

30.09.2024

An ellipse, anywhere, is given. Have we coordinates, yeah?

But can you count on, say, your cat's circle not letting you down?

Just space the nodes evenly AFTER skewing the matrix it's in.

Balance an elliptical disc between two points on a concave down dome: a favour for pants later man.

LIKE SO:

The thing would have some energy at the beginning, dampening on the dome's magnetic frequencies, a kind of coin on a mushroom head after birth of acquisition, or an mushroom of bonzo, the clown punch bag boy.

I take notes:

13:34 30-Sept-2024

The copper thing is elliptical in shape, otherwise flat, uniform am I right?

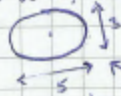
No. But I suggest calculation in spite of it.

The weight is 20 mg, the long & short of it: $\frac{3}{5}$, 1 (cm).

A point along the long axis and to a distance about $\frac{3}{5}$ cm from the centre of it is in contact with a point the same $\frac{3}{5}$ cm from the apex in true length (it's so special) of a continuous and symmetric dome so that the coin rocks back and forth over it, but what's gonna happen as it goes on rocking? It changes direction, and the peaks of the Potential Energy graph curve for the coin correspond to points which are vertices to some star shape, but if you were to trace the point of contact along the dome from start to finish, what shape is left when I remove the coin? You have a star!

rough given an ellipse, find the circle of equal area:

$$a = 5, b = 3 \text{ cm}$$



It's $\frac{3}{5} A_1$, where A_1 is the area of a circle of diameter 5. That's $6\frac{1}{2}\pi$

$$= 10\pi$$

$$\text{then radius} = \frac{10\pi}{\pi} = 10$$

in general:

$$\frac{\pi}{b} \left(\frac{a}{b} \right)^2 = ab\pi$$

so it's like with rectangles and squares!

If you have a list of material substances, and their densities up as far as platinum, but no radioactive stuff on the list, Plot an algorithm to select random combinations, including container weights, but not filling all of them, to make a quota of a certain mass, which is simple enough! The idea is that, lets say you've a few boxes now:
 3cm^3 full of steel, 4cm^3 full of ice, 20cm^3 of coal, 2cm^2 of gold, and the densities are 7.8, 0.920, 1.2, and 19.3 resp. with containers of planar density $5\text{g}/\text{cm}^2$ of regular cube shape. What's the total weight?

$$3\text{cm}^3 \times 7.8\text{g}/\text{cm}^3 + 3\text{cm}^2 \times 6 \times 5\text{g}/\text{cm}^2 = 234\text{g} + 90\text{g} = 324\text{g}$$

$$4\text{cm}^3 \times 0.920\text{g}/\text{cm}^3 + 4\text{cm}^2 \times 6 \times 5\text{g}/\text{cm}^2 = 3.680\text{g} + 120\text{g} = 123.680\text{g}$$

$$20\text{cm}^3 \times 1.2\text{g}/\text{cm}^3 + 20\text{cm}^2 \times 6 \times 5\text{g}/\text{cm}^2 = 24\text{g} + 12\text{kg}$$

$$(32 \times \frac{100}{70})^2 \times 6 \times 5\text{g}/\text{cm}^2 =$$

2 is 70% of x n volume
 the length of the unfilled cube's side is $3 \sqrt{\frac{10}{7} \times 2} = \sqrt{\frac{60}{7}}$ which is about the cubed root of 3 or close enough to it about 1.44
 $\approx 1.41883412...$
 then squared $\times 6 \times 5\text{g}/\text{cm}^2 = 6.0405417...$
 and + the material itself: 2cm^3 of gold at $19.3\text{g}/\text{cm}^3$

and let's say, the algorithm uses up to 5 containers but may only take one of them as partially filled, and the total must be exactly 12kg of weight, but the volumes may only be discrete integer lengthed cubes.

So in the example, to make it right, how much gold do you really take? Well, you gotta probably change the volume of coal down to 19cm^3 behind cube.

Sub total
 $324\text{g} + 123.68\text{g} + 12,024.5 + 44.64\text{g}$
 $= 12,476.82\text{g}$

12845.72g
 $= 12,845.72\text{g}$

only have $\sqrt{\quad}$ function, not $\sqrt[3]{\quad}$:
 $(\sqrt{2})^3 = 2.828427125...$

$f(3) = f(5, 12, 13, 2)$
 that's, if 12 a combination of cubes, squares, cubes and the number 3.
 $\frac{13}{19} = 0.684210526...$
 $1.45^3 \approx 3.048625$
 $1.44^3 \approx 2.985984$
 that's pretty close!
 $1.4421957...$
 $1.4421957^3 = 2.985984$
 $1.4421957^3 = 2.985984$
 but closer!

$3^{1(1+3)} = 3^4 = 81$
 $3^{1(1+3)} = 3^4 = 81$

There was an error: the 3cm volume to use was written as 3cm, but it's actually 27cm³, and that's a lot more!

shell: 293.4g
 ke: 483.68g
 amendment: (3cm)³ of steel at 7.8g/cm³

is $7.8 \times 27 = 70.2g$
 and $3^2 \times 6 \times 5g/cm^2 = 270g$, tot: 340.2g

ke: $64 \times 6 \times 5 = 1920g + 64 \times 0.980 = 62.72g$
 tot: 254.72g

$= 381$
 $\frac{2}{16}$
 $516859 \times 1.2 =$
 $\frac{1371.8}{1371.8} + \frac{6859}{576}$
 $\frac{2}{16}$
 $8230.8g + 205770g$
 $= 214000.8g$
 $= 214kg$

doing the other properly:

steel: $3^2 cm^3 \times 7.8g/cm^3 = 27 \times 7.8 = 70.2g$

$6 \times 3^2 + 12 \times 3 + 8 = 48g$
 tot: 168.2g

ke: $4^3 cm^3 \times 0.980g/cm^3$
 $= 64 \times (1 - \frac{1}{50})$
 $= 62.72$

$6 \times 4^2 + 12 \times 4 + 8 = 96 + 48 + 8 = 152g$

tot: 214.72g

coal: 10.6328kg

remaining weight: 5528

168.2

214.72

10632.8

11015.72g

so 984.28g left, for gold, so what are the concrete box weights?

{1, ...} of box:

1: $26cm^2 \times 5g/cm^3 = 130g$

2: $(6 \times 4 + 12 \times 2 + 8) \times 5 = 56 \times 5 = 280g$

3: $(6 \times 4 + 12 \times 3 + 8) \times 5 = 98 \times 5 = 490g$

and 494.28g of gold
 takes up $\frac{494.28}{19.3}$ of volume

$193/494.28 = 25.66$

143 336

765 1082.8

143 965

316 117.8

1158 1158

94.85319516...%

so it's very nearly filled in this case. well done! 2.0

2cm³ which is about 27cm³

but clearly that's too heavy.
 19cm side cube whose capacity is filled by coal at 1.2g/cm³ and whose surface is of planar density 5g/cm² and in more it's only a rough estimate for surface area (it's only 6 squares 1cm thick of some 5g/cm³ material)
 instead of $8 \times 1g/cm^3 + 12 \times 7g/cm^3 + 6 \times 7g/cm^3$
 which for $n=19$ is $8 + 228 + 246g = 2402g$
 2.402kg
 which is about right empty.

You will get exactly 12 kg left with gold because 193 > 5 for density

$$6n^2 + 12n + 8$$

$$= 6(n+1)^2 + 2$$

$$x^3 = 6(x+1)^2 + 2$$

$$x^3 - 6x^2 - 12x - 8 = 0$$

x is about 7.8
hmm, this is about steel's density too!

Another example:

get 10 kg of {copper 8.5 g/cm³, gold 19.3 g/cm³, milk 1.03 g/cm³, air 0.001293 g/cm³}
container made of steel 7.8 g/cm³.

trying for curiosity just the air:

$$(30 \text{ cm})^3 \times 7.8 \text{ g/cm}^3 = 27000 \times 7.8 = 70100 \text{ g} = 70 \text{ kg } 100 \text{ g}$$

try 1 cm³ air so, it's still heavy cause of the container and we want that gold.
But what if it's important air?

... > sniff <. 2 cm³.

steel cases:

$$1: 26 \times 7.8 = 13^2 \times 2 \times 6 = \frac{6 \cdot 169}{5} = 202.8 \text{ g}$$

$$8: 56 \times 7.8 = 436.8 \text{ g}$$

$$27: 3: 98 \times 7.8 = 7.8(100-2) = 780-156 = 764.4 \text{ g}$$

$$64: 4: 152 \times 7.8 = 1185.6 \text{ g}$$

$$125: 5: 218$$

$$216: 6: 296$$

$$343: 7: 384$$

$$512: 8: 488$$

$$\begin{array}{r} 152 \\ \times 7.8 \\ \hline 1216 \\ 11856 \\ \hline 1185610 \end{array}$$

$$2^3 \times 0.001293 = 0.010344 \text{ g}$$

$$\text{AIR: TOT: } 436.810344 \text{ g}$$

$$\text{milk: } 2 \text{ cm}^3 = 103 \text{ g}$$

$$436.8 \text{ g container}$$

$$\text{and } 8 \times 1.03 \text{ g/cm}^3 \text{ milk}$$

$$= 449.04 \text{ g}$$

$$\text{copper: } 3 \text{ cm}^3 \times 8.5 \text{ g/cm}^3$$

$$\text{gold: } 4 \text{ cm}^3 \times 19.3 \text{ g/cm}^3$$

$$= 64 \times 19.3 = 773 \text{ cm}^3$$

$$\begin{array}{r} 193 \\ 386 \\ 772 \\ 1544 \\ 3088 \\ 6176 \\ 23552 \end{array}$$

$$5 \text{ cm}^3$$

$$125 \times 19.3$$

$$6 \text{ cm}^3$$

$$216 \times 19.3$$

$$= 19.3$$

$$\begin{array}{r} 1158 \\ 386 \\ \hline 6948 \\ \times 3246 \\ \hline 41688 \end{array}$$

$$7 \text{ cm}^3$$

$$343 \times 19.3$$

$$\begin{array}{r} 627 \\ 1351 \\ 2357 \\ \hline 9457 \\ \times 2292 \\ \hline 66199 \end{array}$$

$$6 \text{ cm container:}$$

$$296 \times 7.8$$

$$\begin{array}{r} 296 \\ \times 7.8 \\ \hline 2368 \\ 2368 \\ \hline 2308.8 \end{array}$$

$$343 \times 7.8$$

$$384 \times 7.5$$

$$\begin{array}{r} 384 \\ \times 7.5 \\ \hline 1920 \\ 3072 \\ \hline 2880 \end{array}$$

$$5 \text{ cm container: } 10$$

$$\begin{array}{r} 218 \times 7.8 = \frac{218 \cdot 78}{10} \\ \hline 1700.4 \end{array}$$

$$4477.6 \text{ g}$$

$$6 \text{ cm cube}$$

So, (7 cm)³ of gold leaves 384.9 g for copper left. $\therefore 9615.1 \text{ g}$
then it's 1 cm copper and there's some room left.

Will you guess say the challenge failed then. You're to
got it exactly or the computer'll
beat you.

Using 6477.6g of the gold container lenses

3522.4g left.

we use 436.810344g air can and 445.04g milk can

881.850344

2640.539656g left for copper. 8.5 g/cm^3

4cm full: $64 \times 8.5 = 544 \text{ g}$ can: 1185.6g
 17, 39, 68, 136, 272, tot 1729.6g

or 5cm partial: 1700.4g can lenses
 $125 \times 8.5 = 425 \times 35 = 212.5 \times 5 = 1062.5 \text{ g}$ for the copper
 so it's 88.52137939...% filled.

then what if you were making a ball with shells of different substances, they're to be integer widths, let's go: gold, steel, copper, aluminum at thicknesses: {5, 4, 1, 1} (or such increments)

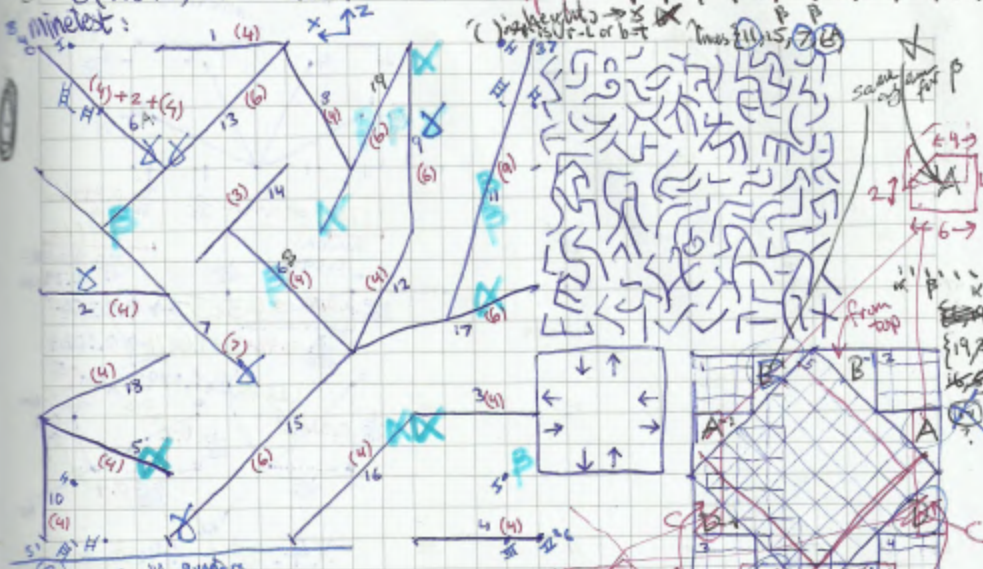
Vol of sphere is $\frac{4}{3}\pi r^3$ so $\frac{4}{3}\pi(5)^3 \times 19.3 \text{ g/cm}^3$ is $\frac{9850\pi}{3} \text{ g}$
 $\frac{4\pi}{3}(9^3 - 5^3) \times 7.8 = \frac{4\pi}{3} \cdot \frac{39}{5} \cdot (729 - 125) = \frac{156\pi}{15}(604) \text{ g}$
 $(10^3 - 9^3) \frac{4\pi}{3} \times 8.5 = 271 \times \frac{34}{3} \pi \text{ g}$
 $(11^3 - 10^3) \frac{4\pi}{3} \times 2.7 = 331 \cdot (36) \pi \text{ g}$

11cm radius metal shell
 with gold in the middle
 $\pi \left(\frac{9850}{3} + \frac{94224}{15} + \frac{31408}{5} \right) = \pi (48250 + 94224 + 46070 + 23674) = \pi (218218)$
 $\frac{14547}{15} \times \frac{22}{7} = \frac{14547}{160017} \times 22 = \frac{320034}{160017} = \frac{13 \times 22}{15 \times 286} = \frac{19}{15} \times \frac{22}{286} = \frac{2286}{160034} = \frac{2286 \times \frac{3}{10}}{160034} \approx 23 \text{ kg}$
 $45721 \frac{1}{105} \text{ g} \approx 45.7 \text{ kg}$
 $14547 \frac{13}{15} \pi \text{ g}$

$$7.2^3 = 474.552$$

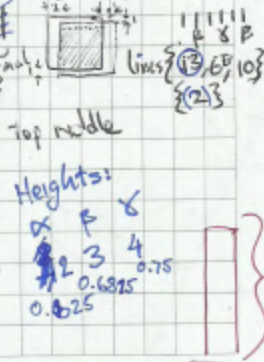
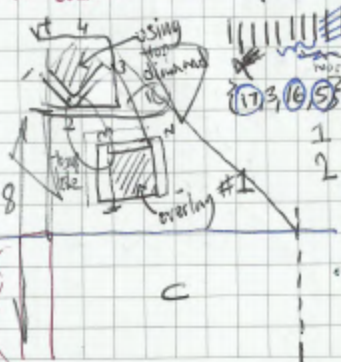
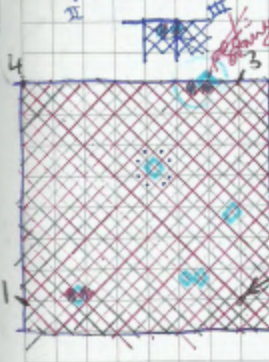
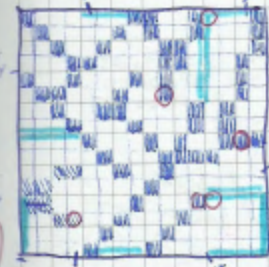
$$6(7.8+1)+2 = 466.64$$

mine test:



If I go in 2 or 3' have range.

0: w/ 1 white/yellow/orange
1st set: w/ 2 yellow (bot above white the same bottom color)
2nd set: w/ 3 bottom color





tk130.png

$\alpha \beta \gamma$

