

Tableau périodique des éléments

	I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII
1	<div>Hydrogène 2, 2 1s¹ 1H 1,00</div>	<div>Électronégativité → 8,88</div> <div>Symbole → X</div> <div>Nombre de protons → Z</div> <div>Masse molaire (g/mol) → 888,88</div> <div>Nom élément ← [Y]</div> <div>Élément ← <i>asⁱ</i> <i>bp^j</i> <i>nd^k</i> <i>mf^l</i></div> <div>Structure électronique [Y] désigne la structure du gaz noble antérieur.</div>																<div>Hélium 1s². 2He 4,00</div>
2	<div>Lithium 0, 98 [He] 2s¹ 3Li 6,94</div>	<div>Béryllium 1, 57 [He] 2s² 4Be 9,01</div>																
3	<div>Sodium 0, 93 [Ne] 3s¹ 11Na 22,99</div>	<div>Magnésium 1, 31 [Ne] 3s² 12Mg 24,31</div>																
4	<div>Potassium 0, 82 [Ar] 4s¹ 19K 39,10</div>	<div>Calcium 1 [Ar] 4s² 20Ca 40,08</div>	<div>Scandium 1, 36 [Ar] 4s² 3d¹ 21Sc 44,96</div>	<div>Titane 1, 54 [Ar] 4s² 3d² 22Ti 47,87</div>	<div>Vanadium 1, 63 [Ar] 4s² 3d³ 23V 50,94</div>	<div>Chrome 1, 66 [Ar] 4s² 3d⁴ 24Cr 52,00</div>	<div>Manganèse 1, 55 [Ar] 4s² 3d⁵ 25Mn 54,94</div>	<div>Fer 1, 83 [Ar] s² 3d⁶ 26Fe 55,85</div>	<div>Cobalt 1, 88 [Ar] 4s² 3d⁷ 27Co 58,93</div>	<div>Nickel 1, 91 [Ar] 4s² 3d⁸ 28Ni 58,69</div>	<div>Cuivre 1, 9 [Ar] 4s² 3d⁹ 29Cu 63,55</div>	<div>Zinc 1, 65 [Ar] 4s² 3d¹⁰ 30Zn 65,38</div>	<div>Gallium 1, 81 [Ar] 4s² 3d¹⁰ 4p¹ 31Ga 69,72</div>	<div>Germanium 2, 01 [Ar] 4s² 3d¹⁰ 4p² 32Ge 72,64</div>	<div>Arsenic 2, 18 [Ar] 4s² 3d¹⁰ 4p³ 33As 74,92</div>	<div>Sélénium 2, 55 [Ar] 4s² 3d¹⁰ 4p⁴ 34Se 78,96</div>	<div>Brome 2, 96 [Ar] 4s² 3d¹⁰ 4p⁵ 35Br 79,90</div>	<div>Krypton 3 [Ar] 4s² 3d¹⁰ 4p⁶ 36Kr 83,80</div>
5	<div>Rubidium 0, 82 [Kr] 5s¹ 37Rb 85,47</div>	<div>Strontium 0, 95 [Kr] 5s² 38Sr 87,62</div>	<div>Yttrium 1, 22 [Kr] 5s² 4d¹ 39Y 88,91</div>	<div>Zirconium 1, 33 [Kr] 5s² 4d² 40Zr 91,22</div>	<div>Niobium 1, 6 [Kr] 5s¹ 4d⁴ 41Nb 92,91</div>	<div>Molybdène 2, 16 [Kr] 5s¹ 4d⁵ 42Mo 95,96</div>	<div>Technétium 1, 9 [Kr] 5s¹ 4d⁶ 43Tc 98</div>	<div>Ruthénium 2, 2 [Kr] 5s¹ 4d⁷ 44Ru 101,07</div>	<div>Rhodium 2, 28 [Kr] 5s¹ 4d⁸ 45Rh 102,91</div>	<div>Palladium 2, 2 [Kr] 5s¹ 4d¹⁰ 46Pd 106,42</div>	<div>Argent 1, 93 [Kr] 5s¹ 4d¹⁰ 47Ag 107,87</div>	<div>Cadmium 1, 69 [Kr] 5s² 4d¹⁰ 48Cd 112,41</div>	<div>Indium 1, 78 [Kr] 5s² 4d¹⁰ 5p¹ 49In 114,82</div>	<div>Étain 1, 96 [Kr] 5s² 4d¹⁰ 5p² 50Sn 118,71</div>	<div>Antimoine 2, 05 [Kr] 5s² 4d¹⁰ 5p³ 51Sb 121,76</div>	<div>Tellure 2, 1 [Kr] 5s² 4d¹⁰ 5p⁴ 52Te 127,6</div>	<div>Iode 2, 66 [Kr] 5s² 4d¹⁰ 5p⁵ 53I 126,90</div>	<div>Xénon 2, 6 [Kr] 5s² 4d¹⁰ 5p⁶ 54Xe 131,29</div>
6	<div>Césium 0, 79 [Xe] 6s¹ 55Cs 132,91</div>	<div>Baryum 0, 89 [Xe] 6s² 56Ba 137,33</div>	<div>Lanthane 1, 1 [Xe] 6s² 5d¹ 57La 138,91</div>	<div>Hafnium 1, 3 [Xe] 6s² 4f¹⁴ 5d² 72Hf 178,49</div>	<div>Tantale 1, 5 [Xe] 6s² 4f¹⁴ 5d³ 73Ta 180,95</div>	<div>Tungstène 2, 36 [Xe] 6s² 4f¹⁴ 5d⁴ 74W 183,84</div>	<div>Rhénium 1, 9 [Xe] 6s² 4f¹⁴ 5d⁵ 75Re 186,21</div>	<div>Osmium 2, 2 [Xe] 6s² 4f¹⁴ 5d⁶ 76Os 190,23</div>	<div>Iridium 2, 2 [Xe] 6s² 4f¹⁴ 5d⁷ 77Ir 192,22</div>	<div>Platine 2, 28 [Xe] 6s¹ 4f¹⁴ 5d⁹ 78Pt 195,08</div>	<div>Or 2, 54 [Xe] 6s¹ 4f¹⁴ 5d¹⁰ 79Au 196,97</div>	<div>Mercure 2 [Xe] 6s² 4f¹⁴ 5d¹⁰ 80Hg 200,59</div>	<div>Thallium 1, 62 [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p¹ 81Tl 204,38</div>	<div>Plomb 2, 33 [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p² 82Pb 207,2</div>	<div>Bismuth 2, 02 [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p³ 83Bi 208,98</div>	<div>Polonium 2 [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p⁴ 84Po 209</div>	<div>Astate 2, 2 [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p⁵ 85At 210</div>	<div>Radon [Xe] 6s² 4f¹⁴ 5d¹⁰ 6p⁶ 86Rn 222</div>
7	<div>Francium 0, 7 [Rn] 7s¹ 87Fr 223</div>	<div>Radium 0, 9 [Rn] 7s² 88Ra 226</div>	<div>Actinium 1, 1 [Rn] 7s² 6d¹ 89Ac 227</div>	<div>Rutherfordium [Rn] 7s² 6d² 5f¹⁴ 104Rf 265</div>	<div>Dubnium [Rn] 7s² 6d³ 5f¹⁴ 105Db 268</div>	<div>Seaborgium [Rn] 7s² 6d⁴ 5f¹⁴ 106Sg 271</div>	<div>Bohrium [Rn] 7s² 6d⁵ 5f¹⁴ 107Bh 272</div>	<div>Hassium [Rn] 7s² 6d⁶ 5f¹⁴ 108Hs 270</div>	<div>Meitnérium [Rn] 7s² 6d⁷ 5f¹⁴ 109Mt 276</div>	<div>Darmstadtium [Rn] 7s² 6d⁸ 5f¹⁴ 110Ds 281</div>	<div>Roentgenium [Rn] 7s² 6d⁹ 5f¹⁴ 111Rg 280</div>	<div>Copernicium [Rn] 7s² 6d¹⁰ 5f¹⁴ 112Cn 285</div>	<div>Nihonium [Rn] 7s² 6d¹⁰ 5f¹⁴ 7p¹ 113Nh 284</div>	<div>Flévorium [Rn] 7s² 6d¹⁰ 5f¹⁴ 7p² 114Fl 289</div>	<div>Moscovium [Rn] 7s² 6d¹⁰ 5f¹⁴ 7p³ 115Mc 288</div>	<div>Livervorium [Rn] 7s² 6d¹⁰ 5f¹⁴ 7p⁴ 116Lv 293</div>	<div>Tennessine [Rn] 7s² 6d¹⁰ 5f¹⁴ 7p⁵ 117Ts N/A</div>	<div>Oganesson [Rn] 5f¹⁴ 6d¹⁰ 7s² 7p⁶ 118Og 294</div>

Famille des lanthanides →

6	Cérium 1, 12 [Xe] 6s ² 4f ¹ 5d ¹ 58Ce 140,12	Praséodyme 1, 13 [Xe] 6s ² 4f ³ 59Pr 140,91	Néodyme 1, 14 [Xe] 6s ² 4f ⁴ 60Nd 144,24	Prométhéum [Xe] 6s ² 4f ⁵ 61Pm 145	Samarium 1, 17 [Xe] 6s ² 4f ⁶ 62Sm 150,36	Europium [Xe] 6s ² 4f ⁷ 63Eu 151,96	Gadolinium 1, 2 [Xe] 6s ² 4f ⁷ 5d ¹ 64Gd 157,25	Terbium [Xe] 6s ² 4f ⁹ 65Tb 158,93	Dysprosium 1, 22 [Xe] 6s ² 4f ¹⁰ 66Dy 162,5	Holmium 1, 23 [Xe] 6s ² 4f ¹¹ 67Ho 164,93	Erbium 1, 24 [Xe] 6s ² 4f ¹² 68Er 167,26	Thullium 1, 25 [Xe] 6s ² 4f ¹³ 69Tm 168,93	Ytterbium [Xe] 6s ² 4f ¹⁴ 70Yb 173,05	Lutéций 1, 27 [Xe] 6s ² 4f ¹⁴ 5d ¹ 71Lu 174,97
7	Thorium 1, 3 [Rn] 7s ² 6d ² 90Th 232,04	Protactinium 1, 5 [Rn] 7s ² 6d ¹ 5f ² 91Pa 231,04	Uranium 1, 38 [Rn] 7s ² 6d ¹ 5f ³ 92U 238,03	Neptunium 1, 36 [Rn] 7s ² 6d ¹ 5f ⁴ 93Np 237	Plutonium 1, 28 [Rn] 7s ² 6d ¹ 5f ⁶ 94Pu 244	Américium 1, 3 [Rn] 7s ² 5f ⁷ 95Am 243	Curium 1, 3 [Rn] 7s ² 6d ¹ 5f ⁷ 96Cm 247	Berkélium 1, 3 [Rn] 7s ² 5f ⁹ 97Bk 247	Californium 1, 3 [Rn] 7s ² 5f ¹⁰ 98Cf 251	Einsteinium 1, 3 [Rn] 7s ² 5f ¹¹ 99Es 252	Fermium 1, 3 [Rn] 7s ² 5f ¹² 100Fm 257	Mendélévium 1, 3 [Rn] 7s ² 5f ¹³ 101Md 258	Nobélium 1, 3 [Rn] 7s ² 5f ¹⁴ 102No 259	Lawrencium [Rn] 7s ² 5f ¹⁴ 7p ¹ 103Lw 262

- Famille I : Colonne des alcalins
- Famille II : Colonne des alcalinoterreux
- Famille XI : Colonne des métaux nobles
- Famille XVII : Colonne des halogènes
- Famille XVIII : Colonne des gaz nobles
- Famille III à XII : Colonnes des métaux de transition