











Figure 1. The effect of the concentration of the solution on the rate of the reaction. The reaction was carried out at 25°C in a 0.1 M HCl solution. The concentration of the solution was varied from 0.01 M to 0.1 M. The rate of the reaction was measured by the change in the absorbance of the solution at 440 nm. The results are shown in the figure. The rate of the reaction increases with the concentration of the solution.











