

Decompression wave speed in CO₂ mixtures: CFD modelling with the GERG-2008 equation of state



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HIGHLIGHTS

- CFD models for decompression simulation of CO₂ mixtures.
- Incorporation of GERG-2008 EOS into CFD code for decompression modelling.
- Predicted decompression wave speed validated by measurements.
- Studies of effects of initial temperature and impurities on decompression wave speed.

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ABSTRACT

The development of CO₂ pipelines for Carbon Capture and Storage (CCS) raises new questions regarding the control of ductile fracture propagation and fracture arrest toughness criteria. The decompression behaviour in the fluid must be determined accurately in order to estimate the proper pipe toughness. However, anthropogenic CO₂ may contain impurities that can modify the fluid decompression characteristics quite significantly. To determine the decompression wave speed in CO₂ mixtures, the thermodynamic properties of these mixtures must be determined by using an accurate equation of state. In this paper we present a new decompression model developed using the Computational Fluid Dynamics (CFD) package ANSYS Fluent. The GERG-2008 Equation of State (EOS) was implemented into this model through User Defined Functions (UDF) to predict the thermodynamic properties of CO₂ mixtures. The model predictions were in good agreement with the experimental data of two 'shock tube' tests. A range of representative CO₂ mixtures was examined in terms of the changes in fluid properties from the initial conditions, with time and distance, immediately after a sudden pipeline opening at one end. Phase changes that may occur within the fluid due to condensation of 'impurities' in the fluid were also investigated. Simulations were also conducted to examine how the initial temperature and impurities would affect the decompression wave speed.

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1. Introduction

The burning of fossil fuels and biomass continues to be the main source of energy worldwide [1,2]. Such processes emit significant quantities of Green House Gases (GHG), particularly Carbon Dioxide (CO₂), which has been identified as the major contributor to global warming and climate change [3,4]. Carbon Capture and Storage (CCS) technology was introduced as a key CO₂ abatement option to mitigate emissions of GHG by 50% by 2050, while populations and economies are expected to continue to grow globally [5]. This technology will necessitate substantial quantities of CO₂

to be conveyed, predominantly by pipelines, over long distances from source to storage sites [6]. In terms of operational and economic motivations, the best way to transport CO₂ mixtures via pipes will be in a liquid and/or supercritical state because a purely gaseous phase transmission would necessitate significantly larger diameter pipelines for the same mass flow rate [7,8]. Under these operational conditions, the possibility of running fractures in the pipeline is a major concern, so arresting and/or preventing them is important for the integrity and safety of the pipeline's operation [5,7].

Fracture propagation in gas pipelines is commonly treated using the semi-empirical Battelle Two-Curve Model (BTCM) [9,10] where the aim is to estimate the required toughness to arrest crack propagation. This method involves the superposition

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Nomenclature

c	speed of sound (m s^{-1})
E	fluid energy (kJ)
h	enthalpy (kJ kg^{-1})
M	molecular weight (kg)
p	pressure (Pa)
s	entropy ($\text{kJ kg}^{-1} \text{K}^{-1}$)
t	time (s)
T	temperature (K)
u	outflow velocity (m s^{-1})
P_i	initial pressure (Pa)
T_i	initial temperature (K)
V_c	fracture velocity (m s^{-1})
W_{ave}	average decompression wave speed (m s^{-1})
W_{exp}	measured decompression wave speed (m s^{-1})
W_{local}	local decompression wave speed (m s^{-1})
k_{eff}	effective thermal conductivity ($\text{W m}^{-1} \text{K}^{-1}$)
c_p	specific heat ($\text{kJ kg}^{-1} \text{K}^{-1}$)
ρ	fluid density (kg m^{-3})
v_x	axial velocity (m s^{-1})
v_y	radial velocity (m s^{-1})
μ	dynamic viscosity (Pa s)

Abbreviations

2D	two-dimensional
AGA	American Gas Association
AS	Australian Standard
AUSM	Advection Upstream Splitting Method
BTCM	Battelle Two-Curve Model
BWRS	Benedict–Webb–Rubin–Starling
CCS	Carbon Capture and Storage
CFD	Computational Fluid Dynamics
CO_2	Carbon dioxide
CVN	Charpy V-Notch
EOS	equation of state
FDM	Finite Difference Method
FVM	Finite Volume Method
GERG	Groupe Européen de Recherches Gazières
GHG	Green House Gases
ID	internal diameter
UDF	User Defined Functions
MOC	Method of Characteristics
PR	Peng–Robinson
RKS	Redlich–Kwong–Soave

of two independently determined curves: the fluid decompression wave speed and the fracture propagation speed (the ‘J curve’), each expressed as a function of pressure. Fig. 1 shows a schematic representation of the BTCM. The shape of the fluid decompression wave speed curve depends on the phase of the fluid, as shown by the red and green curves in Fig. 1. Curves 1, 2, and 3 represent the fracture speed curves for different toughness values.

When fracture curves 2 and 3 intersect with the two-phase decompression characteristic, the fracture and the gas decompression wave move at the same speed, but here the gas pressure at the tip of the fracture no longer decreases, implying that the fracture will continue to propagate. The boundary between arrest and propagation of a running fracture is represented by tangency between the gas decompression wave speed curve and the fracture speed curve (curve 1 with the two-phase decompression wave speed curve and curve 3 with the single phase decompression wave speed curve). According to the BTCM, the minimum toughness required to arrest the propagation of fracture is the value of toughness corresponding to this tangency condition [9,11].

Several numerical models have been proposed to predict the decompression wave speed, mainly in natural gas mixtures. One of these models is GASDECOM [12]. This model uses an analytical

expression for the propagation of an infinitesimal decompression front to determine the decompression wave speed. The main assumptions in such models include: one-dimensional, frictionless, isentropic, and homogeneous-equilibrium fluid flow. GASDECOM uses the Benedict–Webb–Rubin–Starling (BWRS) equation of state (EOS) [13] with modified constants to estimate the thermodynamic properties during isentropic decompression. GASDECOM has suffered from numerical instabilities when dealing with mixtures containing higher fractions of CO_2 . It should be mentioned that the instabilities are due to the implementation in the code, and are not fundamental. GASDECOM cannot be used for mixtures containing hydrogen, oxygen and argon, which are often mixed with CO_2 in CCS-related operations. This is because these components were not originally included in the BWRS EOS [14,15]. Several other models also use assumptions similar to GASDECOM [16], and only differ in the choice of EOS. DECOM [17] was developed to predict the decompression wave speed in CO_2 mixtures, and is also based on assumptions similar to those in GASDECOM. The only difference was use of the NIST Standard Reference Database 23 (REFPROP version 9.0) [18], along with the built-in Span and Wagner EOS [19] for pure CO_2 and the GERG-2004 EOS [20] for multi-component CO_2 mixtures.

Other more complex decompression models that can account for non-isentropic effects using the Computational Fluid Dynamics (CFD) technique have been developed. Examples include: Picard and Bishnoi [21,22], PipeTech [23,24] and CFD-DECOM [25]. These are based entirely on assumptions of one-dimensional homogeneous-equilibrium fluid flow. In these models the effects of friction, heat transfer, and pipe diameter can be considered, which is particularly relevant for smaller diameter and longer pipelines where friction could lead to a range of complex effects on local flow conditions, temperature, and pressure within the pipeline [24,26–29]. CFD-based techniques involve discretising the governing partial differential equations of fluid flow. The Finite Difference Method FDM [30,31], the Method of Characteristics (MOC) [32], and the Finite Volume Method (FVM) [25] are examples of discretisation methods. The MOC solves the fluid flow conservation equations by following the Mach-line characteristics inside the pipe. It is claimed that numerical diffusion related to the finite difference

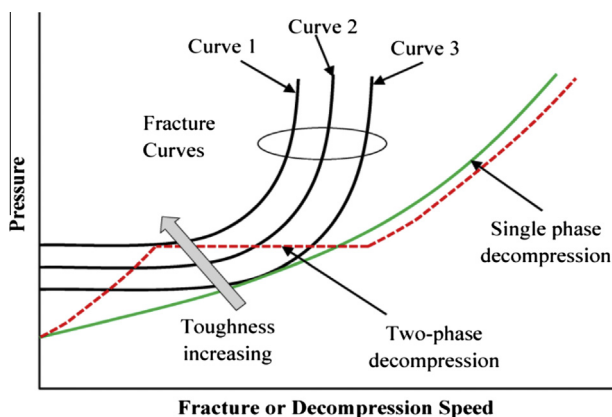


Fig. 1. Schematic of the BTCM [11].

approximation of partial derivatives is reduced by this method [33,34], but the MOC needs much longer computation runtimes and cannot predict non-equilibrium or heterogeneous flows [25,35], while the FVM is better at dealing with multi-dimensional flow. In the existing CFD models, the cubic Peng–Robinson (PR) EOS [36] is often used due to its relatively simple mathematical form compared to other more complex (but more accurate) EOS such as AGA-8 [37], BWRS [13] and GERG [20].

To accurately predict the decompression behaviour of CO₂ mixtures, accurate means of predicting the thermodynamic properties of these mixtures using accurate EOS is essential. To date, no EOS is specifically recommended for CO₂ mixtures, but the ability to accurately predict the Vapour Liquid Equilibrium (VLE), density and speed of sound is considered the best way to gauge any weaknesses or strengths of EOS [2,8,38,39]. Li et al. [2,8] have evaluated eight cubic EOS, including Peng–Robinson (PR) [36], Patel–Teja (PT) [40], Redlich–Kwong (RK) [41], Redlich–Kwong–Soave (SRK) [42], modified SRK (MRK) [43], modified PR (MPR) [44], 3P1T EOS [45], and Improved SRK (ISRK) [46], in terms of predicting the VLE and specific volumes of binary CO₂ mixtures containing CH₄, H₂S, SO₂, Ar and N₂, based on the comparisons with experimental data. Generally, PR is recommended for calculations involving CO₂/CH₄ and CO₂/H₂S; PT is recommended for CO₂/O₂, CO₂/N₂ and CO₂/Ar; 3P1T is recommended for CO₂/SO₂. Liu et al. [47] have implemented the PR EOS into ANSYS Fluent using real gas User-Defined Functions (UDFs) in order to simulate the dispersion of pure CO₂ releases from high-pressure pipelines. Reasonable results were obtained when using the real gas models in conjunction with the CFD method. Botros [48,49] conducted a comparative study of five different EOSs: GERG, AGA-8, BWRS, PR and Redlich–Kwong–Soave (RKS) and compared the predicted densities in the dense phase region using those EOS with measured values for different hydrocarbon mixtures. It was determined that the GERG EOS outperformed the other EOS in the region up to $P = 30$ MPa and $T > -8$ °C. However, the GERG EOS has not been implemented in CFD models of decompression or outflow models to date, though it is currently the reference EOS for natural gas [50].

In this paper we present a CFD model for a full-bore depressurisation of a CO₂ mixture pipeline developed using the versatile CFD software ANSYS Fluent (v 14.5). The built-in EOS in ANSYS Fluent cannot predict the fluid properties of CO₂ mixtures accurately. The GERG-2008 EOS was successfully implemented into ANSYS Fluent to accurately predict the thermodynamic properties of CO₂ mixtures, for the first time. The method used to implement the GERG-2008 EOS into ANSYS Fluent is described. The results were validated against experimental data from two separate ‘shock tube’ tests, and a number of simulations were also conducted to examine the effect of different initial conditions and different components in the CO₂ mixture.

2. Methodology

The CFD package ANSYS-Fluent was used to develop the CFD decompression model because it satisfies the three main demands required for gas decompression analysis:

- Ability to solve transient flows.
- Possibility of invoking an accurate EOS through user-defined subroutines.
- Ability to handle multi-dimensional geometries.

2.1. Computational domain and boundary conditions

The physical flow domain in the shock tube tests consisted of the initially pressurised gas in a horizontal pipe, which undergoes a ‘full-bore’ opening at one end using a rupture disc as schemati-

cally depicted in Fig. 2. The axial symmetry made it possible to construct a two-dimensional computational domain and thus reduce the computational runtime.

The following assumptions are made to develop this model: unsteady, two-dimensional flow; the rupture is instantaneous and represented by a full bore opening, non-isentropic flow conditions (the friction effect is considered); the gas velocity before the rupture of the pipe is negligible compared with the conditions post-rupture; the fluid is considered homogenous so equilibrium conditions prevail during condensation; and the ‘no slip’ condition is satisfied at the pipe wall.

Four boundary conditions were defined: two ‘wall’ boundaries defined at the top ($y = D/2$) and the end ($x = L$) of the computational domain; a ‘symmetry’ boundary (at $y = 0$) on the axis and a pressure outlet (zero gauge pressure) to model the rupture disk (sudden opening to the ambient) at $x = 0$. Based on the above assumptions, the unsteady, two-dimensional form of the governing differential equations of conservation of mass, momentum and energy are solved in this model.

2.2. Numerical method

The FVM is used in ANSYS Fluent to discretise the fluid conservation equations. The implicit first order spatial and temporal formulations were used with the Advection Upstream Splitting Method (AUSM) for the density-based solver [51]. This solver is designed for high-speed compressible flows and allows the use of a user defined real gas model. The governing flow equations of mass, momentum, and energy conservation, supplemented by the auxiliary equation (i.e. EOS) were solved simultaneously while the turbulence equations were treated sequentially. In the density-based solver, the momentum equations were used to obtain the velocity field, while the continuity equation was used to determine the density field and the pressure field was determined from the EOS.

The computational grid conformed to the physical dimensions of the shock tube used in the tests. A ‘symmetry’ boundary condition was used on the axis. At the rupture end ($x = 0$), the fluid was considered to be exposed to ambient pressure at time $t = 0^+$. A no-slip wall was set as the boundary condition for the closed end of the pipe and the pipe wall. Adjacent to both wall boundaries 5 cells were generated to span the boundary layer. The cell adjacent to the wall and the outlet was set at 0.05 mm from the wall with a mesh-growth factor of 1.25. Beyond the 5th cell, the dimensions of the cells (Δx) and (Δy) remained constant at 2 mm in both axial and radial directions. The initial conditions for the flow variables were prescribed based on the operating conditions of each shock tube test whereas the time step size was fixed to $1e^{-6}$ s. The above mesh and time step size were the best setting to obtain an accurate and converged solution. A detail of the mesh near the outlet is shown in Fig. 3.

The speed of the decompression wave was obtained by first calculating the *local* decompression wave speed using Eq. (1), i.e., by monitoring the speed of sound ‘ c ’ and the ‘outflow’ velocity ‘ u ’ against time during the decompression process. The decompression wave speed was then determined by subtracting the outflow velocity from the speed of sound for several pressures below the initial pressure.

$$W_{local} = c - u \quad (1)$$

However, experimental tests such as the shock tube test did not provide the *local* gas decompression wave speed directly because the gas decompression wave speed w was calculated by determining the times at which a certain pressure level was recorded at several pressure transducers at known locations on the pipe wall. By plotting these locations against time, the decompression wave speed was obtained by performing a linear regression of each

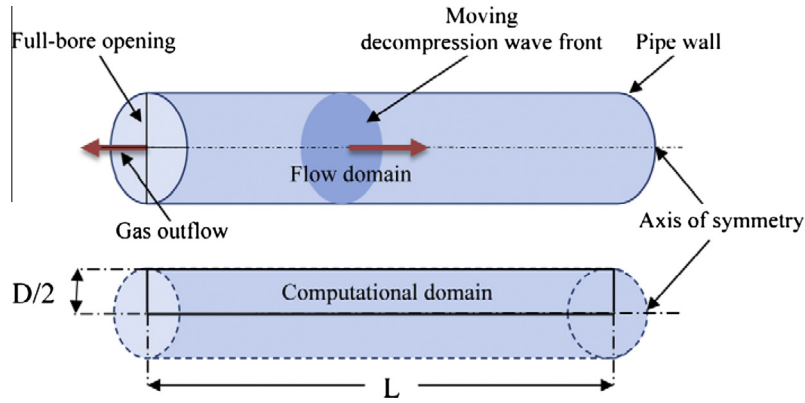


Fig. 2. Flow domain and computational domain – schematic.

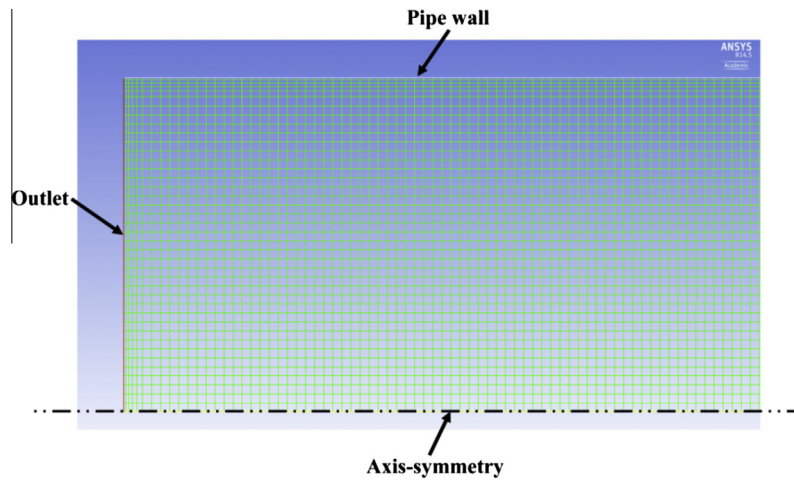


Fig. 3. Two-dimensional computational grid.

isobar curve. The slope of each regression represents the average decomposition wave speed for each isobar.

$$W_{ave} = \frac{dx}{dt} \quad (2)$$

3. Implementation of GERG-2008 EOS into ANSYS fluent

To simulate the real behaviour of gas flow, the thermodynamic properties must be predicted using an accurate real gas EOS. The modern multi-component GERG-2008 EOS [20,52] was used to provide the thermodynamic properties of CO₂ mixtures. This EOS covers the gas phase, liquid phase, supercritical region, and vapour-liquid equilibrium states for mixtures consisting of up to 21 components: methane, nitrogen, **carbon dioxide**, ethane, propane, n-butane, isobutane, n-pentane, isopentane, n-hexane, n-heptane, n-octane, hydrogen, oxygen, carbon monoxide, water, helium, argon, n-nonane, n-decane, and hydrogen sulphide. The normal range of validity of this EOS covers temperatures from 90 K to 450 K and pressures up to 35 MPa. Currently, GERG-2008 EOS is considered to be a reference EOS for natural gas pipelines [14].

The GERG-2008 EOS must be implemented in Fluent using a User-Defined Real Gas Model (UDRGM) using a library of functions written by the end user in the C programming language. These functions represent several thermodynamic properties required by Fluent to solve the system of governing equations. The thermodynamic properties required for Fluent calculation are shown in Table 1.

These properties were supplied to Fluent for given values of pressure and temperature, but because GERG-2008 cannot be programmed within the UDF, the exported functions and subroutines of the dynamic link library 'GERG-2008.DLL' [52] had to be defined within UDF instead. The EOS library is called to calculate the properties at each node in the flow domain. The cost of a direct call to the library during simulation can be a major limitation, and occasionally the library failed to produce some properties at certain P-T values and entered an infinite optimisation loop that caused the library to crash. Moreover, some properties (e.g. speed of sound) were not defined in the two-phase region, so an error was reported. Most modern multi-component EOSs suffer from this drawback. The most frequent error encountered during the simulated decompression was related to the speed of sound in the

Table 1
Thermodynamic properties required for a real gas model in ANSYS Fluent.

Property	Symbol
Density	ρ
Enthalpy	h
Entropy	s
Speed of sound	c
Specific heat at constant pressure	c_p
Molecular weight	M
Partial derivative of ρ w.r.t. T	$\partial\rho/\partial T$
Partial derivative of ρ w.r.t. P	$\partial\rho/\partial P$
Partial derivative of h w.r.t. P	$\partial h/\partial P$

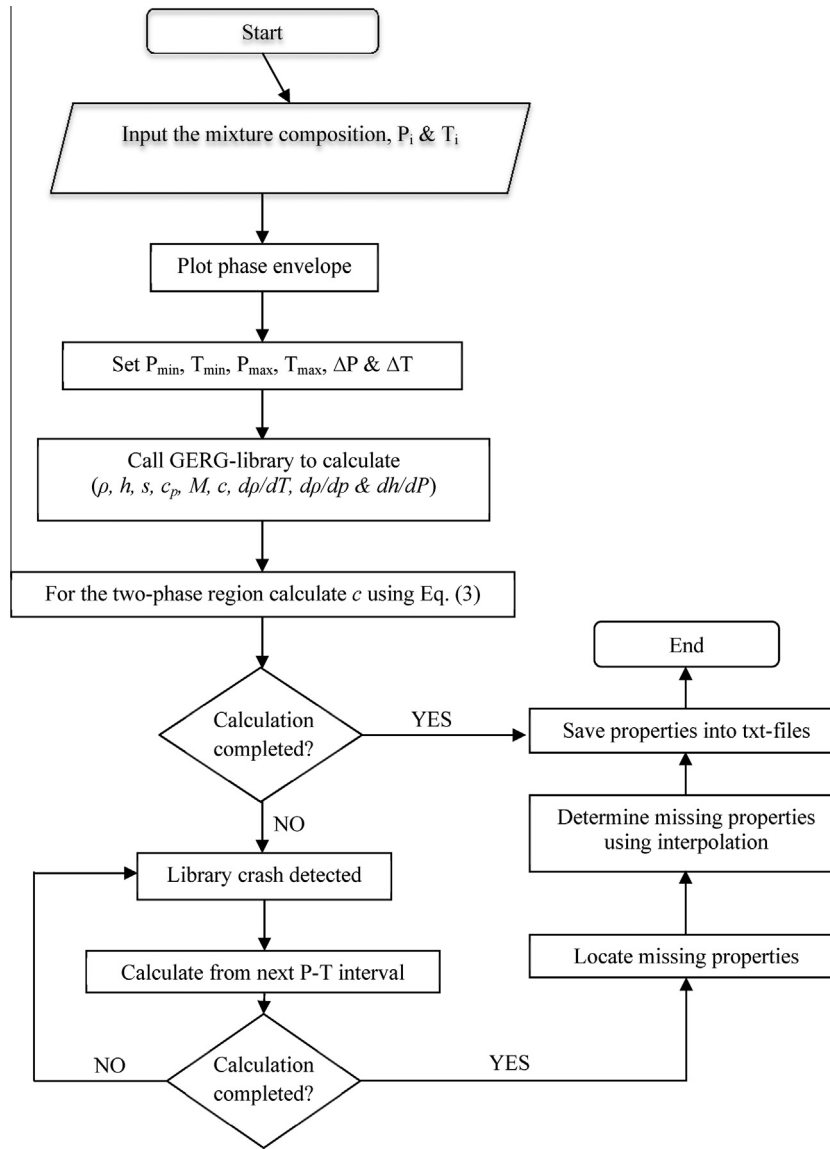


Fig. 4. Property calculation flow chart.

two-phase region. In this model we assumed a homogenous-equilibrium fluid, so the definition of the speed of sound for a single phase fluid could be used in the UDF to overcome the problem. The speed of sound in the two-phase region was defined as:

$$c = \sqrt{\left. \frac{dp}{d\rho} \right|_s} \quad (3)$$

Despite not always being able to calculate the requested property, the above obstacles did not mean the decompression wave velocity could not be accurately predicted. We circumvented those issues by using the EOS library indirectly such that reference to pre-compiled tables of the relevant thermodynamic properties generated by the GERG-2008 EOS replaced a direct call to the dynamic link library 'GERG-2008.DLL [53]'. A linear interpolation scheme was also implemented within the UDF to extract values of the other thermodynamic parameters based on the P - T values solved for by Fluent. This method has proved to be 300 times faster than direct calls to EOS [54] and could save up to 70% of the total computational run time [55]. In this study the performance of the UDF was tested using both methods and for all properties, the search in tables during the simulation was found to be about 20 times faster than a direct call to the library.

A structured two-dimensional array for the chosen ranges of pressures and temperatures was established. The initial conditions and the phase envelope were the key parameters used to establish the boundary of the main P - T table. The EOS library was called for each pressure-temperature node in the 2D-table to produce tables of the properties listed in Table 1. Where the EOS library failed to produce data, hole(s) were displayed in the corresponding table cell(s) and a code was developed to begin the calculation from the next P - T increment and complete the rest of the tables so the remaining properties were displayed normally. The corresponding gaps in the table grid were then filled using interpolation based on the values at the neighbouring nodes. The calculated properties were then saved into readable files linked to ANSYS Fluent through the UDF as LOOK-UP tables. Fig. 4 shows schematically the computing strategy of fluid properties using the GERG-2008 library.

4. Model results and validation

The following paragraphs present the results of the 2D CFD decompression model using the GERG-2008 EOS. This model was validated by a comparison with the results of two separate shock tube tests. The first test (Case A) was conducted at the TransCanada

Table 2

Model parameters setting for the current study.

Case	Pipe length (m)	Diameter (mm)	Surface roughness (μm)	Turbulence model
Case A	42	38.1	Smooth	Realisable $k-\epsilon$
Case B	144	146.36	5	Realisable $k-\epsilon$

pipeline Gas Dynamics Test Facility in Didsbury, Alberta, Canada [56]. The second test (Case B) was commissioned by the National Grid at GL Noble Denton's Spadeadam Test Site in Cumbria, UK [17]. In the first test, the main section of the shock tube was 42 m long, the internal diameter (ID) was 38.1 mm and the tube wall thickness was 11.1 mm. In the second test the pipe was 144 m long, the ID was 146.36 mm, and the pipe wall thickness was 10.97 mm. In Case A, a 'smooth' pipe surface was used, while in Case B the pipe has an average surface roughness ranging between 5 and 6.3 μm . The smoothest pipe was placed nearest the rupture disk. Table 2 lists model parameters used in the current simulations.

CFD simulations were carried out for two mixtures: a binary mixture for Case A and a 5-component mixture for Case B. Table 3 shows the gas compositions and initial conditions of the two tests.

A mesh-dependence study was carried out for both cases using several element sizes (2, 3, 5, 10, 20, 50, 100 mm). An optimum element size was found to be 2 mm, although for decompression wave speed calculation, an element size up to 10 mm was found acceptable.

The pressure and temperature were monitored as a function of time at several locations along the axial direction, near the exit plane. These locations corresponded to where the pressure transducers and temperature probes were in the shock tubes tests. Other properties such as the speed of sound and 'outflow' velocity were monitored at the same locations to determine the local decompression wave speed. Table 4 shows the locations of pressure and temperature transducers mounted on both shock tube tests. The highlighted cells in Table 4 represent locations used for the determination of decompression wave speed.

The thermodynamic properties of each mixture were first produced using the GERG-2008 EOS and then saved into readable files. Table 5 shows the structure of the P - T table established for the mixture in Case A. The properties were calculated for all P - T nodes in the Table. Note that the minimum and maximum values of P and T in the main table will vary depending on the initial conditions and phase envelope of each mixture.

A MATLAB code was written to generate plots of the required properties as a function of pressure and temperature. The calculated properties for Case A are presented in Fig. 5. A smooth distribution was observed for all properties, including the region under the two-phase boundary. This occurred because the main P - T table was made dense enough to account for changes near the phase boundary. This makes for very large files, but it ensured that the calculations were accurate. An acceptable accuracy was achieved using the property tables: the interpolated properties deviated from values obtained directly using the EOS library by approximately 0.001% outside the two-phase region, and 0.1% within the two-phase region.

Table 3

Mixture composition and initial conditions of shock tube tests.

Shock Tube Test	Mixture components (mole %)					P_i (MPa)	T_i (K)
	CO ₂	H ₂	N ₂	O ₂	CH ₄		
Case A	72.6	0	0	0	27.4	28.568	313.65
Case B (T31)	91.03	1.15	4	1.87	1.95	14.95	283.15

Table 4

Monitor point locations.

TEST 1 (Case A)		TEST 2 (Case B)	
Location	Distance from rupture disc (m)	Location	Distance from rupture disc (m)
PT1	0.0295	P2	0.0864
PT1A	0.0924	P4, T4	0.34
PT1B	0.1028	P6	0.54
PT2	0.2	P8	0.74
PT3	0.35	P10	0.94
PT4	0.5	P12	1.24
PT5	0.7	P14, T14	1.84
PT6	0.9	P16	2.44
PT7	1.1	P18	3.64
PT8	3.1	P19	4.84
PT9	5.1	T20	6.04
PT10	7.1	P21	9.04
PT11	9.1	P22	13.54
PT12	13	T23	18.04
PT13	19	P24	22.54
PT14	25	T25	30.04

Table 5 P - T table.

	Pressure (MPa)	Temperature (K)
Min	0.05	180
Max	30	320
Increment	0.1	0.5
No. of nodes	300	281

Decompression of the mixture in Case A was simulated first, with a flow domain compatible with the shock tube test described in [56]. As Fig. 6 shows, the simulated pressure-time histories compared well with the measurements at different points near the exit, but as the decompression wave front reached each location, the pressure at each point dropped rapidly before levelling off at about 9 MPa. There was a slight discrepancy between the measured and predicted pressure at pressures between 27 and 26 MPa. Apart from that, the predicted change in pressure agreed satisfactorily with the experimental results.

Fig. 7 shows the transient behaviour of the fluid temperature at the four locations closest to the outlet boundary (rupture disc). The variations in the speed of sound and the outflow velocity are shown in Fig. 8 and Fig. 9 respectively. The forms of the pressure-time and temperature-time curves were similar. The fluid temperature suddenly dropped from its initial value to 276 K. The temperature remained steady at this value for several time steps, creating a temperature plateau, before continuing to drop steadily.

The predicted speed of sound at the initial conditions was 516.28 m/s. Fig. 8 shows that the speed of sound gradually decreased to a value close to 258 m/s and then dropped to their lowest level of 105 m/s.

Before the rupture disk ruptured, the entire body of gas in the pipeline was at rest. In the simulation, as the outlet

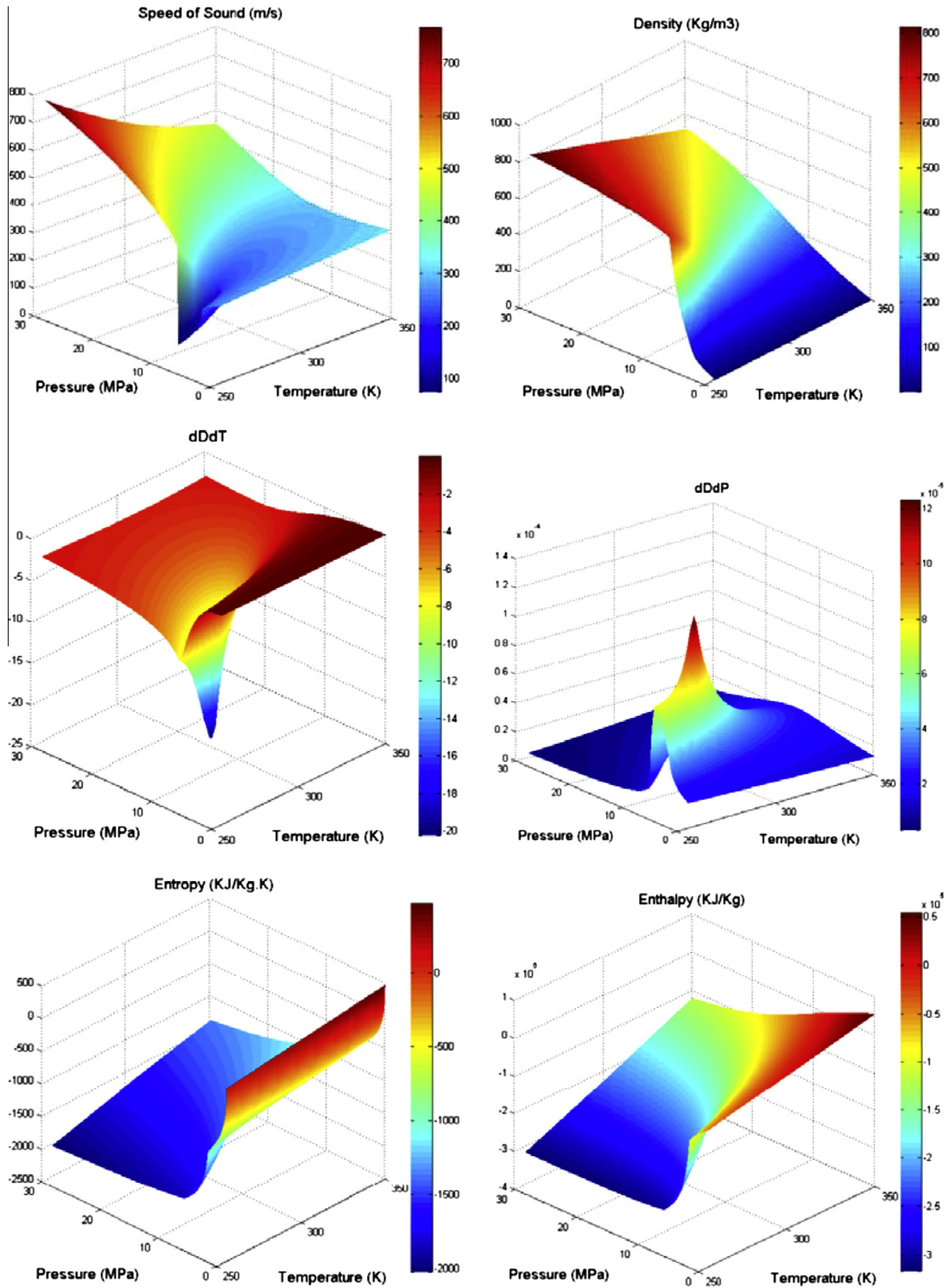


Fig. 5. 3-D plots of thermodynamic properties calculated by GERG-2008 (Case A).

boundary was subjected to ambient pressure at time $t = 0+$, an expansion (decompression) wave was set off. As the wave propagated away from the opening, the exit velocity was seen to

increase. Like the other properties, the outlet velocity remained steady for a short time at 85 m/s before continuing to increase again.

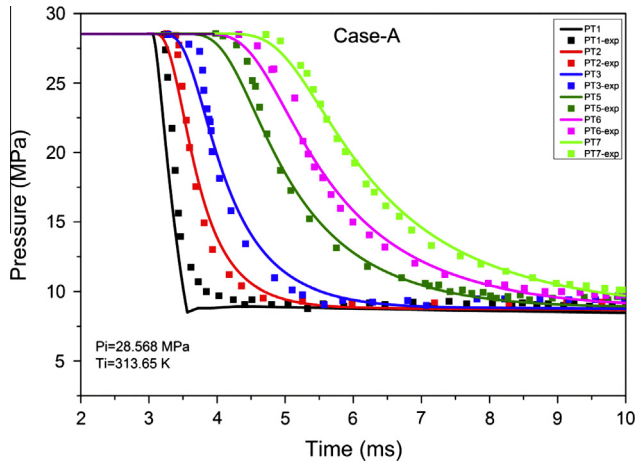


Fig. 6. Comparison between predicted and measured pressure-time traces (Case A).

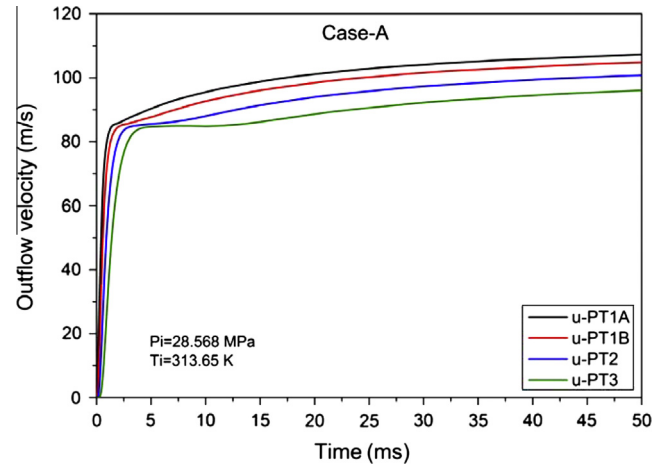


Fig. 9. Predicted 'outflow' velocity versus time (Case A).

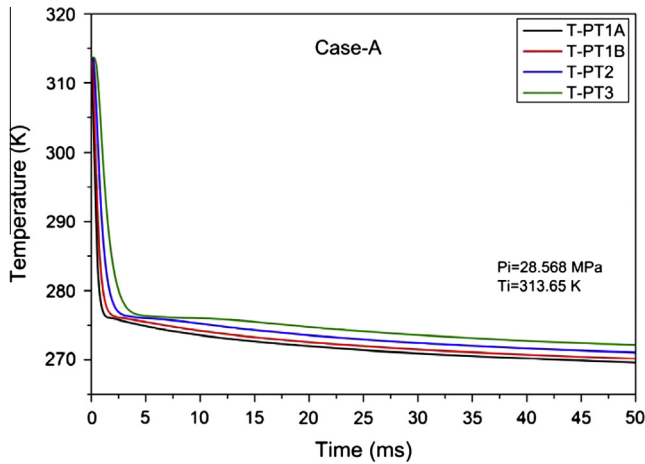


Fig. 7. Predicted fluid temperature versus time (Case A).

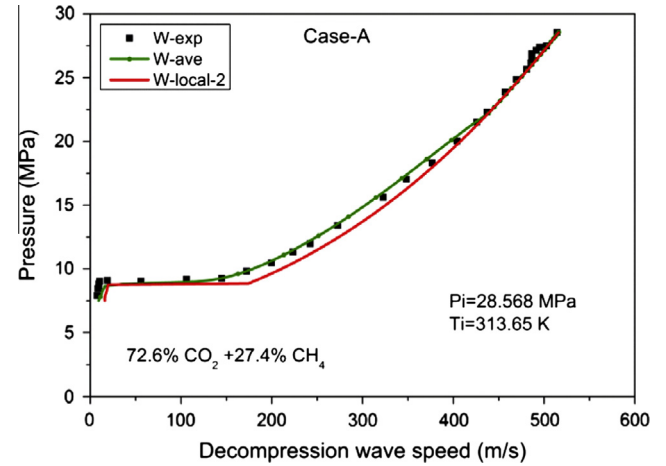


Fig. 10. Comparison of the predicted and the measured decomposition wave speed (Case A).

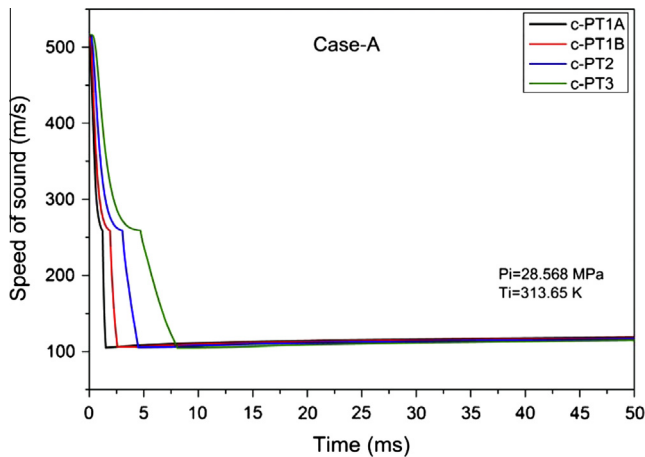


Fig. 8. Predicted speed of sound vs time (Case A).

Fig. 10 shows a comparison between the predicted and experimentally obtained decomposition wave speed. The predicted average decomposition wave speed was obtained based on readings at the 6 pressure transducers listed in Table 3, whereas the local decomposition wave speed was determined using the predicted

speed of sound and the 'outflow' velocity at 200 mm from the exit. Initially (before the flow commenced), the speed of the decomposition wave was equal to the predicted speed of sound in the mixture because the 'outflow' speed was zero. The model predicted the initial decomposition wave speed well, differing by only 0.4% from the measured data. As the pressure decreased the predicted average decomposition wave speed agreed with the measured data, while the local decomposition wave speed varied slightly to the right of the experimental curve because the 'local' decomposition wave speed was obtained using the formulation in Eq. (1), while the average decomposition wave speed was calculated using a similar approach to the measured data (based on the pressure-time traces).

More importantly, the abrupt drop in the measured decomposition wave speed curve which created a long pressure plateau was predicted successfully. According to the BTCM, an accurate determination of the pressure plateau in the decomposition wave speed curve is crucial to guarantee an accurate prediction of the required arrest toughness. The current model under-predicted the plateau level slightly. As seen in Fig. 6, a discrepancy is noticed on the predicted pressure-time curves at the same pressure level. The reason for the discrepancy and its influence is discussed later.

The appearance of the plateau can be explained by superimposing the pressure-temperature gradient on the phase envelope as depicted in Fig. 11. As the fluid crosses the phase boundary (at

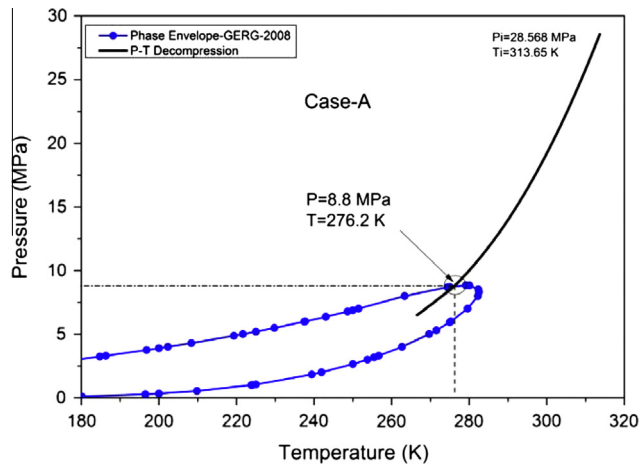


Fig. 11. The pressure–temperature curve and the phase envelope (Case A).

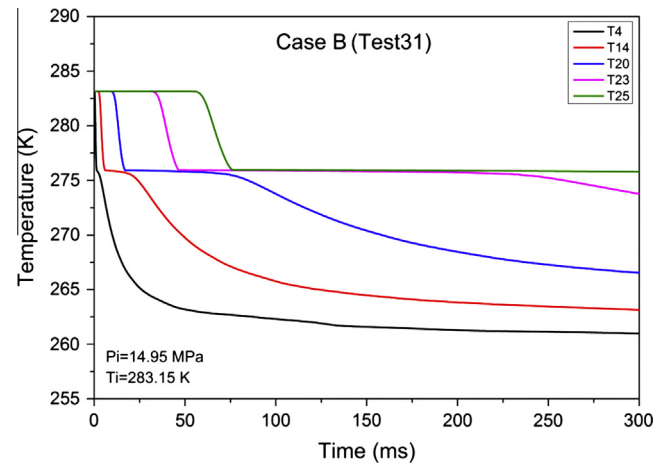


Fig. 13. Predicted temperature–time traces (Case B).

$T = 276 \text{ K}$, $P = 8.8 \text{ MPa}$), the decompression wave speed experiences a sharp drop which can be attributed to the drop in the speed of sound, while simultaneously the monitored properties remained constant for several time steps. Clearly, the trend that appeared in all properties stemmed from the discontinuity at the phase boundary. Such outcomes demonstrate that the current CFD model can successfully deal with the phase change predicted implicitly in the property tables.

The second simulation was for the mixture in Case B. The computational domain here was based on the physical dimensions of the shock tube test described in [17]. Fig. 12 shows the CFD prediction of pressure–time traces at 8 different pressure transducer locations along the pipe. A rapid drop in pressure occurred as the decompression wavefront passed each location. The appearance of a plateau at about 8 MPa can be ascribed to the phase change that occurred due to the decompression process.

Fig. 13 shows the drop in fluid temperature as a function of time at five different locations on the tube. The temperature dropped rapidly from its initial value before flattening out for several time steps at 277 K, creating a plateau in all curves. After this stage, the temperature steadily decreased to its lowest value of 260 K which is predicted at the closest location towards the rupture disc. A comparison with Fig. 14 shows that the plateaus occurred at the

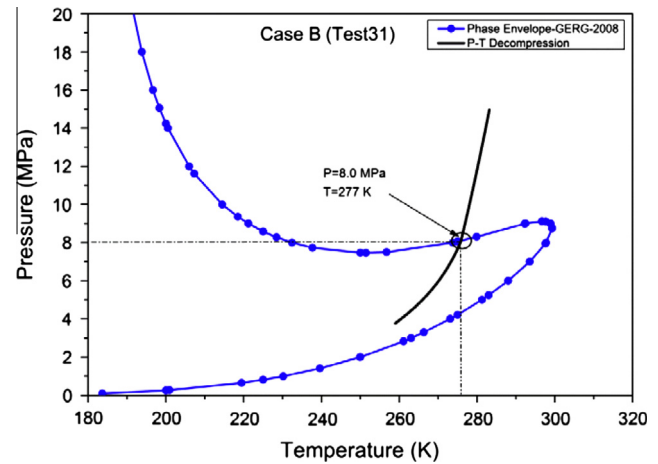


Fig. 14. The decompression of pressure–temperature compared to phase envelope (Case B).

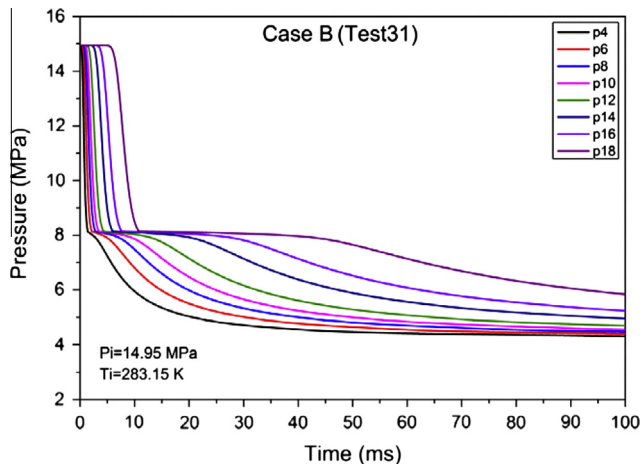


Fig. 12. Predicted pressure–time traces (Case B).

same pressure level as the point of intersection of the pressure–temperature curve with the phase boundary.

The speed of sound and the outflow velocity were both predicted in order to obtain the local decompression wave speed. The predicted speed of sound versus time for five locations close to the outlet is shown in Fig. 15, while the predicted outflow velocity is shown in Fig. 16. At the initial pressure and temperature, the current model predicted the speed of sound as 522 m/s, while the outflow velocity was 0 m/s anywhere inside the tube (before flow commenced). A similar trend that occurred in the outflow velocity of Case A occurred here where a kink appeared on all the curves due to phase change. Referring back to the speed of sound curves, the phase change caused a decrease in the speed of sound, and this overall drop in speed of sound due to discontinuity at the phase boundary was $\sim 350 \text{ m/s}$.

Fig. 17 shows a comparison between the predicted and experimentally obtained decompression wave speed of Case B where the initial decompression wave speed predicted by the current model was 521 m/s. This value deviated by approximately +2.4% from the measured result, but the predicted decompression wave speed was consistent with the experimentally obtained value for pressure levels above and below the plateau level. At the plateau there was a discrepancy between the predicted and measured decompression wave speed even though the plateau began to form close

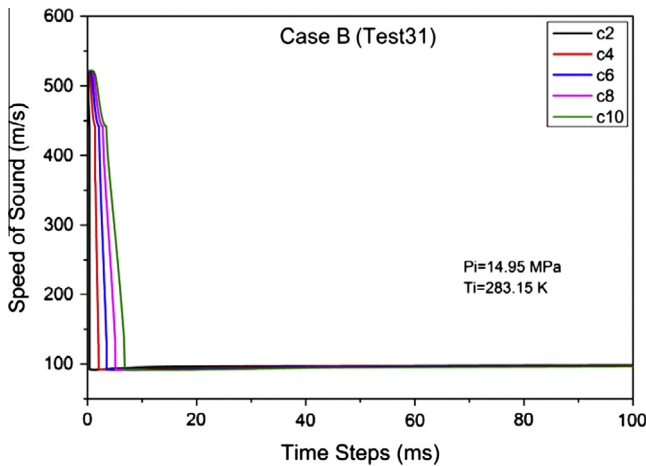


Fig. 15. The predicted speed of sound versus time (Case B).

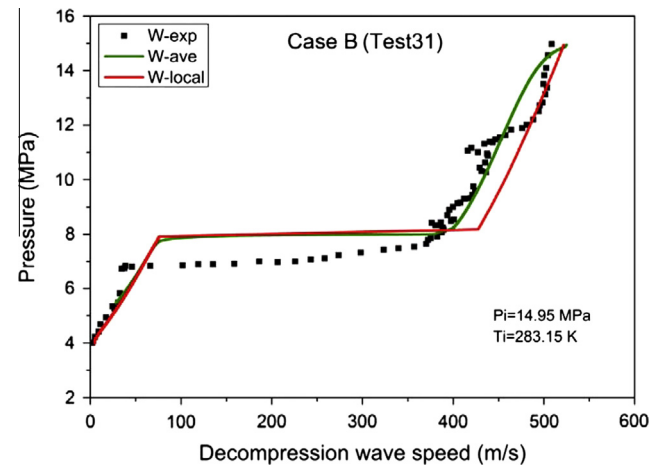


Fig. 17. Comparison of the predicted decomposition wave speed with the measured results (Case B).

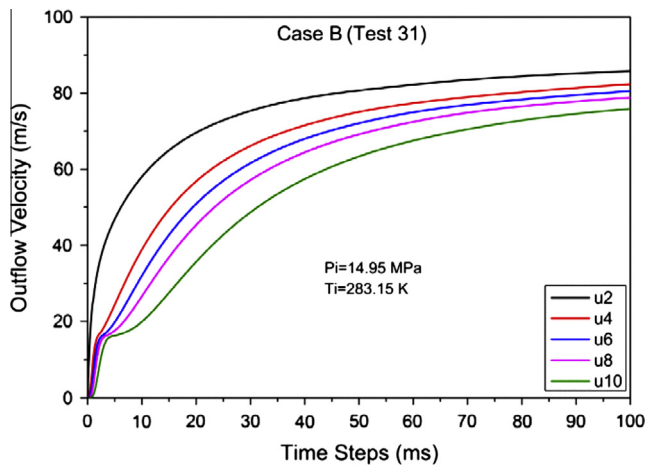


Fig. 16. The predicted outflow velocity versus time (Case B).

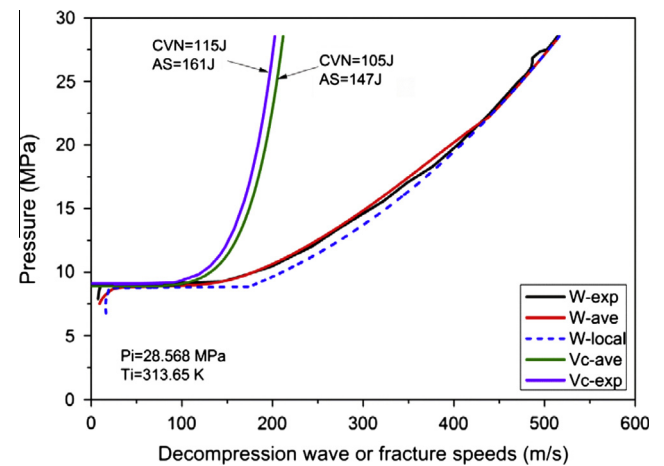


Fig. 18. Arrest toughness prediction for Case A.

to the pressure level of the measured data. Notably, the length of the predicted plateau in the average decompression wave speed curve was consistent with the measured data. Further discussion will be made hereafter.

5. Discussion

If the variation in the simulated pressure matches the experimental results (Fig. 6), the predicted average value of the decompression wave speed W should agree with the measured curve (Fig. 10), but as Fig. 10 shows, there was a slight discrepancy at the plateau between the predicted and experimentally obtained decompression wave speed. This variation appeared at the same pressure levels on the pressure–time curves, as Fig. 6 shows. There was major difference at the plateau level on the decompression wave speed in the second case, as Fig. 17 shows. Such a variation may result from uncertainties inherent in the numerical method and/or the way of implementing the GERG-2008 EOS, although factors such as delayed nucleation and/or rapid phase change dynamics (not considered here) can influence the results to various degrees. Another possible reason for this discrepancy was the actual amount of impurities in the experimental tests which could be slightly different from the listed composition.

The speed of sound in the current model can be tracked as a function of time so its relationship with the decompression wave

speed can be clearly understood. For instance, Fig. 17 shows that the ‘length’ of the pressure plateau ($\sim 348 \text{ m/s}$) was almost equal to the sharp drop in the speed of sound due to the phase change, as seen in Fig. 15.

Figs. 10 and 17 show long pressure plateaus that correspond to a significant drop in the decompression wave speed. This would surely influence the ductile fracture propagation control, as outlined in the BTCM. An example is shown in Fig. 18, where the BTCM was used to predict the CVN value of pipe, grade 480 (X70). The diameter and wall thickness of the pipe was 609.6 mm and 19.1 mm respectively. Based on the predicted average decompression wave speed, the corresponding CVN was $\sim 105 \text{ J}$ while the CVN value based on the experimentally determined decompression wave speed was $\sim 115 \text{ J}$ [56]. The difference between prediction and measurement can be attributed to the difference in the plateau level in the decompression wave speed, because the current CFD model slightly under-predicted the pressure plateau level.

For modern higher grade steels, if the predicted CVN value is greater than $\sim 95 \text{ J}$ [57], then the CVN value should be corrected using a certain correction factor to match the results of full-scale burst tests [58,59]. The Australian Standard (AS 2885.1), states that the predicted toughness should be multiplied by a factor of at least 1.4. Fig. 18 shows the decompression wave speed and the fracture propagation speed as functions of pressure. By applying the correction factor, the predicted CVN becomes 147 J whereas the

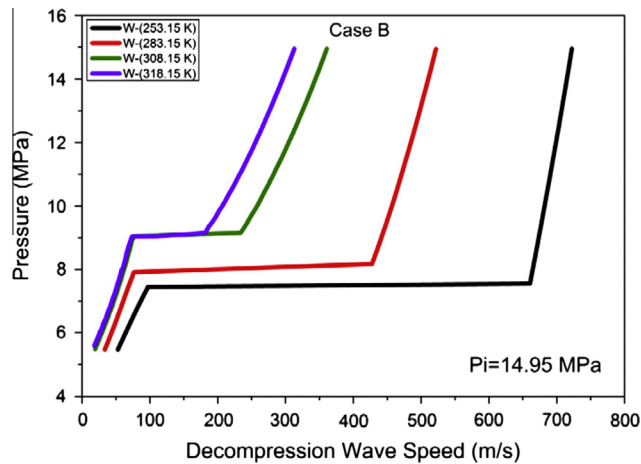


Fig. 19. Initial temperature effect on decompression wave speed (Case A).

measured value was 161 J. Note that the accuracy of the plateau level in the decompression wave speed was within ± 0.1 MPa, the size of the pressure step used in the calculation W_{ave} .

The pressure plateau level which represents the consequence of phase change on decompression wave speed is an important aspect in determining the required fracture toughness to suppress ductile fracture propagation, so investigating factors that could be sensitive to accurately predict the plateau in decompression wave speed was essential. Further simulations were performed to discuss the influences of initial temperature and impurities on the decompression of CO_2 mixtures.

5.1. The effect of initial temperature

The influence of initial temperature on the decompression of CO_2 mixture was examined for Case B. Three different initial temperatures (-20 , 35 and 45 °C) were used while the initial pressure remained the same as the actual case. These temperatures represent three different phases: liquid, dense liquid and supercritical. Fig. 19 shows how changing the initial temperature affects the decompression wave speed. Because the initial temperature of Case B was 10 °C, the main effect of increasing the initial temperature (i.e. 35 and 45 °C) was decreasing the initial decompression wave speed from 521 to 360 and 312 m/s respectively, but lowering the initial temperature caused the initial decompression wave speed to increase to 722 m/s. Moreover, the length and level of the pressure plateaus were affected due to changing the initial temperature; increasing the initial temperature decreased the length of the plateau in the decompression wave speed, and vice versa. Those observations were consistent with the predicted results of pure CO_2 conducted by [60] and for mixtures e.g. [14,56]. However, this effect was different in terms of plateau levels for CO_2 mixtures because it depended on the shape of the bubble curve on phase envelope, which in turn depended on the amount and type of impurities in the CO_2 mixture.

Increasing the initial temperature to 35 and 45 °C raised the level of plateaus by a value of 1 MPa above the main test. Interestingly, as Fig. 19 shows, the apparent plateaus in these two cases occurred at approximately the same level. This can be further explained by representing the pressure–temperature profiles on the phase envelope of the mixture, as depicted in Fig. 20, but note that the phase change occurred at approximately the same pressure level despite different intercept temperatures with the phase boundary which were clearly due to the effect of impurities that rose up the bubble curve on the phase envelope. Such a situation cannot occur for pure CO_2 .

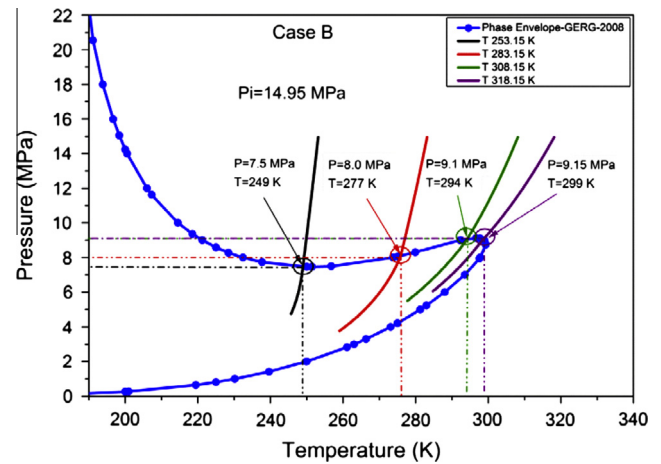


Fig. 20. Intersection points with the phase envelope for different initial temperatures (Case B).

Table 6

The initial conditions of the predominantly CO_2 mixtures.

Case No.	Mixture components (mol %)					P_i (MPa)	T_i (K)
	CO_2	H_2	N_2	O_2	CO		
Case1	95	5	0	0	0	15	283.15
Case2	95	0	5	0	0	15	283.15
Case3	95	0	0	5	0	15	283.15
Case4	95	0	0	0	5	15	283.15

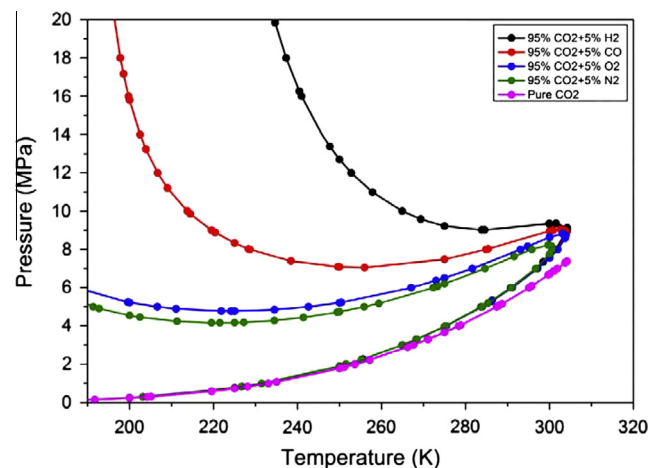


Fig. 21. Phase envelope of CO_2 mixtures calculated by GERG-2008 EOS.

Where the initial temperature was -20 °C, despite the initial decompression wave speed being much higher than in the main test, the plateau level was predicted at a lower pressure level than the main test by 0.5 MPa. Although this was consistent with the trend in the results of pure CO_2 conducted by [60], it cannot be taken as a role for CO_2 mixtures because of the shape of the phase boundary. For instance, if the initial temperature was less than $(-20$ °C), the intersection with the phase boundary would take place at a higher pressure levels because the bubble curve increased again at temperature level below that value. So the trend in the results of pure CO_2 which states that as the initial temperature decreases the plateau level in the decompression wave speed decreases cannot be applied for CO_2 mixtures.

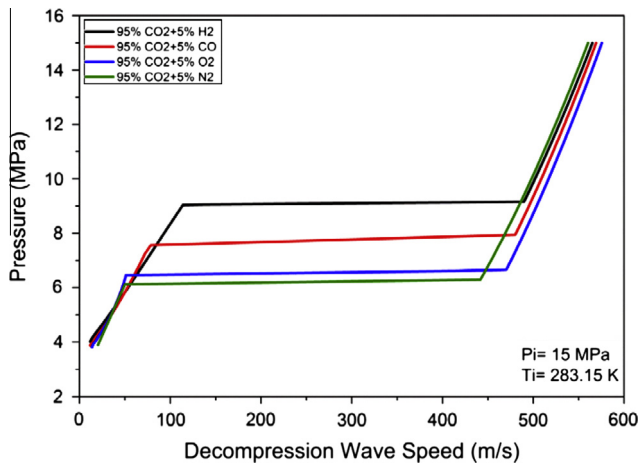


Fig. 22. Impurities effect on CO₂ decompression wave speed.

5.2. Influence of impurities

The effects of several impurities (components other than CO₂) on the decompression of CO₂ pipelines were examined. The impurities that were most likely to exist in carbon dioxide capture technologies were used [61]. Table 6 lists the four binary CO₂ mixtures studied, with the initial conditions.

Fig. 21 illustrates the effect of impurities on the phase envelope of CO₂, and show that adding impurities to pure CO₂ shifts the critical point and the bubble curve in the phase envelope. Notably, an addition 5% of hydrogen to the CO₂ had more effect on the phase equilibrium than the other impurities because it shifted the critical pressure to a value close to 10 MPa.

Simulations of decompression with these binary mixtures were conducted using the same flow domain as in Case A. Fig. 22 shows the influence on the decompression wave speed such that at the same initial conditions and for a fixed fraction of CO₂, each impurity resulted in a different initial decompression wave speed and different pressure plateau level that was clearly related to the phase envelope of the mixture. Adding 5% H₂ to the CO₂ resulted in the highest pressure plateau level (~9 MPa). Adding 5% N₂ resulted in a pressure plateau of about 6 MPa. These changes in the decompression wave speed could influence the fracture propagation/arrest requirements for CO₂ pipelines.

6. Conclusion

Transporting CO₂ mixtures by pipelines is a challenge. In order to improve our knowledge it is important for the modelling tools to handle CCS CO₂ mixtures efficiently. The feasibility of complex and possibly large simulations of fluid-pipe interactions, hydraulic transients and dispersion will otherwise be restricted. This paper has described a CFD model developed using ANSYS Fluent to simulate the decompression behaviour of CO₂ mixtures. For the first time ever, GERG-2008 EOS was successfully implemented into ANSYS Fluent using UDFs based on an indirect use of the GERG-2008 EOS library. This was done by using pre-compiled thermodynamic property tables ("lookup tables") linked to Fluent during simulation time. Several obstacles related to the EOS library were avoided using this method.

The predicted results were validated against two separate 'shock tube' tests. The results mostly agreed with the experimental results available. The following observations were made:

- The CFD model successfully tracked the rapid drop in pressure and accounted for the phase change during decompression.
- The decompression wave speed curves in CO₂ mixtures exhibited long pressure plateaus.
- At the same initial pressure, increasing the initial operating temperature decreases the initial decompression wave speed; and lowering the initial temperature increases the initial decompression wave speed.
- A drop in the initial temperature did not always result in a lower pressure plateau level for CO₂ mixtures.
- The existence of hydrogen in CO₂ stream had a maximum impact on decompression, compared to the other impurities tested; CO, O₂, and N₂.

Overall, the current work shows that the CFD technique can be used to predict rapid and severe gas decompression by solving the governing flow equations, in conjunction with the GERG-2008 EOS. This is an effective tool for determining the decompression wave speeds for several CO₂-based mixtures and it is also applicable in two- or three-dimensional geometries so the effect of pipe diameter, surface roughness and the shape of fracture outlet can be investigated. The implementation of GERG-2008 allows modelling the real behaviour of CO₂ mixture under failure events. This brought about the possibility of using the CFD to investigate several areas related CCS (i.e. the dispersion of CO₂).

Future work will focus on developing a 3D decompression model so the effects of pipe opening and the pressure drop behind the crack tip can be identified. A 3D coupled fracture-decompression model is also a target to understand the interaction between the fracturing pipe and decompressing fluid.

Acknowledgements

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