

**PYL100: Electromagnetic Waves and Quantum Mechanics (II Semester, 2016-17)**

**Exercise Sheet No. 6**

Part 2: QM basics -- Wave function, Probability, Schrödinger Eq., Uncertainty principle

1. The wave function of a particle at  $t=0$  is given by:  $\psi(x) = Ce^{-\alpha^2 x^2}$  ( $-\infty < x < \infty$ ) where  $C$  and  $\alpha$  are constants. Calculate the probability of finding the particle in the region ( $0 < x < \infty$ ).

[Ans. 1/2]

2. The wave function of the particle at a certain instant is given as

$$\psi(x) = A \exp\left(-\frac{x^2}{a^2} + ikx\right)$$

If  $P_1$  and  $P_2$  denote the probability densities of finding the particle in the range  $a$  to  $a+da$  and  $2a$  to  $2a+da$  respectively. Find the ratio  $P_1 : P_2$ .

[Ans.  $e^6$ ]

3. Check whether the following functions represents acceptable wave function of a particle in the range ( $-\infty < x < \infty$ ) or not?

- (a)  $\Psi(x) = 3 \sin \pi x$       (b)  $\Psi(x) = 4 - |x|$   
(c)  $\Psi(x) = x^2$               (d)  $\Psi(x) = 7 \tan x$

[Ans. Only (a)]

4. An one-dimensional wave function of a particle is given as

$$\psi(x) = Ae^{-x^2/a^2 + ikx} \quad (A \text{ is the normalization constant})$$

Calculate the expectation value of position and momentum of the particle.

[Ans. (a) 0, (b)  $\hbar k$ ]

5. Consider a system whose state is given in terms of an orthonormal set of three wave functions:  $\Phi_1(x)$ ,  $\Phi_2(x)$ ,  $\Phi_3(x)$  as:

$$\Psi(x) = \frac{\sqrt{3}}{3} \Phi_1 + \frac{2}{3} \Phi_2 + \frac{\sqrt{2}}{3} \Phi_3.$$

- (a) Verify that  $\Psi$  is normalized.  
(b) Calculate the probability of finding the system in any one of the states  $\Phi_1$ ,  $\Phi_2$ ,  $\Phi_3$ . Verify that the total probability is equal to one.  
(c) Consider now an ensemble of 810 identical systems, each one of them in the state  $\Psi$ . If measurements are done on all of them, how many systems will be found in each of the states  $\Phi_1$ ,  $\Phi_2$ ,  $\Phi_3$ ?

[Ans. (b)  $P_1 = \frac{1}{3}$ ,  $P_2 = \frac{4}{9}$ ,  $P_3 = \frac{2}{9}$ ; (c)  $N_1 = 270$ ,  $N_2 = 360$ ,  $N_3 = 180$ ]

6. The wave function corresponding to a particle is given by:

$$\psi(x) = \frac{1}{\sqrt{a}} \exp\left(-\frac{|x|}{a}\right)$$

(a) Calculate the probability of finding the particle in the region  $-a < x < a$ .

(b) Find the value of 'b' such that probability of finding the particle between  $-b < x < b$  is 0.5.

**[Ans. (a)  $1 - \frac{1}{e^2}$ , (b)  $\frac{a \ln 2}{2}$ ]**

7. In a region of space, a particle with mass m and with zero energy has a time-independent wave function given by

$$\psi(x) = A e^{-x^2/L^2}$$

where A and L are constants. Determine the potential energy U(x) of the particle.

**Ans:**  $U(x) = \frac{2\hbar^2}{mL^4} \left(x^2 - \frac{3L^2}{2}\right)$

8. An electron is described by the wave function

$$\psi(x) = \begin{cases} C e^{-x}(1 - e^{-x}); & x > 0 \\ 0; & x < 0 \end{cases}$$

where x is in nm and C is a constant.

(a) Determine the value of C that normalizes  $\psi(x)$ . **[Ans:  $2\sqrt{3} \text{ nm}^{-1/2}$ ]**

(b) Where is the electron most likely to be found? That is, for what value of x is the probability of finding the electron the largest? **[Ans: 0.693 nm]**

(c) Calculate the average position  $\langle x \rangle$  for the electron. **[Ans: 1.083 nm]**