

$$Q_{Alhol} = Q_{+} \times 1.09$$

$$Q_{+} = \frac{8}{15} \cdot C_{14} \cdot \sqrt{29} \left( H^{\frac{5}{2}} - h^{\frac{7}{2}} \right) + \frac{2}{3} C_{74} \sqrt{29} \cdot (2h_{1}) h^{\frac{7}{2}}$$

$$= \frac{8}{15} \times 0.159 \times 4.43 \left( H^{\frac{7}{2}} - h^{\frac{7}{2}} \right) + \frac{2}{3} \times 0.58 \times 4.43 \left( 84021 \right)$$

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$$= \frac{1.39}{15} \times \left[ H^{\frac{5}{2}} - h^{\frac{5}{2}} \right]$$

$$= \left\{ 1.39 \times \left[ H^{\frac{5}{2}} - h^{\frac{5}{2}} \right] \times \left[ H^{\frac{5}{2}} -$$