

Investigation of the Higher Order Zeeman Effect

By

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DEDICATION

I would like to dedicate this thesis to ...

ACKNOWLEDGEMENTS

...

TABLE OF CONTENTS

DECLARATION OF ORIGINALITY	III
DEDICATION	IV
ACKNOWLEDGEMENTS	V
LIST OF TABLES	VIII
LIST OF FIGURES	IX
LIST OF ABBREVIATIONS	X
1 Introduction	1
1.1 Background and Motivation	1
1.2 Overview of Research Projects	1
1.3 Methodology Overview	1
1.4 Research Significance	1
1.5 Dissertation Structure	1
2 Theoretical Methods	2
2.1 Overview	2
2.2 Atomic Units	3
2.3 One-electron Shrödinger equation	5
2.4 Integration Techniques	7
2.4.1 The angular part	8
2.4.2 The radial part	10
2.5 Perturbation Theory	11
2.6 Recursion relations	11
2.7 The Dalgarno Interchange Theorem	11
3 The Higher Order Zeeman Effect	12
3.1 Overview	13
3.2 History	13
3.3 Motivation	13
3.3.1 The $g - 2$ experiment	13
3.3.2 High-precision magnetometry	13
3.3.3 Connection to Atomic Physics	13
3.4 The Zeeman effect in Hydrogen	13
3.5 The Zeeman effect in Helium-3	13
3.6 The higher-order Zeeman effect	13
3.6.1 The quadratic Zeeman effect	13
3.6.2 The magnetic dipole moment operator	14
3.6.3 The relativistic correction to $^3\text{He}^+$	15

4	Conclusion and Future Work	16
4.1	Overview	16
4.2	Consolidation of Results	16
4.3	Synthesis of Conclusions	16
4.4	Future Work	16
4.5	Theoretical and Practical Impact	16
A	Methods of solving for higher-order perturbations programmatically	17
B	Derivation of the Hydrogen Atom	18
C	Derivation of the Recursion Relations	19
D	Special Functions	20
D.1	Overview	20
D.2	Gamma Function	20
D.3	Laguerre Polynomials	20
D.4	Legendre Polynomials	20
D.5	Spherical Harmonics	20
	REFERENCES	21
	VITA AUCTORIS	21

LIST OF TABLES

LIST OF FIGURES

LIST OF ABBREVIATIONS

CHAPTER 1

Introduction

- 1.1 Background and Motivation
- 1.2 Overview of Research Projects
- 1.3 Methodology Overview
- 1.4 Research Significance
- 1.5 Dissertation Structure

CHAPTER 2

Theoretical Methods

2.1 Overview

This chapter will provide an overview of all theoretical methods used throughout my dissertation. It will provide the theoretical building blocks upon which the dissertation is constructed, ranging from the set of units used in the problem, to perturbation equation solutions and methods for verifying calculations¹. As stated in the introduction, this thesis is based upon calculating matrix elements between two sets of wavefunctions connected by an operator. These matrix elements arise from perturbation theory and are necessary for computing energy shifts due to external fields such as those present in the higher-order Zeeman effect. The foundation of theory will be presented here, but the more specific calculations related to solving for the Zeeman effect will be presented in the relevant chapter below.

I will begin by first discussing atomic units in Sec. 2.2, then move on and introduce the Schrödinger equation for the one-electron problem as well as its solution for hydrogenic wavefunctions in Sec. 2.3. In this section I will also discuss the significance and impact of the canonical momentum, and emphasize its importance in solving for the Zeeman effect. I will then introduce Perturbation Theory in Sec. 2.5, and describe how it can be used to approximate the new eigenvalues and eigenstates of the Zeeman-perturbed Hamiltonian. Sec. 2.4 will discuss various integration techniques that will be implemented both analytically and programmatically in order to simplify the calculation process, followed by an introduction to solving recursion re-

¹Majority of the material discussed throughout this chapter are inspired and shaped by G. W. F. Drake. References will be included where needed.

lations in Sec. 2.6. This is the core step in determining the perturbed wavefunctions used throughout the rest of the study. Finally, this chapter will conclude with the discussion of the Dalgarno Interchange Theorem and how it can serve as a critical check on the correctness of the obtained perturbed wavefunctions.

2.2 Atomic Units

First proposed by Hartree in November of 1928, atomic units have since become the standard across all of atomic physics. Atomic units were designed with the purpose of *”eliminating various universal constants from equations and also to avoid high powers of 10 in numerical work”* ?? . Hartree proposed that we measure the mass in terms of the mass of the electron m_e , charge in terms of the electron charge, e , the vacuum permittivity in terms of $4\pi\epsilon_0$, and the angular momentum in terms of \hbar . Written mathematically we can say

$$\hbar = m_e = e = 4\pi\epsilon_0 = 1 \tag{1}$$

Scaling these units out of the problem greatly simplifies our calculation, and once a result is found, one can multiply the unscaled units back into the answer to retrieve the desired result in standard SI units. While the use of atomic units was proposed long before the field of computational physics emerged, it is of key importance that one implements this technique for any atomic physics simulation. Machine error is an unavoidable reality of programmed computation and working in units which avoids extreme values both large and small can help mitigate the uncertainty present in high precision calculations.

A consequence of utilizing atomic units is the need to redefine key fundamental physical quantities in terms of these scaled values. An example of key importance in the investigation of the higher-order Zeeman effect is the atomic unit of magnetic field strength, otherwise referred to as the atomic unit of magnetic flux density ?? .

$$B_{\text{a.u.}} = \frac{\hbar}{ea_0^2} \quad (2)$$

This value is composed of the Bohr radius a_0 , electron charge e , and \hbar , the unit of action. The Bohr radius is defined as the most probable distance between an electron in the ground state of hydrogen and its nucleus. It can be described by the following equation ??

$$a_0 = \frac{4\pi\epsilon_0\hbar^2}{e^2m_e} = \frac{\hbar}{m_e c \alpha} \quad (3)$$

The Bohr radius itself is defined using a similar combination of fundamental constants which after applying the rules for atomic units, should equal 1. We can see now that all units in the atomic unit of field strength also become 1 as intended. At the end of our problem, we can simply multiply the determined result by the atomic unit of magnetic field strength when its constants are not scaled to 1, and retrieve our desired result in SI units. This is equivalent to multiplying the result in atomic units by ??.

$$B_{\text{a.u.}} = 2.35051757077(73) \times 10^5 \text{ T} \quad (4)$$

The numbers in parenthesis here denote the uncertainty of the quantity. Another key benefit in atomic units is that it allows us to define almost all fundamental quantities in terms of two constants; the speed of light, c and the fine structure constant, α .²

The fine structure constant can be written as the following dimensionless quantity

²The fine structure constant was introduced in 1916 by Arnold Sommerfeld when he was investigating the gap in the spectral lines of the hydrogen atom. He compared his work with the Michelson Morley experiment from 1887 where the spectral lines were shown and it was deduced that the universe is not permeated by the aether ??

$$\alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = 7.2973525643(11) \times 10^{-3} \quad (5)$$

which in atomic units is just the following relationship with the speed of light

$$\alpha = \frac{1}{c} \quad (6)$$

The fine structure constant is often used as an essential expansion parameter, treating relativity as a perturbation in α .³ Applying the atomic unit scaling to the fine structure constant shown in equation (6) confirms that the second equivalence in equation (3) remains equal to 1.

Applying this technique to problems in atomic physics gives a powerful way of simplifying calculations while providing an easy conversion factor to SI units for comparison of theory and experiment. This dissertation will discuss all equations and quantities in atomic units unless further specified otherwise.

2.3 One-electron Schrödinger equation

Discovered by Schrödinger in 1926, the Schrödinger equation is a non-relativistic solution to the two body problem involving an orbiting electron and its nucleus⁴. The electron interacts with the nucleus via the Coulomb interaction described by the following potential

$$V(\vec{r}) = -\frac{Ze^2}{4\pi\epsilon_0\vec{r}} \quad (7)$$

We can write the energy of the system using the Hamiltonian, where our potential is the Coulomb potential. We then come up with

³More on this will be discussed in Sec. 3.6.3

⁴It is assumed here that the nucleus is a point particle with infinite mass.

$$H = \frac{p^2}{2m} - \frac{Ze^2}{4\pi\epsilon_0\vec{r}} \quad (8)$$

However, quantum mechanics dictates that all observable quantities must be determined by hermitian operators that act on the wavefunction ψ . The wavefunction ψ is described as a normalized vector in Hilbert space, \mathcal{H} containing all information pertaining to the system. This means that the momentum needs to be a hermitian operator that acts on the wavefunction, and that our Hamiltonian is also a hermitian operator that satisfies the following eigenvalue problem

$$\hat{H}|\psi\rangle = E_n|\psi\rangle \quad (9)$$

This is what is known as the Schrödinger equation. It can be applied to our two-body problem by substituting the momentum for its quantum operator analog. This results in the following expression

$$\hat{H}\psi = \frac{-\hbar^2}{2m_e}\nabla^2\psi - \frac{Ze^2}{4\pi\epsilon_0\vec{r}}\psi = E_n\psi \quad (10)$$

We now have a second-order partial differential equation. Assuming a separable solution allows one to split the solution to any hydrogenic system into two parts; The radial part and the angular part.

$$\psi_{nlm} = R_{nl}(r)\mathcal{Y}_m^l(\cos\theta) \quad (11)$$

For our two body atomic system the solutions for each can be written in terms of Laguerre polynomials and Spherical Harmonics⁵. The radial equation has the

⁵The derivation will be shown in the appendix for both the angular and radial parts of the wavefunction.

following solutions ??

$$R_{nl}(r) = \frac{2Z}{n^2} \sqrt{\frac{Z(n-l-1)!}{(n+l)!}} \left(\frac{2Zr}{n}\right)^l e^{-\frac{Zr}{n}} L_{n-l-1}^{(2l+1)}(2Zr/n) \quad (12)$$

Where Z is the nuclear charge atomic number, l is the angular momentum quantum number, n is the principle quantum number, and $L_{n-l-1}^{(2l+1)}(2Zr/n)$ is the generalized Laguerre polynomial (which I have defined in Appendix D.3). Additionally, the spherical harmonics can be written as

$$\mathcal{Y}_l^m(\theta, \phi) = (-1)^m \sqrt{\frac{(2l+1)(l-m)!}{4\pi(l+m)!}} P_l^m(\cos \theta) e^{im\phi} \quad (13)$$

Where l is the angular momentum quantum number, m is the magnetic quantum number, and $P_l^m(\cos \theta)$ is the associated Legendre polynomial (which I will discuss further in Appendix D.4). A key feature of the spherical harmonics which I will utilize in Sec. 2.4 is that the spherical harmonics form a normalized complete set of orthogonal basis functions. The ability to separate the solution into two independent parts plays a crucial role in simplifying the task at hand and will be exploited when evaluating integrals involving these wavefunctions in Chapter 3.

2.4 Integration Techniques

In this section of the dissertation, I will discuss the standard approach to solving the integrals pertaining to the various matrix elements required throughout. As discussed in the previous section, the ability to separate the solution into a radial part as well as an angular part significantly simplifies the solution process. Instead of having to integrate a single function over all three dimensions in spherical coordinates, we can split the solution into two steps which can be calculated independently.

$$\langle \psi^n | V | \psi_m \rangle = \int_0^\infty r^2 \psi^n(r) V(r) \psi^m(r) dr \int_\Omega \sin \theta \psi^n(\theta, \phi) V(\theta, \phi) \psi^m(\theta, \phi) d\Omega \quad (14)$$

Where Ω represents the solid angles for θ and ϕ . The total problem can then be solved with the simple combination of calculated integrals.

$$\langle \psi^n | V | \psi_m \rangle = I_{\text{radial}} \cdot I_{\text{angular}} \quad (15)$$

While this may look obvious, it is only from the fact that the wavefunctions are separable that allows us to split the problem apart. The section serves as a general introduction to the process of calculating matrix elements, but Sec. 2.4.1 and Sec. 2.4.2 will discuss further the specifics and techniques used to calculate each piece of the solution.

2.4.1 The angular part

As mentioned previously, the angular part of the solution to Hydrogen is as follows

$$I_{\text{angular}} = \int_0^\pi \int_0^{2\pi} \sin \theta \mathcal{Y}_l^m(\theta, \phi) d\theta d\phi \quad (16)$$

Since we will be using the Hydrogenic wavefunctions to calculate matrix elements for the higher order Zeeman effect, we expect to see integrals of this form with one or more spherical harmonics within the integrand. Depending on the angular momentum quantum number and the magnetic quantum number, these spherical harmonics may be easy or quite difficult to integrate. However, we can simplify the calculation of the angular integrals by exploiting the nature of the spherical harmonic function. Since it forms a complete set of orthogonal basis functions, for any integral involving two spherical harmonics we can write:

$$\int_0^\pi \int_0^{2\pi} \mathcal{Y}_l^m(\theta, \phi) \mathcal{Y}_{l'}^{m'}(\theta, \phi) d\theta d\phi = \delta_{ll'} \delta_{mm'} \quad (17)$$

So for all angular integrals where the spherical harmonics contain different angular momentum and magnetic quantum numbers, the result will be zero. However, if given a scenario where we need to couple three angular momenta together (i.e, we have three spherical harmonic functions to integrate), we can use the Wigner 3j symbols to couple the angular momenta in replacement of the integral. The Wigner 3j symbols are defined as ??

$$\int_0^\pi \int_0^{2\pi} \sin \theta d\theta d\phi \mathcal{Y}_{l_1}^{m_1}(\theta, \phi) \mathcal{Y}_{l_2}^{m_2}(\theta, \phi) \mathcal{Y}_{l_3}^{m_3}(\theta, \phi) = \quad (18)$$

$$\sqrt{\frac{(2l_1+1)(2l_2+1)(2l_3+1)}{4\pi}} \begin{pmatrix} l_1 & l_2 & l_3 \\ 0 & 0 & 0 \end{pmatrix} \begin{pmatrix} l_1 & l_2 & l_3 \\ m_1 & m_2 & m_3 \end{pmatrix}$$

Where the set of 6 parameters inside the brackets at the end of the expression is called a 3j Symbol, which has the following definition:

$$\begin{pmatrix} j_1 & j_2 & j_3 \\ m_1 & m_2 & m_3 \end{pmatrix} = \delta(m_1 + m_2 + m_3, 0) (-1)^{j_1 - j_2 - m_3} \times \quad (19)$$

$$\sqrt{\frac{(j_1 + j_2 - j_3)!(j_1 - j_2 + j_3)!(-j_1 + j_2 + j_3)!}{(j_1 + j_2 + j_3 + 1)!}} \times$$

$$\sqrt{(j_1 - m_1)!(j_1 + m_1)!(j_2 + m_2)!(j_3 - m_3)!(j_3 + m_3)!} \times$$

$$\sum_N^K \frac{(-1)^k}{k!(j_1 + j_2 - j_3 - k)!(j_1 - m_1 - k)!(j_2 + m_2 - k)!(j_3 - j_1 + m_1 + k)!(j_3 - j_1 - m_1 - k)!}$$

Where $K \equiv \max(0, j_2 - j_3 - M - 1, j_1 - j_3 + m_2)$, and $N \equiv \min(j_1 + j_2 - j_3, j_1 -$

$m_1, j_2 + m_2)$

2.4.2 The radial part

The radial part of the hydrogenic wavefunction will exploit the general form of $R_{nl}(r)$, stated in equation 12. The perturbed wavefunctions will thus also resemble a similar form since they serve as small corrections to the original solution⁶. Removing the terms not dependant on r outside of the integrand (which are dependant on the perturbed wavefunction being investigated as well as the angular momentum and principle quantum numbers), we are always left with a series of integrals that resemble the following form⁷

$$I_{\text{radial}} = \int_0^\infty r^j e^{-\alpha r} \quad (20)$$

Where $j \in \mathbb{R}$, and $\alpha \in \mathbb{R}$. This integral is a special one, being in the form of the Gamma function integral⁸, $\Gamma(z)$. The Gamma function integral has the following solution

$$\int_0^\infty r^j e^{-\alpha r} = \frac{\Gamma(j+1)}{\alpha^{j+1}} \quad (21)$$

Where as long as $j+1 \in \mathbb{Z}^+$, we can write

$$\int_0^\infty r^j e^{-\alpha r} = \frac{j!}{\alpha^{j+1}} \quad (22)$$

Fitting the integrals to the Gamma function not only makes the task of computing the integrals significantly easier, but also makes the integration computationally

⁶See Sec. 2.5 for more details.

⁷This is shown explicitly in Sec. D.3.

⁸This function is technically Gamma function-esque, but can be manipulated into the form of the Gamma function. This is shown in greater detail in Sec. D.2.

stable. The solution provides a simple formula to implement when calculating radial integrals, and allows one to skip the implementation of well known integration methods such as Simpson's rule or any Runge-Kutta methods. While these methods are powerful, they do have significant drawbacks compared to the derived formula due to error propagation and machine-precision.

The ability to compute these radial hydrogenic wavefunction integrals exactly is not only a satisfying result mathematically, but opens the door for more difficult problems in atomic physics where machine error is the determining step in the accuracy of a high-precision calculation⁹.

2.5 Perturbation Theory

To quantify how a magnetic field affects atomic energy levels, we must determine the shifts induced by the Zeeman interaction. Since this interaction modifies the Hamiltonian slightly compared to the field-free case, we can utilize perturbation theory to provide a systematic way to approximate the new eigenvalues and eigenstates.

2.6 Recursion relations

2.7 The Dalgarno Interchange Theorem

⁹Such problems include the calculation of Hylleraas wavefunctions for three body atomic systems such as H^- , He , or Li^+ , where the radial integrals follow a similar but more complex closed form solution.

CHAPTER 3

The Higher Order Zeeman Effect

3.1 Overview

3.2 History

3.3 Motivation

3.3.1 The $g - 2$ experiment

3.3.2 High-precision magnetometry

3.3.3 Connection to Atomic Physics

3.4 The Zeeman effect in Hydrogen

3.5 The Zeeman effect in Helium-3

3.6 The higher-order Zeeman effect

3.6.1 The quadratic Zeeman effect

The non-relativistic Hamiltonian for an n electron system in atomic units can be expressed as follows

$$\hat{H} = \sum_{i=1}^n \frac{(\vec{p}_i - e\vec{A}_i)^2}{2m} + V \quad (1)$$

using the canonical momentum instead of the classical momentum is essential to account for electromagnetic interactions. The vector potential \vec{A} can be described in terms of the magnetic field

$$\vec{A} = \frac{B}{2} (y\hat{x} - x\hat{y}) \quad (2)$$

which when expanded gives us the operator corresponding to the quadratic Zeeman effect

$$\hat{H}_Z^{(2)} = \frac{B^2 e^2}{8m} \sum_{i=1}^n (x_i^2 + y_i^2) \quad (3)$$

This operator can be expressed in spherical coordinates, written in terms of Legendre Polynomials

$$\hat{H}_Z^{(2)} = \frac{B^2 e^2}{12m} \sum_{i=1}^n r_i^2 (P_0(\cos \theta) - P_2(\cos \theta)) \quad (4)$$

${}^3\text{He}^+$ is a system which contains only one electron, so we can drop the summation which accounts for all electrons to get our final quadratic Zeeman operator

$$\hat{H}_Z^{(2)}({}^3\text{He}^+) = \frac{B^2 e^2}{12m} r^2 (P_0(\cos \theta) - P_2(\cos \theta)) \quad (5)$$

3.6.2 The magnetic dipole moment operator

The magnetic dipole moment operator represents the interaction of a magnetic dipole moment with an external magnetic field. It is described via the following relation

$$Q_{M1} = \mu_B \left(1 - \frac{2p^2}{3m^2 c^2} + \frac{Ze^2}{3mc^2 r} \right) \vec{\sigma} \cdot \vec{B} \quad (6)$$

Where μ_B is the Bohr magneton

$$\mu_B = \frac{e\hbar}{2mc} \quad (7)$$

The second term in the brackets of the magnetic dipole moment operator accounts

for the relativistic correction to the kinetic energy of the electron, and the third term is the potential energy due to the Coulomb interaction between the electron and the nucleus. The first term corresponds to the ordinary Zeeman Effect, which does not contribute to the sum over states due to orthogonality.

The ordinary Zeeman effect contributes to Q_{M1} in ${}^3\text{He}^+$ because it has non-zero spin due to the missing electron. For systems such as ${}^3\text{He}$, the ordinary Zeeman effect will not contribute.

3.6.3 The relativistic correction to ${}^3\text{He}^+$

Combining the magnetic dipole moment with the quadratic Zeeman operator, we can write down the relativistic corrections for ${}^3\text{He}^+$. Written in terms of pseudostates the relativistic correction is

$$C_{\text{rel}}^{(2)} = \sum_{\substack{n=-\infty \\ n \neq 0}}^{\infty} \frac{\langle \psi_0 | H_Z^{(2)} | \psi_n \rangle \langle \psi_n | Q_{M1} | \psi_0 \rangle}{E_0 - E_n} \quad (8)$$

CHAPTER 4

Conclusion and Future Work

4.1 Overview

4.2 Consolidation of Results

4.3 Synthesis of Conclusions

4.4 Future Work

4.5 Theoretical and Practical Impact

APPENDIX A

*Methods of solving for
higher-order perturbations
programmatically*

APPENDIX B

Derivation of the Hydrogen Atom

APPENDIX C

Derivation of the Recursion Relations

APPENDIX D

Special Functions

D.1 Overview

D.2 Gamma Function

D.3 Laguerre Polynomials

D.4 Legendre Polynomials

D.5 Spherical Harmonics

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