

# An alternative reference fluid for the COSMO-SAC model

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In the COSMO-SAC [1] model of Lin and Sandler, the *activity coefficient* of a segment  $\Gamma_m$  is obtained using a reference fluid where all segments have an identical chemical potential  $\mu^0(0)$ . In the present work we show a new derivation for the activity coefficient of a segment using the reference of an *ideal* fluid with zero interaction energies but not identical chemical potentials.

We start from the Eq. (9) of [1], which is also similar to the one originally derived by Klamt[2]:

$$\frac{\mu_m}{kT} = -\ln \left[ \sum_n \exp \left( \frac{-E_{m,n} + \mu_n}{kT} \right) \right] + \ln p_m \quad (1)$$

where  $\mu_m$  is the chemical potential of a segment  $m$ ;  $E_{m,n} = E_{n,m}$  is the interaction energy for the contact between the segments  $m$  and  $n$ ;  $p_m$  is the probability of finding a segment  $m$  in the mixture; and  $T$  is the temperature.

Now let us consider an *ideal* fluid where all the interactions are zero,  $E_{m,n} = 0$ , resulting in:

$$\frac{\mu_m^0}{kT} = -\ln \left[ \sum_n \exp \left( \frac{\mu_n^0}{kT} \right) \right] + \ln p_m \quad (2)$$

One could argue that, if there is no interaction, the chemical potential of a segment  $\mu_m^0$  should not depend on the chemical potential of other segments  $\mu_n^0$ . Indeed, since  $\sum_n p_n = 1$ , one can verify that  $\mu_m^0/kT = \ln p_m$  is a solution of Eq. (2).

Then, if we subtract  $\mu_m^0/kT = \ln p_m$  from Eq. (1) and define the activity coefficient of a segment as  $\ln \Gamma_m \equiv (\mu_m - \mu_m^0)/kT$ , we get:

$$\ln \Gamma_m = -\ln \left[ \sum_n \exp \left( \frac{-E_{m,n} + \mu_n}{kT} \right) \right] \quad (3)$$

Finally, if we use  $\mu_n/kT = \ln \Gamma_n + \mu_n^0/kT = \ln \Gamma_n + \ln p_n$ , it is possible to show that:

$$\ln \Gamma_m = -\ln \left[ \sum_n p_n \Gamma_n \exp \left( \frac{-E_{m,n}}{kT} \right) \right] \quad (4)$$

which is equivalent to Eq. (10) of [1].

With this derivation it is more clear that many contributions can be added directly to the interaction energy  $E_{m,n}$ , not only the usual electrostatic and hydrogen bond contributions. Further, that the activity coefficient of a segment  $\ln \Gamma_m$  can be used to compute the residual Helmholtz energy, with respect to a non-interacting fluid.

## References

- [1] S.-T. Lin, S. I. Sandler, A Priori Phase Equilibrium Prediction from a Segment Contribution Solvation Model, Ind. & Eng. Chem. Res. 41 (5) (2002) 899–913.
- [2] A. Klamt, Conductor-like Screening Model for Real Solvents: A New Approach to the Quantitative Calculation of Solvation Phenomena, The Journal of Physical Chemistry 99 (7) (1995) 2224–2235.

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