

Section 4: Kinetic Theory

AE435
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3 Velocity Distribution Function

Not all particles move with the same velocity. Also, the velocity of a particle doesn't remain the same over time. We need a statistical way to describe this; this is the velocity distribution function.

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3.1 Mass Distribution Function

To illustrate this idea of distribution function, consider mass density. Essentially what the mass distribution function expresses is a distribution across physical space. A change in the density of the gas in physical space. We will make the analogy of distribution in physical space and distribution in velocity space.

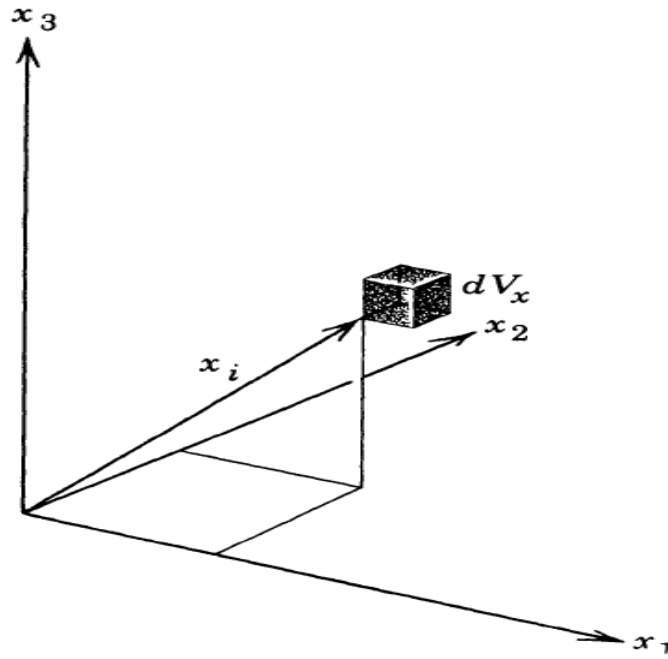


Fig. 1. Volume element in physical space.

Consider a gas of N particles with mass m in a volume V .

The density is:

$$\rho = m \frac{N}{V} = mn \quad (17)$$

If the gas is nonuniform, the density in a differential volume

$$dV_x = dx_1 dx_2 dx_3$$

At a position vector

$$\vec{x}_i = [x_1 \quad x_2 \quad x_3]$$

then the density at that location is:

$$\rho(\vec{x}_i) = \lim_{\Delta V_x \rightarrow 0} m \frac{\Delta N}{\Delta V_x} \quad (18)$$

We still operate under the assumption that the volume contains a large number of particles. Note that this assumes that ΔV_x is large enough to contain a large number of particles. Since the particle mass doesn't change,

$$\rho(\vec{x}_i) = m \lim_{\Delta V_x \rightarrow 0} \frac{\Delta N}{\Delta V_x}$$

Which we can write as:

$$\rho(\vec{x}_i) = m n(\vec{x}_i) \quad (19)$$

Meaning that the mass density can be different depending on where we stand in physical space.

The function $n(\vec{x}_i)$ gives the number of particles per unit volume as a function of position; a.k.a. a "position distribution function"

The number of particles in the differential volume dV_x is

$$dN = n(\vec{x}_i) dV_x$$

So the mass within that differential volume is

$$dM = m dN = \rho(\vec{x}_i) dV_x$$

We can define a "normalized distribution function" as

$$w(\vec{x}_i) = \frac{n(\vec{x}_i)}{N} \quad (20)$$

Which is essentially just the percent of particles at position x_i

So that the total number of particles in dV_x is

$$dN = N w(\vec{x}_i) dV_x \quad (21)$$

This normalized distribution function can be interpreted as a **probability density function**, that is, the probability that a given randomly-chosen particle will be in dV_x . In other words, it is the probability that a particle will be at a specific location.

Integrating over the entire volume,

$$\int_V w(\vec{x}_i) dV_x = 1$$

If we integrate over the entire volume, the probability that a particle within the volume is within the volume is 100%. As expected.

We can generalize this idea to state:

A distribution function gives the concentration of some quantity per unit "volume" as a function of position in some kind of "space".

3.2 Velocity Distribution Function

Now consider particle with velocity $\vec{c} = [c_1 \ c_2 \ c_3]$

We can define a differential volume in this velocity space

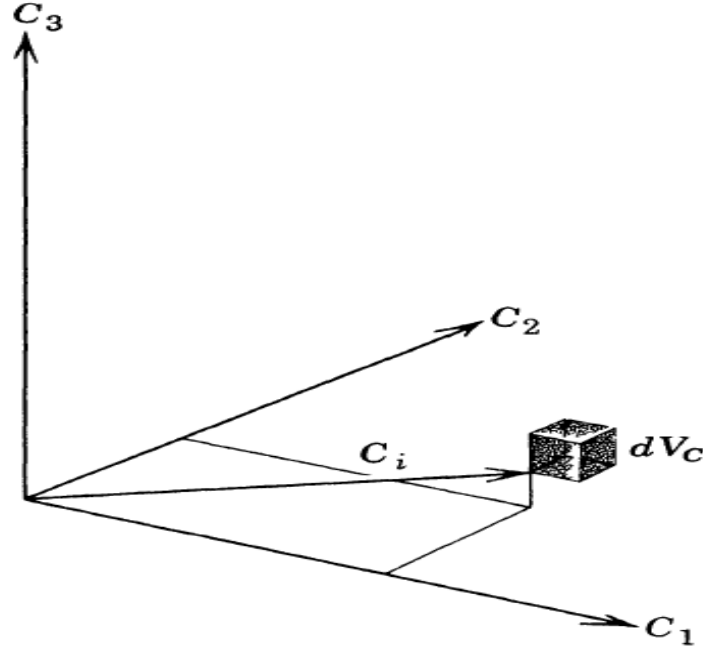


Fig. 2. Volume element in velocity space.

Define local point density $F(\vec{c}_i)$ such that the number of particles within velocity range:

$$\begin{aligned} c_1 &\rightarrow c_1 + dc_1 \\ c_2 &\rightarrow c_2 + dc_2 \\ c_3 &\rightarrow c_3 + dc_3 \end{aligned} \tag{22}$$

which we would write that as $F(\vec{c}_i)dV_c$. This is the **velocity distribution function**. Like the position distribution function $n(\vec{x}_i)$ you have to multiply by a volume to get a real quantity (the number of particles). Essentially telling us how particles are distributed across velocity space.

Define a normalized velocity distribution function

$$f(\vec{c}_i) = \frac{F(\vec{c}_i)}{N} \tag{23}$$

This is the probability that a particle will be within the specific velocity range. The number of particles within dV_c is

$$dN = N f(\vec{c}_i) dV_c \tag{24}$$

In terms of number density $n = \frac{N}{V}$

$$dn = n f(\vec{c}_i) dV_c$$

where the integral over all possible velocities is:

$$\int_{-\infty}^{\infty} N f(\vec{c}_i) dV_c = N \quad \rightarrow \quad \int_{-\infty}^{\infty} f(\vec{c}_i) dV_c = 1 \quad (25)$$

Essentially saying that if you look over the entirety of velocity space, you will find N particles since that is the amount you have in the system. In other words, if we integrate over all of velocity space, the fraction of particles we will find is 1, 100%.

The velocity distribution of particles is important for determining average quantities. For instance, if we have some quantity Q that depends on velocity $Q = Q(\vec{c}_i)$. The mean or expectation value of Q is then:

$$\begin{aligned} \bar{Q} &= \frac{1}{N} \int_N Q dN = \frac{1}{N} \int_{-\infty}^{\infty} Q(\vec{c}_i) N f(\vec{c}_i) dV_c \\ &= \int_{-\infty}^{\infty} Q(\vec{c}_i) f(\vec{c}_i) dV_c \end{aligned} \quad (26)$$

$$\langle \vec{c} \rangle = \int_{-\infty}^{\infty} \vec{c} f(\vec{c}_i) dV_c$$

Integrating over the distribution function. If Q is a velocity, then it is called taking the moment of the velocity distribution function.

For example if we were to do it of $\langle C \rangle$ would be the first moment which is useful for finding the average collision rate, whether Q is the collision cross section which is tied with the particle velocity or if we want to find the average of some property that we are interested in finding, velocity, energy, over the entire system.

3.3 Maxwellian Velocity Distribution Function

A gas at equilibrium has a special velocity distribution function. Called the Maxwellian velocity distribution. This is not often the case for electric propulsion systems since plasmas are rarely in equilibrium.

Basic idea is that:

- Stationary velocity distribution
- Collisions deplete and add to population at same rate
- Thus, no net change.

Collision dynamics with simple billiard-ball model leads to:

Maxwellian Velocity Distribution Function

$$f_M(\vec{c}_i) = \left(\frac{m}{2\pi k T} \right)^{\frac{3}{2}} \exp \left[-\frac{m}{2kT} (c_1^2 + c_2^2 + c_3^2) \right] \quad (27)$$

where

m = Mass of Particle

T = Temperature

k = Boltzmann Constant

We can break this into components along each axis

$$\begin{aligned} \Xi_m(c_1) &= \left(\frac{m}{2\pi k T} \right)^{\frac{1}{2}} \exp \left[-\frac{m}{2kT} c_1^2 \right] \\ \Xi_m(c_2) &= \left(\frac{m}{2\pi k T} \right)^{\frac{1}{2}} \exp \left[-\frac{m}{2kT} c_2^2 \right] \\ \Xi_m(c_3) &= \left(\frac{m}{2\pi k T} \right)^{\frac{1}{2}} \exp \left[-\frac{m}{2kT} c_3^2 \right] \end{aligned} \quad (28)$$

The probability that a particle will be within the velocity space $d\vec{c}_i = dV_c$

$$f_m(\vec{c}_i) dc_1 dc_2 dc_3 = \Xi_m(c_1) dc_1 \Xi_m(c_2) dc_2 \Xi_m(c_3) dc_3 \quad (29)$$

Which is simply the probability density function of the x, y and z component.

The Maxwellian VDF looks like this:

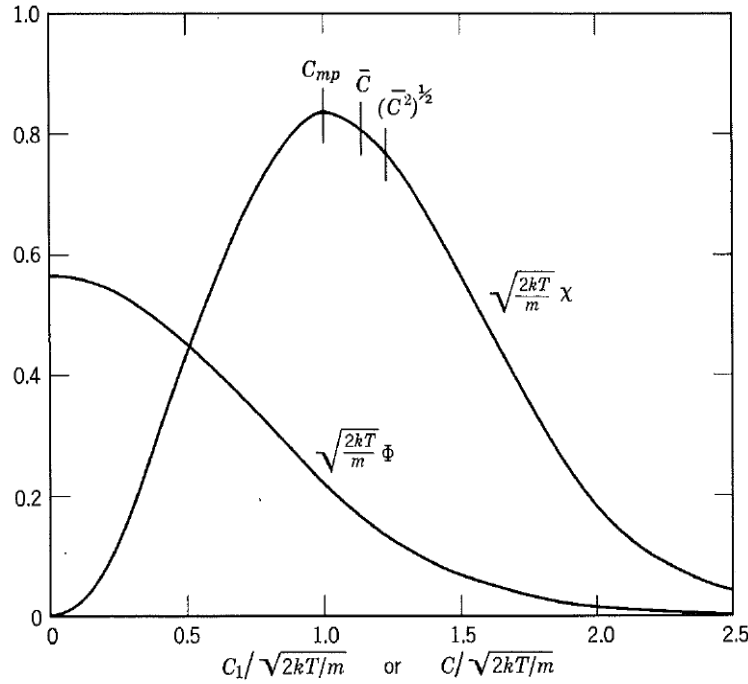


Fig. 8. Distribution functions $\Phi(C_1)$ and $\chi(C)$.

Shown in the plot, the smaller curve is the velocity distribution function. The negative x-axis isn't shown but it is symmetric. Particles can have negative velocity which is why we have this symmetry.

The largest probability is at $c_1 = 0$ which corresponds to particles moving perpendicular to the x_1 -axis with $\vec{c}_i \cdot \hat{x}_1 = 0$.

We can transform the Maxwellian Velocity Distribution Function $f_m(\vec{c}_i)$ into a:

Maxwellian Speed Distribution Function

$$\chi_m(c) = 4\pi \left(\frac{m}{2\pi kT} \right)^{\frac{3}{2}} c^2 \exp \left[-\frac{mc^2}{2kT} \right] \quad (30)$$

Where

$$c^2 = c_1^2 + c_2^2 + c_3^2$$

The speed distribution is not symmetric since speed cannot be negative. Recalling the plot above, the larger curve shown on the plot is the speed distribution recently defined.

Note that the most probable SPEED is not zero. This figure calls out some important points. The

peak is the most probable speed. \bar{c} is the average speed, \bar{c}^2 is the root-mean square mean.

$$c_{mp} = \left(\frac{2kT}{m} \right)^{\frac{1}{2}} \quad (31)$$

Also, the mean speed is not the same as the most probable-speed (which is always the peak).

$$\bar{c} = \int_0^\infty c \chi_m(c) dc = \left(\frac{8kT}{\pi m} \right)^{\frac{1}{2}} = 1.13 c_{mp} \quad 13 \% \text{ Higher} \quad (32)$$

Now we want the average speed squared so we integrate over the function. Finally, the mean squared speed is

$$\bar{c}^2 = \int_0^\infty c^2 \chi_m(c) dc = \frac{3kT}{m}$$

Which works out to give the root-mean-square speed.

$$\sqrt{\bar{c}^2} = \left(\frac{3kT}{m} \right)^{\frac{1}{2}} = 1.22 c_{mp} \quad 22 \% \text{ Higher} \quad (33)$$

What this is saying is that particles have a mean square root speed if and only if our gas is in equilibrium.

Other Distributions for Electric Propulsion:

In electric propulsion, specifically Ion Propulsion, we have thermal electrons inside the discharge change. Thermal ions = equilibrium. What kind of distribution would these particles have? They would have a Maxwellian distribution.

In addition to thermal ions, we have a cathode that is emitting high energy electrons resulting a bump on tail distribution due to the high energy particles. The electrons inside this discharge chamber are not in equilibrium and cannot be described by the Maxwellian distribution.

Wave-Particle Interactions The two stream instability can be thought of as the inverse of Landau damping, where the existence of a greater number of particles that move slower than the wave phase velocity v_{ph} as compared with those that move faster, leads to an energy transfer from the wave to the particles. In the case of the two stream instability, when an electron stream is injected to the plasma, the particles' velocity distribution function has a "bump" on its "tail". If a wave has phase velocity in the region where the slope is positive, there is a greater number of faster particles ($v > v_{ph} > v_{ph}$) than slower particles, and so there is a greater amount of energy being transferred from the fast particles to the wave, giving rise to exponential wave growth.

