

Tutorial 2: A simple charged system

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1 Introduction

This tutorial introduces some of the basic features of ESPResSo for charged systems by constructing a simulation script for a simple salt crystal. In the subsequent task, we use to a more realistic force-field for a NaCl crystal. Finally, we introduce constraints and 2D-Electrostatics to simulate a molten salt in a parallel plate capacitor. We assume that the reader is familiar with the basic concepts of Python and MD simulations. The code pieces can be copied step by step into a python script file, which then can be run using `pypresso <file>` from the ESPResSo build directory. Note that you might have to correct erroneous line breaks. Compile espresso with the following features in your *myconfig.hpp* to be set throughout the whole tutorial:

```
#define EXTERNAL_FORCES
#define CONSTRAINTS
#define MASS
#define ELECTROSTATICS
#define LENNARD_JONES
```

2 Basic set up

The script for the tutorial can be found in your build directory at `\doc\tutorials\python\02-charged_system\nacl.py`. We start with importing numpy and the espressomd features and setting up all the relevant simulation parameters in one place:

```
from espressomd import System, assert_features, electrostatics
import numpy

assert_features(["ELECTROSTATICS", "LENNARD_JONES"])

# System parameters
n_part = 200
n_ionpairs = n_part/2
density = 0.7
time_step = 0.01
temp = 1.0
gamma = 1.0
```

```

l_bjerrum = 1.0

num_steps_equilibration = 1000
num_configs = 100
integ_steps_per_config = 1000

# Particle parameters
types      = {"Anion": 0, "Cation": 1}
numbers    = {"Anion": n_ionpairs, "Cation": n_ionpairs}
charges    = {"Anion": -1.0, "Cation": 1.0}
lj_sigmas  = {"Anion": 1.0, "Cation": 1.0}
lj_epsilon = {"Anion": 1.0, "Cation": 1.0}

WCA_cut = 2.**(1. / 6.)
lj_cuts  = {"Anion": WCA_cut * lj_sigmas["Anion"],
            "Cation": WCA_cut * lj_sigmas["Cation"]}

```

These variables do not change anything in the simulation engine, but are just standard Python variables. They are used to increase the readability and flexibility of the script. The box length is not a parameter of this simulation, it is calculated from the number of particles and the system density. This allows to change the parameters later easily, e. g. to simulate a bigger system. We use dictionaries for all particle related parameters, which is less error-prone and readable as we will see later when we actually need the values. The parameters here define a purely repulsive, equally sized, monovalent salt.

The simulation engine itself is modified by changing the `espressomd.System()` properties. We create an instance `system` and set the box length, periodicity and time step. The skin depth `skin` is a parameter for the link-cell system which tunes its performance, but shall not be discussed here. Further, we activate the Langevin thermostat for our NVT ensemble with temperature `temp` and friction coefficient `gamma`.

```

# Setup System
system = System()
box_l = (n_part / density)**(1. / 3.)
system.box_l = [box_l, box_l, box_l]
system.periodicity = [1, 1, 1]
system.time_step = time_step
system.cell_system.skin = 0.3
system.thermostat.set_langevin(kT=temp, gamma=gamma)

```

We now fill this simulation box with particles at random positions, using type and charge from our dictionaries. Using the length of the particle list `system.part` for the id,

we make sure that our particles are numbered consecutively. The particle type is used to link non-bonded interactions to a certain group of particles.

```
# Place particles
for i in range(n_ionpairs):
    system.part.add(
        id=len(system.part),
        type=types["Anion"],
        pos=numpy.random.random(3) * box_l,
        q=charges["Anion"])
for i in range(n_ionpairs):
    system.part.add(
        id=len(system.part),
        type=types["Cation"],
        pos=numpy.random.random(3) * box_l,
        q=charges["Cation"])
```

Before we can really start the simulation, we have to specify the interactions between our particles. We already defined the Lennard-Jones parameters at the beginning, what is left is to specify the combination rule and to iterate over particle type pairs. For simplicity, we implement only the *Lorentz-Berthelot* rules. We pass our interaction pair to `system.non_bonded_inter[*,*]` and set the pre-calculated LJ parameters `epsilon`, `sigma` and `cutoff`. With `shift="auto"`, we shift the interaction potential to the cutoff so that $U_{LJ}(r_{cutoff}) = 0$.

```
def combination_rule_epsilon(rule, eps1, eps2):
    if rule=="Lorentz":
        return (eps1*eps2)**0.5
    else:
        return ValueError("No combination rule defined")

def combination_rule_sigma(rule, sig1, sig2):
    if rule=="Berthelot":
        return (sig1+sig2)*0.5
    else:
        return ValueError("No combination rule defined")

# Lennard-Jones interactions parameters
for s in [ ["Anion", "Cation"],
           ["Anion", "Anion"],
           ["Cation", "Cation"] ]:
```

```

lj_sig = combination_rule_sigma(
    "Berthelot",
    lj_sigmas[s[0]],lj_sigmas[s[1]])
lj_cut = combination_rule_sigma(
    "Berthelot",
    lj_cuts[s[0]],lj_cuts[s[1]])
lj_eps = combination_rule_epsilon(
    "Lorentz",
    lj_epsilon[s[0]],lj_epsilon[s[1]])

system.non_bonded_inter[types[s[0]], types[s[1]]].
lennard_jones.set_params(epsilon = lj_eps, sigma = lj_sig,
cutoff = lj_cut, shift = "auto")

```

3 Equilibration

With randomly positioned particles, we most likely have huge overlap and the strong repulsion will cause the simulation to crash. The next step in our script therefore is a suitable LJ equilibration. This is known to be a tricky part of a simulation and several approaches exist to reduce the particle overlap. Here, we use a highly damped system (huge gamma in the thermostat) and cap the forces of the LJ interaction. We use `system.analysis.mindist` to get the minimal distance between all particles pairs. This value is used to progressively increase the force capping. This results in a slow increase of the force capping at strong overlap. At the end, we reset our thermostat to the target values and deactivate the force cap by setting it to zero.

```

print("\n--->Lennard Jones Equilibration")
max_sigma = max(lj_sigmas.values())
min_dist = 0.0
cap = 10.0
#Warmup Helper: Cold, highly damped system
system.thermostat.set_langevin(kT=temp*0.1, gamma=gamma*50.0)

while min_dist < max_sigma:
    #Warmup Helper: Cap max. force, increase slowly for
    overlapping particles
    min_dist = system.analysis.mindist(
        [types["Anion"],types["Cation"]],
        [types["Anion"],types["Cation"]])

```

```

    cap += min_dist
    system.non_bonded_inter.set_force_cap(cap)
    system.integrator.run(10)

#Don't forget to reset thermostat and force cap
system.thermostat.set_langevin(kT=temp, gamma=gamma)
system.non_bonded_inter.set_force_cap(0)

```

ESPResSo uses so-called **actors** for electrostatics, magnetostatics and hydrodynamics. This ensures that unphysical combinations of algorithms are avoided, for example simultaneous usage of two electrostatic interactions. Adding an actor to the system also activates the method and calls necessary initialization routines. Here, we define a P³M object with parameters Bjerrum length and rms force error. This automatically starts a tuning function which tries to find optimal parameters for P³M and prints them to the screen:

```

print("\n--->Tuning Electrostatics")
p3m = electrostatics.P3M(bjerrum_length=l_bjerrum,
                          accuracy=1e-3)
system.actors.add(p3m)

```

Before the production part of the simulation, we do a quick temperature equilibration. For the output, we gather all energies with `system.analysis.energy()`, calculate the "current" temperature¹ from the ideal part and print it to the screen along with the total and Coulomb energies.

We integrate for a certain amount of steps with `system.integrator.run(100)`.

```

print("\n--->Temperature Equilibration")
system.time = 0.0
for i in range(num_steps_equilibration/100):
    energy = system.analysis.energy()
    temp_measured = energy['kinetic'] / ((3.0 / 2.0) * n_part)
    print("t={0:.1f}, E_total={1:.2f}, E_coulomb={2:.2f},
          T={3:.4f}".format(system.time, energy['total'],
                             energy['coulomb'], temp_measured))

```

¹Note that for the ideal gas the temperature is given via $1/2m\sqrt{\langle v^2 \rangle} = 3/2k_B T$, where $\langle \cdot \rangle$ denotes the ensemble average. Calculating some kind of "current temperature" via $T_{\text{cur}} = \frac{m}{3k_B} \sqrt{v^2}$ you do not obtain the temperature in the system. Only when averaging the squared velocities first one would obtain the temperature for the ideal gas. T is a fixed quantity and does not fluctuate in the canonical ensemble.

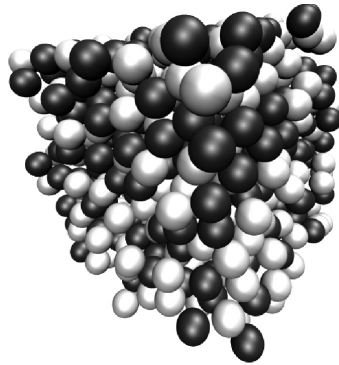


Figure 1: VMD Snapshot of the salt system

```
system.integrator.run(100)
```

4 Running the simulation

Now we can integrate the particle trajectories for a couple of time steps. Our integration loop basically looks like the equilibration:

```
print("\n--->Integration")
system.time = 0.0
for i in range(num_configs):
    energy = system.analysis.energy()
    temp_measured = energy['kinetic'] / ((3.0 / 2.0) * n_part)
    print("t={0:.1f}, E_total={1:.2f}, E_coulomb={2:.2f},
          T={3:.4f}".format(system.time, energy['total'],
                             energy['coulomb'], temp_measured))
    system.integrator.run(integ_steps_per_config)

# Internally append particle configuration
system.analysis.append()
```

Additionally, we append all particle configurations in the core with `system.analysis.append()` for a very convenient analysis later on.

5 Analysis

Now, we want to calculate the averaged radial distribution functions $g_{++}(r)$ and $g_{+-}(r)$ with the `rdf()` command from `system.analysis`:

```
print("\n--->Analysis")
# Calculate the averaged rdfs
rdf_bins = 100
r_min = 0.0
r_max = system.box_l[0]/2.0
r,rdf_00 = system.analysis.rdf(rdf_type='<rdf>',
                              type_list_a=[types["Anion"]],
                              type_list_b=[types["Anion"]],
                              r_min=r_min,
                              r_max=r_max,
                              r_bins=rdf_bins)
r,rdf_01 = system.analysis.rdf(rdf_type='<rdf>',
                              type_list_a=[types["Anion"]],
                              type_list_b=[types["Cation"]],
                              r_min=r_min, r_max=r_max, r_bins=
                              rdf_bins)
```

The shown `rdf()` commands return the radial distribution functions for equally and oppositely charged particles for specified radii and number of bins. In this case, we calculate the averaged rdf of the stored configurations, denoted by the chevrons in `rdf_type='<rdf>'`. Using `rdf_type='rdf'` would simply calculate the rdf of the current particle configuration. The results are two NumPy arrays containing the r and $g(r)$ values. Finally we write the data into a file with standard python output routines.

```
with open('rdf.data', 'w') as rdf_fp:
    for i in range(rdf_bins):
        rdf_fp.write("%1.5e %1.5e %1.5e\n" %
                     (r[i], rdf_00[i], rdf_01[i]))
print("\n--->Written rdf.data\n--->Done")
```

Fig. 2 shows the resulting radial distribution functions. In addition, the distribution for a neutral system is given, which can be obtained from our simulation script by simply not adding the P³M actor to the system. To view your own results, run `gnuplot` in a Terminal window and plot the simulation data:

```
p 'rdf.data' u 1:2 w lp t 'g(r) ++' , '' u 1:3 w lp t 'g(r) +-'
```

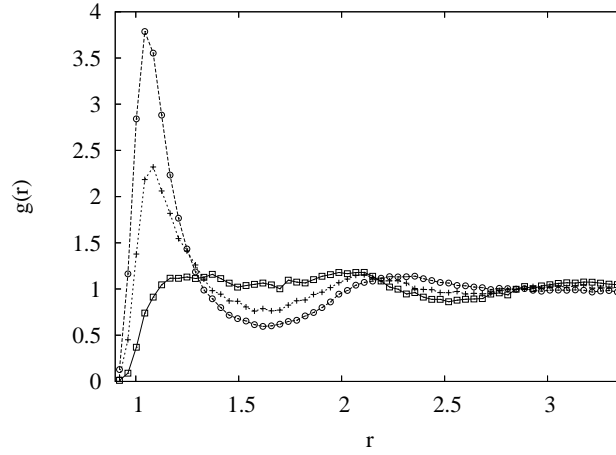


Figure 2: Radial distribution functions $g_{++}(r)$ between equal charges (rectangles) and $g_{+-}(r)$ for opposite charges (circles). The plus symbols denote $g(r)$ for an uncharged system.

6 Task - Real units

So far, the system has arbitrary units and is not connected to any real physical system. Simulate a proper NaCl crystal with the force field parameter taken from:

R. Fuentes-Azcatl and M. Barbosa, *Sodium Chloride, NaCl/ ϵ : New Force Field*, J. Phys. Chem. B, 2016, 120(9), pp 2460-2470

	q/e	$\sigma/\text{\AA}$	$(\epsilon/k_B)/K$	m/u
Na	+1	2.52	17.44	22.99
Cl	-1	3.85	192.45	35.453

Use the following system parameters:

Temperature	298 K
Friction Coeff.	10 ps ⁻¹
Density	1.5736 u \AA^{-3}
Bjerrum length(298 K)	439.2 \AA
Time step	2 fs

To make your life more easy, don't try to equilibrate randomly positioned particles, but set them up in a crystal structure close to equilibrium. If you do it right, you don't even need the Lennard-Jones equilibration. To speed things up, don't go further than 1000 particles and use a P³M accuracy of 10^{-2} . Your RDF should look like the plot in figure 3. When you get stuck, you can look at the solution script `\doc\tutorials\python\02-charged_system\nacl_units.py` (or `nacl_units_vis.py` with visualization).

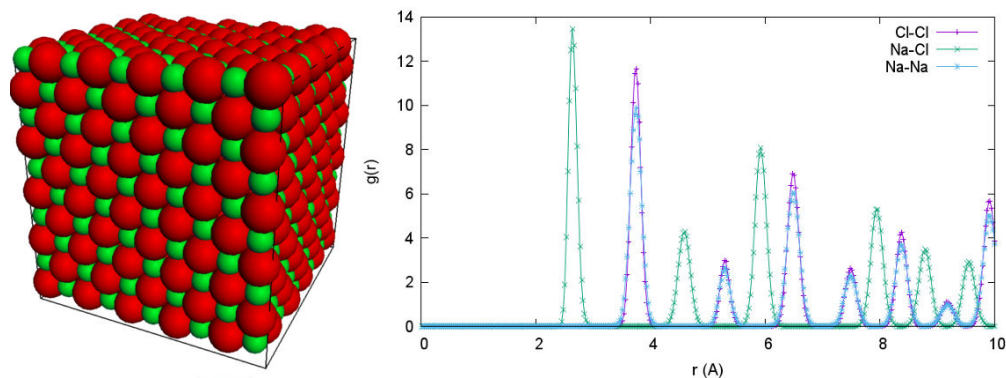


Figure 3: Snapshot and RDF of the parametrized NaCl crystal.

7 2D Electrostatics and Constraints

In this section, we use the parametrized NaCl system from the last task to simulate a molten salt in a parallel plate capacitor with and without applied electric field. We have to extend our simulation by several aspects:

Confinement

ESPResSo features a number of basic shapes like cylinders, walls or spheres to simulate confined systems. Here, we use two walls at $z = 0$ and $z = L_z$ for the parallel plate setup (L_z : Box length in z -direction).

2D-Electrostatics

ESPResSo also has a number of ways to account for the unwanted electrostatic interaction in the now non-periodic z -dimension. We use the 3D-periodic P³M algorithm in combination with the *Electrostatic Layer Correction* (ELC). ELC subtracts the forces

caused by the periodic images in the z-dimension. Another way would be to use the explicit 2D-electrostatics algorithm MMM2D, also available in ESPResSo.

Electric Field

The simple geometry of the system allows us to treat an electric field in z-direction as a homogeneous force. Note that we use inert walls here and don't take into account the dielectric contrast caused by metal electrodes.

Parameters

For our molten NaCl, we use a temperature 100 K above the melting point (1198.3 K) and an approximated density of $\rho = 1.1138 \text{ u}\text{\AA}^{-3}$ found in Janz, G. J., Thermodynamic and Transport Properties of Molten Salts: Correlation Equations for Critically Evaluated Density, Surface Tension, Electrical Conductance, and Viscosity Data, *J. Phys. Chem. Ref. Data*, 17, Suppl. 2, 1988.

Let's walk through the python script. We need additional imports for the wall shapes and the ELC algorithm:

```
from espressomd import System, assert_features, electrostatics,
    electrostatic_extensions, shapes
import numpy
```

If we target a liquid system, we should not set up the particles in a lattice, as this introduces unwanted structure in the starting configuration. We define our system size by the number of particles and the density. The system parameters lead to the following values:

```
system = System()

print("\n--->Setup system")

# System parameters
n_part = 500
n_ionpairs = n_part/2
density = 1.1138
time_step = 0.001823
temp = 1198.3
gamma = 50
#l_bjerrum = 0.885^2 * e^2/(4*pi*epsilon_0*k_B*T)
```

```

l_bjerrum = 130878.0 / temp
wall_margin = 0.5
Ez = 0

num_steps_equilibration = 3000
num_configs = 200
integ_steps_per_config = 100

```

We save the force field parameters in python dictionaries, now with parameters for the walls:

```

# Particle parameters
types      = {"Cl": 0, "Na": 1, "Electrode": 2}
numbers    = {"Cl": n_ionpairs, "Na": n_ionpairs}
charges    = {"Cl": -1.0, "Na": 1.0}
lj_sigmas  = {"Cl": 3.85, "Na": 2.52, "Electrode": 3.37}
lj_epsilons = {"Cl": 192.45, "Na": 17.44, "Electrode": 24.72}

lj_cuts    = {"Cl": 3.0 * lj_sigmas["Cl"],
              "Na": 3.0 * lj_sigmas["Na"],
              "Electrode": 3.0 * lj_sigmas["Electrode"]}

masses     = {"Cl": 35.453, "Na": 22.99, "Electrode": 12.01}

```

To finally calculate the box size, we take into account the diameter of the electrode interaction. Additionally, ELC needs a particle-free gap in the z-direction behind the wall.

```

# Setup System
box_l = (n_ionpairs * sum(masses.values()) / density)**(1. / 3.)
box_z = box_l + 2.0 * (lj_sigmas["Electrode"]+wall_margin)
box_volume = box_l * box_l * box_z
elc_gap = box_z * 0.15
system.box_l = [box_l, box_l, box_z + elc_gap]
system.periodicity = [1, 1, 1]
system.time_step = time_step
system.cell_system.skin = 0.3
system.thermostat.set_langevin(kT=temp, gamma=gamma)

```

In the next snippet, we add the walls to our system. Our constraint takes two arguments: First the `shape`, in our case a simple plane defined by its normal vector and the distance from the origin, second the `particle_type`, which is used to set up the interaction between particles and constraints.

```
# Walls
system.constraints.add(shape=Wall(dist=wall_margin,
                                normal=[0,0,1]),particle_type=types["Electrode"])
system.constraints.add(shape=Wall(dist=-(box_z-wall_margin),
                                normal=[0,0,-1]),particle_type=types["Electrode"])
```

Now we place the particles at random position without overlap with the walls:

```
# Place particles
for i in range(n_ionpairs):
    p = numpy.random.random(3)*box_l
    p[2] += lj_sigmas["Electrode"]
    system.part.add(id=len(system.part), type=types["Cl"], pos=
    p, q=charges["Cl"], mass=masses["Cl"])
for i in range(n_ionpairs):
    p = numpy.random.random(3)*box_l
    p[2] += lj_sigmas["Electrode"]
    system.part.add(id=len(system.part), type=types["Na"], pos=
    p, q=charges["Na"], mass=masses["Na"])
```

The scheme to set up the Lennard-Jones interaction is the same as before, extended by the Electrode-Ion interactions:

```
# Lennard-Jones interactions parameters

def combination_rule_epsilon(rule, eps1, eps2):
    if rule=="Lorentz":
        return (eps1*eps2)**0.5
    else:
        return ValueError("No combination rule defined")

def combination_rule_sigma(rule, sig1, sig2):
    if rule=="Berthelot":
        return (sig1+sig2)*0.5
    else:
        return ValueError("No combination rule defined")
```

```

for s in [["Cl", "Na"], ["Cl", "Cl"], ["Na", "Na"], ["Na", "
Electrode"], ["Cl", "Electrode"]]:
    lj_sig = combination_rule_sigma("Berthelot",
        lj_sigmas[s[0]], lj_sigmas[s[1]])
    lj_cut = combination_rule_sigma("Berthelot",
        lj_cuts[s[0]], lj_cuts[s[1]])
    lj_eps = combination_rule_epsilon("Lorentz",
        lj_epsilon[s[0]], lj_epsilon[s[1]])

    system.non_bonded_inter[types[s[0]], types[s[1]]].
        lennard_jones.set_params(epsilon=lj_eps,
            sigma=lj_sig, cutoff=lj_cut, shift="auto")

```

Next is the Lennard-Jones Equilibration. Here we use an alternative way to get rid of the overlap: ESPResSo features a routine for energy minimization with similar features as in the manual implementation used before. Basically it caps the forces and limits the displacement during integration.

```

energy = system.analysis.energy()
print "Before Minimization: E_total=", energy['total']
system.minimize_energy.init(f_max = 10, gamma = 10.0,
    max_steps = 1000, max_displacement= 0.2)
system.minimize_energy.minimize()
energy = system.analysis.energy()
print "After Minimization: E_total=", energy['total']

```

As described, we use P³M in combination with ELC to account for the 2D-periodicity. ELC is also added to the **actors** of the system and takes *gap size* and *maximum pairwise errors* as arguments.

```

print("\n-->Tuning Electrostatics")
p3m = electrostatics.P3M(bjerrum_length=l_bjerrum,
    accuracy=1e-2)
system.actors.add(p3m)
elc = electrostatic_extensions.ELC(gap_size = elc_gap,
    maxPWerror = 1e-3)
system.actors.add(elc)

```

For now, our electric field is zero, but we want to switch it on later. Here we run over all particles and set an external force on the charges caused by the field:

```
for p in system.part:
    p.ext_force = [0,0,Ez * p.q]
```

This is followed by our standard temperature equilibration:

```
print("\n--->Temperature Equilibration")
system.time = 0.0
for i in range(num_steps_equilibration/100):
    energy = system.analysis.energy()
    temp_measured = energy['kinetic'] / ((3.0 / 2.0) * n_part)
    print("t={0:.1f}, E_total={1:.2f}, E_coulomb={2:.2f},
          T={3:.4f}".format(system.time, energy['total'],
                              energy['coulomb'], temp_measured))
    system.integrator.run(100)
```

In the integration loop, we like to measure the density profile for both ion species along the z-direction. We use a simple histogram analysis to accumulate the density data.

```
print("\n--->Integration")
bins=100
z_dens_na = numpy.zeros(bins)
z_dens_cl = numpy.zeros(bins)
system.time = 0.0
cnt = 0

for i in range(num_configs):
    energy = system.analysis.energy()
    temp_measured = energy['kinetic'] / ((3.0 / 2.0) * n_part)
    print("t={0:.1f}, E_total={1:.2f}, E_coulomb={2:.2f},
          T={3:.4f}".format(system.time, energy['total'],
                              energy['coulomb'], temp_measured))
    system.integrator.run(integ_steps_per_config)

    for p in system.part:
        bz = int(p.pos[2]/box_z*bins)
        if p.type == types["Na"]:
            z_dens_na[bz] += 1.0
        elif p.type == types["Cl"]:
            z_dens_cl[bz] += 1.0
    cnt += 1
```

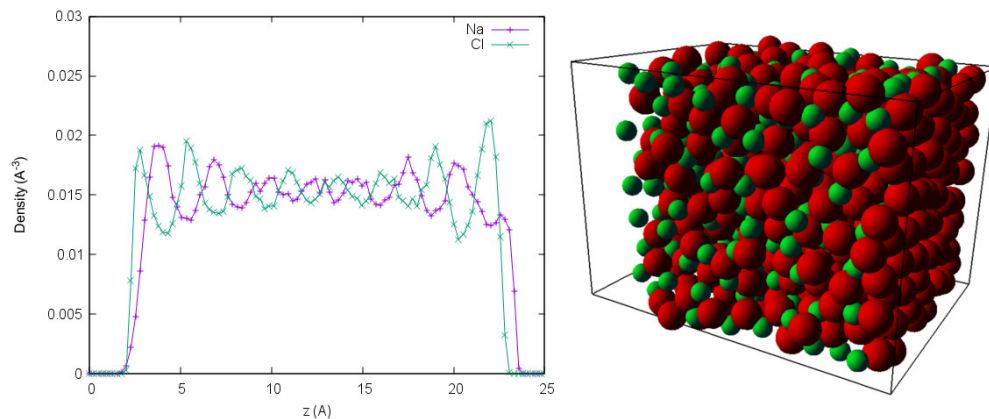


Figure 4: Snapshot and densities along the z-Axis with applied electric field for the ion species.

Finally, we calculate the average, normalize the data with the bin volume and save it to a file using NumPy's `savetxt` command.

```
print("\n--->Analysis")
#Average / Normalize with Volume
z_dens_na /= (cnt * box_volume/bins)
z_dens_cl /= (cnt * box_volume/bins)
z_values = numpy.linspace(0,box_l,num=bins)
res = numpy.column_stack((z_values,z_dens_na,z_dens_cl))
numpy.savetxt("z_density.data",res,
              header="#z rho_na(z) rho_cl(z)")
print("\n--->Written z_density.data")
print("\n--->Done")
```

The resulting density plot is very noisy due to insufficient sampling, but should show a slight depletion of the smaller Na atoms at the walls. Now try to put in an electric field that represents an applied voltage of 15 V between the walls and compare the results. The density data (see figure 4) should show strong layering at the walls, decaying towards the system center. The complete script is at `\doc\tutorials\python\02-charged_system\nacl_units_confined.py`. In the interactive script `nacl_units_confined_vis.py`, you can increase/decrease the electric field with the keys `u/j` (at your own risk).