

$$K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14}$$

$$\text{pH} = -\log[\text{H}_3\text{O}^+]$$

$$\text{pOH} = -\log[\text{OH}^-]$$

$$\text{pH} + \text{pOH} = 14$$

$$K_a \times K_b = K_w$$

$$\text{p}K_a = -\log K_a$$

$$\text{p}K_b = -\log K_b$$

$$\text{pH} = \text{p}K_a + \log \left(\frac{[\text{CB}]}{[\text{acid}]}\right)$$

hydrogen 1 H 1.0079																	helium 2 He 4.0026	
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	fluorine 9 F 18.998
sodium 11 Na 22.990	magnesium 12 Mg 24.305	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	aluminum 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948							potassium 19 K 39.098
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	krypton 36 Kr 83.80							cesium 55 Cs 132.91
barium 56 Ba 137.33	57-70 ★		lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	copper 29 Cu 63.546	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	xenon 54 Xe 131.29							radon 86 Rn [222]
francium 87 Fr [223]	89-102 ★ ★		lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	nickel 28 Ni 58.693	cadmium 48 Cd 112.41	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]							ununquadium 114 Uuq [289]
						iron 26 Fe 55.845	mercury 80 Hg 200.59											unbinilium 111 Uun [272]
						manganese 25 Mn 54.938	unnilium 110 Uun [271]											untrium 113 Uut [283]
						chromium 24 Cr 51.996	unquadium 114 Uuq [289]											unpentium 115 Uup [289]
						vanadium 23 V 50.942	unhexium 116 Uuh [291]											unseptium 117 Uus [293]
						titanium 22 Ti 47.867	unoctium 118 Uuo [295]											unenneium 119 Uue [297]
						scandium 21 Sc 44.956	unnonium 119 Uun [297]											unbihium 120 Uub [301]