



Radiation Detection and Measurement

Lecture 2

Chapter 1: Radiation Sources

Sources of electromagnetic radiation: Beta Decay

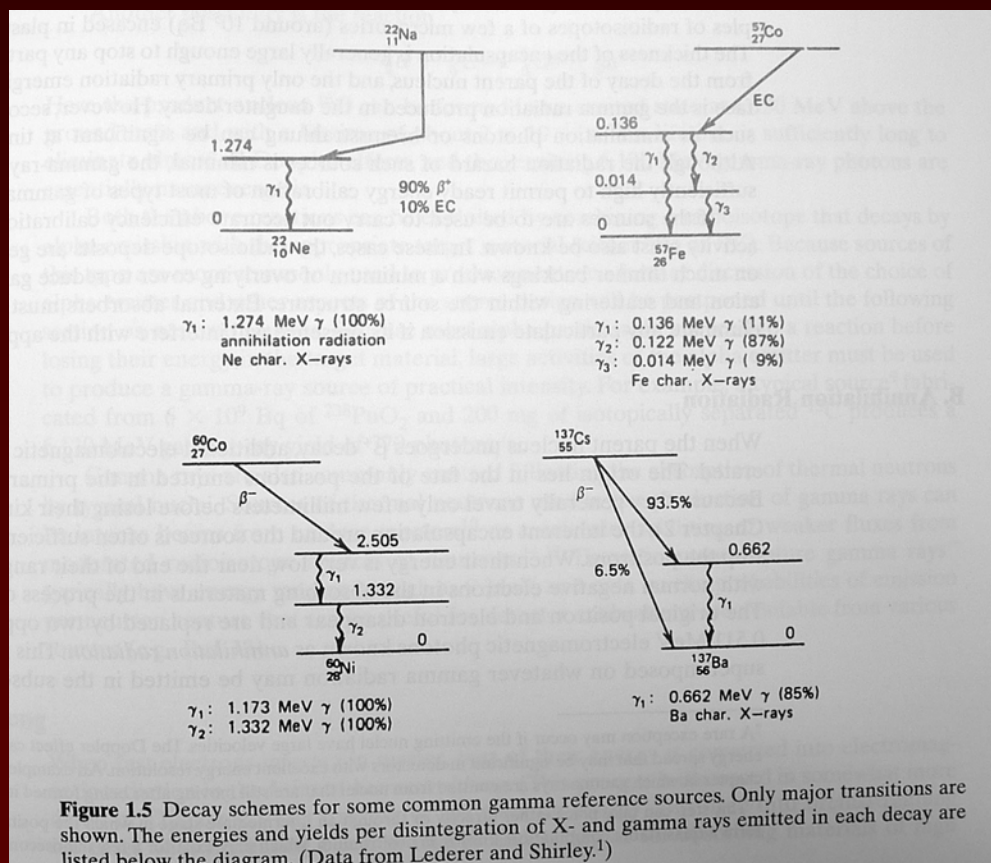
- Includes β^- , β^+ (positron) and electron capture (competes with β^+)
- γ 's are emitted after particulate radiation, where daughter nucleus is left in an excited state
- γ 's are quasi-mono energetic and have a defined energy equal to the difference between the excited and stable states
- General transition reactions

Sources of electromagnetic radiation: Beta Decay

$$\beta^- : n \rightarrow p^+ + e^-$$

$$\beta^+ : p \rightarrow n + e^+$$

$$ec : p + e^- \rightarrow n$$



Sources of electromagnetic radiation: Annihilation

- after a β^+ decay, the β^+ particle will only travel a short distance (small MFP)
- when low on T the β^+ will find an electron and the two will combine to form two γ 's with $E = 511$ keV each
- thus there is a threshold on β^+ emission of 1.2 MeV (to allow for both photons)

Sources of electromagnetic radiation: following a nuclear reaction

- i.e. ${}^4_2\alpha + {}^9_4\text{Be} \rightarrow {}^{12}_6\text{C}^* + {}^1_0n$
- the de-excitation of ${}^{12}_6\text{C}^*$ will be in the form of a γ
- also found after the absorption of thermal neutrons

Sources of electromagnetic radiation: bremsstrahlung

- produced when fast electron interact with matter
- fraction of energy converted to brems is proportional to the energy of the electron
- a monoenergetic group of electron produce a spectrum of γ 's
- found in x-ray tubes, β particles, electrons in accelerators, cosmic rays

Sources of electromagnetic radiation: bremsstrahlung

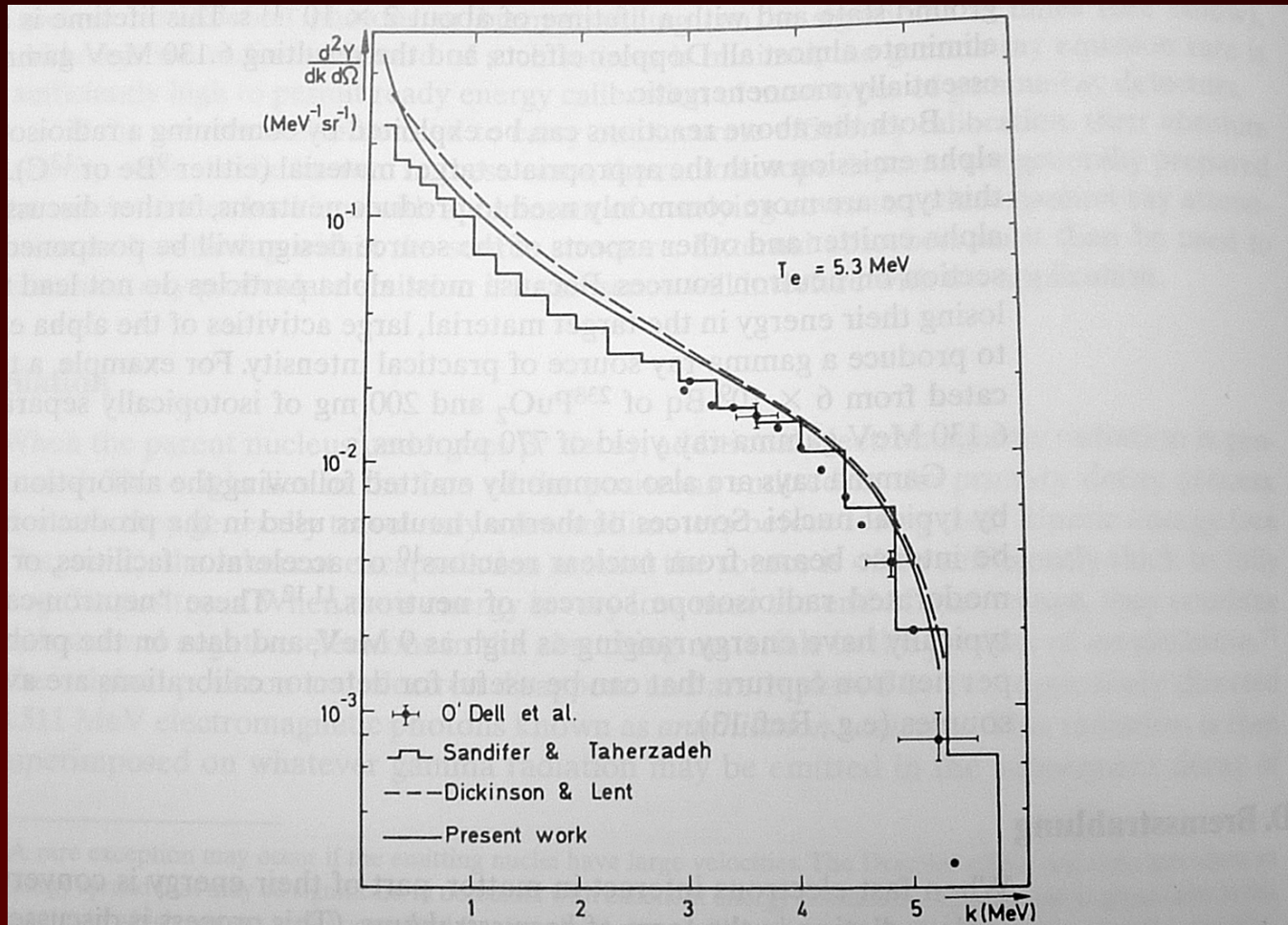


Figure 1.6 The bremsstrahlung energy spectrum emitted in the forward direction by 5.3 MeV electrons incident on a Au-W target. A 7.72 g/cm² aluminum filter also was present. (From Ferdinande et al.¹⁴)

Sources of electromagnetic radiation: bremsstrahlung

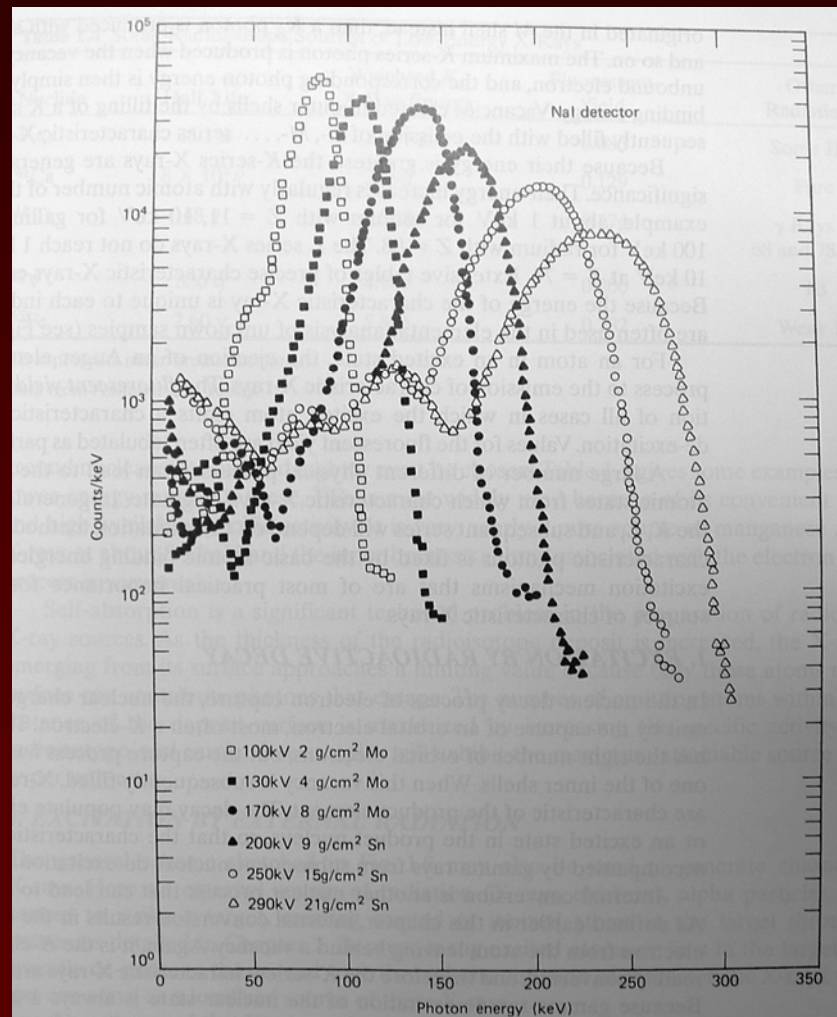


Figure 1.7 Examples of measured pulse height spectra [using a NaI(Tl) scintillator] after filtration an X-ray tube output using the indicated absorbers and tube voltages. (From Storm et al.¹⁵)

Sources of electromagnetic radiation: characteristic x-rays

- classified by the shell with a vacancy being filled in spectroscopic notation
- i.e. K_{α} is the energy emitted from a vacancy filled in the K shell from an electron in the L shell
 - K_{β} would be an electron from the m shell
 - L_{α} is a vacancy filled in the L shell from an electron in the M shell, etc.
- X_{α} will have less energy than X_{β} (higher shell, more E)
- Fluorescent yield is the fraction of all cases in which the excited atom emits a characteristic γ in its de-excitation

Characteristic x-rays: Excitation by radioactive decay

- E.C. : the nucleus captures an electron (generally a K shell) leaving a K shell vacancy
- I.C. : the nucleus emits energy which ejects an electron, leaving a vacancy
- Table 1.4 lists some low energy x-ray sources

Characteristic x-rays: Excitation by external radiation

- an external radiation strikes a target creating ionized atoms (sources; x-rays, electron, α^{2+})
- target will determine energies of characteristic radiation, low Z- soft, high Z- harder
- In all cases the energy of the sources must exceed the energy of the γ 's you are trying to produce.
- X-ray fluorescence: where x-ray gammas interact with the target (generally through photoelectric absorption) and emit a characteristic x-ray of the target
- α^{2+} particles are a convenient way to excite a target and keep the spectrum relatively free of brems. contamination, since they tend to not decay through electron modes and have little γ contributions of their own, see table 1.5 for the branching ratios of two common sources ^{210}Po & ^{244}Cm .

Characteristic x-rays: Excitation by external radiation

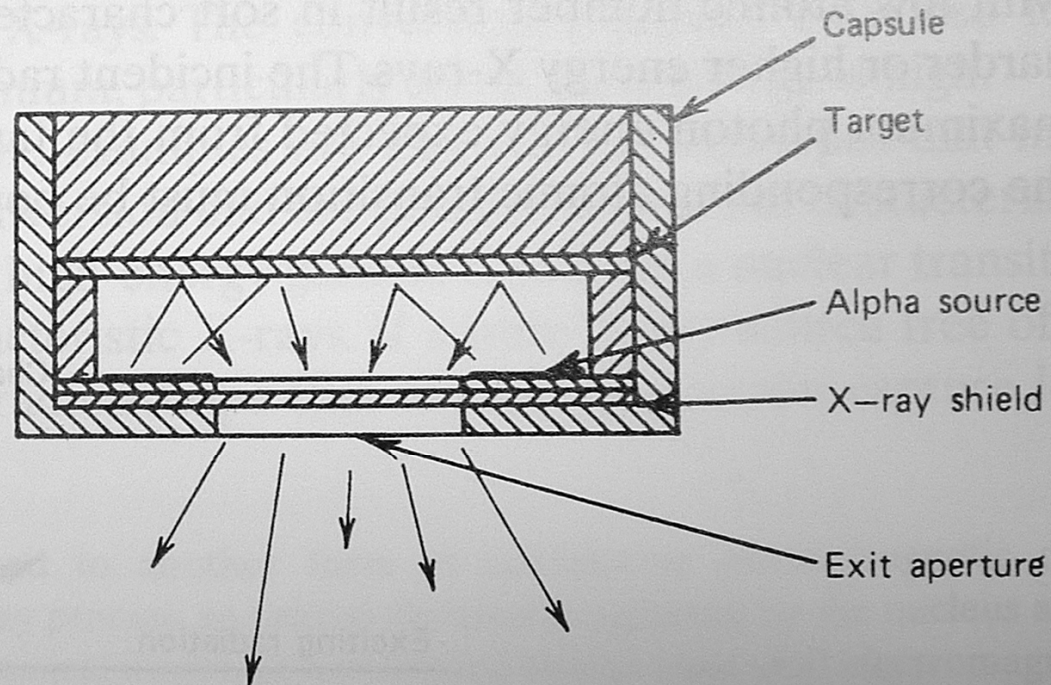


Figure 1.9 Cross-sectional view of a compact source of characteristic X-rays with particle excitation of a target. (From Amlauer and Tuohy.¹⁷)

Sources of electromagnetic radiation: synchrotron

- energetic electron are bent into circular orbits, where the change in momentum (direction changes) causes the beam to lose some energy during each cycle.
- Allows for highly mono- chromatic beams, with monochromators, with a wide energy range $\sim \text{eV}-10^4 \text{ eV}$

Neutron sources: spontaneous fission

- most common is ^{252}Cf producing ~ 3.8 neutrons on average and 9.7γ 's
- the spectrum of neutrons is governed by:

$$\frac{dN}{dE} = E^{1/2} e^{-E/T}, \text{ with } T = 1.3 \text{ MeV}$$

Neutron sources: radioisotope (α ,n) sources

- (α ,n) refers to the incident particle & resultant particle in this case $\alpha^{2+} + X \rightarrow Y + {}_0^1n$
- ${}_2^4\alpha + {}_4^9Be \rightarrow {}_6^{12}C + {}_0^1n$, this reaction produces a maximum neutron yield (irradiating Be with α) , Q value = +5.71 MeV
- All the α emitters of practical interest are actinide elements and a stable metal of the actinide can be formed by MBe_{13} , where M is the actinide element
- this metallurgical combo allows the α 's to interact with the Be nucleus without intermediate energy losses

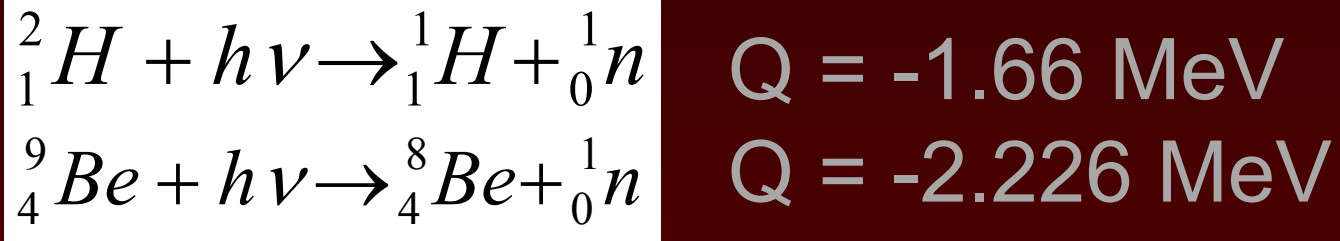
Neutron sources: radioisotope (α ,n) sources

Table 1.6 shows the characteristics of several MBe₁₃ sources

- ²³⁹Pu/Be is the most widely used but ²⁴¹Am/Be is an alternative
- when used for calibration, caution must be taken to account for impurities which may have shorter $T_{1/2}$, and α emitting daughters; this may lead to an increase in the n yield

Neutron sources: photoneutron sources

- Photoneutron emitters are sources that emit neutrons after absorbing γ 's



- Having $(-Q)$ means $h\nu \geq Q$ to allow the reaction

Neutron sources: photoneutron sources

- For γ 's with more than this minimum, the energy of the neutron can be calculated from:

$$E_n(\theta) \cong \frac{M(E_\gamma + Q)}{m + M} + \frac{E_\gamma [2mM(m + M)(E_\gamma + Q)]^{1/2}}{(m + M)^2} \cos \theta$$

Where

θ is the angle between the γ and n

E_γ is the γ energy

M is the mass of the recoil nucleus $\times c^2$

m is the mass of the neutron $\times c^2$

Neutron sources: reactions from accelerated particles



There are a number of the other charged particle reactions with (-Q's), but these require an accelerator to allow the reaction.

Aside: This is one of the only ways to get (${}^3\text{He}$) which is important for cooling applications