

# ASTR 792 HW 1

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## 1a

Let

$$\begin{aligned} M_{gal} &= 4 \cdot 10^9 M_{\odot} & M_{\odot} &\approx 2 \cdot 10^{30} kg \\ r_{disk} &= 15 \text{ kpc} = 15000 \text{ pc} & H_{disk} &= 200 \text{ pc} & \frac{He}{H} &= .1 \end{aligned}$$

$$V_{gas} = \pi \cdot r_{disk} H_{disk} = \pi (1.5 \cdot 10^4)^2 (200 \text{ pc}) = 1.41 \cdot 10^{11} \text{ pc}^3$$

The total mass fraction of hydrogen is

$$\begin{aligned} f(H) &= \frac{N_H}{N_{total}} = \frac{N_H}{N_H + N_{He}} = \frac{N_H}{N_H + .1 N_H} \\ &= \frac{1}{1 + .1} \approx .91 \end{aligned}$$

The mass of H nucleus -  $1.67 \cdot 10^{-27} \text{ kg}$ , so the number density of hydrogen nuclei  $n_H$  is

$$n_H = \frac{7.2 \cdot 10^{39} \text{ kg}}{1.67 \cdot 10^{-27} \text{ kg} \cdot 1.41 \cdot 10^{11} \text{ pc}^3} = 3.1 \cdot 10^{55} \text{ pc}^{-3}$$

## 1b

The total mass of dust =  $.007 M_{gas}$

$$\begin{aligned} &\rightarrow 7 \cdot 10^{-3} \cdot 4 \cdot 10^9 M_{\odot} = 2.8 \cdot 10^7 M_{\odot} \cdot 2 \cdot 10^{30} \text{ kg} \\ &\approx 5.6 \cdot 10^{37} \text{ kg} = 5.6 \cdot 10^{40} \text{ g} \end{aligned}$$

$$\rho_{dust} = 2 \text{ g cm}^{-3} \text{ and } r_{dust} = .1 \text{ } \mu\text{m} = 10^{-5} \text{ cm}$$

$$\begin{aligned} &\rightarrow m_{dust} = V \cdot \rho_{dust} = (10^{-5} \text{ cm})(2 \text{ g cm}^{-3}) \\ &= 2 \cdot 10^{-5} \text{ g} \rightarrow N_{dust} = \frac{M_{total}}{m_{dust}} = \frac{5.6 \cdot 10^{40}}{2 \cdot 10^{-5} \text{ g}} \\ &= 2.8 \cdot 10^{45} \text{ dust grains} \end{aligned}$$

**1d**

$$\begin{aligned}
 V_{clouds} &= \frac{4\pi}{3} (15 \text{ pc})^3 = 1.41 \cdot 10^4 \text{ pc}^3 \\
 V_{gal} &= 1.41 \cdot 10^{11} \text{ pc}^3 \\
 n(H_2) &= 100 \text{ cm}^{-3} \rightarrow n(H) = 200 \text{ cm}^{-3} \\
 n(He) &= .1 \cdot 200 \text{ cm}^{-3} = 20 \text{ cm}^{-3}
 \end{aligned}$$

$$\begin{aligned}
 &\rightarrow \rho_{cloud} = n(H) \cdot m_H + n(He)m_{He} \\
 &200 \text{ cm}^{-3} \cdot (1.67 \cdot 10^{27}) + 20 \text{ cm}^{-3} \cdot 4 \cdot (1.67 \cdot 10^{27}) \\
 &= 4.7 \cdot 10^{-25} \text{ kg cm}^{-3} \rightarrow M_{cloud} = 4.7 \cdot 10^{-25} \text{ kg cm}^{-3} \cdot 1.41 \cdot 10^4 \text{ pc}^3 \\
 &\rightarrow 4.7 \cdot 10^{-25} \text{ kg cm}^{-3} \cdot 4.1 \cdot 10^{59} \text{ cm}^3 \\
 &= 19.3 \cdot 10^{34} \text{ kg} = 1.9 \cdot 10^{35} \text{ kg}
 \end{aligned}$$

The number of cloud  $N_{cloud}$  is

$$\begin{aligned}
 N_{cloud} &= \frac{.3 \cdot M_{gas}}{M_{cloud}} \\
 &= \frac{.3 \cdot 4 \cdot 10^9 \cdot 2 \cdot 10^{30}}{1.9 \cdot 10^{35}} \\
 &= 1.23 \cdot 10^4 \text{ or } \approx 12000 \text{ clouds}
 \end{aligned}$$