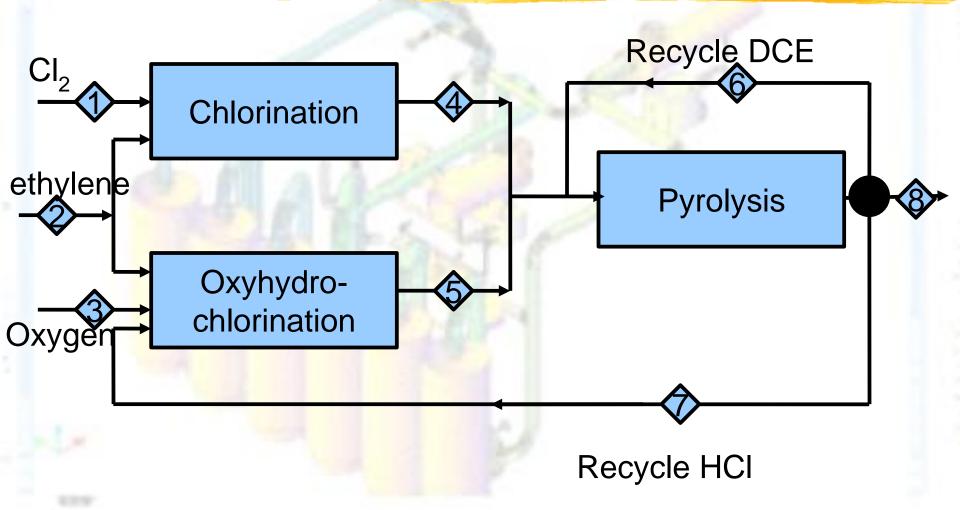
#### Flow Diagrams

- **#In general chemical engineering process are complex**
- **\*\*Model them as a set of flow diagrams**
- **\*These are 2D representation of 3D plant**
- **#Useful throughout lifetime of process** 
  - When process is conceived in lab
  - Design and construction of plant
  - Operational lifetime of plant

## Diagrams for Understanding Chemical Processes

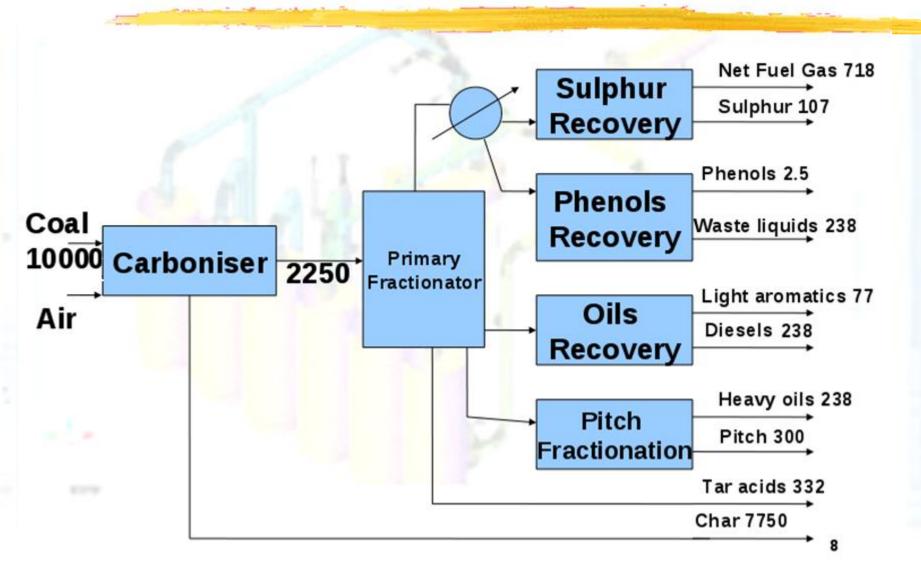
- **#Block Flow Diagrams**
- **#Process Flow Diagrams**
- **#(Piping & Instrumentation Diagram)**



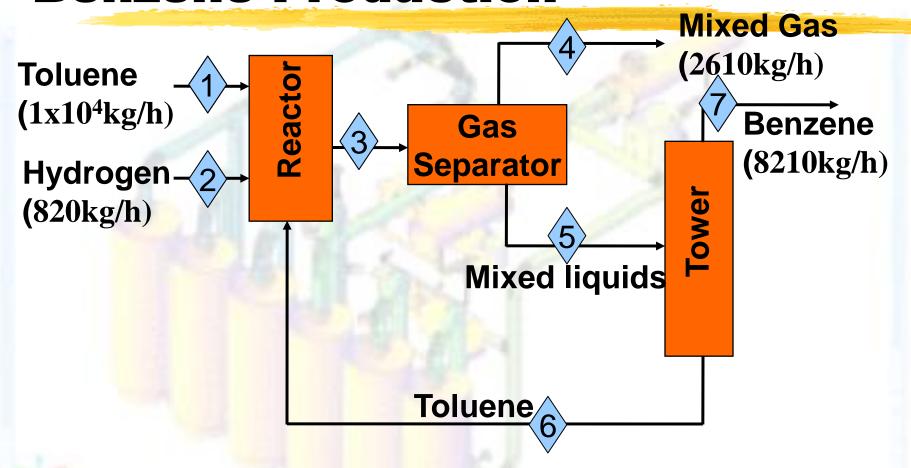
Production of Vinyl Chloride

- **Series of blocks connected with input and output flow streams**
- **\*\*Activity within blocks not important**
- **#Each block is balanced** 
  - Material
  - Energy
- **#Includes operating conditions** 
  - Temperature
  - Pressure

- **\*\*Usually used to set for the preliminary or basic processing concept without details**
- **\*The blocks represent the steps of process without details.**
- **Each block represents** a manufacturing step in the process.
- **XIt is used in survey studies to** management, research summaries and process proposals



# Example of Block Flow Diagram Benzene Production



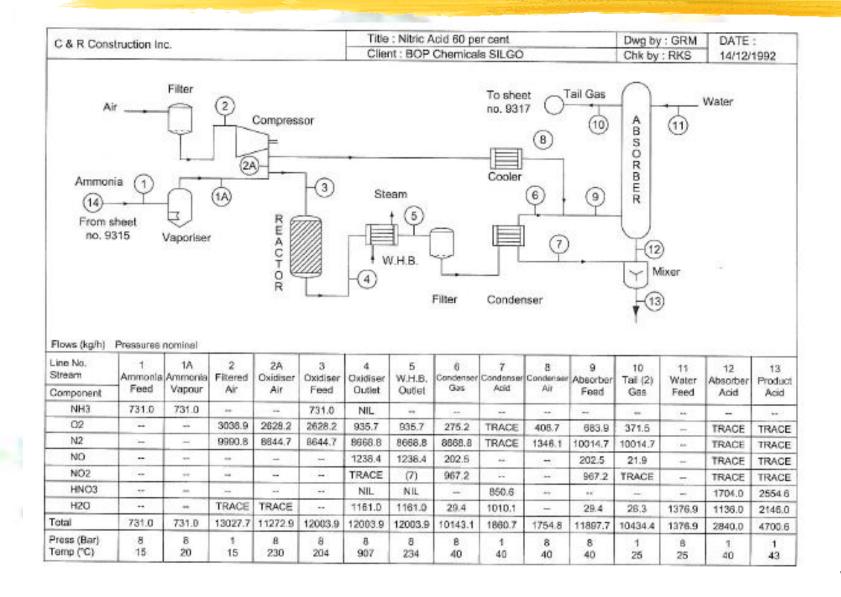
Reaction:  $C_7H_8 + H_2 \rightarrow C_6H_6 + CH_4$ 

- **#Often Microsoft Visio does the job** 
  - Contains enough icons
- **Simulink** is better
  - Based on MatLab
  - Does material and energy balance calculations for you

#### **Process Flow Diagrams**

- **#Contains much more information than a BFD**
- **\*\*No universal standard, but tend to look the same**
- **Should** include
  - △All major pieces of equipment with
    - Description
    - **<b>区Unique number**
  - △All process flow steams
    - ► Description of chemical composition (can be on flow summary table)
    - **区** Description of process conditions (can be on flow summary table)
    - **区**Identified by unique number
  - **△All utility streams**

### **Process Flow Diagrams**

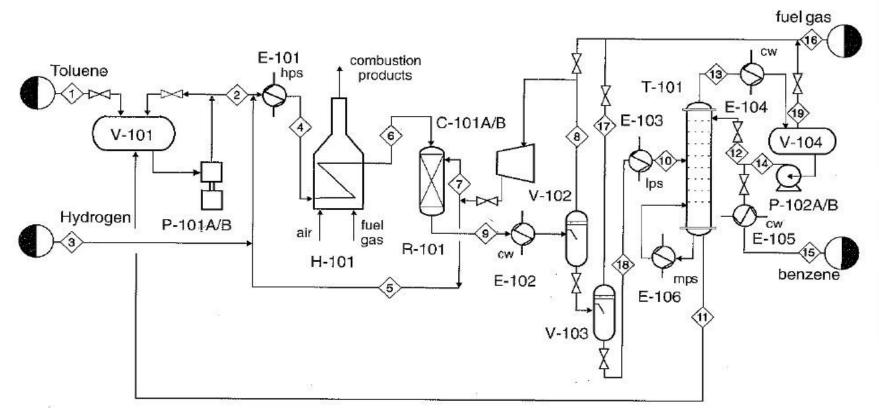


#### **Process Flow Diagrams**

- **# Used to present the heat and mass balances of the process.**
- Show all production steps, starting from raw material to final products.
- **Show the operating conditions of each production step in the process (operating conditions are: Temperature, Pressure, Flowrate, ----).**
- **Show the type and quantities of utilities that required for the process (such as: water, steam, O2, H2, ----)**
- **Shows the process equipments and their connection pipes.**
- **Gives some details for the main process equipments (such as: distillation column: its diameter, height, type of internals (trays or packing types), material of construction, thickness, ---)**
- **Show the main control system (type of instruments) required for the process.**
- **Used as data base for Piping and Instrumentation Diagram (PID)**and for equipment

## **PFD Example – Benzene Production**

V-101 P-101A/B E-103 E-106 E-101 E-102 V-102 V-103 H-101 R-101 C-101A/B V-104 P-102A/B E-105 Reactor RecycleGas Reactor HighPres. Low Pres. Tower Benzene Benzene Benzene Reflux Reflux Toluene Toluene Feed Product Storage Feed Pumps Preheater Heater Compressor Effluent Phase Sep. Phase Sep. Feed Reboiler Column Condenser Drum Cooler Cooler Heater



### **Spreadsheet accompanying PFD**

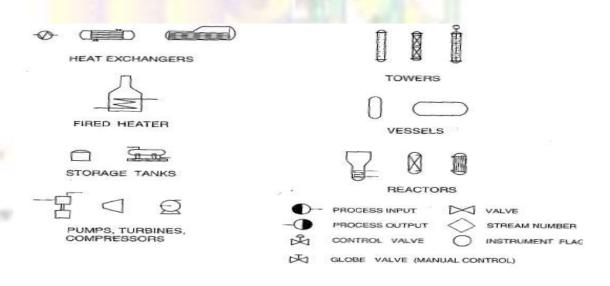
Section 1 Concentualization and Analysis of Chemical Processes

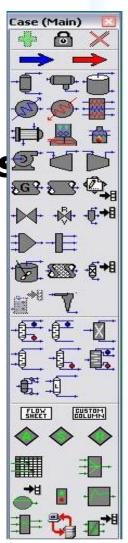
Stream Number	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Temperature (°C)	25	59	25	225	41	600	41	38	654	90	147	112	112	112	38
Pressure (bar)	1.90	25.8	25.5	25.2	25.5	25.0	25.5	23.9	24.0	2.6	2.8	3.3	2.5	3.3	2.3
Vapor Fraction	0.0	0.0	1.00	1.0	1.0	1.0	1.0	1.0	1.0	0.0	0.0	0.0	0.0	0.0	0.0
Mass Flow (tonne/h)	10.0	13.3	0.82	20.5	6.41	20.5	0.36	9.2	20.9	11.6	3.27	14.0	22.7	22.7	8.21
Mole Flow (kmol/h)	108.7	144.2	301.0	1204.4	758.8	1204.4	42.6	1100.8	1247.0	142.2	35.7	185.2	290.7	290.7	105.6
Component Mole Flow (kmol/h)															
Hydrogen	0.0	0.0	286.0	735.4	449.4	735.4	25.2	651.9	652.6	0.02	0.0	0.0	0.02	0.0	- 0.0
Methane	0.0	0.0	15.0	317.3	302.2	317.3	16.95	438.3	442.3	0.88	0.0	0.0	0.88	0.0	0.0
Benzene	0.0	1.0	0.0	7.6	6.6	7.6	0.37	9.55	116.0	106.3	1.1	184.3	289.46	289.46	105.2
Toluene	108.7	143.2	0.0	144.0	0.7	144.0	0.04	1.05	36.0	35.0	34.6	0.88	1.22	1.22	0.4

**From Turton and Bailie** 

#### PFD - Process Topology

- **# Diagram above shows process topology** for the toluene to benzene process
- **#Icons represent equipment**
- **#ISO10628** international standard for icons
- **BS1553** more typically used





#### PFD - Process Topology

- **Each piece of equipment represented by a number, e.g. P-101A/B** 

  - △1 means area 100
  - 01 means this is pump 01 in area 100
  - △A/B means there is a backup pump installed
- **# Other types of equipment** 

  - C = compressor or turbine
  - △ H = heater

#### PFD - Process Topology

- **\*Layout of units pretty much as we'd expect on the plant floor**
- Size of units on PFD should square with their footprints in reality...to some degree
- **Will mean sometimes several sheets** needed
  - Conventions for connecting sheets together

#### **PFD - Codes for Utility Streams**

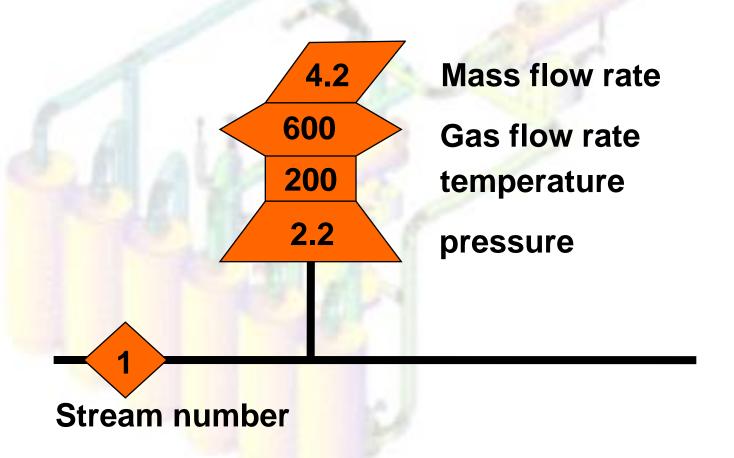
```
#Ips = low pressure steam
#mps = medium pressure steam
#hps = high pressure steam
#cw = cooling water
#rw = refrigerated water
#etc
```

### **PFD - Stream Information**

- **\*Need to note** 

  - Pressures
  - Compositions
  - **△ Flowrates**
  - **△Vapour fraction**
- **#Can be done on diagram**
- **Can be done on stream table (with streams numbered on diagram)**

#### **PFD - Stream Information**



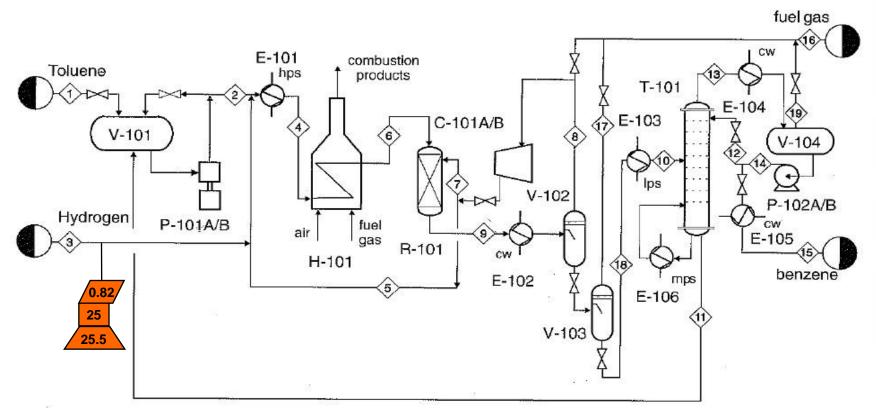
#### **PFD – Equipment Information**

- **#Information necessary for design of equipment**
- **#For example for towers gives:** 
  - Number of trays

  - Height and type of packing
  - materials

## **PFD Example – Benzene Production**

V-101 P-101A/B E-103 E-106 E-101 E-102 V-102 V-103 H-101 R-101 C-101A/B V-104 P-102A/B E-105 Reactor RecycleGas Reactor HighPres. Low Pres. Tower Benzene Benzene Benzene Reflux Reflux Toluene Toluene Feed Product Storage Feed Pumps Preheater Heater Compressor Effluent Phase Sep. Phase Sep. Feed Reboiler Column Condenser Drum Cooler Cooler Heater



					16		3	-2				60						
stream number	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
temperature (celsius)	25	59	25	225	41	600	41	38	654	90	147	112	112	112	38	38	38	38
pressure(bar)	1.9	25.8	25.5	25.2	25.5	25	25.5	23.9	24	2.6	2.8	3.3	2.5	3.3	2.3	2.5	2.8	2.9
vapour fraction	0	0	1	1	1	1	1	1	1	0	0	0	0	0	0	1	1	0
mole flow (kmole/hr)	108.7	144.2	301	1204.4	758.8	1204.4	42.6	1100.8	1247	142.2	35.7	185.2	290.7	290.7	105.6	304.2	4.06	142.2
component mole flow(	moe/hr)				ч	1	ď	J										
nydrogen	0	0	286	735.4	449.4	735.4	25.2	651.9	652.6	0.02	0	0	0.02	0	0	178	0.67	0.02
methane	0	0	15	317.3	302.2	317.3	16.95	438.3	442.3	0.88	0	0	0.88	0	0	123.05	3.1	0.88
<mark>oenzene</mark>	0	1	0	7.6	6.6	7.6	0.37	9.55	116	106.3	1.1	184.3	289.46	289.46	105.2	2.85	0.26	106.3

#### **Developing Process Flow Diagrams**

#### **#Software helps**

- FLOSHEET from Procede the best
- Cocosimulator from Cape Open OK
  - It's free
- Simulator programmes like UniSym, HYSYS that we'll meet in the lab produce stuff that look like PFD's, but they're not really

#### **Developing Process Flow Diagrams**

#### **#Five steps**

- Decide on batch or continuous
- Identify input / output structure
- ☐ Identify and define the recycle structure
- ✓ Identify and design the heat exchanger network or process energy recovery system

#### **Batch or Continuous**

- **#For us, almost always batch (but this changing)**
- **\*Reasons for batch** 
  - Smaller quantity size

  - Flexibility, making several different products
  - Unit flexibility e.g one tank for stirrer, mixer, reactor

  - Better for cleaning

#### **Batch or Continuous**

#### **Reasons for continuous**

- Better efficiency
  - Recycle energies and materials better
  - Use less utilities
- Lower maintenance and labour costs
- Less exposure to chemicals of process