CML100 General Chemistry

Department of Chemistry, IIT Delhi, New Delhi 110 016

Exam: Major

Time: 8.00 to 10.00 a.m. (2 hours)

Day and Date: Sunday, November 22, 2015

Max. Marks: Forty six (46)

Name:	Entry No.:
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Question 1(i): Draw Fischer projection formulas for all stereoisomers of 2,4-dimethyl-3-

hexanol.

Question 1 (ii): Draw the enantiomer of the following chiral molecule.

1 H

Question 1(iii): Explain bonding, orbital structure and hybridization in allene, CH₂=C=CH₂.

2

4

Question 2 (i): (+) tartaric acid has a specific rotation of +12.0. Calculate the specific rotation of a mixture of 68% (+)-tartaric acid and 32% (-)-tartaric acid.

Question 2 (ii): Use Cram's rule to predict the configuration of the major product upon nucleophilic attack on the carbonyl group.

Question 3(i): What product will be obtained from the following enantioselective reactions.

3

Do not write anything on this question paper except your name and entry number

Question 3 (ii): Write the product and energy diagram of the following transformation.

$$\begin{array}{c|c} & H_2SO_4 \\ \hline & 180^{\circ}C \end{array} \qquad \begin{array}{c} H_2SO_4 \\ \hline & 80^{\circ}C \end{array}$$

Question 4 (i): Solvolysis of following compounds in methanol gives a complex product mixture of the following five compounds. Propose mechanism to account for these compounds. 4(2.5+1.5)

(a)
$$CH_3OH$$
 + CH_3OH + CH_3OH + CH_3OH

(b)
$$CH_3OH$$
 OCH_3 + OCH_3 + OCH_3

Question 4 (ii): For the following conversion, provide mechanism for the every product.