

Physics 137A Lectures 1 & 2

The origins of quantum theory

- Successes of classical physics
 - Newtonian mechanics
 - Maxwellian electromagnetism
 - Boltzmann statistical mechanics

⇒ But, catastrophic breakdown when applied to blackbody radiation + atomic spectra

Blackbody Radiation

- 1792: Wedgwood remarks that all objects heated to the same temperature glow the same color
- 1800s: Improvements in spectroscopy show that solids emit continuous spectra; gases emit discrete lines
- 1859: Kirchhoff proposes model of thermal radiation emission
 - ↓ $R(\lambda, T)$ emissive power per unit area

Blackbody:

eg. cavity where light bounces many times off the walls



• Perfect absorber / Perfect emitter

• Equilibrium established between walls at temperature T and light field

Classical Observations:

A. Stefan's Law (1879)

→ experimentally finds that total radiation emitted from a glowing solid $\propto T^4$

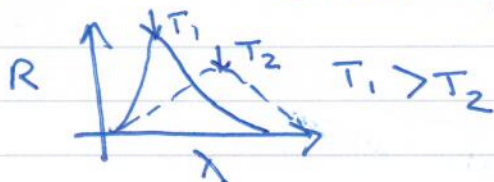


$$R(T) = \int_0^{\infty} R(\lambda, T) d\lambda = \sigma T^4$$

$$5.67 \times 10^{-8} \frac{\text{W}}{\text{m}^2 \text{K}^4}$$

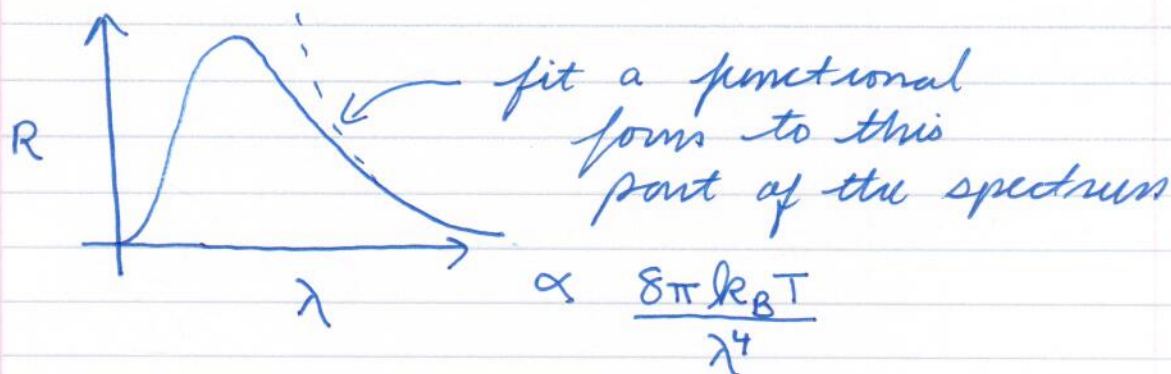
B. Wien's Law

→ From spectroscopic data $\lambda_{\text{max}} T = 2.898 \times 10^{-3} \text{ m} \cdot \text{K}$



c. Rayleigh - Jeans Law

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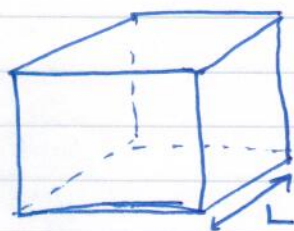


BUT... $\int R d\lambda$ would diverge!! Infinite power emitted!

UV CATASTROPHE

DERIVATION OF CLASSICAL & QUANTUM BLACKBODY RADIATION FORMULAE

- Consider cubic cavity of side L



Need to calculate standing waves inside cavity. Recall:



modes of a fixed string

Must satisfy 3-D wave equation

$$\nabla^2 \psi(\vec{r}, t) = \frac{1}{c^2} \frac{\partial^2}{\partial t^2} \psi(\vec{r}, t) \quad \begin{matrix} \text{wave} \\ \text{amplitude} \end{matrix}$$

$\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$ \rightarrow speed of light

④

Boundary conditions are that

$\psi = 0$ at cavity walls

$$\text{i.e. } \psi(x=0, y, z, t) = \psi(x=L, y, z, t) = \dots = 0$$

$$\rightarrow \text{solution } \psi(\vec{r}, t) = A(t) \sin(k_x x) \sin(k_y y) \cdot \sin(k_z z)$$

Boundary conditions require

$$k_i = \frac{n_i \pi}{L}$$

$i = x, y, z$ n is an integer

Can express as standing wave with a time varying amplitude:

$$A(t) B(x, y, z)$$

Substitute into wave equation

$$-(n_1^2 + n_2^2 + n_3^2) \frac{\pi^2}{L^2} A(t) B(x, y, z) = \frac{1}{c^2} B(x, y, z) \frac{\partial^2}{\partial t^2} A(t)$$

\rightarrow guess solution for $A(t)$

$$A(t) = A_0 \cos(\omega t) + \varphi$$

$$\omega^2 = \frac{c^2 \pi^2}{L^2} (n_x^2 + n_y^2 + n_z^2)$$

Now: We need to count how many different ways (modes) a given ω can be distributed over n_x, n_y, n_z

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Let $g(\omega) = \frac{dN(\omega)}{d\omega}$ ← # of modes

of modes per
unit frequency

Thus:

$$N(\omega) = \int_0^{\omega} g(\omega) d\omega$$

For a given ω , the possible values of n_x, n_y, n_z are bound and must obey

$$n_x^2 + n_y^2 + n_z^2 \leq \frac{\omega^2 L^2}{c^2 \pi^2}$$

- Picture a sphere of radius $\frac{\omega L}{c\pi}$
- All the points inside represent allowed modes



Consider $1/8$ of the whole sphere to account for all positive values of n_x, n_y, n_z

$$\begin{aligned} \Rightarrow N(\omega) &= \frac{1}{8} \left(\frac{4}{3} \pi \frac{\omega^3 L^3}{c^3 \pi^3} \right) \\ &= \frac{\omega^3 V}{6 c^3 \pi^2} \sim \text{volume} = L^3 \end{aligned}$$

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Can convert to linear frequency

$$\omega = 2\pi f$$

$$N(f) = \frac{8\pi^3 f^3 V}{6c^3 \pi^2}$$

$$g(f) = \frac{dN(f)}{df} = \frac{4\pi f^2 V}{c^3}$$

Each mode has 2 polarization directions

$$g(f) = \frac{8\pi f^2}{c^3} V$$

*** Up to this point, we have not invoked any assumptions from classical or quantum physics. The derivation is correct up to this point. Now, we will see differences.

CLASSICAL CALCULATION

• Each mode has energy $k_B T$ (equipartition)

→ Thus, the energy density between f and $f+df$ is $g(f)k_B T df = \frac{8\pi}{c^3} f^2 V k_B T df$

⑦

We get the energy density by dividing by V

$$\rho(f) = \frac{8\pi}{c^3} f^2 k_B T$$

To convert to λ units, use $f = c/\lambda$

$$|df| = \frac{cd\lambda}{\lambda^2}$$

Note: (-) Sign in derivative changes direction of \int

$$\rho(\lambda) = \frac{8\pi}{c^3} \frac{c^2}{\lambda^2} \frac{c}{\lambda^2} k_B T = \boxed{\frac{8\pi k_B T}{\lambda^4}}$$

* This is precisely the Rayleigh-Jeans result!

The formula works for for $k_B T \gg \omega$

but as $f \rightarrow \infty$

$f \rightarrow \infty$

UV Catastrophe

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QUANTUM CALCULATION

* Planck postulated that each oscillator cannot take on all values of energy

• Energy is gained/lost in units of hf

$$E_n = nhf$$

of quanta

Planck's
constant

$$\bar{E} = \sum_{n=0}^{\infty} nhf \frac{e^{-nhf/k_B T}}{\sum_{n=0}^{\infty} e^{-nhf/k_B T}}$$

normalized probability
of occupation

$$\text{let } x = e^{-hf/k_B T}$$

$$\bar{E} = hf \sum_{n=0}^{\infty} n \frac{x^n}{\sum_{n=0}^{\infty} x^n} = hf \frac{x}{1-x}$$

$$= hf \frac{e^{-hf/k_B T}}{1 - e^{-hf/k_B T}} = hf \frac{1}{e^{hf/k_B T} - 1}$$

$$\underline{\text{Thus:}} \quad \rho(f, T) = \frac{g(f)}{V} \bar{E} = \frac{8\pi hf^3}{c^3} \frac{1}{e^{hf/k_B T} - 1}$$

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Convert to λ $f\lambda = c$ $df = -\frac{c d\lambda}{\lambda^2}$

$$\rho(\lambda, T) = \frac{8\pi hc}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda k_B T}} - 1}$$

This is the correct formula and matches $R(\lambda, T)$ perfectly!!