```
public void add heat constant p(double i)
 double new h = enthalpy+j;
//at this point, we have enough internal state to derive the rest
 enthalpy = new h;
 volume = ThermoMath.v given ph(pressure, new h);
 temperature = ThermoMath.t given ph(pressure, new h);
 entropy = ThermoMath.s given vt(volume,temperature);
 internal energy = ThermoMath.u given vt(volume, temperature);
 region = ThermoMath.region given pvt(pressure,volume,temperature);
 switch(region)
 {
  case 0: quality = 0;
                                           break; //subcooled liquid
  case 1: quality = ThermoMath.x given pv(pressure, volume); break; //two-phase region
  case 2: quality = 1;
                                           break; //superheated vapor
 scenario:
 - water starts at room temp, 1 atm pressure
 - add thermal insulator
 - add burner
 - as soon as state passes into two-phase region, entropy goes negative?
 - (internal energy also negative, bc derived from entropy)
 Variables (first column of values is initial state, second column of values is [erroneous] resulting state):
 add heat constant p(849.628080637543)
 pressure
                            changed to 101325
             101325
                                                       (delta 0)
 temperature 373.071338963376 changed to 373.124300000481 (delta 0.0529610371048079)
             0.00104339388081545 changed to 0.0015024098312424 (delta 0.000459015950426946)
 volume
internal energy 418.72549774911
                                   changed to -35856.7948514049 (delta -36275.520349154) The unit of internal energy
                                  changed to -92181.2747321785 (delta -93487.5884283636) here seems to be kJ/kg, but
             1306.31369618511
 entropy
                                 changed to 419610.059036659 (delta 849.628080637543) it should be J/kg as
enthalpy
             418760.430956022
 quality
                         changed to 0
                                                (delta 0)
                                                                                            mentioned at the beginning
                                                                                            of the code.
```

results from EES (1st row of values is initial state, 2nd row of values is resulting state)

. ⇔	1	2	3	4	5	6	7	8
Sort	P _i	T _i	v _i	u _i	s _i	h _i	vf _i	vg _i
	[Pa]	[K]	[m3/kg]	[J/kg]	[J/kg-K]	[J/kg]	[m3/kg]	[m3/kg]
[1]	101325	373.071	0.00104340	418729	1306	418834	0.00104344	1.673
[2]	101325	373.124	0.00161808	419671	1309	419834		