

1.6.1: Measurements, Quantities, and Unity Factors Lecture Demonstrations

Convert Mass

Convert 12 lb weight of bowling ball to g, showing unity factors

$$12.00 \text{ lb} \times 453.59237 \text{ g/lb} = 5443 \text{ g}$$

Calculate and convert volumes

$$V = \frac{4}{3} \pi r^3$$

$$V = \frac{4}{3} * 3.1416 * (1/2 * 8.59 \text{ in})^3 =$$

$$\frac{4}{3} * 3.1416 * 79.23 \text{ in}^3$$

$$= 331.9 \text{ in}^3$$

Unity Factor: $2.54 \text{ cm} = 1 \text{ in}$

$$331.9 \text{ in}^3 \times (2.54 \text{ cm} / 1 \text{ in})^3 \text{ Note!!!}$$

$$331.9 \text{ in}^3 \times 16.39 \text{ cm}^3/\text{in}^3$$

$$= 5439 \text{ cm}^3$$

Densities

Will the bowling ball float in water? Demo ^[1]

$D = 5443 \text{ g} / 5439 \text{ cm}^3$ Too close to call. See [Errors in Measurement Lecture Demonstrations](#)

Mass vs. Weight

What is the mass of hydrogen?

Density of hydrogen at room temperature and 1 Atm = 0.082 g/L

What is the volume in L, assuming same size as bowling ball? Unity Factors?

$$1 \text{ cm}^3 = 1 \text{ mL} = 10^{-3} \text{ L} \text{ (Note: } 1 \text{ mL} = 1 \text{ cm}^3 = \text{"1 cc"})$$

$$V (\text{L}) = 5500 \text{ cm}^3 \times (1 \text{ L} / 1000 \text{ cm}^3)$$

$$m (\text{g}) = V (\text{L}) * D (\text{g/L}) = 5.500 \text{ L} \times 0.082 \text{ g/cm}^3$$

$$= 0.451 \text{ g}$$

Why does the Hydrogen balloon float? $F = W = m g$

$$F = W = (0.451 \text{ g} \times 1 \text{ kg} / 1000 \text{g}) * 9.8 \text{ m} \cdot \text{s}^{-2} =$$

$$= 0.0044 \text{ N}$$

Force Upward: (Archimedes)

D of air = 1.2 g/L ; Archimedes Principle: buoyancy = mass of air displaced (6.6 g)

$$F = m g$$

$$F = (6.6 \text{ g} \times 1 \text{ kg} / 1000 \text{g}) * 9.8 \text{ m} \cdot \text{s}^{-2} =$$

$$= 0.065 \text{ N}$$

Net force = $0.065 \text{ N} - 0.044 \text{ N}$ upwards.

References

1. J. Chem. Educ., 2004, 81 (9), p 1309

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