

## 6.11.1: Lecture Demonstrations

## Conductivity of Molten Salt

Many salts can be melted in test tubes, and the conductivity of their melts measured with a low voltage device and either graphite "leads" from pencils, or TIG welding rods<sup>[1]</sup>.

## Conductivity of Salt Solutions

Compare solution conductivity of ionic and covalent compounds<sup>[2]</sup>. Besides the traditional 120 V light bulb with probes in series (available from scientific suppliers), many conductivity devices have been suggested.<sup>[3][4][5][6]</sup>

## References

- 1. J. Chem. Educ., 2001, 78 (8), p 1052
- 2. J. Chem. Educ., 1962, 39 (10), p 531
- 3. Daniel T. Haworth, Mark R. Bartelt and Michael J. Kenney, Reed Howald. J. Chem. Educ., 1999, 76 (5), p 625
- 4. J. Chem. Educ., 1995, 72 (8), p 728
- 5. J. Chem. Educ., 1987, 64 (7), p 628
- 6. John W. Havrilla. J. Chem. Educ., 1991, 68 (1), p 80

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