### **Radiation Interactions**

- 2 classes: uncharged undergoing a catastrophic event and charged particulate
  - 1) Uncharged: Neutrons ( $\lambda \cong 10^{-1}$  m)  $\Rightarrow$  heavy charged particles x-rays or  $\gamma$ -rays ( $\lambda \cong 10^{-1}$  m)  $\Rightarrow$  fast e<sup>-</sup>.
  - 2) Charged: Heavy charged particles ( $\lambda \cong 10^{-5}$  m); Fast e<sup>-</sup> ( $\lambda \cong 10^{-3}$  m).

### **Heavy Charged Particles**

#### Nature of interaction:

- Primary interaction through coulomb forces with the orbital e<sup>-</sup>'s of the absorber.
- Nuclei interactions are much less likely (Rutherford scattering).
- Results in: Excitation (raise E of e); ionization (remove an e).
- Maximum deliverable energy from a particle of mass m with kinetic energy E to electron of mass m<sub>o</sub> is: 4Em<sub>o</sub>/m. This is about 1/500 of the particle energy per nucleon so many such interactions must occur to stop the particle.
- Particles tend to follow a straight path and have a definite range beyond which the particles do not penetrate.
- May excite e<sup>-</sup>'s with enough energy to allow more interactions. These secondary e<sup>-</sup>'s are known as delta rays.
- Delta rays tend to ionize close to the primary track and have smaller ranges than the primary particle.

### Stopping Power:

- The differential energy loss per unit length in the absorber (S), also called the specific energy loss:

$$S = -\frac{dE}{dx}$$

 A charged particle will lose more energy with decreasing velocity: The relation is described by the Bethe formula:

$$-\frac{dE}{dx} = \frac{4\pi e^4 z^2}{m_o v^2} NB \; ; \quad B = Z \left[ \ln \frac{2m_o v^2}{I} - \ln \left( 1 - \frac{v^2}{c^2} \right) - \frac{v^2}{c^2} \right]$$

where v & ze are the velocity and charge of the primary particle. N & Z are the number density and atomic number of the absorber.  $m_o \& e$  are the rest mass and charge of the  $e^-$ .

*I* is the average ionization potential of the absorber and is considered an experimentally determined #.

For particles with v << c:</li>

$$B = Z \cdot \ln \left( \frac{2m_o v^2}{I} \right)$$

- B varies slowly with E, so behavior of dE/dx is in the multiplicative factor.
- dE/dx ∝  $1/v^2$  (inversely with particle energy), so the slower the particle the e<sup>-</sup>'s feel the influence of the particle and the larger the energy transfer.
- For particles of the same velocity and differing charge the  $z^2$  term dominates, showing more energy transfer for more highly charged particles (i.e.  $\alpha$ 's > protons).
- For varying the absorber, NZ dominates (outside the ln() term), this is the
  electron density, thus higher density materials and higher atomic # will have
  better stopping power.
- For large v/c (particles with velocities close to c) and light materials:  $\frac{dE}{dx} \approx 2 \frac{MeV}{g \cdot cm^2}$ , also known as "minimally ionizing particles".
- Fast electrons fall into the min. ionizing particles at  $\sim$ 1 MeV due to light mass, allowing v to approach c at low energies.
- Bethe formula fails at low energies, where the particle picks up e<sup>-</sup>'s from the absorber and becomes a neutral atom at the end of its range (attracts z e<sup>-</sup>'s).

## **Energy loss Characteristics:**

#### Bragg Curve:

- A plot of specific energy loss with distance into the absorber.
- Note the energy loss goes as 1/E (fig. 2.2) until it drops off at the end of the range, where it begins to acquire neutralizing charge.

### **Energy Straggling:**

 Due to the statistical nature of the interactions within an absorber, there is a spread of energies produced in a monoenergetic beam after passing through a given thickness of absorber - this is termed "energy straggling".

### Range:

#### Definitions of range:

- Experimental set-up of  $\alpha$  emitter and detector with a varying thickness of absorber in between.
- By varying the thickness measuring the intensity of the  $\alpha$  beam.
- We define 2 ranges from the variation in thickness.
  - 1) Mean range ( $R_m$ ) is the thickness that reduces the  $\alpha$  intensity to ½ the initial value.
  - 2) Extrapolated range (R<sub>e</sub>) is the linear extrapolation along the linear portion of the end of the transmission curve to 0.
- Fig. 2.6 shows the range in air for  $\alpha$  particles. Note that, for radiation safety purposes, you need less than 4 cm of air to stop a 4 MeV  $\alpha$  particle.
- Note in fig. 2.7 that the range for various particles is suggested to have an empirical relation of R=aE<sup>b</sup>, where b is relatively constant.
- Note also that to collect all the energy from a particle, a detector would need to have at least the thickness of the range of the particle in the detector material

### Range Straggling:

- The fluctuation in path length is called range straggling and for charges particles is ~ a few percent.
- Can be qualitatively found by taking the derivative w.r.t. x in the dE/dx plot.
- The largest change (peak) occurs at the drop off of the dE/dx curve and the range straggling can be characterized by the width of the peak.
- The wider the peak, the more of an issue the range straggling is for that particle in the absorber.

## Stopping Time

- For non-relativistic particles of mass m, and energy E:

$$v = \sqrt{\frac{2E}{m}} = c\sqrt{\frac{2E}{mc^2}} = (3.0 \times 10^8 \ m/s^2)\sqrt{\frac{2E}{(931 \ MeV/amu) \cdot m_A}}$$

where m<sub>A</sub> is the particle mass in atomic mass units (amu).

- We assume the average particle velocity (as it slows) is  $\langle v \rangle = Kv$ , where v is evaluated at the initial energy.
- Since the particle loses most of its energy at the end K > 0.5 (K=0.5 for a uniform deceleration), and we will assume 0.6:

$$T = \frac{R}{\langle v \rangle} = \frac{R}{kv} = \frac{R}{kc} \sqrt{\frac{mc^2}{2E}} = \frac{R}{k(3.0 \times 10^8 \ m/s)} \sqrt{\frac{931 \ MeV/amu}{2}} \cdot \sqrt{\frac{m_A}{E}}$$

$$T \approx 1.2 \times 10^{-7}$$
  $R \cdot \sqrt{\frac{m_A}{E}}$ ;  $T[s]$ ,  $R[m]$ ,  $m_A[amu]$ ,  $E[MeV]$ 

- T ranges from picoseconds (in solids & liquids) to nanoseconds (in gases).
- Only the fastest detectors would require attending to these stopping times.

## Energy loss in thin absorbers

- Particles that penetrate an absorber (thin) lose energy as:

$$\Delta E = -\left(\frac{dE}{dx}\right)_{avg} \Delta x$$

where  $\Delta x$  is the thickness of the absorber and  $(dE/dx)_{avg}$  is the linear stopping power averaged over the energy of the particle while in the absorber.

- One can use this equation and the linear portion of a (-dE/dx vs. E) plot to determine the thickness of an absorber from a given range.
- In some cases it may be simpler to use Range and Energy plots to determine the thickness of an absorber.
- Figure 2.12 shows increasing energy loss of protons in Si, where a
  discontinuity arises when the energy of the proton exceeds the range of the
  thickness of the detector.
- Once the energy exceeds the range, less energy is deposited in the Si (since the bragg curve shows most energy is deposited at the end of the range).

## Scaling Laws

 For compound absorbers, one can approximate the stopping power of the compound, assuming it is additive (Bragg-Kleeman rule):

$$\frac{1}{N_c} \left( \frac{dE}{dx} \right) = \sum_i W_i \frac{1}{N_i} \left( \frac{dE}{dx} \right)_i;$$

where *N* is the atomic density, dE/dx is the linear stopping power, and  $W_i$  is the atom fraction of the  $i^{th}$  component of the compound (subscript *c*).

- Expect 10-20% differences from actual values for some components.
- Range scaling:

$$R_c = \frac{m_c}{\sum_{i} n_i \cdot \left(\frac{A_i}{R_i}\right)}$$

where  $R_i$  is the range of element i,  $n_i$  is the number of atoms of element i in the molecule,  $A_i$  is the atomic weight of element i, and  $m_c$  is the molecular weight of the compound.

 If range data is not available, estimates can be based on semi-empirical formula (also called Bragg-Kleeman rule):

$$\frac{R_1}{R_0} \approx \frac{\rho_1 \sqrt{A_1}}{\rho_0 \sqrt{A_0}},$$

where  $\rho \& A$  are the density and atomic weight of the absorbers.

- The ratio of R's is only really reasonable for materials of similar atomic weight.
- By integrating the Bethe formula, it can be shown that the range of the particle of mass m and charge z is:

$$R(v) = \frac{m}{v^2} \cdot F(v)$$

where F(v) is a unique function of initial velocity for the particle.

- For particles of the same initial velocity:

$$R_a(v) = \frac{m_a z_a^2}{m_b z_b^2} \cdot R_b(v),$$

where a and b refer to different charged particles.

# **Behavior of Fission Fragments**

- Large effective charge (stripped of electrons) yield high initial specific energy loss.
- Range is  $\approx \frac{1}{2}$  that of a 5 MeV  $\alpha$  particle.
- Picking up electrons on path then decreases the specific energy loss for the particle.

### Secondary e<sup>-</sup> Emission from Surfaces

- e<sup>-</sup>'s may be given enough energy to reach the surface and escape, these are secondary e<sup>-</sup>'s (one type of secondary and different from those produced in  $\gamma$ -ray interactions these are much lower energy).
- Generally the number produced will be proportional to the energy deposited at the surface of the absorber.
- One application is the photomultiplier tube, where a fast e<sup>-</sup> hits a plate and excites 5-10 secondary e<sup>-</sup>'s.