Molecule ID: 25

Keyword: paracetamole

User ID: 8

Date of Creation: 2024-01-20 15:40:43

PubChem

Compound Name: Acetaminophen

Molecular Form: C8H9NO2, HOC6H4NHCOCH3

Molecular weight:151.16 g/mol CAS registration: 103-90-2

ATC code: N - Nervous system / N02 - Analgesics / N02B - Other analgesics and antipyretics / N02BE

- Anilides / N02BE01 - Paracetamol

IUPAC name: N-(4-hydroxyphenyl)acetamide

Solubility: 14 mg/mL at 25 °C **Physical description:** Solid **Melting point:** 170 °C

Decomposition: Decomposition not found

Half life: The half-life for adults is 2.5 h after an intravenous dose of 15 mg/kg. After an overdose, the

half-life can range from 4 to 8 hours depending on the severity of injury to the liver, as it heavily

metabolizes acetaminophen.

Reactivity: Reactivity not found

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Pharmacodynamics: Additionally, more studies on phenacetin in the 1940s showed that paracetamol is one of its major metabolites and thus its pharmacological effects are attributed to paracetamol [Google Scholar][2] Brodie BB, Axelrod J. The fate of acetophenetidin (phenacetin) in man and methods for the estimation of acetophenetidin and its metabolites in biological material. [Ref list][2] Brodie BB, Axelrod J. The fate of acetophenetidin (phenacetin) in man and methods for the estimation of acetophenetidin and its metabolites in biological material.

Overview of Efficacy: "paracetamole" and "Overview of Efficacy" pubmed "free article" - Google Search (function(){var b=window.addEventListener;window.addEventListener=function(a,c,d){"unload"! ==a&b;(a,c,d)};}).call(this);(function(){var

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0!==window.google.kOPI&0!==window.google.kOPI?window.google.kOPI:null};var m,n=[];function p(a){for(var b;a&&(!a.getAttribute||!(b=a.getAttribute("eid")));)a=a.parentNode;return b||m}function q(a){for(var b=null;a&&(!a.getAttribute||!(b=a.getAttribute("leid")));)a=a.parentNode;return b}function r(a){/^http:/i.test(a)&&"https:"===window.location.protocol&&(google.ml&&google.ml;(Error("a"),!1,{src:a,gl mm:1}),a="");return a} function t(a,b,c,d,k){var e="";-1===b.search("&ei;=")&&(e="&ei;="+p(d),-1===b.search("&ei;=")&&(d=q(d))&&(e+="&lei;="+d));d="";var g=-1===b.search("&cshid;=")&&"slh"!==a,f=[];f.p ush(["zx",Date.now().toString()]);h._cshid&&g;&&f.push;(["cshid",h._cshid]);c=c();null!=c&&f.push;(["opi",c.toString()]);for(c=0;c**Pharmacodynamics Drug Interaction:** Acetaminophen absorption and metabolism in an intestine/liver microphysiological system. Chem Biol Interact 2019; 299: 59-76.

Clinical Studies: Older clinical studies using paracetamol at subtherapeutic doses of ≤10 mg/kg generally show that it is less effective than non-steroidal anti-inflammatory drugs (NSAIDs). However, recent evidence shows that when used at dose of 15 mg/kg for fever and pain management, paracetamol is significantly more effective than placebo, and at least as effective as NSAIDs.

Overview of Safety: As a result of its favorable safety and tolerability record, paracetamol has long

been the most common drug for treating pain. Strikingly, recent reports questioned its therapeutic value and safety.

Marketing Experience: Copyright © 2021 Informa UK Limited, trading as Taylor & Francis Group Go to: ABSTRACT article-meta Paracetamol (acetaminophen) is undoubtedly one of the most widely used drugs worldwide. As an over-the-counter medication, paracetamol is the standard and first-line treatment for fever and acute pain and is believed to remain so for many years to come.

Benefits/Risks: Conclusions Paracetamol has been one of the most recognizable drugs, both on- and off-prescription, and it is likely to remain so in the future. As a result of the global aging, painful and disabling conditions are increasing. Paracetamol has a favorable safety profile that will be of utmost importance across all ages and especially in the elderly. Liver toxicity is a concern, but it is questionable at doses up to 4 g/day.

