused on the substrate peptide and have not explained enzyme conformational changes. Moreover, superposition of 48 CYPA crystal structures, including complexes with peptides, inhibitors, and physiological substrates such as the HIV capsid protein^{11,12}, shows no structural heterogeneity that could rationalize the NMR results⁴ (Supplementary Fig. 1).

In search of the catalytically essential, minor conformer of free CYPA detected in solution, we first extended the resolution of the crystal structure to 1.2 Å under cryogenic conditions. This high-resolution structure, however, yielded no evidence for an alternative conformation of the dynamic network. We then used a novel algorithm, Ringer¹³, to systematically sample the electron density around each dihedral angle to discover additional unmodelled side-chain conformers. In a test set of 402 structures determined at 1.5 Å resolution or higher, >15% of the

residues with unbranched side chains have unmodelled secondary electron-density peaks in the range of $0.3-1\sigma$, below the normal noise threshold of 1σ . Strikingly, these peaks are significantly enriched at low-energy rotameric positions, supporting the interpretation that the peaks reflect true minor populations. While Ringer identified discrete side-chain heterogeneity for active-site residues Met 61 and Arg 55 in the 1.2-Å CYPA electron-density map (Supplementary Fig. 2), these alternative conformations were insufficient to explain the extension of the dynamic network away from the active site, particularly to Leu 98 and Ser 99, which showed the largest differences in NMR chemical shifts between the major and minor conformers 4.14.

To address this discrepancy between the X-ray and NMR analyses, we collected 1.39-Å diffraction data at ambient temperature to explore the possibility that the low-temperature data collection might alter the conformational distribution in the crystal 15,16 . Ringer plots indeed uncovered 0.3–1 σ features for alternative rotamers of several residues including Leu 98 and Ser 99, in addition to the side-chain heterogeneity observed at cryogenic temperature (Fig. 1a). These results emphasize that crystal freezing can alter conformational distributions.

Modelling the minor conformation of Ser 99 produced a clash of the side-chain hydroxyl group with the Phe 113 ring. Inspection of the electron density surrounding Phe 113 revealed an alternative conformation below the 1σ threshold. $F_{\rm o}-F_{\rm c}$ difference density maps calculated without bias from model rebuilding confirmed the alternative conformers of Leu 98, Ser 99 and Phe 113 (Fig. 1b, c, Supplementary Fig. 3). Ringer did not identify the Phe 113 minor rotamer owing to a concomitant shift in the backbone that places $C\gamma$ within the envelope of the major conformer. This finding emphasizes that no algorithm can yet identify with complete fidelity all structural features without further scrutiny of the electron-density distribution. Refinement verified a network of alternative side-chain rotamers covering the entire active site and extending into the buried regions of the dynamic network previously detected by NMR (Fig. 1d).

The room-temperature diffraction data afford a possible structural rationale for the large NMR chemical-shift differences between the substates in free CYPA. The alternative side-chain rotamers, particularly of Phe 113, are predicted to alter the magnetic environment of the surrounding residues. Only a coupled switch of rotamers is consistent with the steric repulsion of major and minor rotamers, such as the Ser 99/Phe 113 clash in the dynamic network (Fig. 1b).

To critically test the idea that these two conformers interconvert during turnover, we designed a mutation distant from the active site to stabilize the minor CYPA substate. Ser 99, a buried residue in the dynamic network located >14 Å from the catalytic Arg 55, was replaced by Thr to fill the space occupied by both Ser 99 rotamers. This conservative change was designed to stabilize Phe 113 in the 'out' position by emulating the steric clash between the minor Ser 99 rotamer and the Phe 113 'in' position. Crystal structures of

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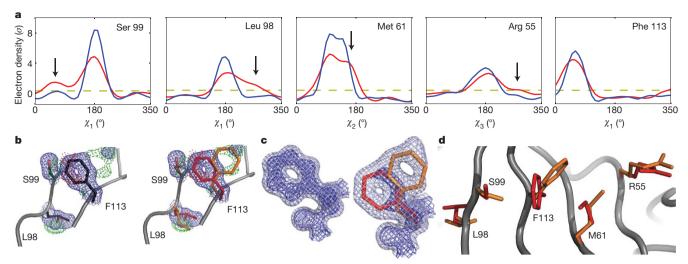


Figure 1 | Room-temperature X-ray crystallography and Ringer analysis detect conformational substates in CYPA. a, Local maxima above the 0.3σ threshold (yellow line) in Ringer plots¹³ of electron density by dihedral angle (χ_i) reveal alternative side-chain conformations in room-temperature (red line) but not cryogenic (blue line) electron density for Ser 99, Leu 98, Met 61 and Arg 55. The peak for the minor conformation of Phe 113 occurred within the envelope of the major conformer due to a shift in the protein backbone. b, Electron-density maps calculated using room-temperature X-ray data define the alternative conformers of Leu 98, Ser 99 and Phe 113. Shown are $2F_0 - F_c$ electron density (blue mesh; 1σ), and positive (green) and negative

(red) $F_{\rm o}-F_{\rm c}$ difference density (3σ) . **c**, $2F_{\rm o}-F_{\rm c}$ composite simulated-annealing omit electron density maps $(1.0\sigma,{\rm dark\,blue};0.3\sigma,{\rm light\,blue})$ show a unique conformation for Phe 113 in the 1.2-Å-resolution cryogenic structure (blue) and distinct major (red) and minor (orange) conformers in the 1.39-Å-resolution room-temperature structure. Electron density around the main chain and the surrounding residues was omitted for clarity. **d**, Steric collisions across the network of major (red) and minor (orange) conformers of Arg 55, Met 61, Phe 113 and Ser 99 explain how side-chain motions link the active site to remote buried residues.

the Ser99Thr mutant, solved at 1.6-Å and 2.3-Å resolution, indeed showed Thr 99 mimicking the alternative Ser 99 conformations, and Phe 113 was detected only in the exposed 'out' rotamer (Fig. 2a, b, and Supplementary Fig. 4). This change in rotamer populations was buttressed by three-bond *J*-coupling solution NMR experiments showing that the dominant Phe 113 χ_1 angle changed from $+60^\circ$ in wild-type CYPA to -60° in the Ser99Thr mutant (Supplementary

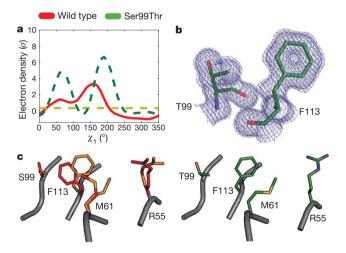


Figure 2 | The structure of the Ser99Thr mutant resembles the minor conformer of wild-type CYPA. a, χ_1 Ringer plot $(0.3\sigma$ threshold is shown as yellow line) of the Ser99Thr mutant (dashed green) and room-temperature, wild-type Ser 99 CYPA structure (red) show that Thr99 occupies both positions populated by the Ser99-OH γ group. The angular offset between the major peaks reflects a backbone shift. b, The $2F_o-F_c$ simulated-annealing omit electron density map of the Ser99Thr CYPA mutant $(1.0\sigma, \text{dark blue}; 0.3\sigma, \text{light blue})$ shows apparently unique conformations for Thr 99 and Phe 113. The structure confirmed the prediction that rotation of Phe 113 to the 'out' position is coupled to rotation of the Ser 99 hydroxyl to the minor rotamer. c, Phe 113 and Met 61 in Ser99Thr CYPA (green, right) are detected exclusively in the position of the minor state of the wild-type enzyme (orange, left).

Table 2). In the Ser99Thr variant structures, Thr 99, Phe 113 and Met 61 occupy the minor rotamers. The positions of Leu 98 and Arg 55 are consistent with either of the rotamers seen in wild-type CYPA. This pattern corroborates the conclusions that Ser 99, Phe 113, Met 61 and possibly Arg 55 are conformationally coupled (Fig. 2c) and that the Ser99Thr mutation severely reduces the population of the major conformation seen in wild-type CYPA.

Connecting this interpretation of the crystal structures to the solution behaviour of the enzyme, NMR detected amide chemical-shift differences between wild-type and Ser99Thr CYPA in most activesite and core residues of the dynamic network (Fig. 3a,b). Although it is tempting to speculate that the Ser99Thr mutation 'traps' the minor state of wild-type CYPA based on our crystallographic data (Fig. 2), NMR relaxation-dispersion analysis¹⁷ of the mutant enzyme revealed conformational exchange in both regions that showed collective motions in the wild-type enzyme (Fig. 3c). These regions include residues coupled to the active site (group I) and residues in the 65-85 loops (group II)⁴. In addition, for most of the group I residues, the peaks in Ser99Thr CYPA shift relative to wild type in the same direction as peaks in the previously characterized Lys82Ala variant, which displays a small increase in the population of the minor state⁴. The Ser99Thr mutation, however, causes much larger shifts (Fig. 3b, Supplementary Figs 5, 6), indicating that the populations are inverted. Together with our crystallographic and NMR J-coupling data, these results show that the Ser99Thr mutation shifts the structural equilibrium strongly towards a conformation that recapitulates key features of the previously undefined minor state of wild-type CYPA.

In addition to this dramatic population shift, the Ser99Thr mutation also specifically slows the motions of only group I residues by at least two orders of magnitude, pushing them into the slow NMR time regime. The slow dynamics of group I are easily identified by the characteristic increase in dispersion amplitude with increasing temperature, while the fast dynamics in the group II loop region, unaffected by the mutation, are characterized by the opposite trend (Fig. 3d, Supplementary Figs 7,8). This result is reinforced by the static magnetic-field (B_0) dependence of relaxation due to chemical exchange ($R_{\rm EX}$) as determined by the α value¹⁸: at 25 °C the $R_{\rm EX}$ of group I residues is independent of B_0 ($<\alpha>=0.16$, Supplementary

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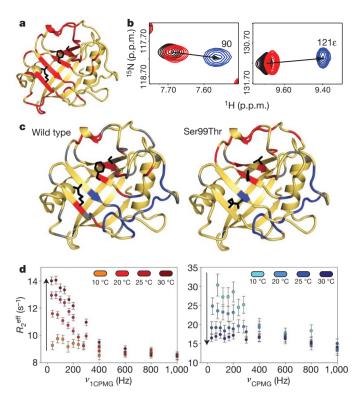


Figure 3 | The Ser99Thr mutation shifts the equilibrium towards the minor wild-type conformation and slows motions in the dynamic network in free CYPA. a, Significant ¹H-¹⁵N chemical-shift differences between Ser99Thr and wild-type CYPA (red) propagate through group I residues (Arg 55, Phe 113 and Ser 99 shown as black sticks), b, Linear amide chemical shift changes (arrows) between wild-type (black), Lys82Ala (red) and Ser99Thr (blue) CYPA reflect the inversion of the major/minor equilibrium due to the Ser99Thr mutation. c, Residues undergoing slow (red) or fast (blue) motions on the NMR timescale in Ser99Thr (right) coincide with previously identified group I (red) and group II (blue) residues in wild-type (left) CYPA (amides in grey are prolines or overlapped peaks). d, Temperature dependence of the apparent 15 N transverse relaxation rate (R_2^{eff}) at increasing refocusing field strength (v_{CPMG}) for group I (left) and group II (right) in Ser99Thr CYPA reveal that the mutation impedes group I conformational dynamics ($R_{\rm EX}\approx k_1$ and $R_{\rm EX}$ increases with temperature). In contrast, group II residues are unaffected by the mutation and display the opposite temperature dependence characteristic of fast motions on the NMR timescale. Dispersion curves were normalized to the intrinsic transverse relaxation rate (R_2^0) at 30 °C. Error bars indicate \pm the root-mean-square deviation of R_2^{eff} calculated from the variance of peak intensities (see Methods).

Table 3), characteristic of slow exchange, while for group II residues the field-dependence is quadratic ($<\alpha>=2.0$), characteristic of fast exchange. Quantitative analysis of the group I residues yields a rate constant of conversion from the major to minor state (k_1) for Ser99Thr CYPA of $1.0\pm0.3~{\rm s}^{-1}$ at $10~{\rm ^{\circ}C}$, in contrast to $\sim60~{\rm s}^{-1}$ for wild-type CYPA⁴. In the slow exchange regime of Ser99Thr, k_1 is well-determined and the major peak represents the true chemical shift of the major state (Fig. 3b).

Given the slower conformational transitions in Ser99Thr CYPA, what is the effect of this mutation on catalytic turnover? In a protease-coupled

enzymatic assay¹⁹, the Ser99Thr mutant showed a 300-fold reduction in the catalytic efficiency of the *cis*-to-*trans* isomerization of the peptide succinyl-AlaAlaProPhe-p-nitroanilide (AAPF) (Fig. 4a). This large reduction in catalytic efficiency resembles the effects of the Arg55Lys mutation, which removes the active-site residue thought to promote the chemical step of the reaction but does not perturb the enzyme dynamics⁴ or global structure (Supplementary Fig. 9). To separate the energetic contributions to the binding and isomerization steps, we measured the dissociation constants of the peptide substrate for the CYPA variants using NMR titration experiments. Peptide affinity (K_D) was weakened

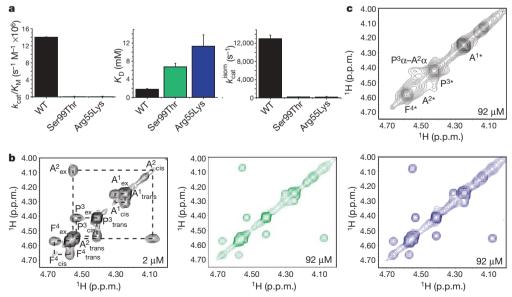


Figure 4 | Impeded motions in the dynamic network severely reduce the catalytic power of a chemically competent enzyme. a, Mutations affecting the enzyme dynamics (Ser99Thr) or the chemical step (Arg55Lys) each drastically reduce $k_{\rm cat}/K_{\rm M}$ (ref. 19) by reducing the bidirectional isomerization step on the enzyme ($k_{\rm cat}$) and not substrate affinity ($K_{\rm D}$) of CYPA. b, $^{\rm 1}{\rm H}-^{\rm 1}{\rm H}$ NOE-exchange spectra at 0.2 s mixing time showing isomerization of the peptide AAPF (1 mM) by catalytic amounts of wild-type (black), Ser99Thr (green) and Arg55Lys (blue) CYPA (concentrations

indicated in the spectra). Assignments and dashed lines connecting exchange (off-diagonal) and auto peaks are included for wild type. Much higher concentrations of the Ser99Thr and Arg55Lys variants are needed relative to wild-type CYPA to obtain similar exchange peaks, reflecting severely reduced catalytic activity. **c**, The *cis*- and *trans*- peaks coalesce (asterisk) for wild-type CYPA at the same enzyme concentration as the mutant forms due to its much greater activity. The only remaining off-diagonal peak is a $P^3 \alpha$ - $A^2 \alpha$ NOE characteristic of a *cis*-prolyl peptide bond.

only 3–6 fold by the Ser99Thr $(6.7\pm0.8\,\mathrm{mM})$ and Arg55Lys $(11.3\pm2.5\,\mathrm{mM})$ mutations (Fig. 4a, Supplementary Fig. 10), suggesting that the major effect of these mutations is on the catalytic turnover number (k_{cat}) and not on the dissociation constant of all enzyme-substrate forms (K_{M}) .

To independently quantify the mutational effects on the rate constant of the isomerization step ($k_{\rm cat}^{\rm isom}$), we measured catalysis of the *cis/trans* isomerization directly by ZZ exchange spectroscopy²⁰ (Fig. 4b). These experiments detect the overall rate of converting the entire substrate pool from the *cis-* to the *trans-*Pro conformation and vice versa. The Ser99Thr and Arg55Lys mutations reduce overall turnover by similar amounts compared to wild-type CYPA (Fig. 4, Supplementary Fig. 11). Quantitative fitting of these exchange data²⁰ indicates that the Ser99Thr mutation severely reduces the rate of the bidirectional isomerization step on the enzyme (190 \pm 20 s $^{-1}$ for Ser99Thr compared to 13,000 \pm 800 s $^{-1}$ for wild-type CYPA). Strikingly, this reduction (\sim 70-fold) parallels the upper bound for the reduction in the rate constant for the major-to-minor conformational change for group I residues (\sim 60-fold) in the free enzyme (Figs 3c, d).

The similar rate reductions in the free enzyme and of substrate turnover strongly suggest that the slowed fluctuations due to the Ser99Thr mutation underlie the reduction of the catalytic rate. The effects of this 'molecular traffic-jam' mutation remote from the active site support the idea that the coupled motions of the dynamic network tied to the rotation of Phe 113 play a key role in turnover. The conformational transition of Phe 113 observed here suggests that the side-chain rotation directly and distinctly impacts cis- and transforms of the proline-substrate during catalysis. Indeed, smaller²¹ or larger⁴ replacements for Phe 113 substantially decrease CYPA activity. suggesting the wild-type enzyme occupies a narrow structural optimum for catalysis. Both the interaction of the Arg 55 side chain with the substrate8,9, which facilitates the chemistry, and sidechain motions throughout the dynamic network, which enable facile interconversion of conformational states, are necessary for catalysis (Fig. 4). Therefore, neither the dynamics nor active-site chemistry that lowers the transition-state energy are sufficient to promote efficient

Only by collecting X-ray diffraction data at ambient temperature did we observe agreement between the conformational substates detected using NMR and X-ray crystallography. This finding underscores previous studies demonstrating that the conformational distributions in crystals of myoglobin 22,23 and RNase 16,24 are restricted at cryogenic temperatures. Unlike these classic studies, which accessed the manifold of crystal structures indirectly through reductions in protein volume, inhibition of ligand binding and reductions in crystallographic *B*-values at cryogenic temperatures, here we use Ringer 13 to directly sample the electron density. This automated, systematic approach reveals alternative conformational substates. Our results emphasize that the common practices of crystal freezing and building models at the 1σ threshold, rather than capturing the conformational distribution in the crystal, can eliminate information about functionally critical conformations.

Our complementary results from NMR and X-ray crystallography for CYPA demonstrate a general strategy to discover structural ensembles of interconverting substates, to identify which regions of enzymes couple to active sites, and to evaluate the roles of hidden, higher energy conformations in catalysis by other enzymes^{3,25}. Particularly by identifying structurally remote residues that couple to the active site, defining the multiple conformations that contribute to function can afford a deeper understanding of the effects of sequence variation in disease and evolution²⁶. Such knowledge may enable progress not only in understanding and manipulating the mechanisms of numerous macromolecular systems, but also in defining the manifold of conformations accessible to inhibitors and therapeutics²⁷.

METHODS SUMMARY

CYPA was purified as described⁴. Crystals were grown at pH 7.5 using PEG3350 or DL-malate as precipitants. All X-ray data sets derive from single crystals. For room-temperature data collection, we used the Rigaku free mounting device at ALS Beamline 12.3.1²⁸. Short exposures and crystal translation every 10 frames limited radiation damage. Ringer analysis was performed by real-space sampling of electron density around side-chain dihedral angles¹³. Inspection of plots of electron density versus dihedral angle revealed Ringer peaks above 0.3σ that guided placement of alternative conformers. This modelling process was performed iteratively using both conventional and simulated-annealing composite omit maps.

Constant-time ¹⁵N backbone CPMG TROSY relaxation-dispersion data¹⁷ were collected in an interleaved manner on Varian INOVA 500 and 600 spectrometers and a Bruker AVANCE-800 equipped with a cryoprobe. Dispersion curves were fitted to the full Carver-Richards equation²⁹, confirming the results for wild-type CYPA where the exchange is in the intermediate to fast time regime⁴. In contrast, for Ser99Thr CYPA, global fits of dispersion of group-I residues at all temperatures, assuming temperature-independent $\Delta\omega$, unambiguously revealed exchange in the slow regime ($R_{\rm EX} \approx k_1$, rate constant for efflux from the major state). Numerical fitting and estimation of the dispersion amplitude gave the same rate constant k_1 at 10 °C within experimental error.

Three-bond, aromatic J couplings were obtained at 25 °C as described³⁰. Dissociation constants for AAPF were obtained by titrating the peptide into a solution of 0.2 mM CYPA at 6 °C and fitting the resulting chemical shifts to single-site exchange. The enzyme-catalysed *cis/trans* isomerization rates at 6 °C were measured using $^{1}\text{H}-^{1}\text{H}$ NOESY spectra of AAPF and fitting the data to equations for ZZ exchange²⁰. The bidirectional k_{cat} isom was calculated from the fitted exchange rate using the K_{D} measured in titrations (see Online Methods).

Full Methods and any associated references are available in the online version of the paper at www.nature.com/nature.

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- Henzler-Wildman, K. & Kern, D. Dynamic personalities of proteins. Nature 450, 964–972 (2007).
- Mittermaier, A. & Kay, L. E. Review—new tools provide new insights in NMR studies of protein dynamics. Science 312, 224–228 (2006).
- Boehr, D. D., McElheny, D., Dyson, H. J. & Wright, P. E. The dynamic energy landscape of dihydrofolate reductase catalysis. Science 313, 1638–1642 (2006).
- Eisenmesser, E. Z. et al. Intrinsic dynamics of an enzyme underlies catalysis. Nature 438, 117–121 (2005).
- Hammes-Schiffer, S. & Benkovic, S. J. Relating protein motion to catalysis. *Annu. Rev. Biochem.* 75, 519–541 (2006).
- Schramm, V. L. & Shi, W. Atomic motion in enzymatic reaction coordinates. Curr. Opin. Struct. Biol. 11, 657–665 (2001).
- Agarwal, P. K. Cis/trans isomerization in HIV-1 capsid protein catalyzed by cyclophilin A: insights from computational and theoretical studies. Proteins 56, 449–463 (2004).
- Hamelberg, D. & McCammon, A. Mechanistic insight into the role of transitionstate stabilization in cyclophilin A. J. Am. Chem. Soc. 131, 147–152 (2009).
- Li, G. H. & Cui, Q. What is so special about Arg 55 in the catalysis of cyclophilin A? Insights from hybrid QM/MM simulations. J. Am. Chem. Soc. 125, 15028–15038 (2003)
- Trzesniak, D. & Van Gunsteren, W. F. Catalytic mechanism of cyclophilin as observed in molecular dynamics simulations: pathway prediction and reconciliation of X-ray crystallographic and NMR solution data. *Protein Sci.* 15, 2544–2551 (2006).
- Howard, B. R., Vajdos, F. F., Li, S., Sundquist, W. I. & Hill, C. P. Structural insights into the catalytic mechanism of cyclophilin A. *Nature Struct. Biol.* 10, 475–481 (2003).
- Ke, H. M. & Huai, Q. Crystal structures of cyclophilin and its partners. Front. Biosci. 9, 2285–2296 (2004).
- 13. Lang, P. T. et al. Automated electron-density sampling reveals widespread conformational polymorphism in proteins. *Protein Sci.* (submitted).
- Eisenmesser, E. Z., Bosco, D. A., Akke, M. & Kern, D. Enzyme dynamics during catalysis. Science 295, 1520–1523 (2002).
- Halle, B. Biomolecular cryocrystallography: structural changes during flashcooling. Proc. Natl Acad. Sci. USA 101, 4793–4798 (2004).
- Rasmussen, B. F., Stock, A. M., Ringe, D. & Petsko, G. A. Crystalline ribonuclease A loses function below the dynamical transition at 220 K. *Nature* 357, 423–424 (1992).
- 17. Loria, J. P., Rance, M. & Palmer, A. G. A TROSY CPMG sequence for characterizing chemical exchange in large proteins. *J. Biomol. NMR* 15, 151–155 (1999).
- Millet, O., Loria, J. P., Kroenke, C. D., Pons, M. & Palmer, A. G. The static magnetic field dependence of chemical exchange linebroadening defines the NMR chemical shift time scale. J. Am. Chem. Soc. 122, 2867–2877 (2000).

LETTERS

- Kofron, J. L., Kuzmic, P., Kishore, V., Colonbonilla, E. & Rich, D. H. Determination of kinetic constants for peptidyl prolyl cis-trans isomerases by an improved spectrophotometric assay. *Biochemistry* 30, 6127–6134 (1991).
- Farrow, N. A., Zhang, O. W., Forman-Kay, J. D. & Kay, L. E. A Heteronuclear correlation experiment for simultaneous determination of ¹⁵N longitudinal decay and chemical exchange rates of systems in slow equilibrium. *J. Biomol. NMR* 4, 727–734 (1994).
- Zydowsky, L. D. et al. Active site mutants of human cyclophilin A separate peptidyl-prolyl isomerase activity from cyclosporine A binding and calcineurin inhibition. Protein Sci. 1, 1092–1099 (1992).
- 22. Frauenfelder, H. et al. Thermal expansion of a protein. *Biochemistry* **26**, 254–261 (1987)
- Frauenfelder, H., Petsko, G. A. & Tsernoglou, D. Temperature-dependent X-ray diffraction as a probe of protein structural dynamics. *Nature* 280, 558–563 (1970)
- Tilton, R. F., Dewan, J. C. & Petsko, G. A. Effects of temperature on protein structure and dynamics: X-ray crystallographic studies of the protein ribonuclease-A at nine different temperatures from 98 to 320K. *Biochemistry* 31, 2469–2481 (1992).
- Beach, H., Cole, R., Gill, M. L. & Loria, J. P. Conservation of μs-ms enzyme motions in the apo- and substrate-mimicked state. J. Am. Chem. Soc. 127, 9167–9176 (2005).
- Tokuriki, N. & Tawfik, D. S. Protein dynamism and evolvability. Science 324, 203–207 (2009).
- 27. Lee, G. M. & Craik, C. S. Trapping moving targets with small molecules. *Science* 324, 213–215 (2009).
- Kiefersauer, R. et al. A novel free-mounting system for protein crystals: transformation and improvement of diffraction power by accurately controlled humidity changes. J. Appl. Crystallogr. 33, 1223–1230 (2000).

- Davis, D. G., Perlman, M. E. & London, R. E. Direct measurements of the dissociation-rate constant for inhibitor-enzyme complexes via the T_{1p} and T₂(CPMG) methods. J. Maan. Reson. Ser. B 104, 266–275 (1994).
- Hu, J. S., Grzesiek, S. & Bax, A. Two-dimensional NMR methods for determining χ₁ angles of aromatic residues in proteins from three-bond J_{C'Cγ} and J_{NCγ} couplings.
 J. Am. Chem. Soc. 119, 1803–1804 (1997).

Supplementary Information is linked to the online version of the paper at www.nature.com/nature.

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Author Contributions J.S.F., S.C.D. and R.E. performed the X-ray experiments, M.W.C. performed the NMR experiments, and M.W.C. and J.S.F. performed the activity and binding assays. J.S.F., M.W.C., D.K. and T.A. analysed data and wrote the paper. All authors contributed to data interpretation and commented on the manuscript.

Author Information Atomic coordinates and structure factors for the reported crystal structures have been deposited in the PDB under accession codes 3KOM, 3KON, 3KOP, 3KOQ and 3KOR. Reprints and permissions information is available at www.nature.com/reprints. Correspondence and requests for materials should be addressed to T.A. (tom@ucxray.berkeley.edu) or D.K. (dkern@brandeis.edu).

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METHODS

Crystallography. The standard purification⁴ was augmented with a Superdex S75 column (GE Healthcare) equilibrated in 20 mM HEPES pH 7.5, 100 mM NaCl and 0.5 mM TCEP. Crystals (wild-type and Ser99Thr) with $P2_12_12_1$ symmetry were grown by hanging-drop vapour diffusion by mixing an equal volume of well solution (100 mM HEPES pH 7.5, 22% PEG 3350, 5 mM TCEP) and protein (wild-type CYPA at 60 mg ml⁻¹ and mutants at 40 mg ml⁻¹). Trigonal ($P3_221$) crystals of Ser99Thr and Arg55Lys CYPA were grown similarly from 1.8 M DL-malic acid. For cryogenic data sets (collected at 100 K), crystals were flash frozen in liquid nitrogen with well solution plus 15% xylitol as a cryoprotectant for the orthorhombic crystals and 10% glycerol for the trigonal crystals. Wild-type CYPA cryogenic X-ray data were collected at Beamline 9-1 at the Stanford Synchrotron Radiation Laboratory. Ser99Thr and Arg55Lys cryogenic data sets were collected at Beamline 8.3.1 at the Lawrence Berkeley National Laboratory Advanced Light Source (ALS).

Room-temperature X-ray data were collected at 15 °C with 96% humidity using a temperature- and humidity-controlled goniometer head² (the Rigaku free-mounting device) at ALS Beamline 12.3.1. To limit the effects of radiation damage while maintaining high signal-to-noise, we collected short exposures (0.1 s) for 180° with 1° oscillations. A large crystal (1 mm × 0.5 mm × 0.3 mm) was translated halfway through this rapid collection protocol. Following this initial low-exposure pass, an additional complete data set of 90° with longer exposures (1 s) was collected while translating the crystal every 10 frames. During data processing, we ensured that radiation damage had not degraded the data quality by subdividing the data into subsets of 45 frames and confirming that unit cell parameters, scale factors and χ^2 statistics were consistent throughout the data set 31 . Owing to the short total exposure time (108 s), the limited exposure of each spot on the crystal, the size of the crystal and the finite rate of crystal damage, the data were not significantly influenced by radiation damage.

Reflections were processed using HKL2000³². Structures were solved using PHASER³³ through phenix.automr³⁴ with 2CPL chain A as the search model. To corroborate the features of residues 98, 99 and 113 observed in the initial electron density, reflections were processed using MOSFLM, through ELVES³⁵, and XDS, through XIA2³⁶, and the structures were additionally solved using 1RMH chain A as the search model. The features described in the text were consistent across electron-density maps calculated from data processed with all three programs and both molecular-replacement search models.

Coordinates, atomic displacement parameters, and occupancy (where appropriate) were refined using phenix.refine³⁴ with manual rebuilding using COOT³⁷. Validation was performed using MOLPROBITY³⁸ and PROCHECK³⁹. Data collection, processing, refinement, and validation statistics are listed in Supplementary Table 1. Structural figures were prepared using PyMol⁴⁰.

Ringer. Ringer¹³ analyses electron density in real space around the dihedral angles of the modelled side chains to discover unmodelled alternative conformers. For each residue, each side-chain dihedral angle is rotated in 10° increments, and the electron density value calculated using a cubic spline interpolation is tabulated at the position of the rotated atom. In sp^3 geometry, for example, the sampled position is 1.53 Å away and oriented 111.1° from the preceding side-chain carbon atom. In regions of electron density that are not occupied by modelled atoms, Ringer peaks $> 0.3\sigma$ are significantly enriched at rotameric positions, and χ_1 peaks are strongly correlated with rotameric χ_2 peaks in long side chains, providing evidence that these features correspond to alternative conformers¹³. The contour level of 0.3σ provides an initial working cut-off to detect alternative rotamers in high-resolution electron density maps. Alternative conformers are modelled after inspection of the electron density maps for continuous or correlated electron density for the side chain and subsequently refined.

Comparison of crystal structures. Forty-eight CYPA structures with 100% sequence identity (Supplementary Fig. 1) were superimposed using Theseus⁴¹ and analysed for backbone root-mean-square-deviation (r.m.s.d.) to establish regions of conformational heterogeneity or flexibility.

NMR methods. Unless otherwise specified, NMR samples consisted of 1 mM wild-type or mutant CYPA in 50 mM Na₂HPO₄ pH 6.5, 0.02% NaN₃, 1 mM DTT, 10% D₂O. Constant-time relaxation dispersion spectra^{17,42} were collected in an interleaved manner, processed using NMRPipe⁴³ software, and analysed using NMRView⁴⁴ and custom scripts. Errors in peak intensities were estimated using the variance for non-exchanging peaks, signal to noise, and duplicate points. Errors for the fits were estimated using 200 Monte Carlo simulations. CPMG relaxation dispersion data collected at 25 °C at 500, 600 and 800 MHz not only constrained the fitted parameters but also provided α values for all residues¹⁸.

We note that in the slow exchange regime of Ser99Thr, CPMG relaxation data and fitting them to the Carver–Richards equation is not robust to determine chemical shift differences or populations. The absence of the minor peak, in light of the estimated line-broadening, suggests an upper population limit for this species of about 10%. However, k_1 is well determined in the slow time regime by the value of $R_{\rm ex}$.

Chemical shift differences between mutant and wild-type backbone amides or side-chain indoles (Trp 121) were considered significant when $\Delta\delta$ was greater than 0.05 p.p.m. in the $^1{\rm H}$ dimension or 0.25 p.p.m. in the $^{15}{\rm N}$ dimension. Three-bond J couplings were assessed using difference spectra 30 .

NOESY experiments were performed at 6 °C in a buffer containing 860 μ M AAPF and 100% D₂O using a standard gradient-filtered experiment⁴⁵. Intensity curves were fitted to previously described equations²⁰ in Origin 6.0 using populations for the *cis* and *trans* isomer determined from the intensities of the peaks in a NOESY spectrum with a mixing time of zero. The fits of the data for all three enzyme forms yielded the corresponding exchange rates ($k_{\rm ex}$), but the same R_1 values for the auto-peaks within experimental error. The bidirectional $k_{\rm cat}$ was calculated from this $k_{\rm ex}$ using the following equations:

$$k_{\rm ex} = \frac{v^{\rm isom}}{[S]}$$

$$v^{\text{isom}} = \frac{k_{\text{cat}}^{\text{isom}}[E][S]}{K_{\text{M}} + [S]}$$

where v^{isom} is the reaction velocity, K_{M} is the Michaelis-Menten constant, and [S] and [E] are substrate and enzyme concentrations, respectively. For fast dissociation relative to the rate of the isomerization, $K_{\mathrm{M}} \approx K_{\mathrm{D}}$, so:

$$k_{\text{cat}}^{\text{isom}} = \frac{k_{\text{ex}}(K_{\text{D}} + [S])}{[E]}$$

This yields a bidirectional $k_{\rm cat}^{\rm isom}$ when one uses the $K_{\rm D}$ averaged for the cis and trans isomers obtained from the titration experiments.

Coupled enzymatic assay. $k_{\rm cat}/K_{\rm M}$ for the enzyme catalysed *cis*-to-*trans* isomerization of AAPF was measured at 10 °C using the standard protease coupled assay¹⁹.

- Southworth-Davies, R. J., Medina, M. A., Carmichael, I. & Garman, E. F. Observation of decreased radiation damage at higher dose rates in room temperature protein crystallography. Structure 15, 1531–1541 (2007).
- 32. Otwinowski, Z. & Minor, W. in *Macromolecular Crystallography* Part A, 307–326 (Methods in Enzymology, Vol. 276, Academic, 1997).
- McCoy, A. J. et al. Phaser crystallographic software. J. Appl. Crystallogr. 40, 658–674 (2007)
- 34. Adams, P. D. et al. PHENIX: building new software for automated crystallographic structure determination. Acta Crystallogr. D 58, 1948–1954 (2002).
- Holton, J. & Alber, T. Automated protein crystal structure determination using ELVES. Proc. Natl Acad. Sci. USA 101, 1537–1542 (2004).
- 36. Kabsch, W. Automatic processing of rotation diffraction data from crystals of initially unknown symmetry and cell constants. *J. Appl. Crystallogr.* **26**, 795–800
- 37. Emsley, P. & Cowtan, K. Coot: model-building tools for molecular graphics. *Acta Crystallogr. D* **60**, 2126–2132 (2004).
- Davis, I. W. et al. MolProbity: all-atom contacts and structure validation for proteins and nucleic acids. Nucleic Acids Res. 35, W375–W383 (2007).
- Laskowski, R. A., Macarthur, M. W., Moss, D. S. & Thornton, J. M. Procheck—a program to check the stereochemical quality of protein structures. *J. Appl. Crystallogr.* 26, 283–291 (1993).
- 40. Delano, W. L. The PyMOL Molecular Graphics System (DeLano Scientific, Palo Alto, 2008); (http://www.pymol.org).
- 41. Theobald, D. L. & Wuttke, D. S. THESEUS: maximum likelihood superpositioning and analysis of macromolecular structures. *Bioinformatics* 22, 2171–2172 (2006).
- 42. Mulder, F. A. A., Mittermaier, A., Hon, B., Dahlquist, F. W. & Kay, L. E. Studying excited states of proteins by NMR spectroscopy. *Nature Struct. Biol.* **8**, 932–935 (2001).
- 43. Delaglio, F. et al. NMRPipe—a multidimensional spectral processing system based on UNIX pipes. J. Biomol. NMR 6, 277–293 (1995).
- 44. Johnson, B. A. & Blevins, R. A. NMR View—a computer program for the visualization and analysis of NMR data. *J. Biomol. NMR* 4, 603–614 (1994).
- 45. Jeener, J., Meier, B. H., Bachmann, P. & Ernst, R. R. Investigation of exchange processes by 2-dimensional NMR-spectroscopy. *J. Chem. Phys.* **71**, 4546–4553