

Molecular Dynamics Simulation

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1 Introduction

Molecular dynamics simulation mimics the natural motions of particles in nature. The energy defined by the potential function allows the atoms to move away from and close to neighbouring atoms. And intuitively in different states particles move in a very distinctive fashion, and it is interesting to find out in what ways they differ.

The motivation of this project is to understand what atoms do in real life, assuming a given potential energy function. In order to achieve this goal we run a simulation of Argon gas atoms, and analyse the physical properties of the box in three different states, by varying the temperature and density of the box. Two different initialisation methods were used, where we focus mainly on the default initialisation with lattice configuration, additional investigation on the initialisation was also discussed and demonstrated in the very last section.

This report is laid out as follows: in section 2 we introduce the initialisation techniques and potential function used for the particle-particle interaction; in section 3 we introduce briefly the details of implementing the periodic boundary conditions. In section 4 we introduce the Verlet algorithm which is used particularly in molecular dynamics simulation, and the updating functions. In section 5 we describe the physical properties of our three different simulations, and our expectations. In the last but not least section, we discuss the implications of the results, and present our final results. In the last section, we also presented our extra investigation using a different initialisation method.

2 Initialisation

2.1 Global constants

There are a lot of properties at play during the whole course of the simulation, and we certainly want to keep record of them. A reliable way is to put all the physical constants as globals such that they do not get updated during any step. Below lists all the global constants in this simulation.

Table 1: The physical constants used in this simulation.

Constants	$\sigma[\text{m}]$	$k_B[\text{m}^2 \text{kg s}^{-2} \text{K}^{-1}]$	$T_0[\text{K}]$	$\epsilon[\text{m}^2 \text{kg s}^{-2}]$	$m[\text{kg}]$
Values	3.405×10^{-10}	1.38×10^{23}	100	$k_B T_0$	6.6×10^{-26}

2.2 Potential

We assume a Lennard-Jones potential for the two point interaction, such that when the particles get too close to each other, they feel repulsive force, and when they are too far apart, they attract each other. The Lennard-Jones potential takes the following form

$$U(r) = 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right] \quad (1)$$

Below is an illustration of the Lennard-Jones potential.

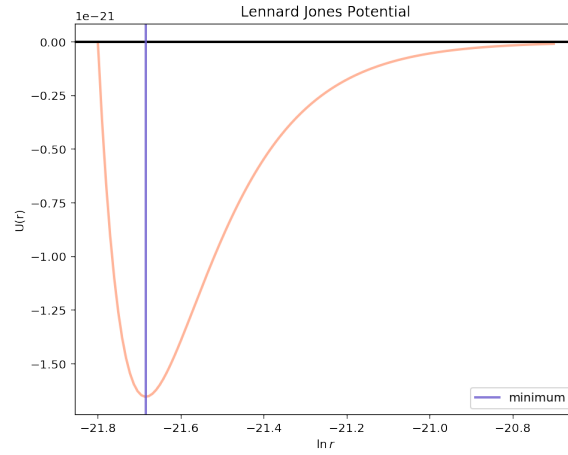


Figure 1: An illustration of the Lennard-Jones potential.

From the above plot it can be seen, the minimum of the potential is at $2^{1/6}\sigma \sim 1.12\sigma$, suggesting that for a pair of particles with such a separation, the force is neither attracting nor repulsing, which is a fixed point. Hence we can learn that to initialise the position of the particles we should avoid such a point.

The L-J potential is a relatively good approximation. Due to its simplicity, it is often used to describe the properties of gases and to model dispersion and overlap interactions in molecular models. It is especially accurate for noble gas atoms and is a good approximation at long and short distances for neutral atoms and molecules.

2.3 Position

Since we have a Lennard-Jones potential for calculating the two-point interactions, the separation between the pairs of particles can not exceed the natural unit σ , below which the potential becomes positive. Hence the positions of the particles are initialised using the lattice cell, for the fact that Argon forms face-centered lattice, where each unit cell consists of eight particles on the vertices and 6 particles on the faces. Below is an illustration of how this unit cell is being copied in space.

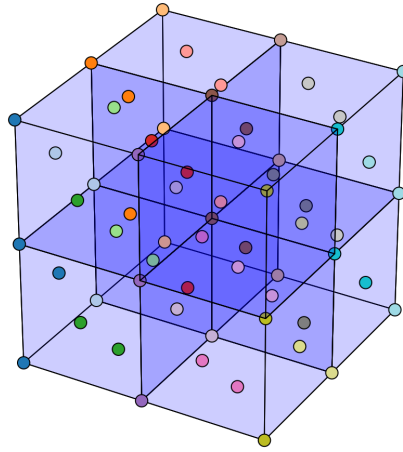


Figure 2: An illustration of a simulation box with a number of cells equals 2^3 , so that there are in total 63 particles inside the box colour coded randomly. Details see 'plot.lattice.py'.

In this simulation we initialise the simulation box with a number of cells of 3^3 , meaning that there are in total 108 particles inside the box. Other ways of initialising the position of the particles have also been

explored. In addition to the lattice generation technique to avoid the positive potential disaster, we can also use a somewhat more random way of generating the particles, such that we perturb the particles slightly from there lattice configuration, and rescale the positions such that, now the two point distances do not go below 1σ . Details of this generation routine please see the listing of the code in the appendix.

2.4 Velocity

The Maxwell-Boltzmann distribution corresponds to the most likely rate distribution in a collision-based system composed of a large number of non-interacting particles, in which quantum effects can be ignored. Since the molecular interactions in gas are generally quite small, the Maxwell-Boltzmann distribution provides a very good approximation of the state of the gas.

Assuming the system of interest contains a large number of particles, the fraction of the particles within a infinitesimal element of three- dimensional velocity space d^3v , centered on a velocity vector of magnitude v , is $f(v)d^3v$, in which

$$f(v)d^3v = \left(\frac{m}{2\pi k_B T}\right)^{\frac{3}{2}} e^{-\frac{mv^2}{2k_B T}} d^3v \quad (2)$$

where m is the particle mass, k_B is Boltzmann constant and T is thermodynamic temperature. For velocity in a standard Cartesian coordinate system, the element of velocity space can be written as $d^3v = dv_x dv_y dv_z$, here $f(v)$ is given as a probability distribution function, normalized so that $\int f(v)d^3v$ over all velocities equals one.

The mean speed $\langle v \rangle$ and root-mean speed $\sqrt{\langle v^2 \rangle}$, which is the expected value of the speed distribution and the second-order raw moment of the speed distribution, can be obtained from properties of the Maxwell distribution. For particles with same mass, the most probable speed v_p is the speed most likely to be possessed by any particles in system and corresponds to the maximum value of $f(v)$

$$\frac{df(v)}{dv} = 0 \quad \Rightarrow \quad v_p = \sqrt{\frac{2k_B T}{m}} \quad (3)$$

and the mean speed reads

$$\langle v \rangle = \int v f(v) dv = \sqrt{\frac{8k_B T}{m}} = \frac{2}{\sqrt{\pi}} v_p \quad (4)$$

the root mean square speed corresponding to the speed of a particle with median kinetic energy,

$$v_{rms} = \sqrt{\langle v^2 \rangle} = \left(\int v^2 f(v) dv \right)^{\frac{1}{2}} = \sqrt{\frac{3k_B T}{m}} = \sqrt{\frac{3}{2}} v_p \quad (5)$$

Meanwhile, distribution for the speed follows immediately from the distribution of the velocity vector, note that the speed is

$$v = \sqrt{v_x^2 + v_y^2 + v_z^2} \quad (6)$$

and the volume element in spherical coordinated

$$dv_x dv_y dv_z = v^2 \sin \theta dv d\theta d\phi = v^2 dv d\Omega \quad (7)$$

where θ and ϕ are the spherical coordinate angles of the velocity vector. Integration of the probability density function of the velocity over the solid angles $d\Omega$ yields an additional factor of 4π . The speed distribution with substitution of the speed for the sum of the squares of the vector components:

$$f(v) = \left(\frac{2}{\pi}\right)^{\frac{1}{2}} \left(\frac{m}{k_B T}\right)^{\frac{3}{2}} v^2 e^{-\frac{mv^2}{2k_B T}} \quad (8)$$

By using a 3d multivariate Gaussian located at $(0,0,0)$ and with only diagonal elements in the covariance matrix Σ , we can have our velocities distributed according to the rule described above. For details of the velocity generation routine please refer 'Argon_simulation.py'.

3 Periodic Boundary Condition

During the whole simulation, periodic boundary condition is applied. This is because we want our box volume to be constant, and the number of particles to be the same throughout this simulation, so that the energy is conserved and no heat nor matter is exchanged with the outside reservoir. In this setup, if we have a high velocity particle flying outside of the box, it should automatically reappear on the other side of the box. This setup applies to forces as well. In order to compute forces under the assumption of a box with periodic boundary condition, we make 27 copies of the original simulation box, and where each contain particles with the same velocities. Using this extended box which is 3^3 the size of the original one, we compute the two point forces and update the positions and velocities of the particles. Below is an illustration of how the original simulation box is copied and extended.

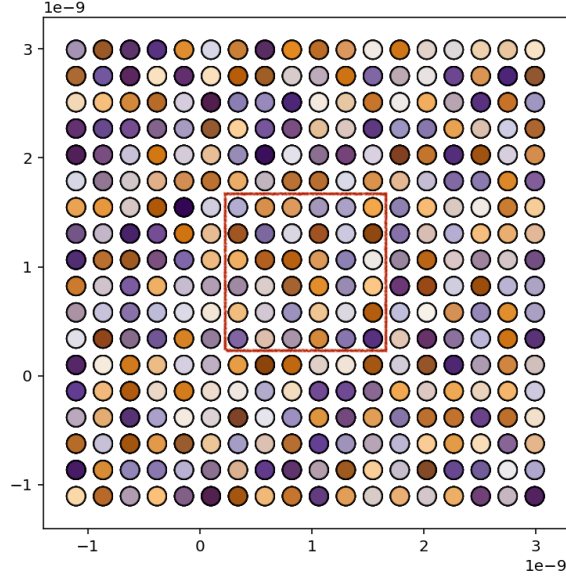


Figure 3: An illustration of the extended box, now colour coded by the velocities of the particles.

The extended box is plotted in the projected xy -plane, and the red square outlines the original projected simulation box. In this way, the particles on the vertices of the simulation box can also feel the forces from the other end of the vertices, thus updating the velocities and positions accordingly. Notice that the extension is made in all three dimensions and above is just an illustration how the extension works.

4 Verlet Algorithm

To update the positions and velocities of the particles we implement Verlet algorithm. To find the two-point interaction potential, we first need to find the update rule for the evolution of the potential, which is given by

$$\mathbf{F}(\mathbf{x}(t+h)) = -\nabla U(\mathbf{x}(t+h)) \quad (9)$$

where U can be found by (1). For each particle, we feed the function \mathbf{F} an array of vectors which are the subtraction of the positions of the surrounding particles and the current particle. Then \mathbf{F} is effectively the sum of all the divergence of the potential given by all the surrounding particles. The position vector and velocity vector for each particle are updated as

$$\begin{aligned} \mathbf{x}(t+h) &= \mathbf{x}(t) + h\mathbf{v}(t) + \frac{h^2}{2m}\mathbf{F}(\mathbf{x}(t)) + \mathcal{O}(h^3) \\ \mathbf{v}(t+h) &= \mathbf{v}(t) + \frac{h}{2m}\mathbf{F}(\mathbf{x}(t+h)) + \mathbf{F}(\mathbf{x}(t)) \end{aligned} \quad (10)$$

Hence if initialised with what was described above, we can expect the lattice configuration of the simulation box to be randomised. Below shows some snapshots during one simulation.

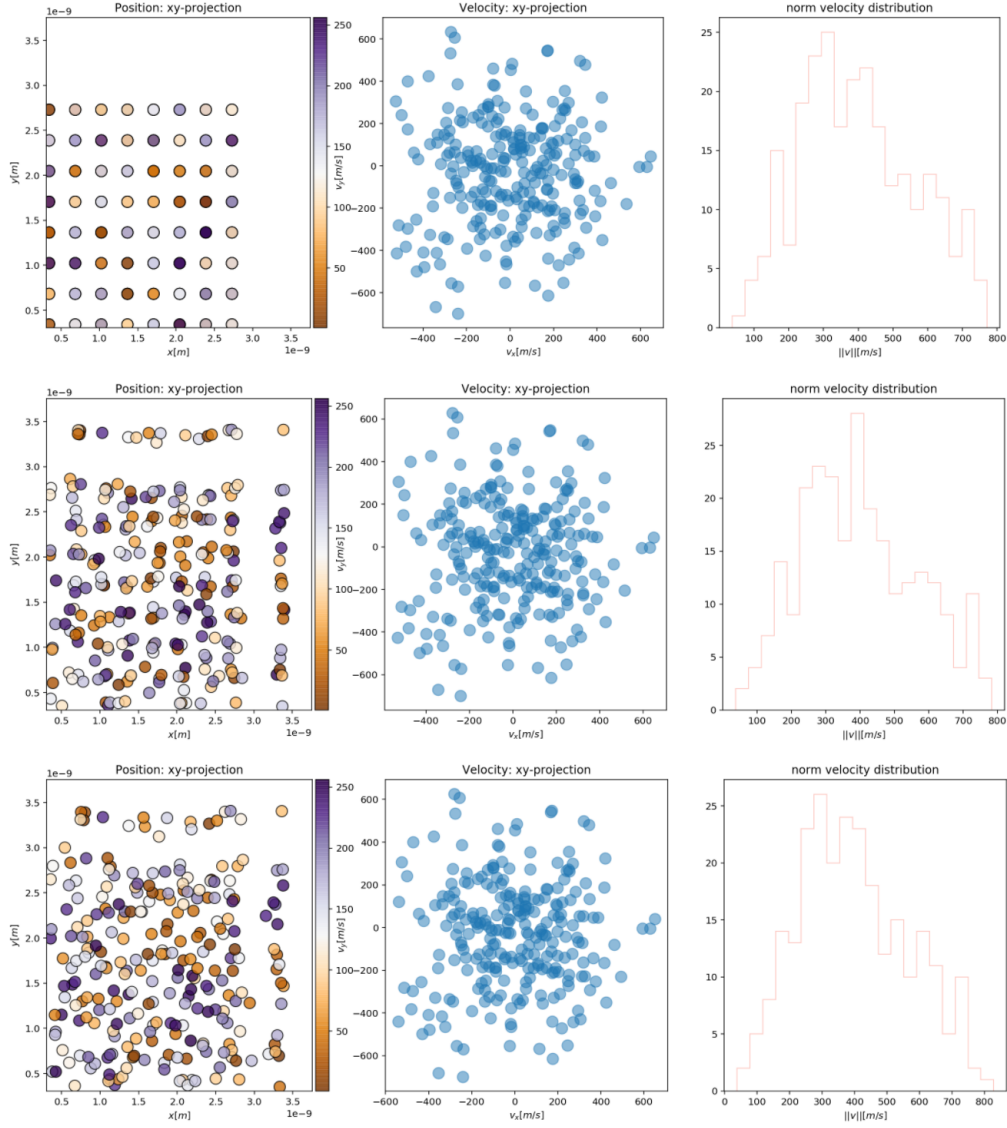


Figure 4: Snapshots for one of the simulation runs (Simulation 2, details see Table 2 and 3). The first column shows the particles on the projected xy -plane, where the particles are colour coded by their velocities; the second column shows the distribution of the velocities on the projected $v_x v_y$ -plane; and the third column shows the histogram of the norm of the velocity vectors. Notice that the empty spaces are there to ensure when making the extended box, two particles do not overlap.

The snapshots were taken every 25 loops, with the first row being the initial state. It can be seen from this simulation, the symmetry given by the lattice configuration is soon broken, and particles more randomly distributed.

5 Pressure and Correlation Function

In a box of particles, pair correlation function $g(r)$ describes how density varies as a function of distance from a reference particle. The pair correlation function is usually determined by calculating the distance between all particle pairs and binning them into a histogram. The histogram is then normalized with respect to an ideal gas, where particle histograms are completely uncorrelated, which means $g(r) = 1$. For three dimensions, this normalization is the number density of the system ρ multiplied by the volume of the spherical shell, which symbolically can be expressed as $\rho 4\pi r^2 dr$.

Table 2: Simulation configurations for three different states of Argon.

simulation configurations			
Simulation	State	ρ	T
Simulation 1	Gas	0.3	3
Simulation 2	Liquid	0.8	1
Simulation 3	Solid	1.2	0.5
Number of particles: 108			

Table 3: Physical properties for three different states of Argon.

Physical properties				
Simulation	State	$\rho[\text{kg m}^{-3}]$	$T[\text{K}]$	$V[\text{m}^3]$
Simulation 1	Gas	5.833×10^2	300	2.896×10^{-26}
Simulation 2	Liquid	1.556×10^3	100	1.086×10^{-26}
Simulation 3	Solid	2.333×10^3	50	7.241×10^{-27}

In this case, the pair correlation function can be computed by

$$g(r) = \frac{2V}{N(N-1)} \frac{\langle n(r) \rangle}{4\pi r^2 \Delta r} \quad (11)$$

where V is the volume of simulation cell, and N is the particle number. $\langle n(r) \rangle$ denotes the average over many configurations.

The radial distribution function is fundamentally important since it can be used to link the microscopic details to macroscopic properties. Hence, we can use the following formula to obtain the pressure of the system:

$$\frac{p}{\rho k_B T} = 1 - \frac{1}{3Nk_B T} \left\langle \sum_i \sum_{j>i} r_{ij} \frac{\partial U(r_{ij})}{r} \right\rangle - \frac{\rho}{6k_B T} \int \mathbf{r} \frac{\partial U}{\partial \mathbf{r}} g(\mathbf{r}) d\mathbf{r} \quad (12)$$

The first term of Eq.12 presents the ideal gas, the second term is the time average of the virial. The last term is a correction term which takes into account the effect on the pressure of the tail of the potential which has been neglected in the dynamics as well as in this simulation.

In order to analyse the properties of the Argon of three different states, we initialise as follows

6 Discussion and Conclusion

In order to set up the box as listed in Table 2, we first calculate the density in the liquid state of Argon, which is $1.3954 \times 10^3 \text{ kg m}^{-3}$. Hence with the given ratio we found the density and temperature in the other two states to be The trajectories for the three simulations are shown as follows. [\[Panel figure 3 x 4\]](#)

Table 4: Mean pressure with error for three different states of Argon.

Pressure values[Pa]			
Simulation	State	\bar{p}	σ_p
Simulation 1	Gas	3.66×10^7	582
Simulation 2	Liquid	3.25×10^7	46
Simulation 3	Solid	2.44×10^7	91

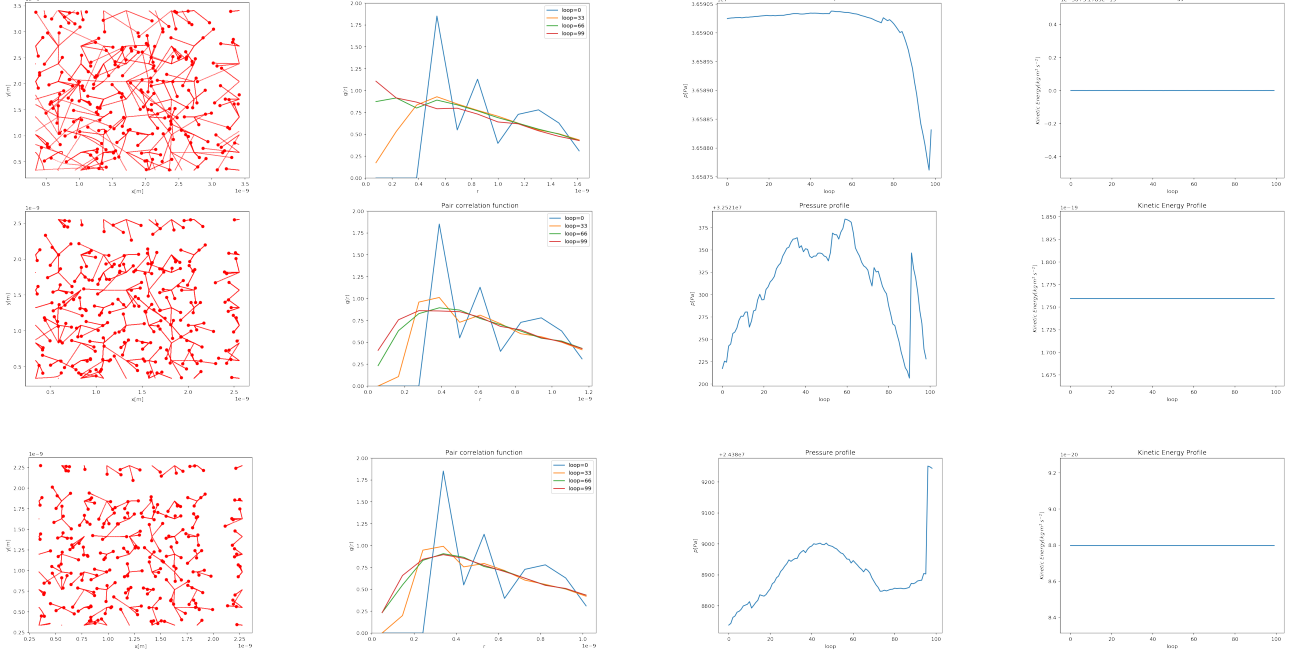


Figure 5: From top to bottom: Simulation 1, Simulation 2, Simulation 3. From left to right: trajectories, correlation function, pressure profile, and conservation of kinetic energy. To get an impression of what this means on the expected values for pressure, we do the dimensional analysis. The SI unit for pressure is Pascal, which is equivalently $\text{kg} \cdot (\text{m} \cdot \text{s}^2)^{-1}$, which is equivalent to the unit of $\frac{k_B T}{m}$, such that

$$[p] = \left[\frac{k_B T}{m} \right] = \frac{\text{kg}}{\text{m} \cdot \text{s}^2} \quad (13)$$

The Boltzmann constant is on the order of 10^{-23} , the temperature is on the order of 10^2 and density 10^3 and mass 10^{-25} , making the pressure on the order of 10^7 , given that the correction term

$$\frac{1}{3Nk_B T} \left\langle \sum_i \sum_{j>i} r_{ij} \frac{\partial U(r_{ij})}{r} \right\rangle \quad (14)$$

is negligible. Hence we can expect our calculated pressure to be on the order of 10^7 Pa, and the values from three states should be roughly on the same order. It can be seen from the above panel, that the three different states have very distinctive correlation functions. For the gas state, the lattice configuration is quickly broken, and the equilibrated result (as depicted by the red solid line) shows that the correlation function goes from ~ 1 and then asymptotically decays to zero. As for the liquid state, the correlation function starts at ~ 0.5 , then peaks at ~ 1 and then decreases. For the solid state, the equilibrated result shows that there is still a strong correlation at distances $r = (\sqrt{2}/2)l, l, 2l$ where l is the side length of the unit lattice cell. These correlations arise because the symmetry of the lattice configuration is not yet fully broken in solid state, yet slightly perturbed. And in this situation particles are oscillating around their initial positions, instead of flying around as seen in the first simulation.

The pressure for the three different simulations are listed as follows. From W. Wagner et al. we know for Argon at ~ 87 K, the pressure is ~ 100 kPa, and with each increment of ~ 10 K, we expect the pressure to increase by one order. The details are presented as follows. This is very different from what we expected as

Table 5: Vapour pressure for Argon.

P[Pa]	1	10	100	1k	10k	100k
T[k]	-	47	53	61	71	87

Table 6: Additional simulation.

Simulation	State	ρ	$V[\text{m}^3]$
Simulation x	Gas	0.176	4.687×10^{-23}

computed from (12), whose expected values are somewhat on the same order. This implies that, it is insufficient to infer an educated guess for pressure, in a simulation concerning only Lennard-Jones potential as two-point interaction potential. To further investigate pressure we have to take other facts into consideration.

In a nutshell, we have simulated a box of Argon particles, initialised with the lattice configuration. We have allowed the box to evolve for sufficiently long enough epochs such that the box is well equilibrated. For all three simulations we found that the symmetry given by the lattice configuration is to some extent broken. We have found that the three states are indeed distinctively correlated, with the solid state appearing a strong localised correlation, and the gas appearing a smoother behaviour in correlation, and liquid lies somewhere in between.

7 Further Investigation

Additionally, we have evolved the box using random initialisation. In this case, in order to circumvent positive potential, we compute the pair-wise distances and rescale the position, such that the smallest separation is now greater than 1σ . Below we show the result of this additional experiment. The Argon particles are initialised in gas state, with the following properties The snapshots of Simulation x are shown as follows.

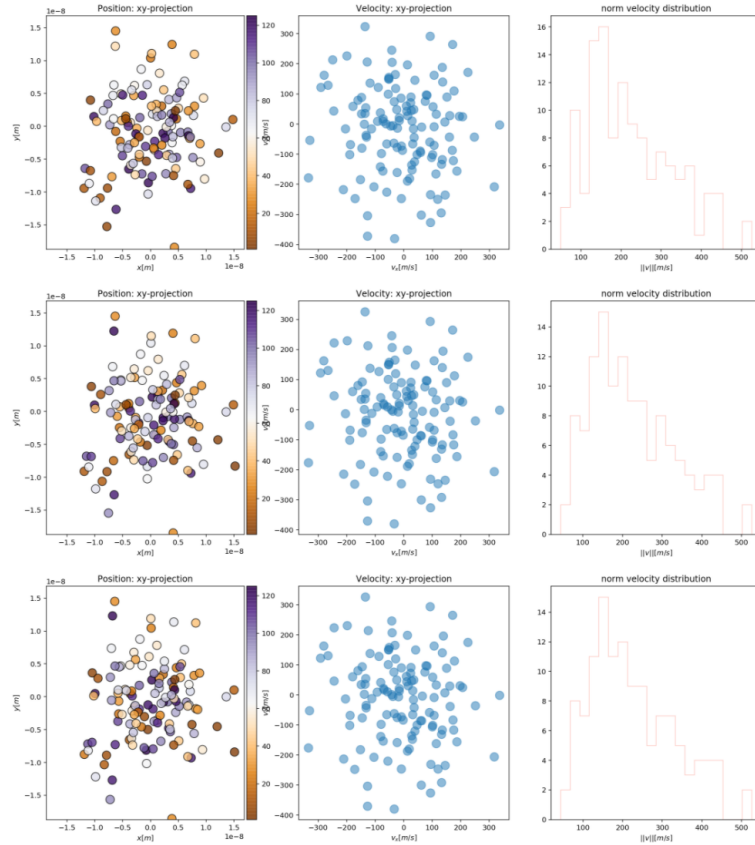


Figure 5: Snapshots for Simulation x, the positions of the particles are now initialised randomly. Details see 'random_generator' in 'Argon_simulation.py'.

The corresponding trajectories, correlation function, pressure profile, and kinetic energy are shown as follows.

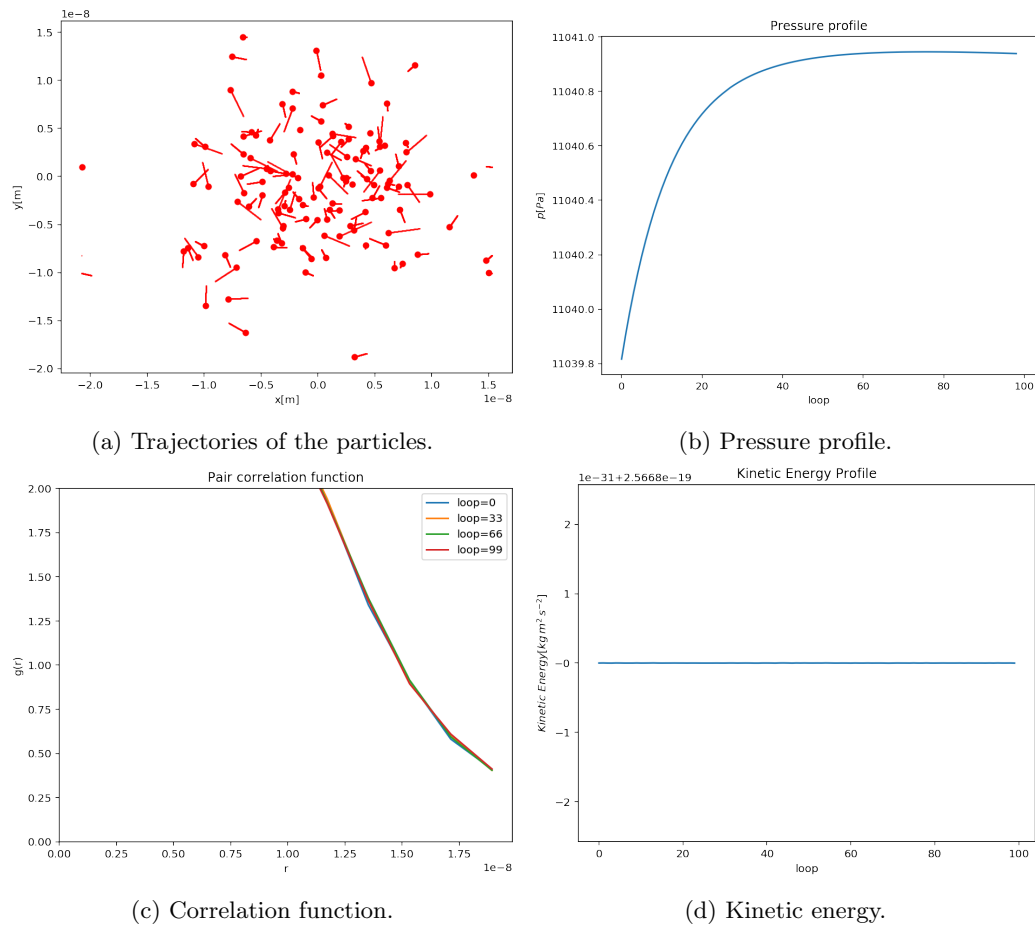


Figure 6: Outputs for Simulation x.

We have found the pressure to be 11040 Pa, with an error of 0.25 Pa. We have also found a smoother behaviour in the pressure profile, with even lower error. In addition to what was found above, this correlation function serves as a complement to what was shown and discussed above.

References

- [1] Wagner, W. (1973), "New vapour pressure measurements for argon and nitrogen and a new method for establishing rational vapour pressure equations", *Cryogenics*, 13 (8): 470–482

8 Appendix

co

```
1 #!/usr/bin/env python
2 # coding: utf-8
3
4 from math import *
5 import numpy as np
6 import tqdm
7 import seaborn as sns
8 from scipy.optimize import newton, bisect
9 import matplotlib.animation as animation
10 import matplotlib.pyplot as plt
11 from matplotlib.pyplot import figure, show
```

```

12 from mpl_toolkits.axes_grid1 import make_axes_locatable
13
14
15 class particle_in_box:
16     global sigma
17     global kB
18     global T0
19     global epsilon
20     global m
21
22     sigma = 3.405e-10 # Angstrom
23     kB = 1.38e-23 # Boltzmann constant
24     T0 = 100 # starting temperature of the box
25     epsilon = T0*kB # initialising the parameter epsilon, it changes w.r.t. the mean
26     # temperature of the box
27     m = 6.6e-26 # mass of an Argon atom
28
29     def __init__(self, Num=8, gentype=1, sep=1.6, T=1, loop=300, tol=.2, hlevel=10, *args,
30     **kwargs):
31
32         self.loop = 0
33         self.sep = sep
34         self.T = T0*T # change temperature
35         self.epsilon = self.T*kB
36         self.loop_end = loop # end loop
37         self.tol = tol # define an effective radius to calculate potential
38         self.N = Num # number of particles
39         self.hlevel = hlevel # parameter that defines the stepsize
40         self.reff = self.r_eff() # updating effective radius
41         self.gentype = gentype
42
43         if gentype == 1:
44             try:
45                 self.Pos_stack = np.zeros((self.loop_end+1, self.N, 3))
46                 self.Vel_stack = np.zeros((self.loop_end+1, self.N, 3))
47                 self.Etarget = (self.N - 1)*1 * kB * self.T
48                 self.Pos, self.Vel, self.Lmin, self.Lmax = self.random_generator()
49                 self.V = (self.Lmax-self.Lmin)**3
50                 self.density = m*self.N/self.V
51             except ValueError:
52                 raise ValueError("input Num should be an int^3!")
53
54         elif gentype == 2:
55             self.Pos_stack = np.zeros((self.loop_end+1, self.N, 3))
56             self.Vel_stack = np.zeros((self.loop_end+1, self.N, 3))
57             self.Etarget = (self.N - 1)*1 * kB * self.T
58             self.Pos, self.Vel, self.Lmin, self.Lmax = self.cor_random_generator()
59             self.V = (self.Lmax-self.Lmin)**3
60             self.density = m*self.N/self.V
61
62         elif gentype == 3:
63             # the initialisation of Pos_stack, Vel_stack, E_target are done inside the function
64             self.Pos, self.Vel, self.Lmin, self.Lmax = self.lattice_generator()
65             self.N = len(self.Pos)
66             self.V = (self.Lmax-self.Lmin)**3
67             self.density = m*self.N/self.V
68
69         else:
70             raise NameError("gentype should be 1,2 or 3!")
71             print('density is: ', '%.3e' % self.density, "kg m^-3.")
72             print("temeperature is: ", np.round(self.T,2), "K.")
73             print("simulation box size: ", '%.3e' % self.V, "m^3")
74
75         self.arr = self.cube_extension()
76
77         # default initialisation, this generator gives initial position on a meshgrid, then give
78         # it a small perturbation
79         def random_generator(self):
80             N = int(np.round(self.N**(1/3)))
81
82             # place the particles on randomly

```

```

81 Pos = np.random.randn(N**3, 3)
82 # the velocities are generated from a 3D normal distribution
83 v0 = sqrt(epsilon/m)
84 mean = (0, 0, 0)
85 cov = [[1, 0, 0], [0, 1, 0], [0, 0, 1]]
86 Vel = np.random.multivariate_normal(mean, cov, N**3)
87 l = np.sqrt(self.Etarget/(m*np.sum(Vel**2)))
88 Vel = l*Vel
89 deltax = np.array([0])
90 xmatrix = Pos
91 # permute xmatrix
92 for i in range(1, N):
93     newx = np.concatenate((xmatrix[i:], xmatrix[0:i]), axis=0)
94     sig = xmatrix-newx
95     D = np.sqrt(np.sum(sig**2, axis=1))
96     deltax = np.concatenate((deltax, D), axis=None)
97 deltax = np.unique(deltax)
98 deltax = np.abs(deltax[deltax > 0])
99 self.dmin = np.min(deltax)
100
101 # rescale the position so that now the smallest separation is < 1.1 sigma
102 Pos *= self.sep/(deltax.min()/sigma)
103
104 # define the edge of the box
105 minarr = np.array([Pos[:, 0].min(), Pos[:, 1].min(), Pos[:, 2].min()])
106 maxarr = np.array([Pos[:, 0].max(), Pos[:, 1].max(), Pos[:, 2].max()])
107 Lmin = np.min(minarr)
108 Lmax = np.max(maxarr)
109
110 self.Pos_stack[0] = Pos
111 self.Vel_stack[0] = Vel
112
113 return Pos, Vel, Lmin, Lmax
114
115 # given N, generates N**3 cubes of lattice partices
116 def lattice_generator(self):
117
118     def lattice_generator(N):
119         n = 2
120         Pos = np.zeros((N**3, 3))
121         for i in range(len(Pos)):
122             Pos[i][0] = i//N**2
123             Pos[i][1] = (i-Pos[i][0]*N**2)//N
124             Pos[i][2] = i-Pos[i][0]*N**2-Pos[i][1]*N
125
126         p0 = Pos[-1]/2
127         R = np.sqrt(np.sum(p0**2))
128         for item in Pos:
129
130             p1 = np.array([item[0]+.5, item[1]+.5, item[2]])
131             p2 = np.array([item[0]+.5, item[1], item[2]+.5])
132             p3 = np.array([item[0], item[1]+.5, item[2]+.5])
133             Pos = np.vstack((Pos, p1, p2, p3))
134
135         Pos = np.unique(Pos, axis=0)
136
137         Dist = np.sqrt((Pos[:, 0]-p0[0])**2 +
138                        (Pos[:, 1]-p0[1])**2+(Pos[:, 2]-p0[2])**2)
139         ind = Dist < 1.1*R
140         Pos = Pos[ind]
141
142         ext = np.zeros((N**3, 3))
143         Post = Pos
144         for i in range(len(ext)):
145             ext[i][0] = i//N**2
146             ext[i][1] = (i-ext[i][0]*N**2)//N
147             ext[i][2] = i-ext[i][0]*N**2-ext[i][1]*N
148
149         Post = np.vstack((Post, Pos+ext[i]))
150
151         Post = np.unique(Post, axis=0)
152         mask = (Post[:, 0] == Post.max()) + (Post[:, 1] ==

```

```

153 Post.max()) + (Post[:, 2] == Post.max())
154
155 Post = Post[~mask]
156
157 return Post
158
159 Pos = lattice_generator(self.N)
160 # rescale position so that the smallest separation is now user-defined sigma, and in
the attractive range
161 Pos *= self.sep*sigma/(sqrt(2)/2)
162 # add a small translation
163 Pos += 1*sigma
164 N = len(Pos)
165 self.Pos_stack = np.zeros((self.loop_end+1, N, 3))
166 self.Vel_stack = np.zeros((self.loop_end+1, N, 3))
167 self.Etarget = (N - 1)*1 * kB * self.T
168
169 # generate velocities
170 v0 = sqrt(epsilon/m)
171 mean = (0, 0, 0)
172 cov = [[1, 0, 0], [0, 1, 0], [0, 0, 1]]
173 Vel = np.random.multivariate_normal(mean, cov, N)
174 for i, a in enumerate(Vel):
175     while sqrt(np.sum(a*a)) > 3:
176         Vel[i] = np.random.multivariate_normal(mean, cov, 1)
177 l = np.sqrt(self.Etarget/(m*np.sum(Vel**2)))
178 Vel = l*Vel
179
180 # define the edge of the box
181 minarr = np.array([Pos[:, 0].min(), Pos[:, 1].min(), Pos[:, 2].min()])
182 maxarr = np.array([Pos[:, 0].max(), Pos[:, 1].max(), Pos[:, 2].max()])
183 sep = sqrt(2) * self.sep * sigma
184 Lmin = np.min(minarr)
185 Lmax = np.max(maxarr)+1*sep
186
187 self.Pos_stack[0] = Pos
188 self.Vel_stack[0] = Vel
189
190 return Pos, Vel, Lmin, Lmax
191
192 # correlated position and velocity, with higher velocities in the box centre
193
194 def cor_random_generator(self):
195     N = self.N
196     mean = (0, 0, 0)
197     cov = [[.1, 0, 0], [0, .1, 0], [0, 0, .1]]
198     Pos = np.random.multivariate_normal(mean, cov, N)
199
200     deltad = np.array([0])
201     xmatrix = Pos
202     # permute xmatrix
203     for i in range(1, N):
204         newx = np.concatenate((xmatrix[i:], xmatrix[0:i]), axis=0)
205         sig = xmatrix-newx
206         D = np.sqrt(np.sum(sig**2, axis=1))
207         deltad = np.concatenate((deltad, D), axis=None)
208     deltad = np.unique(deltad)
209     deltad = np.abs(deltad[deltad > 0])
210     self.dmin = np.min(deltad)
211     # rescale the position so that now the smallest separation is < 1.1 sigma
212     Pos *= self.sep/(deltad.min()/sigma)
213
214     # the velocities are generated from a 3D normal distribution
215     v0 = sqrt(epsilon/m)
216     mean = (0, 0, 0)
217     cov = [[1, 0, 0], [0, 1, 0], [0, 0, 1]]
218     Vel = np.random.multivariate_normal(mean, cov, N)
219     l = np.sqrt(self.Etarget/(m*np.sum(Vel**2)))
220     Vel = l*Vel
221
222     def Cartesian_to_Spherical(x, y, z):
223         r = np.sqrt(x*x+y*y+z*z)

```

```

224         theta = np.arctan2(y, x)
225         phi = np.arccos(z/r)
226         return r, theta, phi
227
228     R, Theta, Phi = Cartesian_to_Spherical(
229         Pos[:, 0]-.5*D, Pos[:, 1]-.5*D, Pos[:, 2]-.5*D)
230     indsort = np.argsort(np.abs(R)) # sort R in ascending order
231     Vel[:, 0][indsort] = np.sort(Vel[:, 0][::-1])
232     Vel[:, 1][indsort] = np.sort(Vel[:, 1][::-1])
233     Vel[:, 2][indsort] = np.sort(Vel[:, 2][::-1])
234
235     # define the edge of the box
236     minarr = np.array([Pos[:, 0].min(), Pos[:, 1].min(), Pos[:, 2].min()])
237     maxarr = np.array([Pos[:, 0].max(), Pos[:, 1].max(), Pos[:, 2].max()])
238     Lmin = np.min(minarr) # *.999
239     Lmax = np.max(maxarr) # *1.001
240     self.Pos_stack[0] = Pos
241     self.Vel_stack[0] = Vel
242
243     return Pos, Vel, Lmin, Lmax
244
245     def r_eff(self):
246         tol_percentage = self.tol
247         def U(r): return 4*self.epsilon*((sigma/r)**12-(sigma/r)**6)
248         Umin = U(2**(1/6)*sigma)
249         def Usol(r): return U(r)-(1-tol_percentage)*Umin
250         reff = bisect(Usol, (2**(1/6))*sigma, (2**(1/6)+3)*sigma)
251         return reff
252
253     def F(self, x, v):
254         # if there is more than one particle in the force computing range
255         if len(x.flatten())/3 > 1:
256             r = np.sqrt(np.sum(x*x, axis=1))
257             vtot = np.sum(v**2)
258             eps = m*vtot/(len(x))
259             U = 4*self.epsilon*((sigma/r)**12-(sigma/r)**6)
260             dUdr = 24*self.epsilon*(-sigma/r**2) * \
261                 (sigma/r)**5*(2*(sigma/r)**6-1)
262             div = -dUdr/r
263             rdU = np.sum(dUdr*r)
264             delvec = (np.zeros(x.shape)+div[:, np.newaxis])*x
265             return np.sum(delvec, axis=0), eps, np.sum(U), rdU
266         # if there is only one particle in the force computing range
267         else:
268             r = np.sqrt(np.sum(x*x))
269             vtot = np.sum(v**2)
270             eps = m*vtot
271             U = 4*self.epsilon*((sigma/r)**12-(sigma/r)**6)
272             dUdr = 24*self.epsilon*(-sigma/r**2) * \
273                 (sigma/r)**5*(2*(sigma/r)**6-1)
274             div = -dUdr/r
275             rdU = dUdr*r
276             delvec = div*x
277             return delvec, eps, U, rdU
278
279     def cube_extension(self):
280         arr = np.zeros((3**3, 3))
281         D = self.Lmax - self.Lmin
282         for i in range(27):
283             arr[i][0] = i//3**2
284             arr[i][1] = (i-arr[i][0]*3**2)//3
285             arr[i][2] = i-arr[i][0]*3**2-arr[i][1]*3
286         arr = D*(arr-np.ones(arr.shape))
287         return arr
288
289     def Verlet(self):
290         self.pressure = np.zeros(self.loop_end)
291         PE = np.zeros((self.loop_end, self.N))
292         for loop_this in tqdm.tqdm(range(self.loop_end)):
293             if loop_this >= self.loop_end:
294                 break
295             h = sigma/(np.max(self.Vel))/self.hlevel

```

```

296 Pos_ext = np.array([[0, 0, 0]])
297 Vel_ext = np.array([[0, 0, 0]])
298 D = self.Lmax - self.Lmin
299 for i in range(len(self.arr)):
300     newbox = self.Pos + self.arr[i][np.newaxis, :]
301     Pos_ext = np.concatenate((Pos_ext, newbox), axis=0)
302     Vel_ext = np.concatenate((Vel_ext, self.Vel), axis=0)
303
304 Pos_ext = Pos_ext[1:]
305 Vel_ext = Vel_ext[1:]
306
307 fig = figure(figsize=(7.5,6))
308 ax = fig.add_subplot(111)
309 ax.plot(Pos_ext[:,0], Pos_ext[:,1], 'o')
310 show()
311
312 correction = np.zeros(self.N)
313 # for each particle, calculate the neighbouring particles inside reff
314 for i in range(self.N):
315
316     P0 = self.Pos[i]
317     V0 = self.Vel[i]
318     Dist = np.sqrt(
319         np.sum((Pos_ext-P0*np.ones(Pos_ext.shape))**2, axis=1))
320     mask = (Dist > 0) & (Dist < D/2)
321
322     Peff = Pos_ext[mask]
323     Veff = Vel_ext[mask]
324
325     delva, self.epsilon, U0, sum1 = self.F(Peff, Veff)
326     xth = P0+h*V0+.5*h**2/m*delva
327     delvah, self.epsilon, Ui, sum2 = self.F(xth, V0)
328     vth = V0+.5*h/m*(delvah+delva)
329
330     if np.any(xth < self.Lmin):
331         ind = np.where(xth < self.Lmin)[0]
332         n = (xth[ind]-self.Lmin)//D
333         xth[ind] -= n*D
334
335     if np.any(xth >= self.Lmax):
336         ind = np.where(xth > self.Lmax)
337         n = (xth[ind]-self.Lmin)//D
338         xth[ind] -= n*D
339
340     correction[i] = sum2
341     self.Pos[i] = xth
342     self.Vel[i] = vth
343     # rescale velocities
344     l = np.sqrt(self.Etarget/(m*np.sum(self.Vel**2)))
345     self.Vel = l*self.Vel
346     self.Pos_stack[self.loop+1] = self.Pos
347     self.Vel_stack[self.loop+1] = self.Vel
348     PE[self.loop][i] = Ui
349     # print("correction", correction)
350     # print("pressure term", (1-.5*np.sum(correction)/(3*self.N*kB*T0)))
351     self.pressure[self.loop] = (
352         1-.5*np.sum(correction)/(3*self.N*kB*self.T))*kB*self.T*self.density/m
353     self.epsilon = m*np.sum(self.Vel**2)/self.N
354     Vel_r = np.sqrt(np.sum(self.Vel**2, axis=1))
355
356     self.loop += 1
357
358     return self.Pos_stack, self.Vel_stack, np.sum(PE, axis=1), self.pressure
359
360 # get the kinetic energy of the particles per snapshot
361
362 def get_KE(self):
363     sum1 = np.sum(.5*m*self.Vel_stack[1:]**2, axis=2)
364     # array of length loop_end+1 which gives the KE per snapshot
365     sum2 = np.sum(sum1, axis=1)
366     KE = sum2
367     fig = figure(figsize=(7.5, 6))

```

```

368 ax = fig.add_subplot(111)
369 ax.plot(np.arange(len(KE)), KE)
370 ax.set_xlabel('loop')
371 ax.set_ylabel(r'$Kinetic \ Energy [kg \, m^2 \, s^{-2}]$')
372 ax.set_title('Kinetic Energy Profile')
373 show()
374
375 def get_plots(self):
376     for i in range(self.loop_end+1):
377         vx = self.Vel_stack[i][:, 0]
378         vy = self.Vel_stack[i][:, 1]
379         vz = self.Vel_stack[i][:, 2]
380         vr = np.sqrt(vx**2+vy**2+vz**2)
381         ind = np.argsort(vr)
382         t = (np.arange(len(vr))+np.ones(len(vr)))[ind]
383         if i % 20 == 0:
384             fig = figure(figsize=(18, 6))
385             ax1 = fig.add_subplot(131)
386             ax2 = fig.add_subplot(132)
387             ax3 = fig.add_subplot(133)
388             ax1.plot(self.Pos_stack[i][:, 0], self.Pos_stack[i][:, 1], marker='o',
389                     markersize=3, ls=' ', color='white', alpha=.1, zorder=1)
390             im1 = ax1.scatter(self.Pos_stack[i][:, 0], self.Pos_stack[i][:, 1],
391                              marker='o', s=140, c=t, cmap='PuOr', edgecolor='k',
392                              alpha=.8, zorder=2)
393             ax2.scatter(
394                 self.Vel_stack[i][:, 0], self.Vel_stack[i][:, 1], marker='o', s=140,
395                 alpha=.5)
396             ax3.hist(np.sqrt(vx**2+vy**2+vz**2), bins=20,
397                     histtype='step', color='salmon', alpha=.4)
398             divider = make_axes_locatable(ax1)
399             cax1 = divider.append_axes('right', size='5%', pad=0.05)
400             fig.colorbar(im1, cax=cax1, orientation='vertical')
401             ax1.ticklabel_format(style='sci', axis='both')
402             ax1.set_xlim([self.Lmin*.9, self.Lmax*1.1])
403             ax1.set_ylim([self.Lmin*.9, self.Lmax*1.1])
404             ax1.set_xlabel(r'$x [m]$')
405             ax1.set_ylabel(r'$y [m]$')
406             ax2.set_xlabel(r'$v_x [m/s]$')
407             ax2.set_ylabel(r'$v_y [m/s]$')
408             ax3.set_xlabel(r'$||v|| [m/s]$')
409             ax1.set_title('Position: xy-projection')
410             ax2.set_title('Velocity: xy-projection')
411             ax3.set_title('norm velocity distribution')
412             show()
413             if i == 60:
414                 break
415
416 def get_correlation(self):
417
418     def pairCorrelation_3D(p, D, rMax, dr):
419         """Compute the three-dimensional pair correlation function for a set of
420         spherical particles contained in a cube with side length D.
421         Arguments:
422             p          3d positions of the particles
423             D          length of each side of the cube in space
424             rMax       outer diameter of largest spherical shell
425             dr         increment for increasing radius of spherical shell
426         Returns a tuple: (g, radii, interior_indices)
427             g(r)       a numpy array containing the correlation function g(r)
428             radii      a numpy array containing the radii of the
429                        spherical shells used to compute g(r)
430         """
431
432         # Find particles which are close enough to the cube center that a sphere of radius
433         # rMax will not cross any face of the cube
434
435         x = p[:, 0]
436         y = p[:, 1]
437         z = p[:, 2]
438         mask1 = (x > rMax) + (x < (D - rMax))
439         mask2 = (y > rMax) + (y < (D - rMax))

```

```

438     mask3 = (z > rMax) + (z < (D - rMax))
439     interior_indices, = np.where(mask1 + mask2 + mask3)
440     num_interior_particles = len(interior_indices)
441
442     if num_interior_particles < 1:
443         raise RuntimeError("No particles found for which a sphere of radius rMax\
444             will lie entirely within a cube of side length D. Decrease rMax\
445             or increase the size of the cube.")
446
447     edges = np.arange(0., rMax + 1.1 * dr, dr)
448
449     num_increments = len(edges) - 1
450     g = np.zeros([num_interior_particles, num_increments])
451     radii = np.zeros(num_increments)
452     numberDensity = len(x) / D**3
453
454     # Compute pairwise correlation for each interior particle
455     for p in range(num_interior_particles):
456         index = interior_indices[p]
457         d = np.sqrt((x[index] - x)**2 + (y[index] - y)
458             ** 2 + (z[index] - z)**2)
459         d[index] = 2 * rMax
460
461         (result, bins) = np.histogram(d, bins=edges, normed=False)
462         g[p, :] = result / numberDensity
463
464     # Average g(r) for all interior particles and compute radii
465     g_average = np.zeros(num_increments)
466     for i in range(num_increments):
467         radii[i] = (edges[i] + edges[i+1]) / 2.
468         rOuter = edges[i+1]
469         rInner = edges[i]
470         g_average[i] = np.mean(
471             g[:, i]) / (4.0 / 3.0 * pi * (rOuter**3 - rInner**3))
472
473     return (g_average, radii)
474 D = self.Lmax - self.Lmin
475 Pos_stack = self.Pos_stack
476 fig = figure(figsize=(7.5, 6))
477 ax = fig.add_subplot(111)
478 for i in range(len(Pos_stack)):
479     #if (i > 0) & (i % (self.loop_end//3) == 0):
480     if i % (self.loop_end//3) == 0:
481         Pos = Pos_stack[i]
482         g, r = pairCorrelation_3D(Pos, D, D/2., D/20.)
483         ax.plot(r, g, label='loop='+str(i))
484         ax.legend()
485     ax.set_ylim([0, 2])
486     ax.set_xlabel('r')
487     ax.set_ylabel('g(r)')
488     ax.set_title('Pair correlation function')
489     show()
490     return fig, ax
491
492 def get_pressure(self):
493     pressure = self.pressure[1:]
494     N = len(pressure)
495     fig = figure(figsize=(7.5, 6))
496     ax = fig.add_subplot(111)
497     ax.plot(np.arange(N), pressure)
498     ax.set_xlabel('loop')
499     ax.set_ylabel(r'$p$[Pa]$')
500     ax.set_title('Pressure profile')
501     show()
502     print("Mean: ", np.mean(pressure), "std: ", np.std(pressure))
503
504 def get_trajectories(self):
505     Pos_stack = self.Pos_stack
506     fig = figure(figsize=(7.5, 6))
507     ax = fig.add_subplot(111)
508     for i in range(len(Pos_stack)):
509         if i == len(Pos_stack)-1:

```



```

510         ax.plot(Pos_stack[i][:, 0], Pos_stack[i]
511                [:, 1], '.', markersize=10, color='r')
512         ax.plot(Pos_stack[i][:, 0], Pos_stack[i]
513                [:, 1], '.', markersize=.5, color='r')
514         ax.set_xlabel('x[m]')
515         ax.set_ylabel('y[m]')
516     show()
517
518
519 if __name__ == "__main__":
520
521     # initialise with lattice generator
522     """
523     below gives the configuration for the liquid state, with rho = 1.39e3 kg / ^3, T = 100 K,
524     using the default generation model
525     """
526
527     loop = 100
528     hlevel = 50
529     tol = 0.1
530
531     # gas
532     print("Gas state: ")
533     Argon_particle = particle_in_box(
534         Num=3, sep=1.022/(3/8.)*(1/3.), T=3, loop=loop, tol=tol, gentype=3, hlevel=hlevel)
535     Argon_particle.Verlet()
536     # Argon_particle.get_plots()
537     Argon_particle.get_trajectories()
538     print("pressure is: ", round(np.mean(self.pressure), 2), "Pa, with error: ",
539           round(np.std(self.pressure), 2))
540     # Argon_particle.get_correlation()
541     # Argon_particle.get_pressure()
542     # Argon_particle.get_KE()
543
544     print("Liquid state: ")
545     Argon_particle = particle_in_box(
546         Num=3, sep=1.022, T=1, loop=loop, tol=tol, gentype=3, hlevel=hlevel)
547
548     Argon_particle.Verlet()
549     # Argon_particle.get_plots()
550     Argon_particle.get_trajectories()
551     print("pressure is: ", round(np.mean(self.pressure), 2), "Pa, with error: ",
552           round(np.std(self.pressure), 2))
553     # Argon_particle.get_correlation()
554     # Argon_particle.get_pressure()
555     # Argon_particle.get_KE()
556
557     # solid
558     print("Solid state: ")
559     Argon_particle = particle_in_box(
560         Num=3, sep=1.022/1.5*(1/3.), T=.5, loop=loop, tol=tol, gentype=3, hlevel=hlevel)
561     Argon_particle.Verlet()
562     # Argon_particle.get_plots()
563     Argon_particle.get_trajectories()
564     print("pressure is: ", round(np.mean(self.pressure), 2), "Pa, with error: ",
565           round(np.std(self.pressure), 2))
566     # Argon_particle.get_correlation()
567     # Argon_particle.get_pressure()
568     # Argon_particle.get_KE()
569
570     """
571     below gives the configuration for the liquid state, with rho = 1.39e3 kg / ^3, T = 100 K,
572     using the default generation model
573     """
574     # initialise with random generator
575     print("Liquid state: ")
576     Argon_particle = particle_in_box(Num=216, sep = 1.0001, T =
577         1, loop=100, tol=.1, gentype=1, hlevel=100)
578     Argon_particle.Verlet()

```

```

576     Argon_particle.get_plots()
577     Argon_particle.get_trajectories()
578     Argon_particle.get_correlation()
579     Argon_particle.get_pressure()
580     Argon_particle.get_KE()
581
582
583     print("Gas state: ")
584     Argon_particle = particle_in_box(Num=216, sep = 1.05/(3/8.)*(1/3.), T =
3, loop=100, tol=.1, gentype=1, hlevel=100)
585     Argon_particle.Verlet()
586     Argon_particle.get_plots()
587     Argon_particle.get_trajectories()
588     Argon_particle.get_correlation()
589     Argon_particle.get_pressure()
590     Argon_particle.get_KE()
591
592     print("Solid state: ")
593     Argon_particle = particle_in_box(Num=216, sep = 1.05/1.5*(1/3.), T =
.5, loop=100, tol=.1, gentype=1, hlevel=100)
594     Argon_particle.Verlet()
595     Argon_particle.get_plots()
596     Argon_particle.get_trajectories()
597     Argon_particle.get_correlation()
598     Argon_particle.get_pressure()
599     Argon_particle.get_KE()
600     '''

```

Argon.Simulation.py

```

1  #!/usr/bin/env python
2  # coding: utf-8
3
4
5  from Argon_simulation import particle_in_box
6  import numpy as np
7
8
9  fac = 1
10
11 np.random.seed(8888)
12 # gas
13 print("Gas state: ")
14 Argon_particle = particle_in_box(
15     Num=4, sep=1.022/(3/8.)*(1/3.), T=fac*3, loop=100, tol=0.1, gentype=3, hlevel=40)
16 Argon_particle.Verlet()
17 print("pressure is: ", round(np.mean(self.pressure), 2), "Pa, with error:
", round(np.std(self.pressure), 2))
18 Argon_particle.get_plots()
19 Argon_particle.get_trajectories()
20 Argon_particle.get_correlation()
21 Argon_particle.get_pressure()
22 Argon_particle.get_KE()
23
24
25
26 np.random.seed(8888)
27 print("Liquid state: ")
28 Argon_particle = particle_in_box(
29     Num=4, sep=1.022, T=fac*1, loop=100, tol=0.1, gentype=3, hlevel=100)
30
31 Argon_particle.Verlet()
32 print("pressure is: ", round(np.mean(self.pressure), 2), "Pa, with error:
", round(np.std(self.pressure), 2))
33 # Argon_particle.get_plots()
34 Argon_particle.get_trajectories()
35 Argon_particle.get_correlation()
36 Argon_particle.get_pressure()
37 Argon_particle.get_KE()
38
39
40

```

```

41 np.random.seed(8888)
42 # solid
43 print("Solid state: ")
44 Argon_particle = particle_in_box(
45     Num=4, sep=1.022/1.5**(1/3.), T=fac*.5, loop=100, tol=0.1, gentype=3, hlevel=300)
46 Argon_particle.Verlet()
47 print("pressure is: ",round(np.mean(self.pressure),2),"Pa, with error:
48     ",round(np.std(self.pressure),2))
49 # Argon_particle.get_plots()
50 Argon_particle.get_trajectories()
51 Argon_particle.get_correlation()
52 Argon_particle.get_pressure()
53 Argon_particle.get_KE()
54
55
56
57
58 # Now with random initialisation
59 print("random, Gas state: ")
60 Argon_particle = particle_in_box(
61     Num=125, sep=1.022/(3/8.)*(1/3.), T=fac*3, loop=100, tol=0.1, gentype=1, hlevel=10)
62 Argon_particle.Verlet()
63 Argon_particle.get_plots()
64 Argon_particle.get_trajectories()
65 Argon_particle.get_correlation()
66 Argon_particle.get_pressure()
67 Argon_particle.get_KE()

```

sim_run.py

```

1 #!/usr/bin/env python
2 # coding: utf-8
3
4 from math import *
5 import numpy as np
6 import matplotlib.pyplot as plt
7 from mpl_toolkits.mplot3d import Axes3D
8 from mpl_toolkits.mplot3d.art3d import Poly3DCollection, Line3DCollection
9
10 def lattice_generator(N):
11     n = 2
12     Pos = np.zeros((n**3,3))
13     for i in range(len(Pos)):
14         Pos[i][0] = i//n**2; Pos[i][1] = (i-Pos[i][0]*n**2)//n; Pos[i][2] =
15         i-Pos[i][0]*n**2-Pos[i][1]*n
16
17     p0 = Pos[-1]/2
18     R = np.sqrt(np.sum(p0**2))
19     for item in Pos:
20
21         p1 = np.array([item[0]+.5,item[1]+.5,item[2]])
22         p2 = np.array([item[0]+.5,item[1],item[2]+.5])
23         p3 = np.array([item[0],item[1]+.5,item[2]+.5])
24         Pos = np.vstack((Pos,p1,p2,p3))
25
26     Pos = np.unique(Pos,axis=0)
27
28     Dist = np.sqrt((Pos[:,0]-p0[0])**2+(Pos[:,1]-p0[0])**2+(Pos[:,2]-p0[0])**2)
29     ind = Dist<1.1*R
30     Pos = Pos[ind]
31
32
33     ext = np.zeros((N**3,3))
34     Post = Pos
35     for i in range(len(ext)):
36         ext[i][0] = i//N**2; ext[i][1] = (i-ext[i][0]*N**2)//N; ext[i][2] =
37         i-ext[i][0]*N**2-ext[i][1]*N
38
39     Post = np.vstack((Post,Pos+ext[i]))

```

```

40     Post = np.unique(Post, axis=0)
41
42     return Post
43
44
45
46
47 def plot_cube(cube_definition):
48     cube_definition_array = [
49         np.array(list(item))
50         for item in cube_definition
51     ]
52
53     points = []
54     points += cube_definition_array
55     vectors = [
56         cube_definition_array[1] - cube_definition_array[0],
57         cube_definition_array[2] - cube_definition_array[0],
58         cube_definition_array[3] - cube_definition_array[0]
59     ]
60
61     points += [cube_definition_array[0] + vectors[0] + vectors[1]]
62     points += [cube_definition_array[0] + vectors[0] + vectors[2]]
63     points += [cube_definition_array[0] + vectors[1] + vectors[2]]
64     points += [cube_definition_array[0] + vectors[0] + vectors[1] + vectors[2]]
65
66     points = np.array(points)
67
68     edges = [
69         [points[0], points[3], points[5], points[1]],
70         [points[1], points[5], points[7], points[4]],
71         [points[4], points[2], points[6], points[7]],
72         [points[2], points[6], points[3], points[0]],
73         [points[0], points[2], points[4], points[1]],
74         [points[3], points[6], points[7], points[5]]
75     ]
76
77
78
79     faces = Poly3DCollection(edges, linewidths=1, edgecolors='k')
80     faces.set_facecolor((0,0,1,0.1))
81
82     ax.add_collection3d(faces)
83
84     # Plot the points themselves to force the scaling of the axes
85     ax.scatter(points[:,0], points[:,1], points[:,2], s=0)
86     ax.axis('off')
87
88
89
90
91
92 fig = plt.figure(figsize=(10,10))
93 n=2
94 Post = lattice_generator(n)
95 m = np.arange(len(Post))
96 ax = fig.add_subplot(111, projection='3d')
97 ax.scatter(Post[:,0], Post[:,1], Post[:,2], c=m, s=140, alpha=1, cmap='tab20', edgecolor='k')
98
99
100 for i in range(n):
101     for j in range(n):
102         for k in range(n):
103             cube_definition = [(i,j,k), (i,j+1,k), (i+1,j,k), (i,j,k+1)]
104             plot_cube(cube_definition)
105
106 plt.show()

```

plot_lattice.py