hC[1] = enthalpy(F\$, T=T[1], P=PC[1])

uC[1] = intenergy(F\$, T=T[1], P=PC[1])

vC[1] = volume(F\$, T=T[1], P=PC[1])

sC[1] = entropy(F\$, T=T[1], P=PC[1])

"Constant Volume heat addition" F\$ = 'Steam IAPWS' "Initial State - part A: subcooled liquid" P[1] = 101.325[kpa]"initial pressure" T[1] = 25[c]"initial temperature" "1 liter vessel - constant volume" vol1 = 1e-3[m3]hA[1] = enthalpy(F\$, T=T[1], P=P[1])"specific enthalpy of subcooled liquid at state 1" uA[1] = intenergy(F\$, T=T[1], P=P[1])"specific internal energy of subcooled liquid at state 1" vA[1] = **volume**(F\$, **T**=T[1], **P**=P[1]) "specific volume of subcooled liquid at state 1" sA[1] = entropy(F\$, T=T[1], P=P[1])"specific entropy of subcooled liquid at state 1" "Add heat - go to state 2: since the volume is constant, there is no work involved" Q 12 = 1[kw]"rate of adding heat" mass A = vol1/vA[1]"mass contained in 1 liter volume" "State 2" uA[2] = uA[1] + Q_12*time/mass_A "specific internal energy at state 2" hA[2] = enthalpy(F\$, u=uA[2], v=vA[1])"specific enthalpy at state 2" sA[2] = entropy(F\$, u=uA[2], v=vA[1])"specific entropy at state 2" vA[2] = vA[1]"no change of volume or mass" PA[2] = pressure(F\$, u=uA[2], v=vA[1])"pressure at state 2" TA[2] = temperature(F\$, u=uA[2], v=vA[1])"temperature at state 2" "Initial State - part B: two phase" xB[1] = 0.6"initial quality" hB[1] = enthalpy(F\$, x=xB[1], P=P[1])"specific enthalpy of two phase fluid at state 1" uB[1] = intenergy(F\$, x=xB[1], P=P[1])"specific internal energy of two phase fluid at state 1" vB[1] = volume(F\$, x=xB[1], P=P[1])"specific volume of two phase fluid at state 1" sB[1] = entropy(F\$, x=xB[1], P=P[1])"specific entropy of two phase fluid at state 1" "Add heat - go to state 2: since the volume is constant, there is no work involved" $mass_B = vol1/vB[1]$ "mass contained in 1 liter volume" "State 2B" uB[2] = uB[1] + Q 12*time/mass B"specific internal energy at state 2" hB[2] = enthalpy(F\$, u=uB[2], v=vB[1])"specific enthalpy at state 2" sB[2] = entropy(F\$, u=uB[2], v=vB[1])"specific entropy at state 2" vB[2] = vB[1]"no change of volume or mass" PB[2] = pressure(F\$, u=uB[2], v=vB[2])"pressure at state 2" TB[2] = temperature(F\$, u=uB[2], v=vB[2]) "temperature at state 2" "Initial state - Part C: superheated vapor" PC[1] = 1[kpa]"initial pressure"

"specific enthalpy of superheated vapor at state 1"

"specific volume of superheated vapor at state 1"

"specific entropy of superheated vapor at state 1"

"specific internal energy of superheated vapor at state 1"

 $mass_C = vol1/vC[1]$

"mass contained in 1 liter volume"

"State 2C"

 $\begin{tabular}{ll} uC[2] = uC[1] + Q_12*time/mass_C \\ hC[2] = enthalpy(F\$, u=uC[2], v=vC[1]) \\ sC[2] = entropy(F\$, u=uC[2], v=vC[1]) \\ vC[2] = vC[1] \\ \end{tabular}$

PC[2] = pressure(F\$, *u*=uC[2], *v*=vC[2]) TC[2] = temperature(F\$, *u*=uC[2], *v*=vC[2]) "specific internal energy at state 2"

"specific enthalpy at state 2"
"specific entropy at state 2"

"no change of volume or mass"

"pressure at state 2"

"temperature at state 2"

SOLUTION

Unit Settings: SI C kPa kJ mass deg

(Case C, Run 100) F\$ = 'steam_iapws' massB = 0.0009956 [kg] Q₁₂ = 1 [kW] vol1 = 0.001 [m³]

massa = 0.997 [kg] massc = 0.000007271 [kg] time = 0.1 [s]

No unit problems were detected.

