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# Can the maturity concept be used to separate the autogenous shrinkage and thermal deformation of a cement paste at early age?

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The influence of temperature on the autogenous shrinkage of cement paste has been studied using the maturity approach based on Arrhenius' law. Application of this law requires knowledge of the apparent activation energy,  $E_a$ , of cement. In this work,  $E_a$  has been determined by the "setting time method." The external volume change of cement paste was measured by hydrostatic weighing. In order to separate the thermal and autogenous deformations, the thermal dilation coefficient (TDC) was determined at both 20 and 30 °C. Investigations have shown that maturity can be used to predict autogenous shrinkage under isothermal and realistic conditions as long as temperatures remain between 10 and 40 °C. Outside of this temperature range, the calculated autogenous deformation and measured isothermal shrinkage are quite different and, as a result, autogenous shrinkage appears to be dependent on more than hydration advancement alone.

**Keywords:** Autogenous shrinkage; Temperature; Cement paste; Early age; Maturity concept

## 1. Introduction

At early age, thermal deformations and autogenous shrinkage occur simultaneously and may, if restrained, lead to the cracking of cementitious materials. These thermal deformations,  $\varepsilon_{th}$ , result from the temperature rise caused by hydration reactions and are proportional to the thermal dilation coefficient (TDC) of the cement paste:  $\varepsilon_{TH} = TDC \times \Delta T$ . Autogenous shrinkage, on the other hand, is a consequence of chemical shrinkage: hydration products are denser than the unhydrated compounds. In the plastic state, this contraction is converted into settlement. Once the solid skeleton has been formed and in the absence of an external source of water, hydration reactions continue through the consumption of capillary water. This phenomenon is called "self-desiccation." According to the Kelvin equation, this self-desiccation gradually increases the tensile stress in pore water, which then leads to a global compressive stress on the

solid skeleton. A deformation, called autogenous shrinkage,  $\varepsilon_{as}$ , thereby occurs.

For the calculation of stresses at early age, autogenous shrinkage is assumed to be dependent solely on the degree of hydration,  $\alpha$ , with  $\varepsilon_{as} = \varepsilon_{as}(\alpha)$ , and total deformation is taken as equal to the sum of thermal and autogenous deformations, i.e.,  $\varepsilon_{total} = \varepsilon_{as} + \varepsilon_{th}$  [1]. However, recent studies [2,3] have shown that the autogenous shrinkage amplitude of concrete is influenced by the temperature history. It seems that for a given degree of hydration, concretes with different temperature histories do not develop the same autogenous shrinkage. These results question the applicability of the maturity concept.

This paper focuses on the influence of temperature on the autogenous shrinkage of cement paste by use of the maturity concept [4]. The rate of chemical reactions is affected by temperature, since cement hydration is thermally activated. Moreover, the microstructure itself will undergo change when formed at different temperatures. The maturity concept serves to predict the effect of temperature on certain material properties with respect to the degree of hydration (e.g., compressive strength). This concept is based on Arrhenius' law, which requires determining the thermal

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sensitivity of hydration reactions, called the apparent activation energy,  $E_a$ .

This study deals first with the determination of apparent activation energy by using the setting time method [5,6]. Deformations of the cement paste at early age were measured for different thermal conditions, both isothermal and “realistic.” The method chosen to assess volume change was the measurement of buoyancy variations for a sample in a flexible latex mold through hydrostatic weighing, as presented in detail elsewhere in the literature [3]. With this method, we also measured the TDC, which has allowed calculating thermal deformations in the realistic cases. Lastly, using the maturity approach, we have attempted to separate thermal deformation from autogenous shrinkage.

## 2. Materials and experimental methods

### 2.1. Materials

An ordinary type I Portland cement (CPA CEM I 52.5 CP2), containing 60%  $C_3S$ , 12%  $C_2S$ , 9%  $C_3A$ , and 8%  $C_4AF$ , was used. The Blaine specific surface was 332  $m^2/kg$ . The water-to-cement (W/C) ratio was 0.25. Cement and water were mixed for 3 min in order to ensure a homogeneous mix.

### 2.2. Experimental methods

#### 2.2.1. Determination of $E_a$ by means of the setting times method

The maturity concept determines the time required at a reference temperature for the cement paste to achieve the same level of development as that under the influence of the actual time–temperature history. This time is called the “equivalent age” and can be calculated by a function based on Arrhenius’ law (Eq. (1)):

$$\tau_{\text{equ}} = \int_0^t \exp\left(\frac{E_a}{R} \left( \frac{1}{273 + \theta_{\text{ref}}} - \frac{1}{273 + \theta(t)} \right)\right) dt \quad (1)$$

where  $\tau_{\text{equ}}$  is the equivalent age at the reference temperature,  $\theta_{\text{ref}}$  (h),  $t$  is the paste age (h),  $\theta(t)$  is the paste temperature at time  $t$  ( $^{\circ}C$ ),  $\theta_{\text{ref}}$  is the reference temperature ( $^{\circ}C$ ),  $E_a$  is the apparent activation energy (J/mol), and  $R$  is the universal gas constant (J/mol/K).

The following proposed relation has been demonstrated theoretically [5]:

$$\ln\left(\frac{\Delta\alpha}{\Delta t}\right) = -\frac{E_a}{R} \left(\frac{1}{T_0}\right) + \text{constant} \quad (2)$$

where  $T_0$  is the temperature of the paste (K) under isothermal conditions and  $\Delta t$  is the time required to increase the degree of hydration by an increment  $\Delta\alpha$  at  $T_0$ . By plotting the time  $\Delta t$  for different temperatures  $T_0$ , we can

deduce  $E_a$ . Nevertheless, a way must be found to characterize the degree of hydration  $\alpha$ . The setting time method presumes that both the initial set time,  $t_{\text{is}}$ , and final set time,  $t_{\text{fs}}$ , are reached for definite levels of hydration (at a given W/C ratio). This method therefore consists of measuring these specific times ( $t_{\text{is}}$  and  $t_{\text{fs}}$ ) for several isothermal conditions and, in the subsequent step, calculating an activation energy for each period, initial setting and setting ( $t_{\text{fs}} - t_{\text{is}}$ ).

Initial set time and setting time of cement pastes were measured using the Vicat needle apparatus, which operates in accordance with European Standard EN 196-3. The samples were placed in a water bath at various temperatures (10, 20, 30, and 40  $^{\circ}C$ ). The temperature inside the samples was measured through embedded thermocouples.

#### 2.2.2. External volume change

Measurements of the external volume change of cement paste were conducted through hydrostatic weighing. Immediately after casting, a prophylactic without lubricant was filled with cement paste in order to form a spherical sample. The latex mold was closed using a thin stainless steel wire. The excess latex was then cut and the sample container was cleaned and weighed. The sample was hung under a balance and immediately immersed in the water bath at the chosen temperature. The sample mass was approximately 90 g. The bath temperature was controlled by a “cryo-thermostat” (Fig. 1) with an accuracy of 0.1  $^{\circ}C$ . The external volume change results from a change in the buoyancy force and serves to alter the weight reading on the balance. Both the bath temperature and measured mass were continuously logged on the computer at 5-min intervals for a period exceeding 24 h.

In realistic tests, the bath temperature is imposed by a cement paste sample placed in a quasi-adiabatic enclosure after casting. The temperature is measured by a thermocouple introduced into the cement paste. The weighed sample therefore undergoes the same temperature history. A detailed description of this system is provided in Ref. [3]. In order to obtain different temperature histories (Fig. 2), the mass of the sample placed in the enclosure is changed.

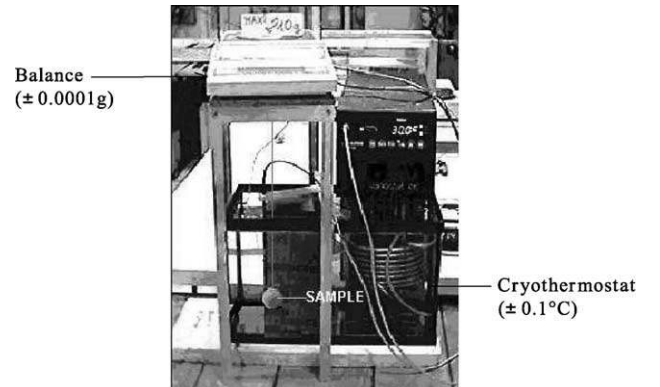


Fig. 1. Experimental system.

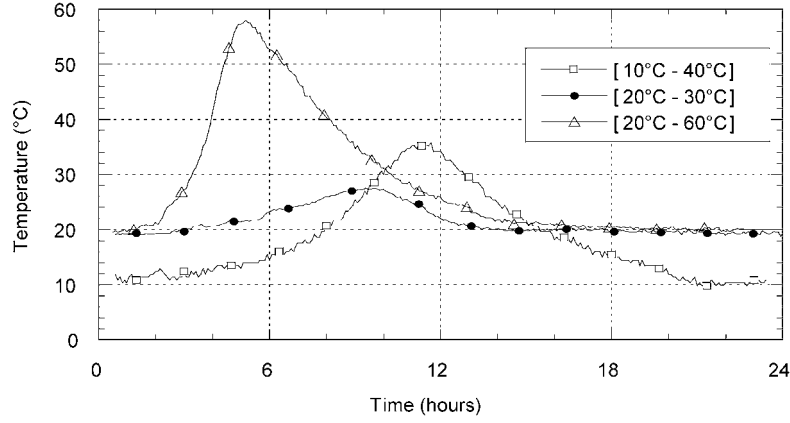


Fig. 2. Different realistic temperature histories.

### 2.3. Calculation of the deformations

The external shrinkage, expressed in cubic millimeters per gram cement, was obtained from Eq. (3) as a result of the calculation already discussed in detail [3]:

$$\frac{\Delta V}{M_c} = \left[ \left( \frac{\rho_{iw}}{\rho_w(T)} - 1 \right) - \frac{M_r}{V_i \rho_w(T)} \right] \frac{V_i}{M_c} \quad (3)$$

where  $T$  is the bath temperature (°C),  $\rho_{iw}$  is the initial density of water (g/cm<sup>3</sup>),  $\rho_w(T)$  is the density of water as a function of temperature (g/cm<sup>3</sup>),  $V_i$  is the initial sample volume (cm<sup>3</sup>),  $\Delta V$  is the volume change (cm<sup>3</sup>),  $M_r$  is the mass reading on the balance (g), and  $M_c$  the cement mass in the sample (g).

## 3. Results and discussion

### 3.1. Determination of $E_a$ by the setting times method

The results related to the initial set and setting times determined by the Vicat needle device have been listed in Table 1. Both the initial set and setting time increase with decreasing temperature. This finding has been previously reported.

Figs. 3 and 4 were drawn by assuming isothermal conditions until the initial set; a linear curve was fitted to the data points. According to Eq. (2), the computed apparent activa-

tion energies were 29 kJ/mol for the time periods until the initial set and 39 kJ/mol between the initial and final sets.

This procedure yields two activation energies, which is not unrealistic since  $E_a$  reflects more of a mathematical than a chemical tool, thereby making it possible to estimate the global thermal sensitivity of all reactions taking place in the hydrating cement paste.

### 3.2. Isothermal tests at 10, 20, 30, and 40 °C

Exposed to each temperature, the external shrinkage was determined as the mean value of three measurements and

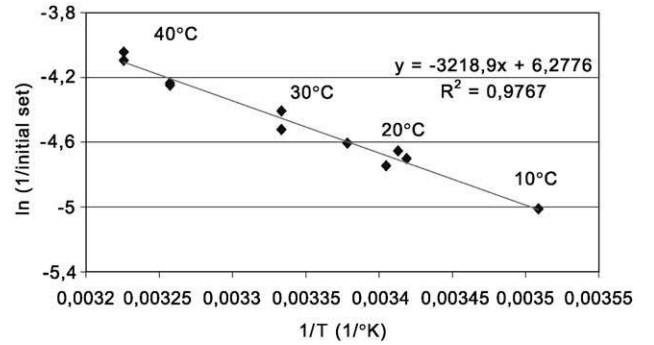


Fig. 3. Arrhenius plot for initial set times.

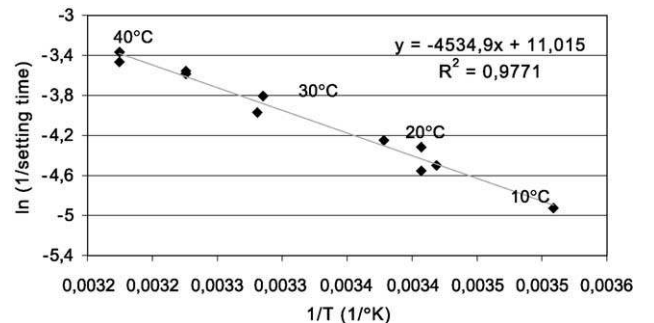


Fig. 4. Arrhenius plot for setting time.

Table 1  
Initial set and setting time at different temperatures

Average temperature (°C)	Initial set time, $t_{is}$ (min)	Setting time, $t_{fs} - t_{is}$ (min)
12	150	138
20	110	90
34	70	50
37	60	30

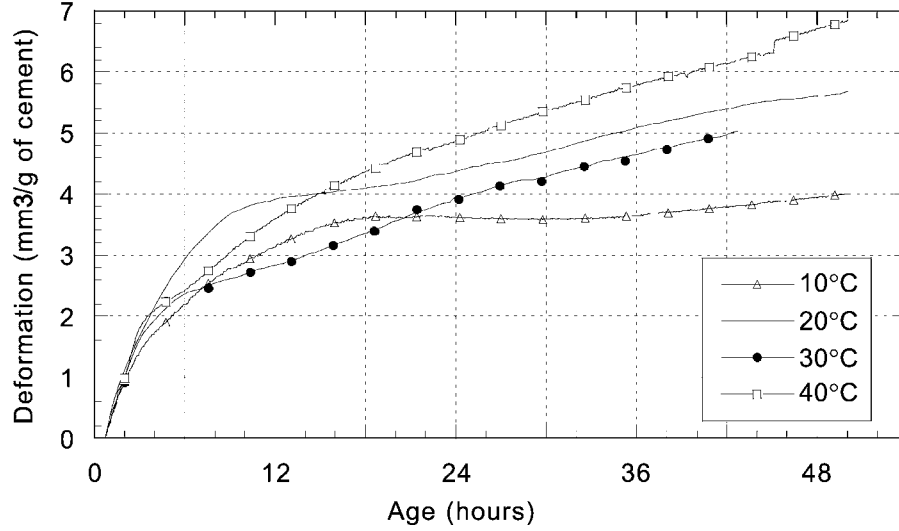


Fig. 5. Comparison of deformation for different temperatures versus paste age.

expressed in cubic millimeters per gram of cement, with a variation of less than 8%.

The external shrinkage at different temperatures versus real time has been plotted in Fig. 5. It can be observed that temperature has an unsystematic effect on the autogenous deformation of cement paste. This result, which has also been reported by Bjøntegaard [2], is not surprising since the comparison was conducted in an inconsistent manner by virtue of not incorporating the temperature effect on the rate of cement hydration, which is responsible for autogenous deformations. Comparing autogenous shrinkage curves versus real age is similar to comparing different materials; in this case, we need to apply the maturity concept.

Fig. 6 shows deformation curves versus equivalent age, which has been calculated for each temperature using Eq. (1).  $E_a$  is taken as equal to 29 kJ/mol before the initial set

and 39 kJ/mol thereafter. The reference temperature is 20 °C (as results obtained at 20 °C do not change with time transformation). It is important to note that deformations must be initialized at the same equivalent age, i.e., for the same theoretical degree of hydration. The selected equivalent time origin was 3.25 h, in correspondence with the final set time (Fig. 6). It can be seen that the four curves follow similar trends with just little scatter, which is caused by measurement accuracy levels. The maturity concept therefore leads to predicting with reasonable accuracy the autogenous shrinkage of cement paste under isothermal conditions over the 10–40 °C range.

### 3.3. External volume change in realistic cases

By considering the behavior of material to be isotropic, linear deformation represents one-third of the volumetric

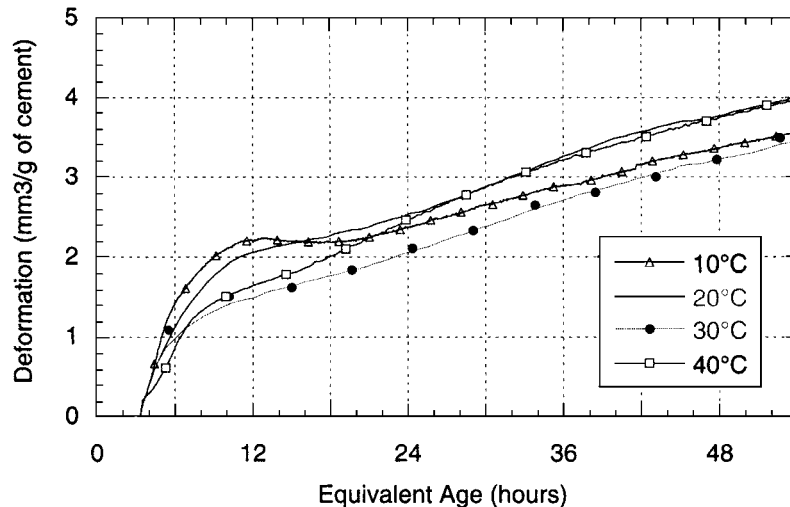


Fig. 6. Autogenous deformation for different temperatures versus maturity, starting after 3.25 h.

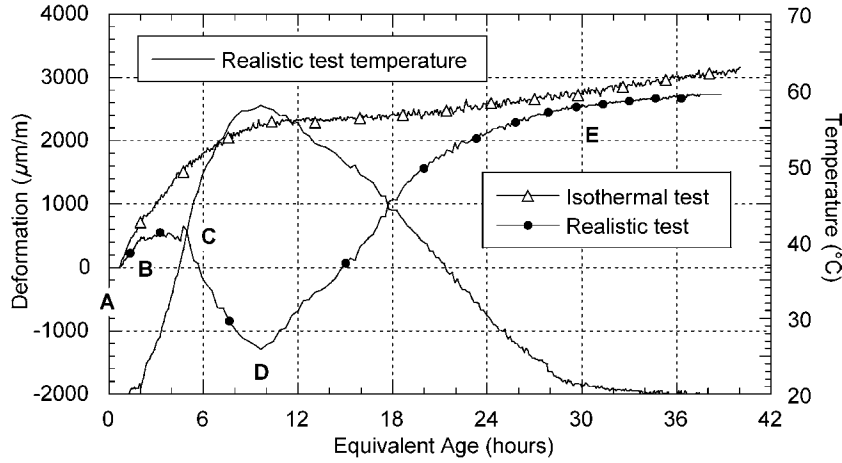


Fig. 7. Comparison of the isothermal test at 20 °C and the 20–60 °C realistic test.

deformation calculated in Eq. (3). In this part, deformations are expressed in micrometers per meter.

### 3.3.1. Deformation kinetics

Three realistic temperature tests with initial temperatures of 10 and 20 °C were performed. A typical curve of a realistic temperature history (20–60 °C) is illustrated in Fig. 7.

The realistic deformation curve can be divided into four phases: AB, BC, CD, and DE. In the first phase, it is observed that the isothermal and realistic deformations display the same shape since the temperature is constant in both tests. When temperature increases in the realistic test (B), the curves diverge; during BC, no “realistic” deformation is recorded. It seems that thermal deformations compensate for autogenous shrinkage. In the case of the “realistic” test, thermal expansion occurs until the temperature peak (CD), while the rate of autogenous shrinkage at 20 °C is decreasing. During the cooling period (DE), a large deformation caused by both thermal contraction and autogenous shrinkage can be observed. Moreover, when the temperature returned to 20 °C, the magnitude of the “realistic” deformation corresponds to the autogenous shrinkage recorded under isothermal conditions. It can thereby be

stated that the total measured deformation under realistic conditions is equal to the sum of the measured deformation in the isothermal test and a thermal component related to temperature history. In order to separate thermal and autogenous deformations, we have determined the evolution of the TDC with respect to time.

### 3.4. TDC measurements

The TDC was measured with the method of spontaneous heating by using hydrostatic weighing [3]. Tests were carried out under two isothermal conditions: 20 and 30 °C. Every hour, the bath temperature was increased by 4 °C within a period of about 7 min and then immediately decreased to the isothermal temperature. Fig. 8 presents the evolution in weight given by the balance without any water density correction. Note that the 4 °C peak effect is different at very early age and after 2 h, which reveals an evolution in the TDC.

By using this method for the TDC calculation, it was assumed that the temperature rise was sufficiently fast for the deformation measured during heating to be consid-

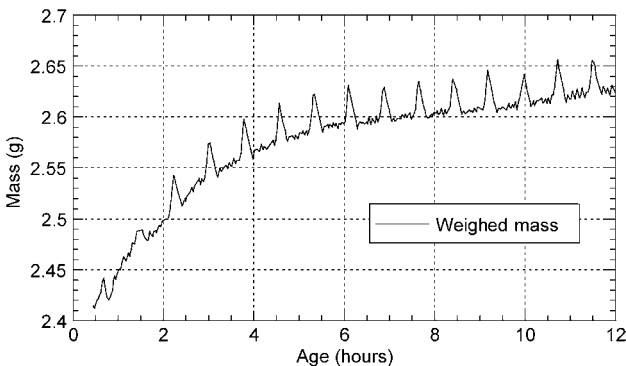


Fig. 8. Evolution in mass readings during the TDC measurement.

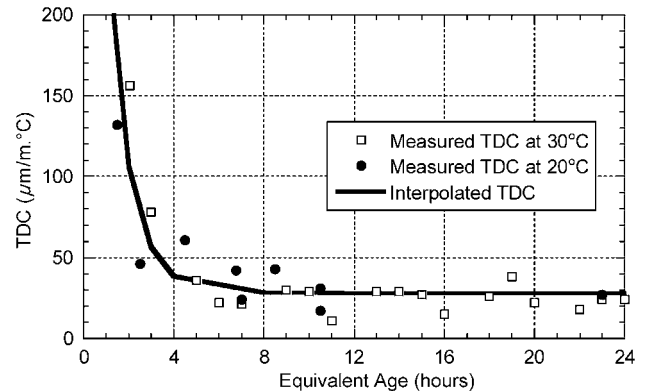


Fig. 9. TDC evolution.

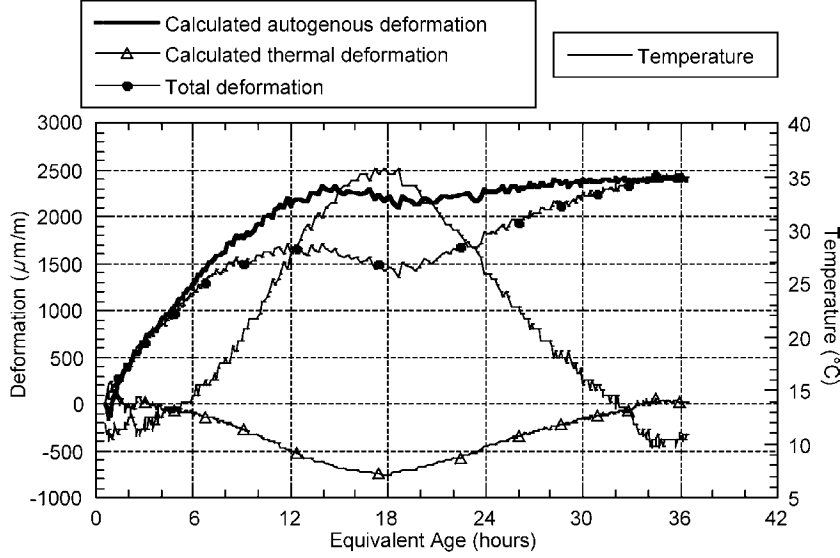


Fig. 10. Separation of thermal and autogenous deformations for the 10–40 °C temperature history.

ered only from a thermal origin. In spite of the dispersion of experimental points (Fig. 9), TDC seems to depend solely on equivalent age (with  $\theta_{ref}=20$  °C). A TDC curve may be interpolated from our results in accordance with Eq. (4) below:

$$TDC(t_{equ}) = 137 * \exp(1.44 - t_{equ}) + 28. \quad (4)$$

It is observed in Fig. 9 that the TDC decreases sharply during the first few hours after casting and then remains constant from 8 h onwards. Its value is high (28  $\mu\text{m}/\text{m}^\circ\text{C}$ ), yet in agreement with other cement paste TDC reported by other researchers [7]. Miao et al. [8] reported a TDC of about 24  $\mu\text{m}/\text{m}^\circ\text{C}$  at 7 h after casting for a concrete with a W/C ratio of 0.28. It should be noted that the TDC value is very sensitive to the experimental measurement procedure used.

### 3.5. Separation of thermal and autogenous deformations

Interpolated TDC allows determining the thermal deformation for any temperature history. We can also deduce autogenous deformation from total deformation by assuming that total deformation is the sum of the two other components. Moreover, deformation separation has to be performed by application of the maturity concept. Fig. 10 presents an example of separation in the case of the 10–40 °C realistic test.

### 3.6. Maturity concept approach results and discussion

Figs. 11 and 12 compare isothermal shrinkage at 10 and 20 °C, respectively, and calculate autogenous deformation. It can be observed that although the determined TDC is rather inaccurate, both the calculated and measured defor-

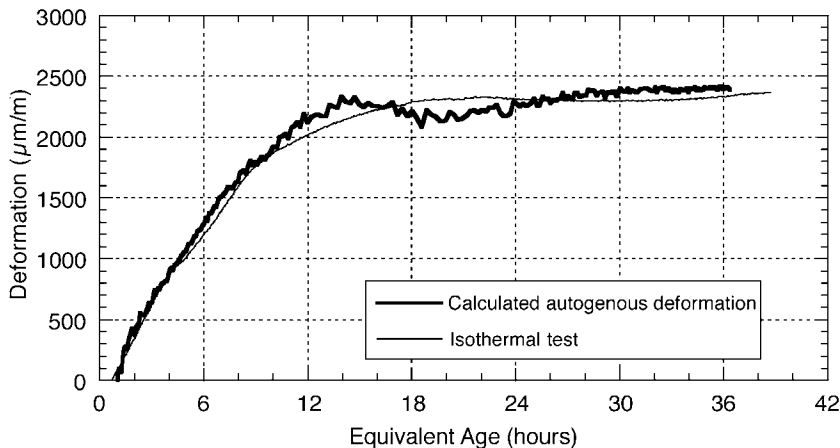


Fig. 11. Comparison of the measured autogenous shrinkage at 10 °C and the calculated autogenous deformation for the 10–40 °C test.

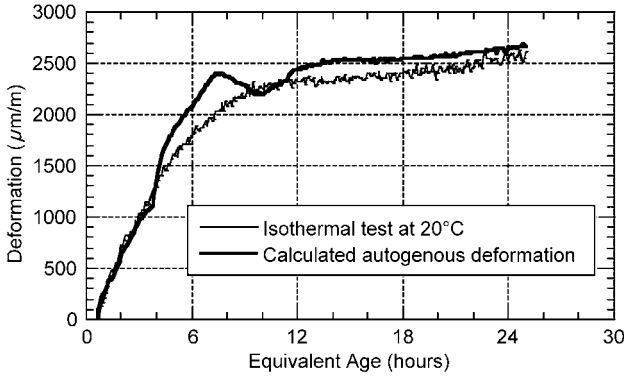


Fig. 12. Comparison of the measured autogenous shrinkage at 20 °C and the calculated autogenous deformation for the 20–30 °C test.

mations are nearly identical. The maturity approach provides a good estimate of the autogenous shrinkage of cement paste.

However, the method used to separate thermal and autogenous deformations yields a different result for the 20–60 °C test (Fig. 13). In this instance, the final amplitude of the calculated autogenous deformation is about 700  $\mu\text{m/m}$  higher than that measured in the isothermal test at 20 °C. Moreover, the calculated deformation displays a significant dilation peak. In other words, the usually expected kinetics of autogenous shrinkage is not present herein.

In Fig. 13, it is observed that the final amplitude of the calculated autogenous deformation is nearly equal to the amplitude of isothermal shrinkage plus the final amplitude of thermal deformation. In fact, thermal deformation is not reversible because of TDC evolution: during the heating period, TDC is higher than the cooling phase and, consequently, temperature variations induce dilation. In the 10–40 °C test (Fig. 11), no dilation can be detected; paste heating and cooling must therefore occur at a time when TDC is constant.

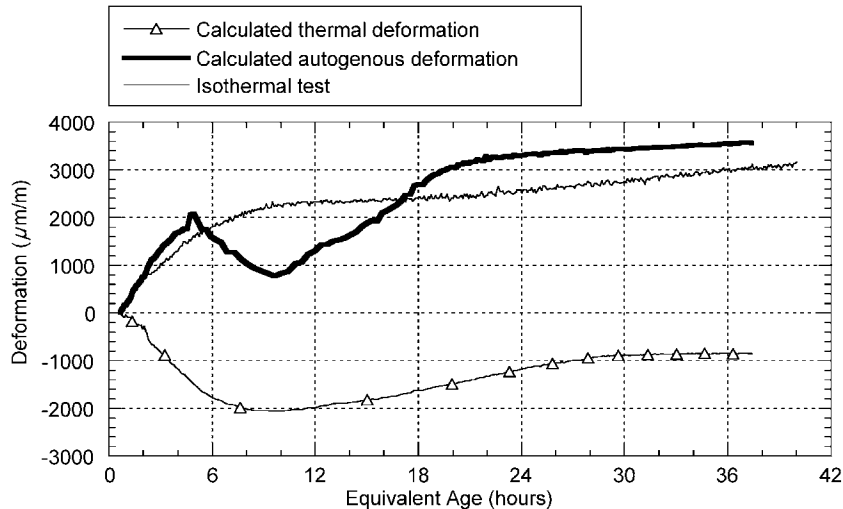


Fig. 13. Comparison of the measured autogenous shrinkage at 20 °C and the calculated autogenous deformation for the 20–60 °C test.

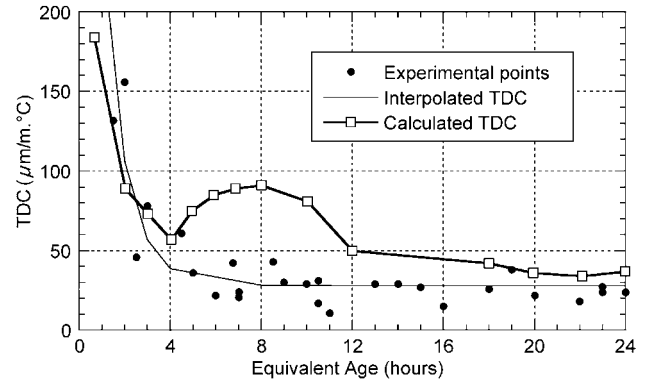


Fig. 14. Comparison of Laplante method TDC and experimental TDC.

At this juncture, two assumptions can be made. Firstly, the maturity concept cannot predict autogenous deformation evolution and amplitude when the thermal history lies beyond a temperature field between 10 and 40 °C. The maturity concept is a mathematical tool that may be too simplistic for describing microstructural changes. For example, the hydration product form, and therefore chemical shrinkage, may be different at 20 °C than at 60 °C.

The validity of the determined TDC can also be questioned. Indeed, this TDC does not allow entirely separating thermal and autogenous deformations in the 20–60 °C test. For example, dilation observed in the calculated autogenous deformation may be from a thermal origin. Afterwards, we determined a theoretical TDC that allows obtaining the same autogenous shrinkage by means of calculation as by isothermal measurement. This method has been developed by Laplante [1]. Thermal deformation is presumed to be obtained by subtracting the measured isothermal deformation from the measured total deformation. TDC is equal to the ratio of this deformation to the temperature difference between isothermal and realistic tests.



Fig. 14 compares experimental TDC from isothermal tests and TDC calculated using the Laplante method. It is shown that the calculated coefficient does not present a typical evolution. TDC may be dependent on thermal history when the temperature of realistic history exceeds a certain limit. This result reveals the need for further work in TDC measurement, for example, in the realistic case.

#### 4. Conclusions and subsequent work

Based on the assumption that the maturity concept is applicable throughout the hydration process, the following conclusions can be drawn:

- Between 10 and 40 °C, the maturity concept allows predicting the isothermal autogenous shrinkage of cement paste measured using the volumetric method: autogenous shrinkage thus seems to be solely dependent on the degree of hydration.
- In realistic tests, total deformation is shown in an initial analysis to be the sum of autogenous and thermal components. Separation of thermal and autogenous deformations is therefore made possible by use of a valuable TDC.
- The TDC is linked to the degree of hydration only between 20 and 30 °C.
- The maturity concept allows estimating the autogenous deformation amplitude for thermal conditions between 10 and 40 °C by measuring the isothermal shrinkage.
- For the 20–60 °C realistic test, both the calculated autogenous deformation and isothermal measured shrinkage differ. Autogenous deformation therefore seems to be affected

by thermal history above 40 °C. Nonetheless, a process must be developed in the future to accurately measure TDC under different thermal conditions: both isothermal and realistic. These results also need to be verified by means of linear tests.

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