Investigation of the Higher Order Zeeman Effect

By

Evan M. R. Petrimoulx

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by Evan M. R. Petrimoulx APPROVED BY: Initial. Last Name Department of Mechanical, Automotive and Materials Engineering Initial. Last Name School of Computer Science

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I would like to dedicate this thesis to ...

AKNOWLEDGEMENTS

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CHAPTER 1

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CHAPTER 2

Theoretical Methods

2.1 Overview

This chapter will provide an overview of all theoretical methods used throughout my dissertation. It will provide the theoretical building blocks upon which the dissertation is constructed, ranging from the set of units used in the problem, to perturbation equation solutions and methods for verifying calculations¹. As stated in the introduction, this thesis is based upon calculating matrix elements between two sets of wavefunctions connected by an operator. These matrix elements arise from perturbation theory and are necessary for computing energy shifts due to external fields such as those present in the higher-order Zeeman effect. The foundation of theory will be presented here, but the more specific calculations related to solving for the Zeeman effect will be presented in the relevant chapter below.

I will begin by first discussing atomic units in Sec. 2.2, then move on and introduce the Schrödinger equation for the one-electron problem as well as its solution for hydrogenic wavefunctions in Sec. 2.3. In this section I will also discuss the significance and impact of the canonical momentum, and emphasize its importance in solving for the Zeeman effect. I will then introduce Perturbation Theory in Sec. 2.5, and describe how it can be used to approximate the new eigenvalues and eigenstates of the Zeeman-perturbed Hamiltonian. Sec. 2.4 will discuss various integration techniques that will be implemented both analytically and programmatically in order to simplify the calculation process, followed by an introduction to solving recursion re-

¹Majority of the material discussed throughout this chapter are inspired and shaped by G. W. F. Drake. References will be included where needed.

lations in Sec. 2.6. This is the core step in determining the perturbed wavefunctions used throughout the rest of the study. Finally, this chapter will conclude with the discussion of the Dalgarno Interchange Theorem and how it can serve as a critical check on the correctness of the obtained perturbed wavefunctions.

2.2 Atomic Units

First proposed by Hartree in November of 1928, atomic units have since become the standard across all of atomic physics. Atomic units were designed with the purpose of "eliminating various universal constants from equations and also to avoid high powers of 10 in numerical work"??. Hartree proposed that we measure the mass in terms of the mass of the electron m_e , charge in terms of the electron charge, e, the vacuum permittivity in terms of $4\pi\epsilon_0$, and the angular momentum in terms of \hbar . Written mathematically we can say

$$\hbar = m_e = e = 4\pi\epsilon_0 = 1 \tag{1}$$

Scaling these units out of the problem greatly simplifies our calculation, and once a result is found, one can multiply the unscaled units back into the answer to retrieve the desired result in standard SI units. While the use of atomic units was proposed long before the field of computational physics emerged, it is of key importance that one implements this technique for any atomic physics simulation. Machine error is an unavoidable reality of programmed computation and working in units which avoids extreme values both large and small can help mitigate the uncertainty present in high precision calculations.

A consequence of utilizing atomic units is the need to redefine key fundamental physical quantities in terms of these scaled values. An example of key importance in the investigation of the higher-order Zeeman effect is the atomic unit of magnetic field strength, otherwise referred to as the atomic unit of magnetic flux density ??.

$$B_{\text{a.u.}} = \frac{\hbar}{ea_0^2} \tag{2}$$

This value is composed of the Bohr radius a_0 , electron charge e, and \hbar , the unit of action. The Bohr radius is defined as the most probable distance between an electron in the ground state of hydrogen and its nucleus. It can be described by the following equation ??

$$a_0 = \frac{4\pi\epsilon_0 \hbar^2}{e^2 m_e} = \frac{\hbar}{m_e c\alpha} \tag{3}$$

The Bohr radius itself is defined using a similar combination of fundamental constants which after applying the rules for atomic units, should equal 1. We can see now that all units in the atomic unit of field strength also become 1 as intended. At the end of our problem, we can simply multiply the determined result by the atomic unit of magnetic field strength when its constants are not scaled to 1, and retrieve our desired result in SI units. This is equivalent to multiplying the result in atomic units by ??.

$$B_{\text{a.u.}} = 2.35051757077(73) \times 10^5 \text{ T}$$
 (4)

The numbers in parenthesis here denote the uncertainty of the quantity. Another key benefit in atomic units is that it allows us to define almost all fundamental quantities in terms of two constants; the speed of light, c and the fine structure constant, α .²

The fine structure constant can be written as the following dimensionless quantity

²The fine structure constant was introduced in 1916 by Arnold Sommerfeld when he was investigating the gap in the spectral lines of the hydrogen atom. He compared his work with the Michelson Morley experiment from 1887 where the spectral lines were shown and it was deduced that the universe is not permeated by the aether ??

$$\alpha = \frac{e^2}{4\pi\epsilon_0 \hbar c} = 7.2973525643(11) \times 10^{-3} \tag{5}$$

which in atomic units is just the following relationship with the speed of light

$$\alpha = \frac{1}{c} \tag{6}$$

The fine structure constant is often used as an essential expansion parameter, treating relativity as a perturbation in α .³ Applying the atomic unit scaling to the fine structure constant shown in equation (6) confirms that the second equivalence in equation (3) remains equal to 1.

Applying this technique to problems in atomic physics gives a powerful way of simplifying calculations while providing an easy conversion factor to SI units for comparison of theory and experiment. This dissertation will discuss all equations and quantities in atomic units unless further specified otherwise.

2.3 One-electron Schrödinger equation

Discovered by Schrödinger in 1926, the Schrödinger equation is a non-relativistic solution to the two body problem involving an orbiting electron and its nucleus⁴. The electron interacts with the nucleus via the Coulomb interaction described by the following potential

$$V(\vec{r}) = -\frac{Ze^2}{4\pi\epsilon_0 \vec{r}} \tag{7}$$

We can write the energy of the system using the Hamiltonian, where our potential is the Coulomb potential. We then come up with

 $^{^3}$ More on this will be discussed in Sec. 3.6.3

⁴It is assumed here that the nucleus is a point particle with infinite mass.

$$H = \frac{p^2}{2m} - \frac{Ze^2}{4\pi\epsilon_0 \vec{r}} \tag{8}$$

However, quantum mechanics dictates that all observable quantities must be determined by hermitian operators that act on the wavefunction ψ . The wavefunction ψ is described as a normalized vector in Hilbert space, \mathcal{H} containing all information pertaining to the system. This means that the momentum needs to be a hermitian operator that acts on the wavefunction, and that our Hamiltonian is also a hermitian operator that satisfies the following eigenvalue problem

$$\hat{H}|\psi\rangle = E_n|\psi\rangle \tag{9}$$

This is what is known as the Schrödinger equation. It can be applied to our twobody problem by substituting the momentum for its quantum operator analog. This results in the following expression

$$\hat{H}\psi = \frac{-\hbar^2}{2m_e} \nabla^2 \psi - \frac{Ze^2}{4\pi\epsilon_0 \vec{r}} \psi = E_n \psi \tag{10}$$

We now have a second-order partial differential equation. Assuming a separable solution allows one to split the solution to any hydrogenic system into two parts; The radial part and the angular part.

$$\psi_{nlm} = R_{nl}(r)\mathcal{Y}_m^l(\cos\theta) \tag{11}$$

For our two body atomic system the solutions for each can be written in terms of Laguerre polynomials and Spherical Harmonics⁵. The radial equation has the

 $^{^{5}}$ The derivation will be shown in the appendix for both the angular and radial parts of the wavefunction.

following solutions??

$$R_{nl}(r) = \frac{2Z}{n^2} \sqrt{\frac{Z(n-l-1)!}{(n+l)!}} \left(\frac{2Zr}{n}\right)^l e^{\frac{-Zr}{n}} L_{n-l-1}^{(2l+1)} \left(2Zr/n\right)$$
(12)

Where Z is the nuclear charge atomic number, l is the angular momentum quantum number, n is the principle quantum number, and $L_{n-l-1}^{(2l+1)}(2Zr/n)$ is the generalized Laguerre polynomial (which I have defined in Appendix D.3). Additionally, the spherical harmonics can be written as

$$\mathcal{Y}_{l}^{m}(\theta,\phi) = (-1)^{m} \sqrt{\frac{(2l+1)}{4\pi} \frac{(l-m)!}{(l+m)!}} P_{l}^{m}(\cos\theta) e^{im\phi}$$
(13)

Where l is the angular momentum quantum number, m is the magnetic quantum number, and $P_l^m(\cos\theta)$ is the associated Legendre polynomial (which I will discuss further in Appendix D.4). A key feature of the spherical harmonics which I will utilize in Sec. 2.4 is that the spherical harmonics form a normalized complete set of orthogonal basis functions. The ability to separate the solution into two independent parts plays a crucial role in simplifying the task at hand and will be exploited when evaluating integrals involving these wavefunctions in Chapter 3.

2.4 Integration Techniques

In this section of the dissertation, I will discuss the standard approach to solving the integrals pertaining to the various matrix elements required throughout. A matrix element is an operator - usually a perturbed Hamiltonian, that is used to approximate the behaviour of quantum systems under small disturbances. Calculating thier expectation values gives insight to the systems behaviour, revealing properties such as transition rates or behaviour in an external field ??. The calculation of these matrix elements requires integration over the states as well as the operator acting on a state,

which can involve radial and angular pieces. As discussed in the previous section, the ability to separate the solution into a radial part as well as an angular part significantly simplifies the solution process. Instead of having to integrate a single function over all three dimensions in spherical coordinates, we can split the solution into two steps which can be calculated independently.

$$\langle \psi^n | V | \psi^m \rangle = \int_0^\infty r^2 \ \psi^n(r) V(r) \psi^m(r) \ dr \int_\Omega \sin\theta \ \psi^n(\theta, \phi) V(\theta, \phi) \psi^m(\theta, \phi) \ d\Omega \quad (14)$$

Where Ω represents the solid angles for θ and ϕ . The total problem can then be solved with the simple combination of calculated integrals.

$$\langle \psi^n | V | \psi^m \rangle = I_{\text{radial}} \cdot I_{\text{angular}}$$
 (15)

The section serves as a general introduction to the process of calculating matrix elements, but Sec. 2.4.1 and Sec. 2.4.2 will discuss further the specifics and techniques used to calculate each piece of the solution.

2.4.1 The angular part

As mentioned previously, the angular part of the solution to Hydrogen is as follows

$$I_{\text{angular}} = \int_0^{\pi} \int_0^{2\pi} \sin\theta \, \mathcal{Y}_l^m(\theta, \phi) \, d\theta d\phi \tag{16}$$

Since we will be using the Hydrogenic wavefunctions to calculate matrix elements for the higher order Zeeman effect, we expect to see integrals of this form with one or more spherical harmonics within the integrand. Depending on the angular momentum quantum number and the magnetic quantum number, these spherical harmonics may be easy or quite difficult to integrate. However, we can simplify the calculation of the angular integrals by exploiting the nature of the spherical harmonic function. Since it forms a complete set of orthogonal basis functions, for any integral involving two

spherical harmonics we can write:

$$\int_0^{\pi} \int_0^{2\pi} \mathcal{Y}_l^m(\theta, \phi) \mathcal{Y}_{l'}^{m'}(\theta, \phi) \ d\theta d\phi = \delta_{ll'} \delta_{mm'} \tag{17}$$

This rule is extendable to any number of coupled spherical harmonics. For all angular integrals where the spherical harmonics contain different angular momentum and magnetic quantum numbers, the result will be zero. However, if given a scenario where we need to couple three angular momenta together (i.e, we have three spherical harmonic functions to integrate), we can use the Wigner 3j symbols to couple the angular momenta in replacement of the integral. The Wigner 3j symbols are defined as ??

$$\int_{0}^{\pi} \int_{0}^{2\pi} \sin \theta \, d\theta d\phi \, \mathcal{Y}_{l_{1}}^{m_{1}}(\theta, \phi) \mathcal{Y}_{l_{2}}^{m_{2}}(\theta, \phi) \mathcal{Y}_{l_{3}}^{m_{3}}(\theta, \phi) =$$

$$\sqrt{\frac{(2l_{1} + 1)(2l_{2} + 1)(2l_{3} + 1)}{4\pi}} \begin{pmatrix} l_{1} & l_{2} & l_{3} \\ 0 & 0 & 0 \end{pmatrix} \begin{pmatrix} l_{1} & l_{2} & l_{3} \\ m_{1} & m_{2} & m_{3} \end{pmatrix}$$
(18)

Where the set of 6 parameters inside the brackets at the end of the expression is called a 3j Symbol, which has the following definiton:

$$\begin{pmatrix} j_1 & j_2 & j_3 \\ m_1 & m_2 & m_3 \end{pmatrix} = \delta(m_1 + m_2 + m_3, 0)(-1)^{j_1 - j_2 - m_3} \times$$

$$\sqrt{\frac{(j_1 + j_2 - j_3)!(j_1 - j_2 + j_3)!(-j_1 + j_2 + j_3)!}{(j_1 + j_2 + j_3 + 1)!}} \times$$

$$\sqrt{(j_1 - m_1)!(j_1 + m_1)!(j_2 + m_2)!(j_3 - m_3)!(j_3 + m_3)!} \times$$

$$\sum_{N} \frac{(-1)^k}{k!(j_1 + j_2 - j_3 - k)!(j_1 - m_1 - k)!(j_2 + m_2 - k)!(j_3 - j_1 + m_1 + k)!(j_3 - j_1 - m_2 + k)!}$$

Where $K \equiv \max(0, j_2 - j_3 - M - 1, j_1 - j_3 + m_2)$, and $N \equiv \min(j_1 + j_2 - j_3, j_1 - m_1, j_2 + m_2)$. This provides a closed form solution to any angular integrals involving three angular momenta. This technique can be applied to systems with a singular spherical harmonic by coupling two more systems with some angular momenta into the problem. This is allowed because of equation (17), where as long as we divide by the constants associated with the angular momenta we are coupling, is equivalent to multiplying the problem by the constant one. The equivalent expression of the problem in equation (16) for the ground state is⁶

$$I_{\text{angular}} = 4\pi \int_0^{\pi} \int_0^{2\pi} \sin\theta \, \mathcal{Y}_0^0(\theta, \phi) \mathcal{Y}_0^0(\theta, \phi) \mathcal{Y}_0^0(\theta, \phi) \, d\theta d\phi \tag{20}$$

which using the Wigner-3j symbols simplifies to:

$$I_{\text{angular}} = 4\pi \sqrt{\frac{(2l+1)}{4\pi}} \begin{pmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \end{pmatrix} \begin{pmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \end{pmatrix}$$

We now have:

⁶Since $\mathcal{Y}_0^0(\theta,\phi) = \sqrt{\frac{1}{4\pi}}$, we multiply the result by 4π to make the multiplication of two more zeroed spherical harmonics equivalent to mulitplying by one, and maintain an equality between the LHS and RHS of the equation.

$$I_{\text{angular}} = \sqrt{4\pi} \tag{21}$$

Which is equivalent to the integration of the singular \mathcal{Y}_0^0 integral by itself! This can be done for any angular momentum and magnetic quantum number, and will always yield the same result. The ability to substitute the angular part of the integral with a closed form solution in terms of Wigner-3j symbols allows for easy computation providing exact analytic solutions to part of the problem being discussed. While this is typically used for systems where the coupling of 3 angular momenta is necessary, this coupling can be exploited into a sort of pseudo-coupling which we can use to greatly simplify a normally unpleasant calculation.

Additionally, we can further exploit the fact that spherical harmonics form a complete set of orthogonal basis functions, we can express any function of θ and ϕ as a linear combination of spherical harmonics. This technique allows us to replace any matrix element containing an angular contribution (such as $\cos \theta$, as seen in Sec. 3) with a linear combination of spherical harmonic functions. We can then apply the same process of integration utilizing the Wigner-3j coefficients to compute the matrix element without having to perform any actual integration.

2.4.2 The radial part

The radial part of the hydrogenic wavefunction will exploit the general form of $R_{nl}(r)$, stated in equation 12. The perturbed wavefunctions will thus also resemble a similar form since they serve as small corrections to the original solution⁷. Removing the terms not dependant on r outside of the integrand (which are dependant on the perturbed wavefunction being investigated as well as the angular momentum and principle quantum numbers), we are always left with a series of integrals that resemble the following form⁸

⁷See Sec. 2.5 for more details.

⁸This is shown explicitly in Sec. D.3.

$$I_{\text{radial}} = \int_0^\infty r^j e^{-\alpha r} \tag{22}$$

Where $j \in \mathbb{R}$, and $\alpha \in \mathbb{R}$. This integral is a special one, being in the form of the Gamma function integral⁹, $\Gamma(z)$. The Gamma function integral has the following solution

$$\int_0^\infty r^j e^{-\alpha r} = \frac{\Gamma(j+1)}{\alpha^{j+1}} \tag{23}$$

Where as long as $j+1 \in \mathbb{Z}^+$, we can write

$$\int_0^\infty r^j e^{-\alpha r} = \frac{j!}{\alpha^{j+1}} \tag{24}$$

Fitting the integrals to the Gamma function not only makes the task of computing the integrals significantly easier, but also makes the integration computationally stable. The solution provides a simple formula to implement when calculating radial integrals, and allows one to skip the implementation of well known integration methods such as Simpson's rule or any Runge-Kutta methods. While these methods are powerful, they do have significant drawbacks compared to the derived formula due to error propagation and machine-precision.

The ability to compute these radial hydrogenic wavefunction integrals exactly is not only a satisfying result mathematically, but opens the door for more difficult problems in atomic physics where machine error is the determining step in the accuracy of a high-precision calculation¹⁰.

⁹This function is technically Gamma function-esque, but can be manipulated into the form of the Gamma function. This is shown in greater detail in Sec. D.2.

 $^{^{10}}$ Such problems include the calculation of Hylleraas wavefunctions for three body atomic systems such as ${\rm H^-}$, He, or ${\rm Li^+}$, where the radial integrals follow a similar but more complex closed form solution.

2.5 Perturbation Theory

To quantify how a magnetic field affects atomic energy levels, we must determine the shifts induced by the Zeeman interaction. Since this interaction modifies the Hamiltonian slightly compared to the field-free case, we can utilize perturbation theory to provide a systematic way to approximate the new eigenvalues and eigenstates.

To start we can write the perturbed Hamiltonian as a sum of two terms

$$\hat{H} = H^0 + \lambda H' \tag{25}$$

Where H^0 is the standard Hamiltonian and H' is the perturbation. λ is a small weighting coefficient which keeps the perturbation small. We can then expand the wavefunction and the energy of the system as a power series in λ

$$\psi_n = \psi_n^0 + \lambda \psi_n^1 + \lambda^2 \psi_n^2 + \dots$$
 (26)

$$E_n = E_n^0 + \lambda E_n^1 + \lambda^2 E_n^2 + \dots$$
 (27)

Where each superscript is a higher order correction to the original wavefunction. Each higher order term is less impactful to the solution (by a factor of λ) but still contributes a small amount. We can write out the first order perturbation equation as the following

$$(H^0 - E^0) |\psi^1\rangle = (V - E^1) |\psi^0\rangle$$
(28)

To find the first order correction to the energy, E^1 , we simply multiply through by $|\psi^0\rangle$

$$\langle \psi^0 | H^0 - E^0 | \psi^1 \rangle = \langle \psi^0 | V | \psi^0 \rangle - E^1 \langle \psi^0 | \psi^0 \rangle \tag{29}$$

Which by orthogonality, makes the first term 0, and the last term 1. We then end up with

$$E^1 = \langle \psi^0 | V | \psi^0 \rangle \tag{30}$$

In general, any higher order energy can be found with the following expression, which stems from the same process performed here for the first order energy

$$E^n = \langle \psi_{n-1} | V | \psi_0 \rangle \tag{31}$$

Similarly to the standard quantum mechanical Hamiltonian, the perturbation equation is a second order partial differential equation. However, the perturbing term in the equation as well as the higher order corrections to the energy serve as inhomogenous terms, meaning we cannot use separation of variables. To solve this, we can assume that the higher order wavefunction is of the form of a power series, and use the method of Frobenius to check if a solution of this form exists. This will be discussed further in the following section.

To calculate a higher order correction to the wavefunction, we repeat a similar process. This time starting with the second order equation

$$(H^0 - E^0) |\psi^2\rangle = (V - E^1) |\psi^1\rangle + E^2 |\psi^0\rangle$$
(32)

As it can be seen above, the second order solution requires the solution from the first order equation. The requirement of solving the previous correction in order to get the current one makes the processing of calculating higher order corrections quite laborious, and developming a method to solve for higher order corrections computa-

tionally removes the burden of solving lengthy equations by hand. The adaptation of solving the higher order perturbation equations computationally will be discussed in Sec. A.

2.6 Recursion relations

When solving the perturbation equation, you end up with an inhomogeneous second order partial differential equation. To solve for the radial piece of the corrected wavefunction, one must use the method of Frobenius. The method of Frobenius involves assuming the solution is of the form of a power series, and then solving for terminating recursion relations to check if a solution exists. The method of solving these recurrence relations is what will be highlighted in the section below.

A recurrence relation is an equation where the n^{th} term of a sequence is dependant on some combination of its previous terms in the sequence. A simple example of a recurrence relation is the Fibonacci numbers

$$F_n = F_{n-1} + F_{n-2} \tag{33}$$

But these equations can only be solved if there is a base case. A base case is the condition that somewhere, the sequence stops, or is no longer defined by its previous term. A good example of this is the factorial function. The $n^{\rm th}$ factorial number is determined by

$$n! = n(n-1)! (34)$$

Where 0! is not dependant on -1!, but is equivalent to 1. The recurrence relation stops here. If the recurrence relation did not have a base case, and kept referring to the previous iteration forever, there would be no solution to n!. Recurrence relations can be written going downwards, as shown by the Fibonacci numbers and the factorial

examples, or they can be written going upwards. This recurrence relation has a similar but less common form, where the n-1th term depends on the nth term. In this case, the recurrence can still be solved, but you must show that the base case lies somewhere above the current index of your recurrence, or you need to show that the sequence terminates. This means that after a certain point, you can prove that the series is equal to zero for every future term. This is best shown by the following recurrence relation

$$\sum_{j=0}^{\infty} \left[Z(j-1)a_{j-1} - \frac{j(j+1)}{2}a_j \right] r^{j-2} = -\left(\frac{1}{r} + Z\right)$$
 (35)

We can identify the recurrence relation by writing out values for different j

$$j = 0$$
 $-Za_{-1} = 0$ (36)
 $j = 1$ $a_1 = -1$
 $j = 2$ $Za_1 - 3a_2 = -Z$
 $-Z + 3a_2 = -Z$
 $a_2 = 0$
 $j = 3$ $2Za_2 + 6a_3 = 0$
 \vdots \vdots (37)

And we can see here that for every term after a_1 , the series gives zero. Our recurrence relation was used for each iteration including the base case for j = 0 and we have shown that the sequence terminates, providing a non-divergent answer. If this was the solution to our PDE given by the Frobenius method, it would mean that the solution is in the form of a power series and the PDE is solved using our closed form solution.

2.7 The Dalgarno Interchange Theorem

Now that we have discussed the setup for solving our perturbation equation and the matrix elements associated with it, as well as laid the ground work for the Hydrogenic wavefunctions, we need a way to confirm our calculations. This presents itself as the Dalgarno Interchange Theorem. The theorem states that given any two perturbations V and W, thier first order equations can be written as follows:

$$(H^0 - E^0) \psi^1 + V \psi^0 = E^1 \psi^0 \tag{38}$$

$$(H^0 - E^0) \varphi^1 + W \psi^0 = F^1 \psi^0$$
(39)

Multiplying equation (38) by φ^1 and equation (39) by ψ^1 and then subtracting (38) from (39) gives us

$$\langle \psi^0 | V | \psi^1 \rangle - \langle \psi^0 | W | \varphi^1 \rangle = 0 \tag{40}$$

Therefore

$$\langle \psi^0 | V | \psi_1 \rangle = \langle \psi^0 | W | \varphi^1 \rangle \tag{41}$$

So given any two perturbations, we can solve thier respective perturbation equations to retrieve φ^1 and ψ^1 . Using these solutions we can compute two matrix elements with the perturbed wavefunction and its opposing perturbation from the other equation, and the result should be the same! This method serves as a check that our solutions are correct. Once I have calculated both perturbing terms for the magnetic dipole operator and the relativistic correction, I can apply the Dalgarno Interchange theorem to verify my calculations are successful.

CHAPTER 3

The Higher Order Zeeman Effect

3.1 Overview

Now that the proper theory has been introduced, we now move to the main section of the thesis. Here I will introduce the Zeeman Effect, talk about the motivation and direct application of my work, and then move on to the higher order Zeeman Effect. The main focus of this chapter is to show the effect of the quadratic Zeeman Effect, and show how using the magnetic dipole operator in conjunction with the relativistic corrections to ${}^{3}\text{He}^{+}$ can yield a Cubic Zeeman Effect. The effects of both the quadratic and cubic corrections will be discussed in great detail, and the impact of the effect on high precision measurements will be displayed for various magnetic field strengths.

I will begin in Sec. 3.2 where I will discuss the history of the Zeeman effect, its origins and discovery. Afterwards I will move towards the motivation for the project in Sec. 3.3. Here I will discuss some current experiments in the field such as the g-2 experiment being conducted at the Max Planck Institute by Klaus Blaum's research group as well as the applications to high-precision magnetometry. I will then discuss further some additional applications in the field of atomic physics such as I will then discuss in Sec. 3.4 the ordinary Zeeman effect, introducing its theory and application to atomic systems such as $^3{\rm He^+}$. After introducing the ordinary Zeeman effect I will move forward to discuss the quadratic Zeeman effect, where I will derive it using the canonical momentum and give a description of its impact on an atom subjected to a magnetic field. Moving towards higher order systems, I will then

introduce the main topic of this thesis, the cubic Zeeman effect. Starting with the effects that contribute to the cubic Zeeman effect such as the magnetic dipole operator in Sec. 3.6.2 and relativistic corrections to the ${}^{3}\text{He}^{+}$ ion in Sec. 3.6.3 I will show how these effects can be combined to yield a B^{3} contribution to the energy splitting within the presence of an external magnetic field.

3.2 History

- 3.3 Motivation
- 3.3.1 The g-2 experiment
- 3.3.2 High-precision magnetometry
- 3.3.3 Connection to Atomic Physics
- 3.4 The Zeeman effect in Hydrogen
- 3.5 The Zeeman effect in Helium-3
- 3.6 The higher-order Zeeman effect
- 3.6.1 The quadratic Zeeman effect

The non-relativistic Hamiltonian for an n electron system in atomic units can be expressed as follows

$$\hat{H} = \sum_{i=1}^{n} \frac{\left(\vec{p}_i - e\vec{A}_i\right)^2}{2m} + V \tag{1}$$

using the cannonical momentum instead of the classical momentum is essential to account for electromagnetic interactions. The vector potential \vec{A} can be described in

terms of the magnetic field

$$\vec{A} = \frac{B}{2} \left(y\hat{x} - x\hat{y} \right) \tag{2}$$

which when expanded gives us the operator corresponding to the quadratic Zeeman effect

$$\hat{H}_Z^{(2)} = \frac{B^2 e^2}{8m} \sum_{i=1}^n (x_i^2 + y_i^2)$$
(3)

This operator can be expressed in spherical coordinates, written in terms of Legendre Polynomials

$$\hat{H}_Z^{(2)} = \frac{B^2 e^2}{12m} \sum_{i=1}^n r_i^2 \left(P_0(\cos \theta) - P_2(\cos \theta) \right) \tag{4}$$

³He⁺ is a system which contains only one electron, so we can drop the summation which accounts for all electrons to get our final quadratic Zeeman operator

$$\hat{H}_Z^{(2)}(^3\text{He}^+) = \frac{B^2 e^2}{12m} r^2 \left(P_0(\cos\theta) - P_2(\cos\theta) \right)$$
 (5)

3.6.2 The magnetic dipole moment operator

The magnetic dipole moment operator represents the interaction of a magnetic dipole moment with an external magnetic field. It is described via the following relation

$$Q_{M1} = \mu_B \left(1 - \frac{2p^2}{3m^2c^2} + \frac{Ze^2}{3mc^2r} \right) \vec{\sigma} \cdot \vec{B}$$
 (6)

Where μ_B is the Bohr magneton

$$\mu_B = \frac{e\hbar}{2mc} \tag{7}$$

The second term in the brackets of the magnetic dipole moment operator accounts for the relativistic correction to the kinetic energy of the electron, and the third term is the potential energy due to the Coulomb interaction between the electron and the nucleus. The first term corresponds to the ordinary Zeeman Effect, which does not contribute to the sum over states due to orthogonality.

The ordinary Zeeman effect contributes to Q_{M1} in ${}^{3}\mathrm{He^{+}}$ because it has non-zero spin due to the missing electron. For systems such as ${}^{3}\mathrm{He}$, the ordinary Zeeman effect will not contribute.

3.6.3 The relativistic correction to ³He⁺

Combining the magnetic dipole moment with the quadratic Zeeman operator, we can write down the relativistic corrections for ${}^{3}\text{He}^{+}$. Written in terms of pseudostates the relativistic correction is

$$C_{\text{rel}}^{(2)} = \sum_{\substack{n = -\infty\\n \neq 0}}^{\infty} \frac{\langle \psi_0 | H_Z^{(2)} | \psi_n \rangle \langle \psi_n | Q_{M1} | \psi_0 \rangle}{E_0 - E_n}$$
(8)

3.7 Results

CHAPTER 4

Conclusion and Future Work

- 4.1 Overview
- 4.2 Consolidation of Results
- 4.3 Synthesis of Conclusions
- 4.4 Future Work
- 4.5 Theoretical and Practical Impact

APPENDIX A

Methods of solving for higher-order perturbations programmatically

APPENDIX B

Derivation of the Hydrogen Atom

APPENDIX C

$egin{aligned} Derivation & of the Recursion \ Relations \end{aligned}$

APPENDIX D

Special Functions

- D.1 Overview
- D.2 Gamma Function
- D.3 Laguerre Polynomials
- D.4 Legendre Polynomials
- D.5 Spherical Harmonics

VITA AUCTORIS

NAME: Evan Petrimoulx

Windsor Ontario,

Canada:

2003:

Highschool Diploma:

University of Windsor, Undergraduate Honours

Physics, Windsor, Ontario, 2025