

ASTR 792 HW 3

Craig Brooks

Due: September 28 2023

Draine 1.3

$$n_H = .22 \text{ cm}^{-3}$$

$$v_{rel} = 26 \pm 1 \text{ km s}^{-1}$$

$$\frac{He}{H} = .1$$

$$F_{dust} = .005 m_{gas}$$

$$a = .1 \mu m = 10^{-5} \text{ cm}$$

$$\rho_{dust} = 2 \text{ g cm}^{-3}$$

First, let's compute the mass of a single particle of dust

$$\begin{aligned} m_{part} &= 2 \text{ g cm}^{-3} \cdot \frac{4}{3} \pi (10^{-5} \text{ cm})^3 \\ &= 8.38 \cdot 10^{-15} \text{ g} \end{aligned}$$

Next, let us calculate the density of the gas

$$\begin{aligned} \rho_{gas} &= n_H m_H + n_{He} m_{He} \\ .22 \text{ cm}^{-3} \cdot 1.67 \cdot 10^{-24} \text{ g} &+ .022 \text{ cm}^{-3} \cdot 4 \cdot 1.67 \cdot 10^{-24} \text{ g} \\ &= 5.14 \cdot 10^{-25} \text{ g cm}^{-3} \end{aligned}$$

The density of the dust is therefore

$$\begin{aligned} \rho_{dust} &= .005 \cdot \rho_{gas} \\ &= .005 \cdot 5.14 \cdot 10^{-25} \text{ g cm}^{-3} \\ &= 2.57 \cdot 10^{-27} \text{ g cm}^{-3} \end{aligned}$$

The number density for the dust then becomes

$$\begin{aligned} n_{dust} &= \frac{2.57 \cdot 10^{-27} \text{ g cm}^{-3}}{8.38 \cdot 10^{-15} \text{ g}} \\ &= 3.05 \cdot 10^{-13} \text{ cm}^{-3} \end{aligned}$$

Therefore we can calculate the area by

$$\begin{aligned}
 1/hr &= n_{dust} A v_{rel} \\
 \frac{1}{3600 \text{ s}} &= 3.05 \cdot 10^{-13} \text{ cm}^{-3} \cdot \frac{2.6 \cdot 10^6 \text{ cm}}{\text{s}} A \\
 &= 350 \text{ cm}^2
 \end{aligned}$$