

①

The splitting $C^+ : 2p_{1/2}^0 ; 2p_{3/2}^0$

low dense } excitation: collision with e^-

de-exc: emission

$$A_{10} = 2.4 \times 10^{-6} \text{ s}^{-1}$$

$$\lambda_{10} = 152,74 \mu\text{m}$$

$$k_{10} = 4.53 \times 10^{-8} T_e^{-1/2} \text{ cm}^3/\text{s}$$

$\rightarrow C^+$ with e^- rare; $n_1 (n_e, T_{\text{gas}})$

In the equilibrium we have that
(excit) (deexcit)

$$n_0 \cdot n_e \cdot k_{01} = n_1 A_{10}$$

Where $n_c = n_0 + n_1$

Be C^+ is main source of e^- $n_e = n_c$

So then $n_e (n_e - n_1) k_{01} = n_1 A_{10}$

$$n_1 = \frac{n_e^2 k_{01}}{A_{10} + n_e k_{01}}$$

we have that $\left\{ \begin{array}{l} k_{01} = \frac{g_1}{g_0} k_{10} e^{-\Delta E / kT} \text{ where } \left\{ \begin{array}{l} g_1 = 2J+1 = 4 \\ g_0 = 2J+1 = 2 \end{array} \right. \\ k_{10} = 4.53 \times 10^{-8} / \sqrt{T} \end{array} \right.$

So then we have $\left\{ \begin{array}{l} k_{01} = \frac{4}{2} \cdot 4.53 \times 10^{-8} / \sqrt{T} \cdot e^{-\Delta E / kT} \\ \Delta E = \frac{hc}{\lambda} = 7.8 \times 10^{-3} \text{ eV} \rightarrow \Delta E \cdot K = 91,21 \text{ K} \end{array} \right.$

Then replace into eq. for $n_1 \rightarrow n_1 = \frac{n_e^2}{n_e + 9,267 \cdot \sqrt{T} \cdot e^{91,21/T}}$

The energy radiated per ($\text{cm}^3 \text{ s}^{-1}$) in the $152,74 \mu\text{m}$ line is:

$$\mathcal{L} = n_1 \cdot h\nu \cdot A_{10} = 2,32 \times 10^{-26} \text{ erg / cm}^3/\text{s}$$

②

H Lyman excitation

$$\sigma = 5 \times 10^{-16} \text{ cm}^2$$

$$n_H \sim 100 \text{ cm}^{-3}$$

$$T \sim 100 \text{ K}$$

We know that $k_{ji} = \langle u \sigma_{ji} \rangle$ units? \rightarrow collision rate coef. ($\text{cm}^3 \text{ s}^{-1}$)

For 2 levels upper and lower (u) (l)

$$k_{lu} = \frac{g_u}{g_l} k_{ul} e^{-\Delta E/kT}$$

We know that $\langle u \rangle = \left(\frac{8kT}{\pi m_r} \right)^{1/2}$ \rightarrow velocity

Then we have that $k_{lu} \sim 1.025 \times 10^{-11} \text{ cm}^3/\text{s}$ $\rightarrow m_H/2$, red. mass

From Boltzmann eq. we have that $\rightarrow k_{ul} = \frac{g_l}{g_u} k_{lu} e^{\Delta E/kT} \sim 3.419 \times 10^{-11} \text{ cm}^3/\text{s}$

$\left(\begin{array}{l} g_u = 3 \\ g_l = 1 \end{array} \right) \quad \Delta E = \frac{hc}{\lambda} = 5.7 \text{ eV}$

\Rightarrow then $\rightarrow \frac{n_l}{n_e} \sim \frac{g_l}{g_u} e^{-\Delta E/kT} \sim 2.997 \sim 3$

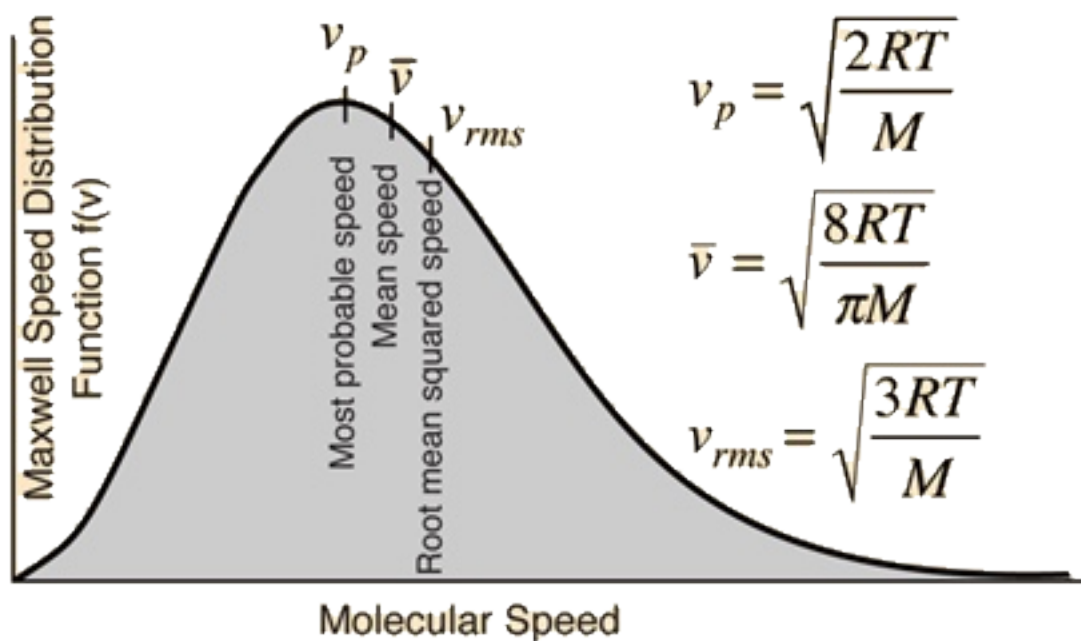
$$n(H) = n_l + n_e = 4 \cdot n_e \quad \left\{ \begin{array}{l} n_l = 75 \text{ cm}^{-3} \\ n_e = 25 \text{ cm}^{-3} \end{array} \right.$$

The rates are $\left\{ \begin{array}{l} \text{exc} \rightarrow k_{lu} \cdot n_e \sim 2.56 \times 10^{-9} \text{ s}^{-1} \\ \text{de-exc} \rightarrow k_{ul} \cdot n_l \sim 2.56 \times 10^{-9} \text{ s}^{-1} \end{array} \right.$

$\sim 25 \text{ cm}^{-3}$
 $\sim 75 \text{ cm}^{-3}$

$\langle u \rangle$ is the mean molecular speed of the Maxwellian velocity distribution for the particles (target+collider), see next page or (<https://farside.ph.utexas.edu/teaching/sm1/Thermalhtml/node87.html>)

$$f(v) = 4\pi \left[\frac{M}{2\pi RT} \right]^{\frac{3}{2}} v^2 \exp \left[\frac{-Mv^2}{2RT} \right]$$



③ $T = 8000 K$ $n_e = 10^3 \text{ cm}^{-3}$ $\beta(\text{CaI}) = 3,8 \times 10^{-12} \text{ s}^{-1}$ $\beta(\text{CaII}) = 4,0 \times 10^{-12} \text{ s}^{-1}$ $\alpha_n(\text{CaI}) = 5,4 \times 10^{-3} \text{ cm}^3/\text{s}$ $\alpha_n(\text{CaII}) = 4,28 \times 10^{-12} \text{ cm}^3/\text{s}$

Eg. conditions: $n(X^i) \beta_{i,ph} = n(X^{i+1}) n_e \sum_j \alpha_j$
 \hookrightarrow photoioniz. \hookrightarrow recomb.

① $n(\text{CaI}) \beta(\text{CaI}) = n(\text{CaII}) n_e \alpha_n(\text{CaI})$

② $n(\text{CaII}) \beta(\text{CaII}) = n(\text{CaIII}) n_e \alpha_n(\text{CaII})$

③ $n(\text{CaI}) + n(\text{CaII}) + n(\text{CaIII}) = n(\text{Ca})$

We can rewrite $\frac{n(\text{CaI})}{n(\text{Ca})} + \frac{n(\text{CaII})}{n(\text{Ca})} + \frac{n(\text{CaIII})}{n(\text{Ca})} = 1$

$\Uparrow e_I = 1 - e_{II} - e_{III}$ ④

From ① $\frac{n(\text{CaI})}{n(\text{Ca})} = \frac{n(\text{CaII})}{n(\text{Ca})} \cdot \frac{n_e \alpha_n(\text{CaI})}{\beta(\text{CaI})}$

$e_I = e_{II} \frac{n_e \alpha_n(\text{CaI})}{\beta(\text{CaI})}$ ⑤

$e_{II} = e_{III} \frac{n_e \alpha_n(\text{CaII})}{\beta(\text{CaII})}$ ⑥

Replacing ⑥ in ⑤ $1 - e_{III} = e_{II} \left(1 + \frac{n_e \alpha_n(\text{CaI})}{\beta(\text{CaI})} \right)$

$e_{II} = (1 - e_{III}) \left(1 + \frac{n_e \alpha_n(\text{CaI})}{\beta(\text{CaI})} \right)^{-1}$ ⑦

And ② into ⑥

$(1 - e_{III}) \left(1 + \frac{n_e \alpha_n(\text{CaI})}{\beta(\text{CaI})} \right)^{-1} = e_{III} \frac{n_e \alpha_n(\text{CaII})}{\beta(\text{CaII})}$

Then $(1 - e_{III}) \cdot 0,299 = e_{III} \cdot 1070 \rightarrow e_{III} = 2,794 \times 10^{-4}$

$$\text{Ans} \left\{ \begin{array}{l} \textcircled{b} \quad E_{II} = 0,299 / \\ \textcircled{c} \quad E_{I} = 0,7 / \end{array} \right.$$

LTE applies inside the stars, NOT in the dilute ISM
 Local thermodynamic equilibrium (LTE) requires all
 species, including ions, electrons neutrals, and
 photons collide with each other sufficiently
 frequently.)

④

The absorption of a photon with optical length $\tau \Rightarrow I = I_0 \cdot e^{-\tau}$

Then we have that $\tau = T_{\text{span}}^{-1} \cdot 1,38 \text{ K}$ (τ is inv. prop. to T_{span} , so $\uparrow T_{\text{span}} \Rightarrow \downarrow \tau$, more absorption)

We know that $\frac{I - I_0}{I_0} < 0,01$

\downarrow

$$e^{-\tau} - 1 < 0,01 \Rightarrow e^{-\tau} < 1,01$$

from that we have $-\tau < \ln(1,01)$

$$-1,38/T < \ln(1,01)$$

$$T > -1,38/\ln(1,01)$$

$$\boxed{T > 128,7 \text{ K}}$$