



Nuclear Fission Using Algebra and Calculus

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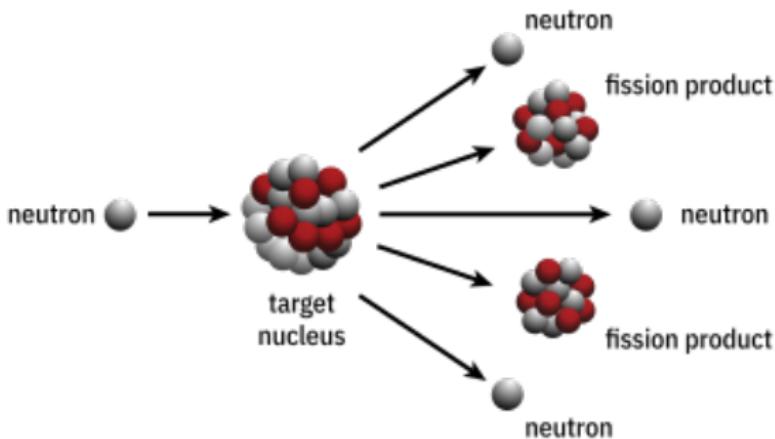
Defense Threat Reduction Information Analysis Center (DTRIAC)

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Compute Energy Release for Fission of ^{235}U





Overview

- 1 Nuclear Models
- 2 Liquid Drop Model: Spherical
- 3 Liquid Drop Model: Non-Spherical
- 4 Backup



Strange Behavior of the Nuclear Force

What Holds the Nucleus Together?

Electrical forces bind the electron to the atom, but they cause nuclear particles to fly apart. The powerful cohesion of protons and neutrons must be explained by a wholly different phenomenon

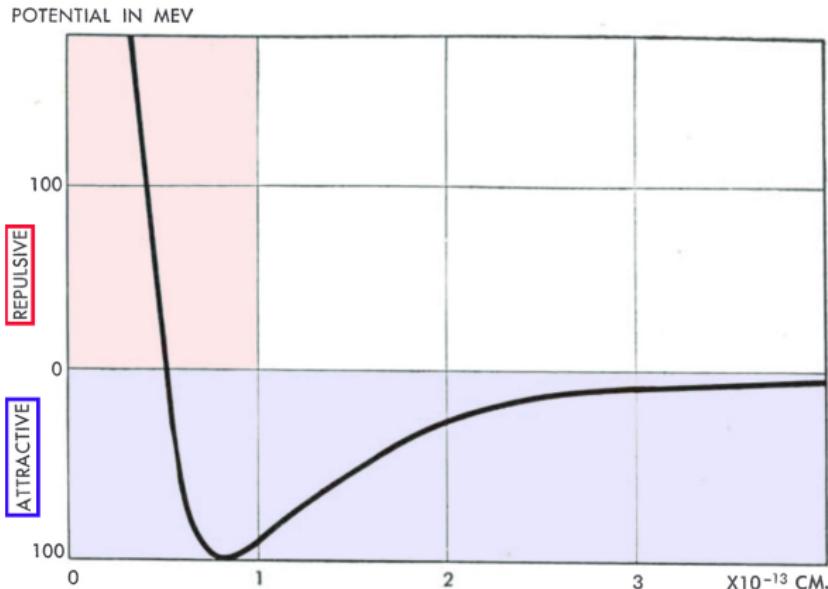
by Hans A. Bethe



Scientific American, Vol. 189, No. 3 (September 1953), pp. 58-63



Strange Behavior of the Nuclear Force



NUCLEAR FORCE (measured in millions of electron volts) is plotted against the distance between particles. When the distance is less than half of 10^{-13} centimeter, the nucleons repel one another. They most strongly attract one another at just under 10^{-13} cm.



The Binding Force of Nucleons

THE ATOMIC NUCLEUS

How do physicists presently visualize it? Curiously, different approaches to the nucleus suggest different pictures, notably the liquid-drop model, the shell model and the optical model

by R. E. Peierls

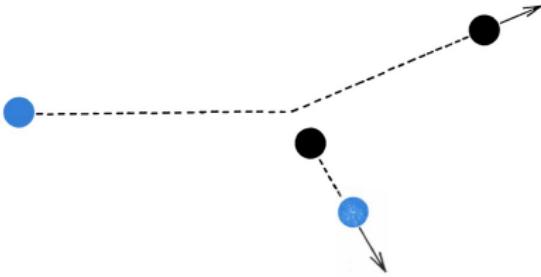
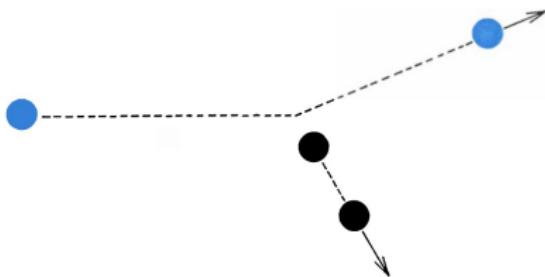
[Scientific American, Vol. 200, No. 1 \(January 1959\), pp. 75-84](#)



Vagaries of the Nuclear Force



NUCLEAR FORCES are dependent on the distance between particles. If the particles are very close, they repel each other (*left*). If they are a certain distance apart, they attract each other (*center*). If they are farther apart, they have little effect on each other (*right*).

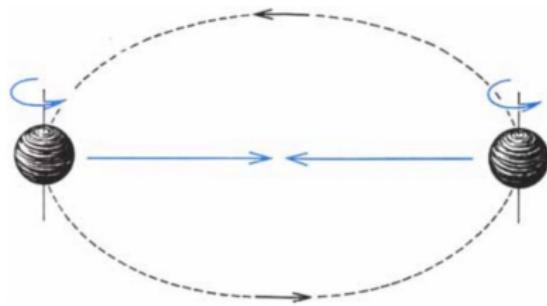


CHARGE EXCHANGE in the nucleus is schematically depicted. When protons (*black balls*) are struck by fast neutrons (*red balls*),

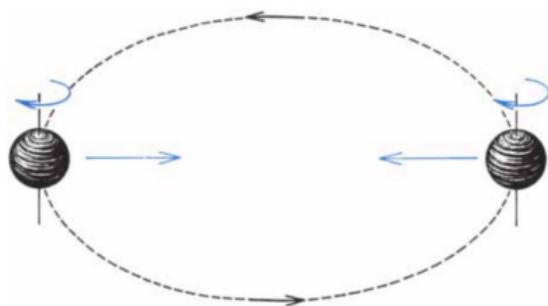
in half the cases (*left*) the neutron continues forward. In the other half (*right*), the proton exchanges its charge with the neutron.



Spin–Orbit Interactions



SPIN-ORBIT FORCE arises from a relationship between spin and orbit. When two particles (*left*) spin in the same direction as that



in which they move on an orbit, the force between them is strong. When they spin in opposite directions (*right*), force is weak.



Popular Models of the Nucleus

- ① Liquid Drop Model: [Fission](#)
- ② Shell Model: [Energy Levels](#)
- ③ Optical Model: Neutron capture [cross section](#)



Popularization of the Optical Model

A Model of the Nucleus

As an aid to understanding the atomic nucleus, physicists visualize it in terms of simplified models. A surprisingly fruitful approach is to regard it as a cloudy crystal ball

by Victor F. Weisskopf and E. P. Rosenbaum

Scientific American, Vol. 193, No. 6 (December 1955), pp. 84-91



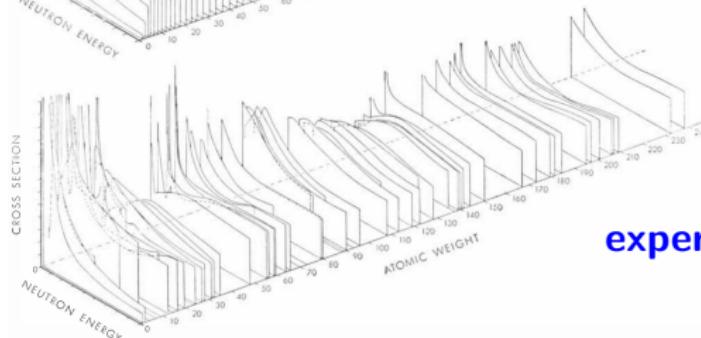
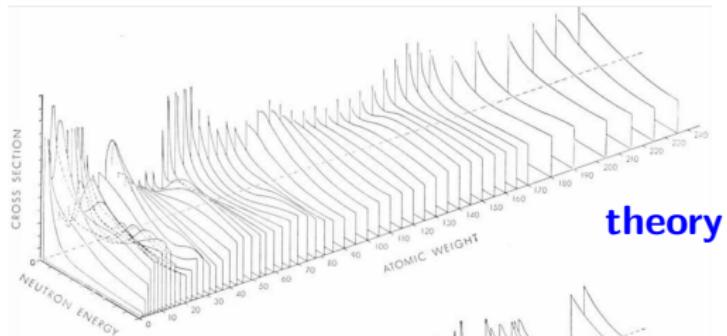
Cloudy Crystal Ball



OPTICAL MODEL pictures the nucleus as a somewhat cloudy crystal ball. The cloudiness represents the tendency of bombarding neutrons to be absorbed by the nucleus.



Neutron Cross Section



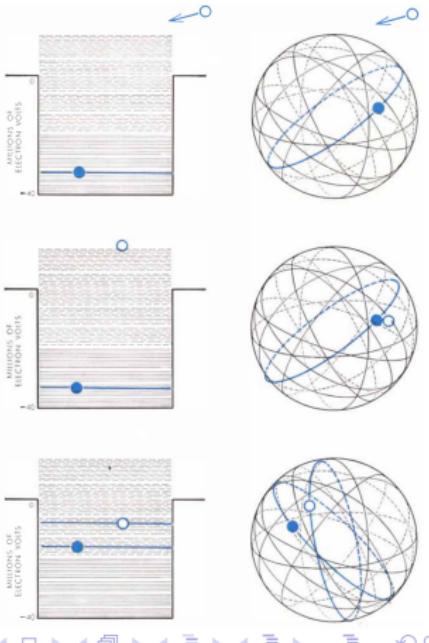
NEUTRON CROSS SECTIONS of all the elements as predicted by the cloudy-crystal-ball model are shown at the top. The actual values found by experiment appear below. The graphs are three-dimensional, with atomic weight increasing horizontally to the

right, the energy of the probing neutrons increasing in the direction from the plane of the page out toward the reader, and the cross section increasing vertically upward. The range of neutron energies shown extends from zero to about three million electron volts.



Neutron Absorption

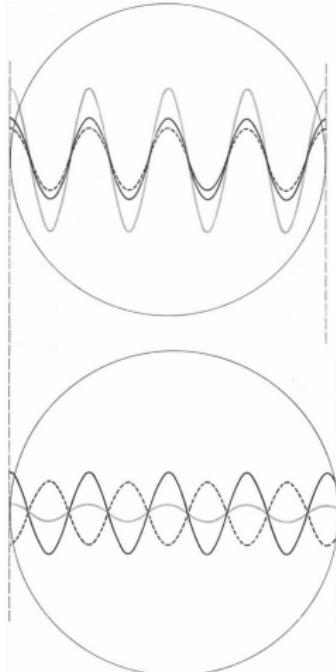
NEUTRON ABSORPTION is illustrated in these drawings. Left-hand diagrams show energies, right-hand diagrams, locations in space. At the top, a bombarding neutron (open dot) approaches the nucleus. Solid dot represents the nucleon destined to be hit, colored lines, its orbit. Middle diagrams show particles just before impact, when no energy has been exchanged. In the final result (bottom) the neutron has lost energy, dropping into a vacant orbit, and the target nucleon has picked up the energy, jumping to a higher orbit. The energy transaction is possible only if both particles find vacant positions. Cloudiness, or tendency to absorption, increases with the energy of the bombarding neutron because high-energy impacts put both particles into the upper, sparsely populated orbit region, while low-energy impacts leave them in the lower region where they are less likely to find vacancies.





Resonance

TOTAL WAVE MOTION inside nucleus is strong (*gray curve in top diagram*) when incoming wave (*solid black curve*) and reflected wave (*broken curve*) reinforce each other. If nuclear radius is wrong length (*bottom*) individual waves interfere giving small total.





Bonus Content: Ramsauer-Townsend Effect

AMERICAN JOURNAL OF PHYSICS

VOLUME 36, NUMBER 8

AUGUST 1968

Demonstration of the **Ramsauer-Townsend Effect** in a Xenon Thyatron

STEPHEN G. KUKOLICH

Department of Physics, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139

(Received 15 January 1968; revision received 15 April 1968)

The anomalously small scattering of electrons near 1 eV energy by noble gas atoms may be easily demonstrated using a 2D21 xenon thyatron. This experiment is suitable for a lecture demonstration or for an undergraduate physics laboratory. The probability of scattering and the scattering cross section may be obtained as a function of electron energy by measuring the grid and plate currents in the tube.



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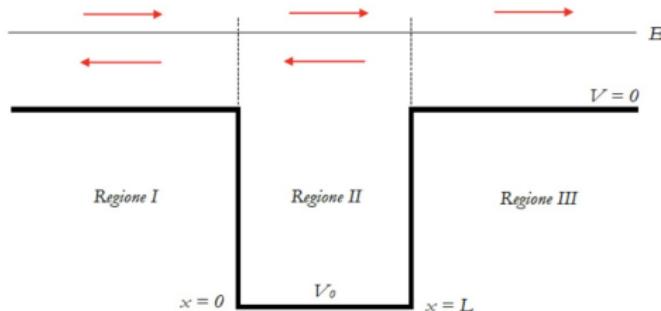
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Quantum Mechanical Scattering

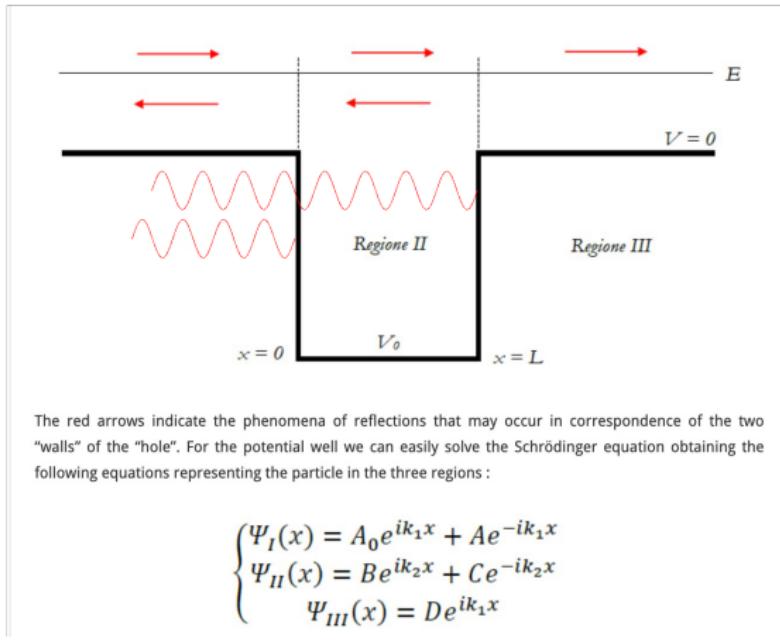


The red arrows indicate the phenomena of reflections that may occur in correspondence of the two "walls" of the "hole". For the potential well we can easily solve the Schrödinger equation obtaining the following equations representing the particle in the three regions :

$$\begin{cases} \Psi_I(x) = A_0 e^{ik_1 x} + A e^{-ik_1 x} \\ \Psi_{II}(x) = B e^{ik_2 x} + C e^{-ik_2 x} \\ \Psi_{III}(x) = D e^{ik_1 x} \end{cases}$$



Quantum Mechanical Scattering





Borrow From Atomic Physics

THE STRUCTURE OF THE NUCLEUS

The electrons in an atom tend to occupy distinct shells. What about particles in the nucleus? The “magic numbers” of the isotope chart suggest that they also have a shell arrangement.

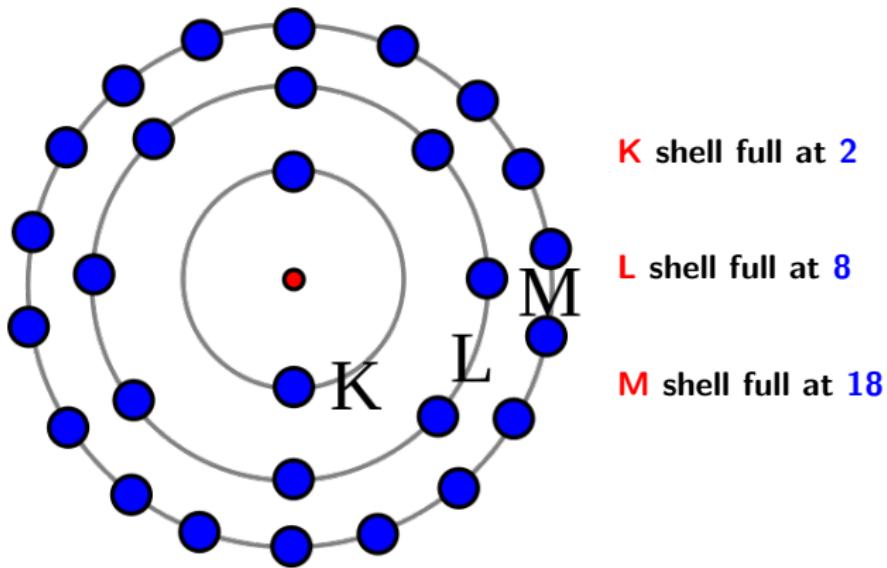
by Maria G. Mayer



Scientific American, Vol. 184, No. 3 (March 1951), pp. 22-27



Atomic Physics: Shell Names



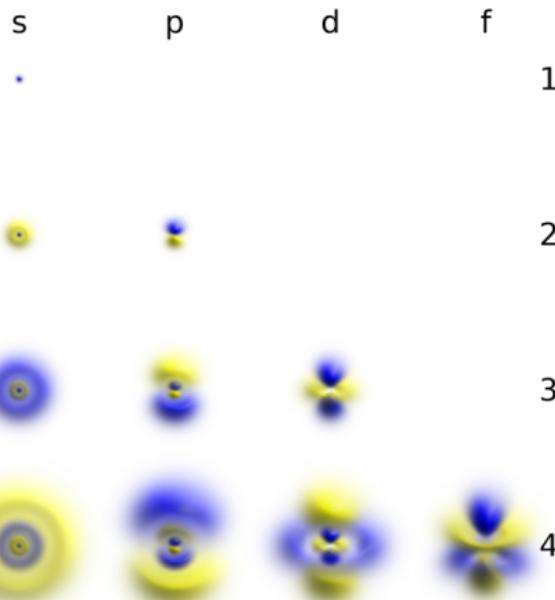


Atomic Physics: Electron Shell, Subshell Names

| Shell name | Subshell name | Subshell max electrons | Shell max electrons |
|------------|---------------|------------------------|-----------------------------|
| K | 1s | 2 | 2 |
| L | 2s | 2 | $2 + 6 = 8$ |
| | 2p | 6 | |
| M | 3s | 2 | $2 + 6 + 10 = 18$ |
| | 3p | 6 | |
| | 3d | 10 | |
| N | 4s | 2 | $2 + 6 + 10 + 14 = 32$ |
| | 4p | 6 | |
| | 4d | 10 | |
| | 4f | 14 | |
| O | 5s | 2 | $2 + 6 + 10 + 14 + 18 = 50$ |
| | 5p | 6 | |
| | 5d | 10 | |
| | 5f | 14 | |
| | 5g | 18 | |



Atomic Physics: Electron Shells



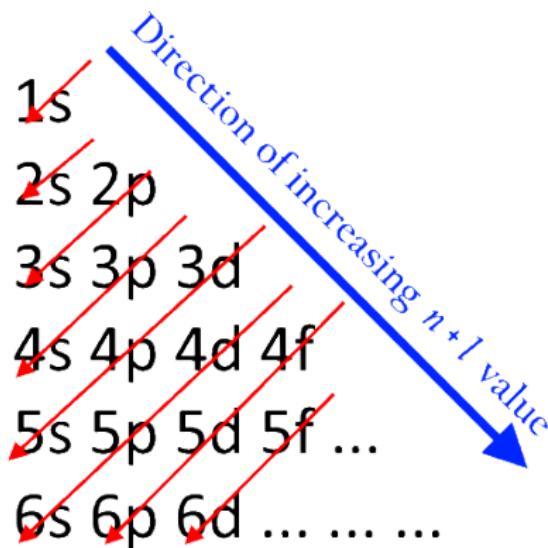


Atomic Physics: Electron Shells

| Subshell label | ℓ | Max electrons | Shells containing it | Historical name |
|----------------|--------|---------------|--------------------------------------|--|
| s | 0 | 2 | Every shell | sharp |
| p | 1 | 6 | 2nd shell and higher | principal |
| d | 2 | 10 | 3rd shell and higher | diffuse |
| f | 3 | 14 | 4th shell and higher | fundamental |
| g | 4 | 18 | 5th shell and higher (theoretically) | (next in alphabet after f) ^[25] |

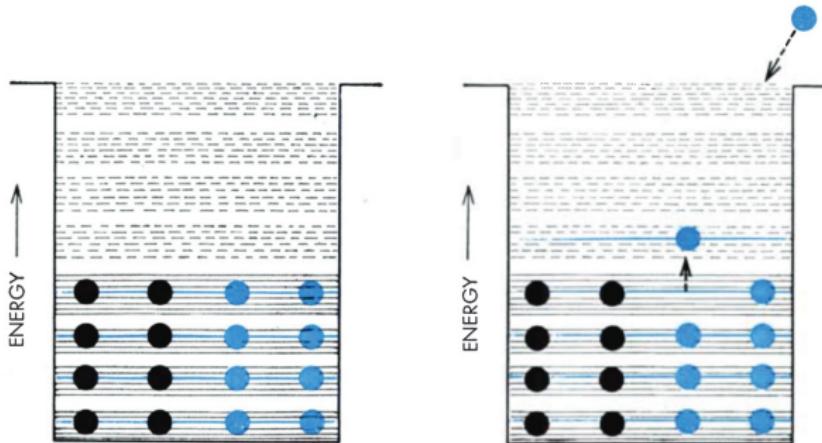


Atomic Physics: Filling the Shells





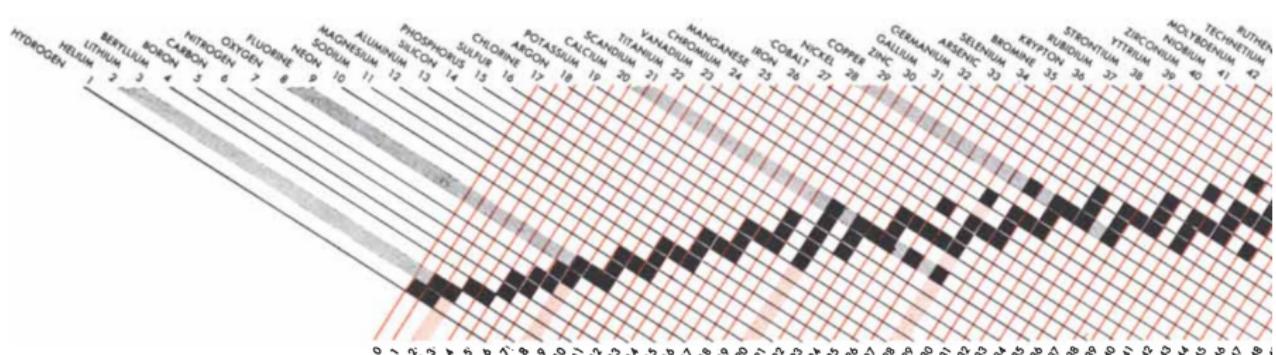
Shell Levels



LOW-ENERGY ORBITS in the shell model of the nucleus may each be occupied by only two neutrons (*colored balls*) and two protons (*black balls*). In the normal state of affairs (left) the low-energy orbits are filled; the particles cannot gain or lose energy, and thus cannot change their orbits. A bombarding particle (*upper right*) has energy to spare; thus it can exchange energy with a particle in nucleus and move it to orbit of higher energy.



Shell Theory Magic Numbers



THE CHART OF THE ISOTOPES reveals seven “magic numbers” (rows shaded red and black). Each isotope is represented by a black square. Each row out-

lined by black lines includes nuclei with the same number of protons, i.e., isotopes; each row enclosed in red lines shows nuclei with same number of neutrons. Since



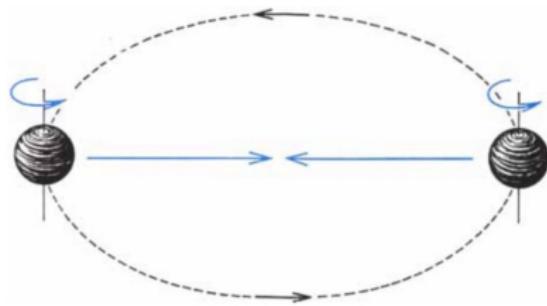
Margenau: Naieve Shell Levels

| ORBITAL ANGULAR MOMENTUM | SHELL | NUMBER OF NUCLEONS IN A LEVEL | NUMBER OF NUCLEONS UP TO A LEVEL |
|--------------------------|-------|-------------------------------|----------------------------------|
| 0 | 1s | 2 | 2 |
| 1 | 1p | 6 | 8 |
| 2 | 1d | 10 | |
| 0 | 2s | 2 | 20 |
| 3 | 1f | 14 | |
| 1 | 2p | 6 | 40 |
| 4 | 1g | 18 | |
| 2 | 2d | 10 | |
| 0 | 3s | 2 | 70 |
| 5 | 1h | 22 | |
| 3 | 2f | 14 | |
| 1 | 3p | 6 | 112 |
| 6 | 1i | 26 | |
| 4 | 2g | 18 | |
| 2 | 3d | 10 | |
| 0 | 4s | 2 | 168 |

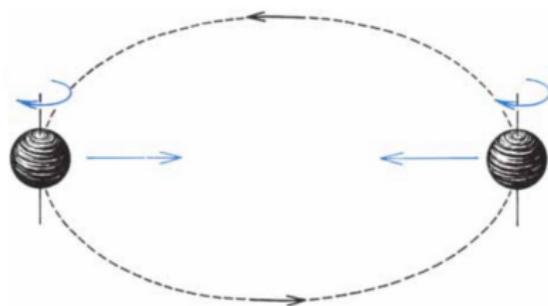
MARGENAU CHART of nuclear energy levels supports the shell-structure hypothesis in part. The first three numbers in column at right match the magic numbers. The next four do not; at this point the scheme breaks down.



Spin–Orbit Interactions



SPIN-ORBIT FORCE arises from a relationship between spin and orbit. When two particles (*left*) spin in the same direction as that



in which they move on an orbit, the force between them is strong. When they spin in opposite directions (*right*), force is weak.



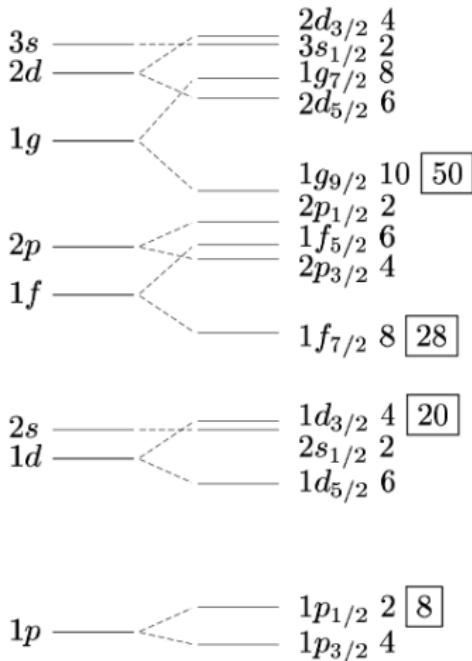
Mayer: Corrected Shell Levels

| ORBITAL ANGULAR MOMENTUM | SHELL | TOTAL ANGULAR MOMENTUM | NUMBER OF NUCLEONS IN A LEVEL | NUMBER OF NUCLEONS UP TO A LEVEL |
|--------------------------|----------|---------------------------------|-------------------------------|----------------------------------|
| 0 | 1s | 1/2 | 2 | 2 |
| 1 | 1p | 3/2 1/2 | 4 2 | 8 |
| 2 | 1d 2s | 5/2 3/2 1/2 | 6 4 2 | 20 |
| 3 | 1f | 7/2 5/2 3/2 1/2 9/2 | 8 6 4 2 10 | 28 |
| 1 | 2p | 7/2 5/2 3/2 1/2 9/2 | 8 6 4 2 10 | 50 |
| 4 | 1g | 7/2 5/2 | 8 6 | |
| 2 | 2d 3s | 3/2 1/2 11/2 | 4 2 12 | |
| 0 | | | | 82 |
| 5 | 1h | 9/2 7/2 | 10 8 | |
| 3 | 2f | 5/2 3/2 | 6 4 | |
| 1 | 3p | 1/2 13/2 | 2 14 | |
| 6 | 1i | 11/2 | 12 | 126 |

MAYER CHART accounts for the discrepancies by introducing the spins of nuclei (*fractions in the center column*). “Spin-orbit coupling” splits the close-lying levels apart and creates energy gaps where the magic numbers occur.



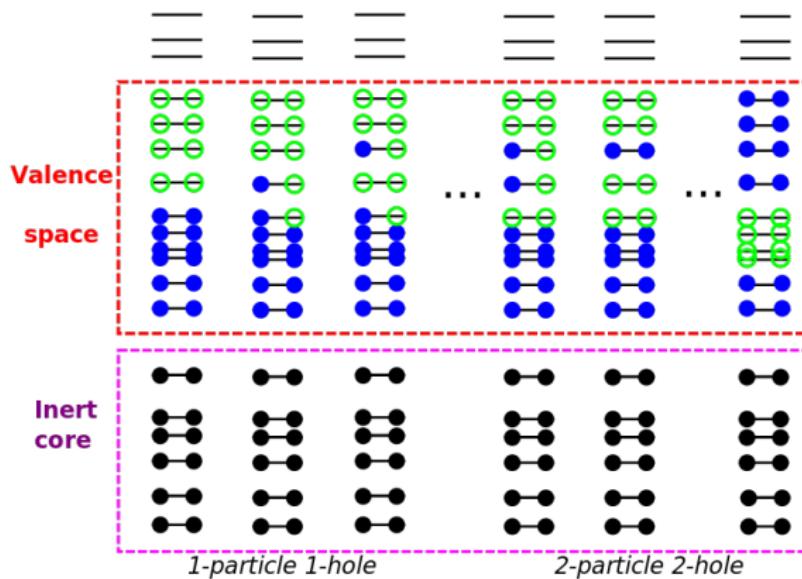
Wikipedia: Nuclear Shell Model





Wikipedia: Nuclear Shell Model

$$|\Psi\rangle = \alpha_1 |\Phi\rangle_1 + \alpha_2 |\Phi\rangle_2 + \alpha_3 |\Phi\rangle_3 + \dots + \alpha_k |\Phi\rangle_k + \alpha_{k+1} |\Phi\rangle_{k+1} + \dots$$



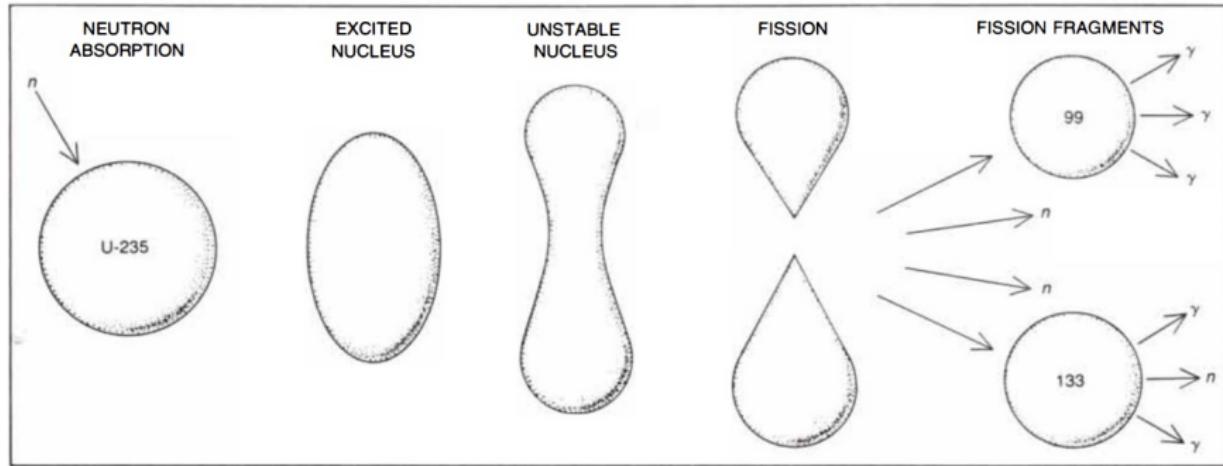


Summary of Liquid Drop Model

- ① Nucleus is liquid drop**
- ② Fluid is incompressible**
- ③ Volume is conserved**



Fission of a Liquid Drop



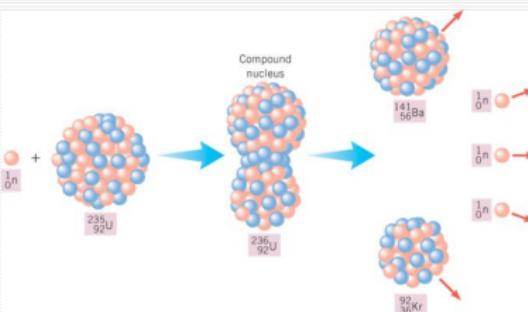
FISSION OF A NUCLEUS of U-235 is induced by the absorption of a thermal, or slow, neutron (n), which excites the nucleus and deforms it. About 85 percent of the time the deformed nucleus becomes unstable and splits into two fragments of unequal size. The fission fragments shown have atomic mass numbers of 99 and 133; many other pairs of fragments are possible and each has a well-defined probabili-

ty. The fragments are themselves unstable and are transformed by their subsequent decay, so that the total spectrum of fission products includes many isotopes of more than 30 elements. At the moment of fission high-energy photons, or gamma rays (γ), are emitted, as are a few neutrons. For a chain reaction to be sustained at least one neutron must be absorbed and must induce fission in another U-235 nucleus.



Deformation of Spherical Drop

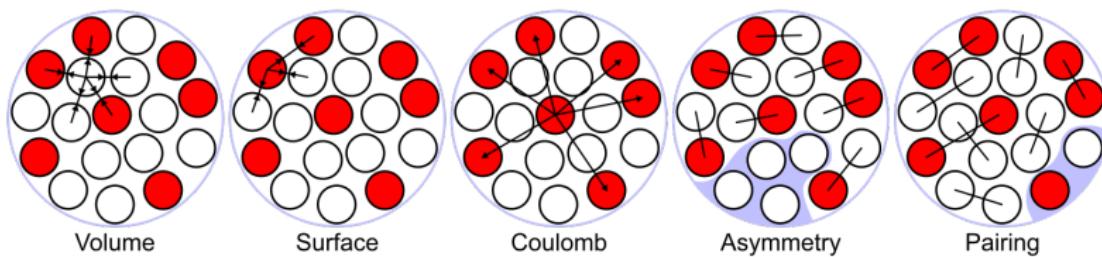
Nuclear Fission



A slowly moving neutron causes the uranium nucleus to fission into barium, krypton, and three neutrons.

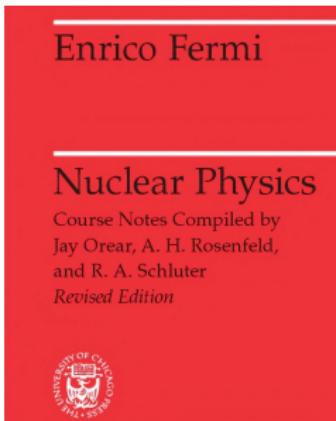


Interactions Within the Liquid Drop





Fission of a Liquid Drop



J. Fission **

The most useful model for explaining fission phenomena is the liquid drop model (see chapter I,C, p. 6). This model permits calculation of the change in potential energy when the nuclear drop suffers an ellipsoidal deformation from spherical shape*. If the potential energy increases, spherical shape is a stable configuration.

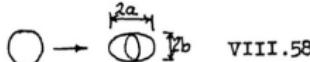


Fission of a Liquid Drop

The two contributions to the potential energy are 1) capillary energy, 2) electrostatic energy. We will calculate the change in these contributions to the potential energy for a constant volume prolate ellipsoidal deformation given by the equations

$$\text{major semi-axis} = a = R(1+\epsilon)$$

$$\text{minor semi-axis} = b = R/\sqrt{1+\epsilon}$$



VIII.58

where R = initial radius, ϵ = parameter giving extent of deformation. (3ϵ approaches the square of the eccentricity of the elliptical section as both approach 0.) Volume is invariant:
 $V = (4\pi/3)ab^2 = (4\pi/3)R^3$.

- 1) The capillary energy is proportional to the surface area.

$$\text{Ellipsoidal surface} = 4\pi R^2 (1 + 2/5 \epsilon^2 + \dots) \quad \text{VIII.59}$$

The capillary energy was computed in Ch. I, p.7, and found to be $0.014 \text{ A}^{2/3}$ for an unexcited (spherical) nucleus, therefore

$$\text{Capillary energy} = 0.014 \text{ A}^{2/3} (1 + 2/5 \epsilon^2 + \dots) \quad (\text{mass units}) \quad \text{VIII.60}$$

2) The electrostatic energy = $(3/5)(e^2 Z^2 / R)(1 - 1/5 \epsilon^2)$. At sphericity the energy is, from Ch. I, p. 6, $0.000627 Z^2 / \text{A}^{1/3}$, therefore

$$\text{Electrostatic energy} = 0.000627 (Z^2 / \text{A}^{1/3}) (1 - 1/5 \epsilon^2) \quad \text{VIII.61}$$

This is evidently maximum at sphericity. The total change is

$$\epsilon^2 (2/5 \times 0.014 \text{ A}^{2/3} - 1/5 \times 0.000627 Z^2 / \text{A}^{1/3}) \quad \text{VIII.62}$$



Nuclear Models

- Liquid Drop Model: Spherical
- Liquid Drop Model: Non-Spherical
- Backup
- References



Mysterious Nuclear Force

- Optical Model
- Shell Model
- Liquid Drop Model
- Fission Limit Using Liquid Drop

Water Drop

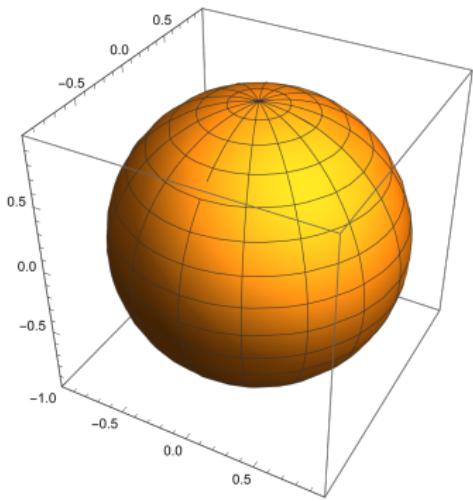


By A7N8X - Own work, CC BY-SA 4.0, <https://commons.wikimedia.org/w/index.php?curid=48828510>

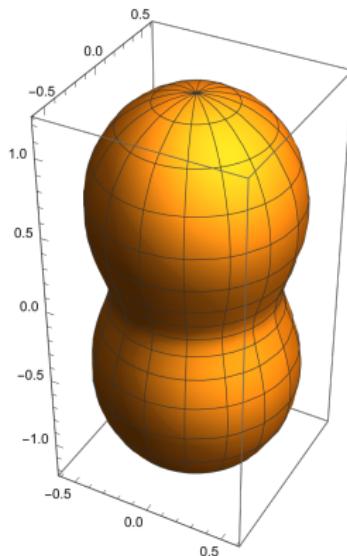


Comparison of Liquid Drop Models

Constant Radius



Variable Radius





Two Schools of Thought

Constant Radius



Serber

Variable Radius



Bohr



Wheeler

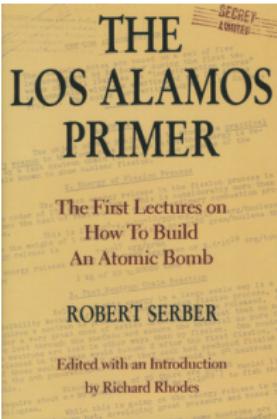


Summary of Liquid Drop Model

- ① Nucleus is liquid drop of **constant radius**
- ② Fluid is incompressible
- ③ Volume is conserved **Hofstadter 1956**



Serber's Ingredients



...based on a set of five lectures given by R. Serber during the first two weeks of April 1943, as an “indoctrination course” in connection with the starting of the Los Alamos Project.



Serber's Ingredients

- ① Volume of Sphere
- ② High School Electrostatics

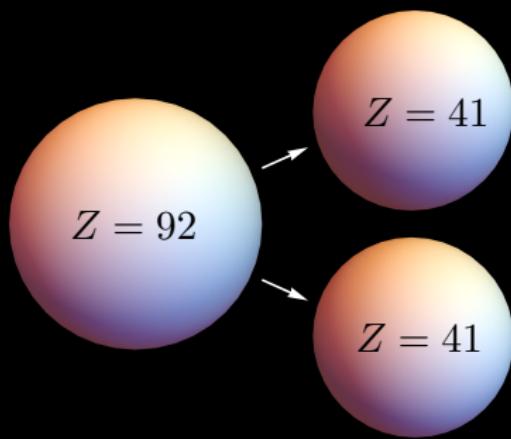


Reaction Model: Charge Bisection





Reaction Model: Charge Bisection

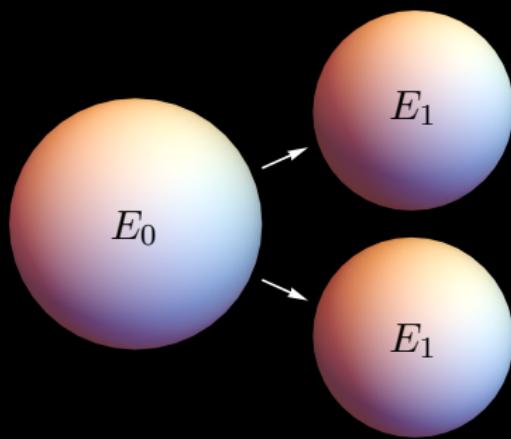




Quantify Change In Electrostatic Energy.

Before

After





Electrostatic Energy: Before and After

Compute **electrostatic energy** for both configurations.



Electrostatic Energy For Sphere of Charge Z

Electrostatic Energy For Sphere of Charge Z :

$$E = \frac{Z^2}{R} \quad (1)$$



Electrostatic Energy: Before and After

Before

After

$$E_0 = \frac{92^2}{R_0} \quad E_1 = 2 \times \frac{41^2}{R_1}$$



Electrostatic Energy: Before and After

Before

$$E_0 = \frac{92^2}{R_0}$$



After

$$E_1 = 2 \times \frac{41^2}{R_1}$$





Nuclear Models

Liquid Drop Model: Spherical
Liquid Drop Model: Non-Spherical
Backup
References

Serber's First Principles
Computation
Fission Energy Release
References
Next Step



Change in Radius

How does R_1 compare to R_0 ?



Change in Radius

How does R_1 compare to R_0 ?

Use Conservation of Nuclear Fluid.



Nuclear Fluid (Volume) is Conserved

Before

$$E_0 = \frac{92^2}{R_0}$$

$$V_0 = \frac{4}{3}\pi R_0^3$$



After

$$E_1 = 2 \times \frac{41^2}{R_1}$$

$$V_1 = 2 \times \frac{4}{3}\pi R_1^3$$





Solve for Radius R_1

$$V_0 = V_1 \quad \Rightarrow \quad R_0^3 = 2R_1^3$$



Solve for Radius R_1

$$V_0 = V_1 \quad \Rightarrow \quad \frac{4}{3}\pi R_0^3 = 2 \cdot \frac{4}{3}\pi R_1^3$$



$$R_0^3 = 2R_1^3$$



$$R_1 = 2^{-\frac{1}{3}} R_0$$



$$R_1 \approx \frac{4}{5} R_0$$



Change in Electrostatic Energy

$$\Delta E = E_0 - 2E_1 = \frac{92^2}{R_0} - 2\frac{41^2}{R_1} = \frac{3}{8}E_0 \quad (2)$$



Change in Electrostatic Energy

$$\Delta E = \frac{3}{8} E_0 \quad (2)$$





Change in Electrostatic Energy: Algebra Details

$$\begin{aligned}\Delta E &= E_0 - 2E_1 \\&= \frac{92^2}{R_0} - 2 \cdot \frac{41^2}{R_1} \\&= \frac{(2 \cdot 41)^2}{R_0} - \frac{2 \times 41^2}{\frac{4}{5}R_0} \\&= \frac{4 \cdot 41^2 - 5/2 \cdot 41^2}{R_0} \\&= \frac{\frac{3}{2}41^2}{R_0} \\&= \frac{3}{8}E_0\end{aligned}\tag{3}$$



Serber's Ingredients and Result

- ① Volume of Sphere
- ② High School Electrostatics

Result: Fission Releases Astonishing Energy

$$\Delta E = \frac{3}{8} E_0 \quad (2)$$



Astonishing Energy Release

If TNT gives you \$1 per reaction, how much will ^{235}U provide?



Astonishing Energy Release

If TNT gives you \$1 per reaction, how much will ^{235}U provide?

A stack of dollar bills 7,160 ft tall.



Astonishing Energy Release

If TNT gives you \$1 per reaction, how much will ^{235}U provide?

A stack of dollar bills 7,160 ft tall.

1.4 miles high.



Nuclear Models

Liquid Drop Model: Spherical

Liquid Drop Model: Non-Spherical

Backup

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Next Step



Astonishing Energy Release





Nuclear Models

Liquid Drop Model: Spherical

Liquid Drop Model: Non-Spherical

Backup

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Serber's First Principles

Computation

Fission Energy Release

References

Next Step



Hiroshima.

Amount of ^{235}U fissioned < 1.5 cubic inches





Internet References: Liquid Drop Model.

- **nuclear-power.com:** Liquid Drop Model of Nucleus
- **University of Saskatchewan:** Liquid Drop Model
- **Colorado School of Mines:** The Liquid Drop Model of the Nucleus
- **Wikipedia:** Semi-empirical mass formula
- **University of Southampton:** The Liquid Drop Model
- **Vendatu:** Liquid Drop Model in Nuclear Physics
- **hyperphysics:** Liquid Drop Model of Nucleus



Next Step: Variable R

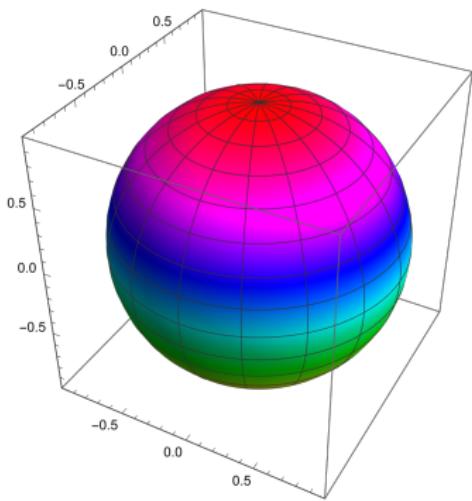
Serber's argument used a constant radius: $R = \text{const}$

Bohr and Wheeler let the radius vary $R = R(\theta)$

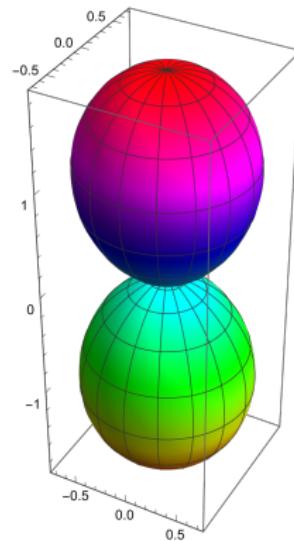


Comparison of Liquid Drop Models

Constant Radius

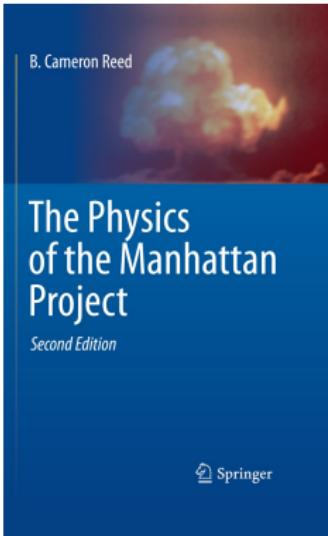


Variable Radius





Historical Survey



Appendix E: Formal Derivation of the Bohr–Wheeler Spontaneous Fission Limit.



Summary of Liquid Drop Model

- ① Nucleus is liquid drop of **variable radius**
- ② Fluid is **incompressible**
- ③ Volume is **conserved**



1939 Paper on Nuclear Fission

SEPTEMBER 1, 1939

PHYSICAL REVIEW

VOLUME 56

The Mechanism of Nuclear Fission

NIELS BOHR

University of Copenhagen, Copenhagen, Denmark, and The Institute for Advanced Study, Princeton,



AND

JOHN ARCHIBALD WHEELER

Princeton University, Princeton, New Jersey

(Received June 28, 1939)

On the basis of the liquid drop model of atomic nuclei, an account is given of the mechanism of nuclear fission. In particular, conclusions are drawn regarding the variation from nucleus to nucleus of the critical energy required for fission, and regarding the dependence of fission cross section for a given nucleus on energy of the exciting agency. A detailed discussion of the observations is presented on the basis of the theoretical considerations. Theory and experiment fit together in a reasonable way to give a satisfactory picture of nuclear fission.



Deformations from Sphericity

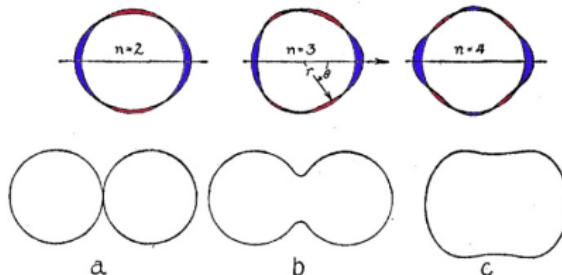


FIG. 2. Small deformations of a liquid drop of the type $\delta r(\theta) = \alpha_n P_n(\cos \theta)$ (upper portion of the figure) lead to characteristic oscillations of the fluid about the spherical form of stable equilibrium, even when the fluid has a uniform electrical charge. If the charge reaches the critical value $(10 \times \text{surface tension} \times \text{volume})^{\frac{1}{2}}$, however, the spherical form becomes unstable with respect to even infinitesimal deformations of the type $n=2$. For a slightly smaller charge, on the other hand, a finite deformation (c) will be required to lead to a configuration of *unstable equilibrium*, and with smaller and smaller charge densities the critical form gradually goes over (c, b, a) into that of two uncharged spheres an infinitesimal distance from each other (a).



Bohr-Wheeler Derivation

to the value

$$r(\theta) = R[1 + \alpha_0 + \alpha_2 P_2(\cos \theta) + \alpha_3 P_3(\cos \theta) + \dots], \quad (8)$$

where the α_n are small quantities. Then a straightforward calculation shows that the surface energy plus the electrostatic energy of the comparison drop has increased to the value

$$\begin{aligned} E_{S+E} = & 4\pi(r_0 A)^2 O [1 + 2\alpha_2^2/5 + 5\alpha_3^2/7 + \dots \\ & + (n-1)(n+2)\alpha_n^2/2(2n+1) + \dots] \\ & + 3(Ze)^2/5r_0 A^3 [1 - \alpha_2^2/5 - 10\alpha_3^2/49 - \dots \\ & - 5(n-1)\alpha_n^2/(2n+1)^2 - \dots], \quad (9) \end{aligned}$$

where we have assumed that the drop is composed of an **incompressible fluid** of volume $(4\pi/3)R^3 = (4\pi/3)r_0^3 A$, uniformly electrified to a charge Ze , and possessing a surface tension O .



Express Radius as a Function of Polar Angle

$$\underline{r} = \underline{R_0} \quad (4)$$



$$r(\theta) = R_0 (1 + \alpha_0 + \alpha_2 P_2(\cos \theta)) \quad (5)$$



Express Radius as a Function of Polar Angle

Adjust α_0 , α_2 so that volume is conserved:

$$r(\theta) = R_0 (1 + \alpha_0 + \alpha_2 P_2(\cos \theta)) \quad (5)$$



Why Legendre Polynomials?

Problem: Planetary Orbits

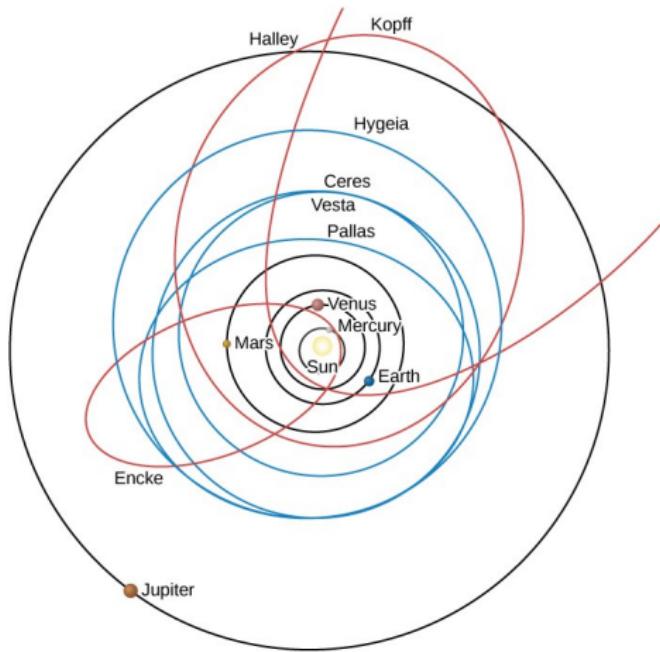
Recherches sur la figure des planètes

AM Legendre

Mémoires de l'Académie Royale des sciences, 1784

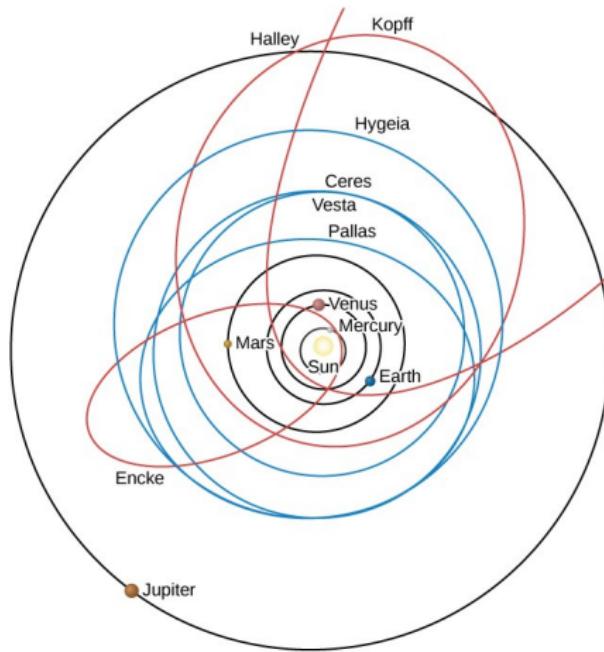


Why Legendre Polynomials?





Why Legendre Polynomials?



Orbit types:
Circular
Elliptical
Parabolic
Hyperbolic



Expansion of the Newtonian Potential

$$\frac{1}{\|d\|_2} = \frac{1}{\sqrt{R^2 + r^2 - 2Rr \cos \gamma}} = \sum_{k=0}^{\infty} \frac{r^k}{R^{k+1}} P_k(\cos \gamma) \quad (6)$$



Expansion of the Newtonian Potential

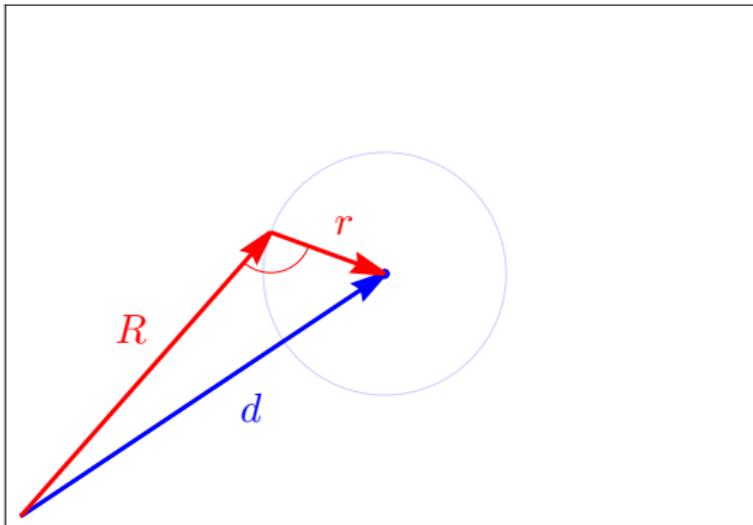
$$\frac{1}{\|d\|_2} = \frac{1}{\sqrt{R^2 + r^2 - 2Rr \cos \gamma}} = \sum_{k=0}^{\infty} \frac{r^k}{R^{k+1}} P_k(\cos \gamma) \quad (7)$$

↑

I am the Law of Cosines



Law of Cosines





Expansion of the Newtonian Potential

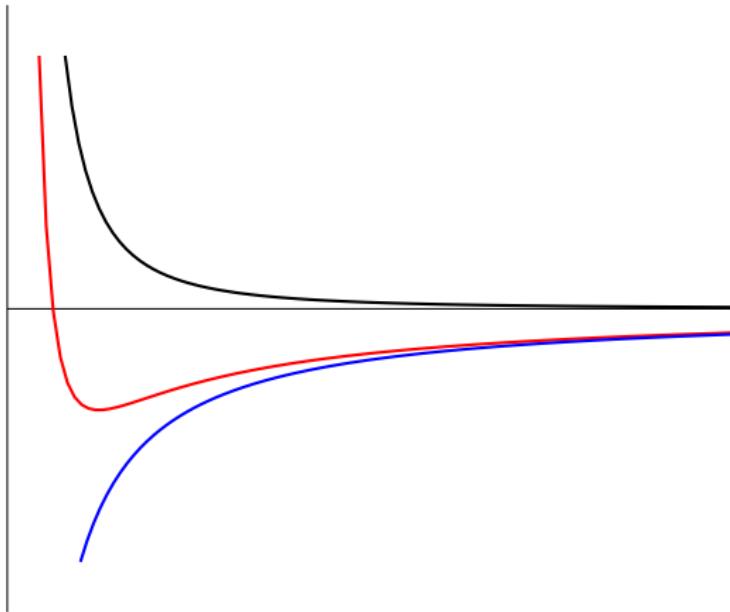
$$\frac{1}{\|d\|_2} = \frac{1}{\sqrt{R^2 + r^2 - 2Rr \cos \gamma}} = \sum_{k=0}^{\infty} \frac{r^k}{R^{k+1}} P_k(\cos \gamma) \quad (8)$$

↑

I am the Law of Cosines

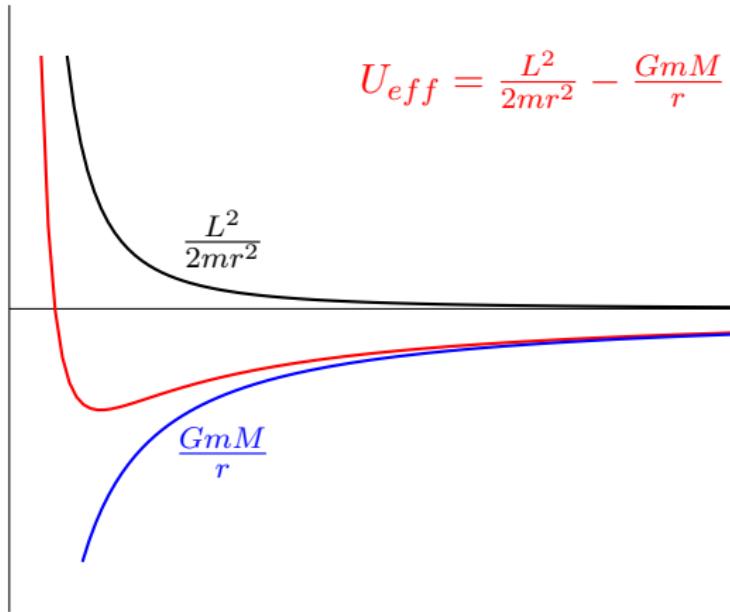


Effective Potential



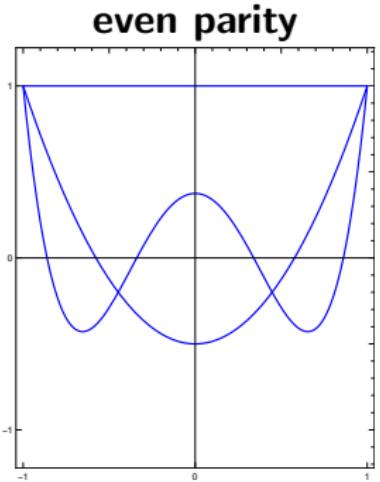


Effective Potential





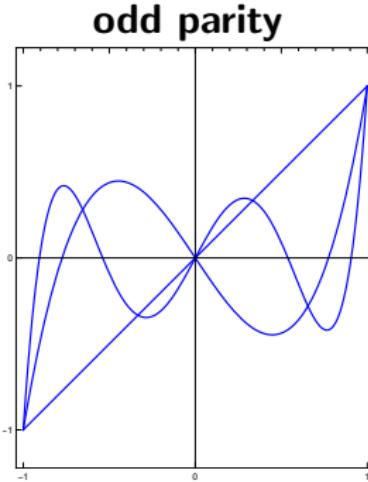
Deformations from Sphericity



$$P_0(x) = 1$$

$$P_2(x) = \frac{1}{2} (3x^2 - 1)$$

$$P_4(x) = \frac{1}{8} (35x^4 - 30x^2 + 3)$$



$$P_1(x) = x$$

$$P_3(x) = \frac{1}{2} (5x^3 - 3x)$$

$$P_5(x) = \frac{1}{8} (63x^5 - 70x^3 + 15x)$$



Mathematical Properties of Legendre Polynomials

- ❶ **Norm:** $\int_{-1}^1 P_j(x)P_k(x)dx = \frac{1}{2k+1}\delta_j^k$
- ❷ **Solution to** $(1-x^2)P_n(x)'' - 2xP_n(x)' + n(n+1)P_n(x) = 0$
- ❸ **Sturm–Liouville Form:** $\frac{d}{dx} \left((1-x^2) \frac{d}{dx} \right) P(x) = -\lambda P(x)$
- ❹ **Rodrigues' formula:** $P_n(x) = \frac{1}{2^n n!} \frac{d^n}{dx^n} (x^2 - 1)^n$
- ❺ **Decomposition:** $f_n(x) = \sum_{k=0}^l a_k P_k(x), a_k = \frac{2k+1}{2} \int_{-1}^1 f(x)P_k(x) dx$
- ❻ **Completeness:** $\sum_{k=0}^{\infty} \frac{2k+1}{2} P_k(x)P_k(y) = \delta(x-y), x, y \in [-1, 1]$



Mathematical Properties of Legendre Polynomials

① **First Use:** $\frac{1}{\sqrt{R^2+r^2-2Rr \cos \theta}} = \sum_{k=0}^{\infty} \frac{r^k}{R^{k+1}} P_k(\cos \theta)$

② **Generating function:** $\frac{1}{\sqrt{1-2xt+t^2}} = \sum_{k=0}^{\infty} P_k(x)t^n$

③ **Monic Gram-Schmidt on** $(1, x, x^2, x^3, \dots)$ **over** $x \in [-1, 1]$

④ **Bonnet's Recursion Formula:**

$$(n+1)P_{n+1}(x) = (2n+1)xP_n(x) - nP_{n-1}(x)$$

⑤ **a.k.a. Legendre functions of the first kind**

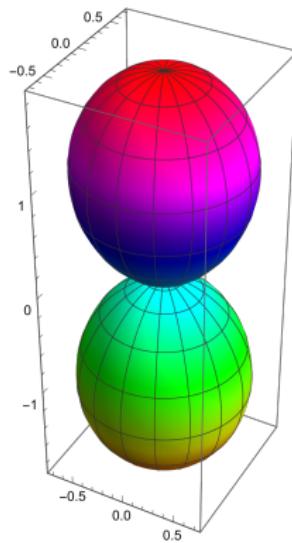
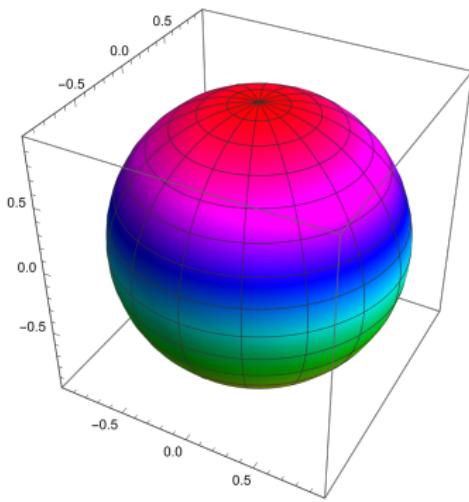
⑥ **a.k.a. zonal harmonics**



How To Compute Volume of Deformed Nucleus?

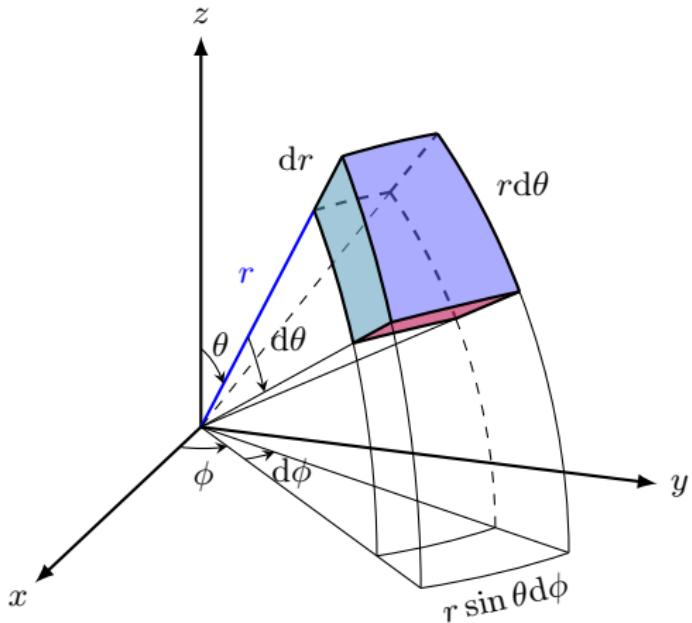
$$V = \frac{4}{3}\pi R_0^3$$

$$V = ???$$



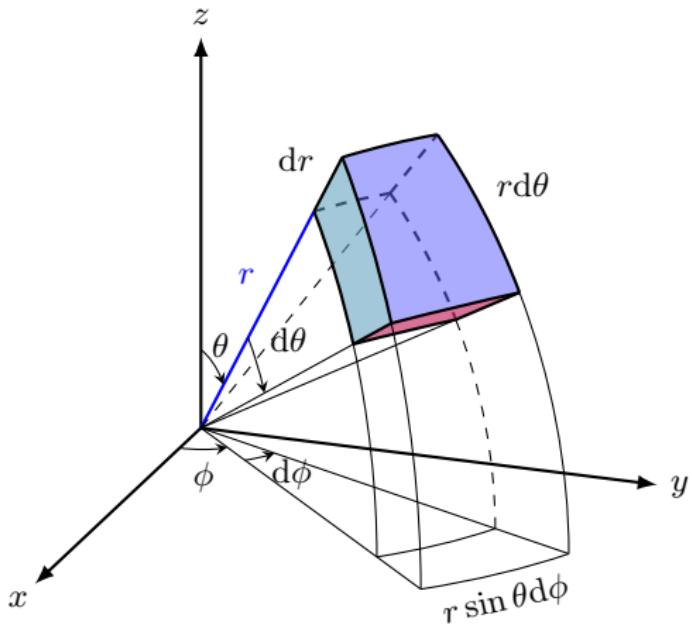


Draw Spherical Volume Element





Define Angular Range

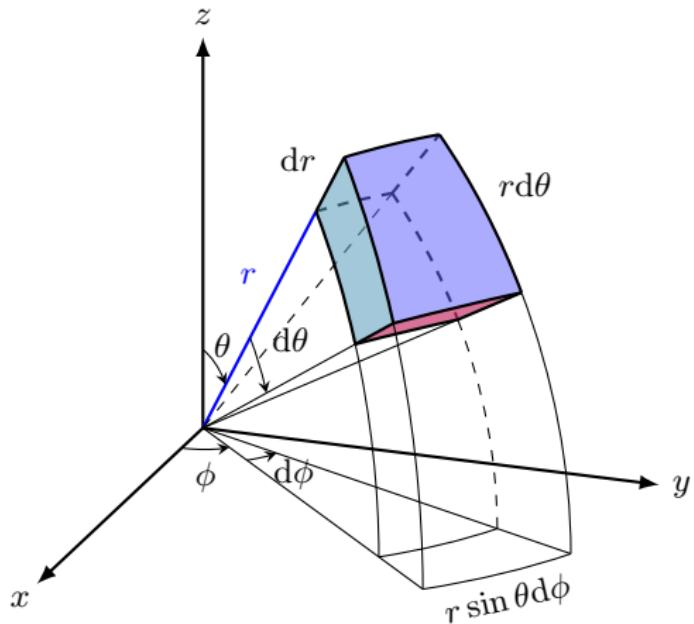


$$\theta \in [0, \pi]$$

$$\phi \in [0, 2\pi]$$



Spherical Volume Element $dV = r^2 \sin \theta d\theta dr d\phi$



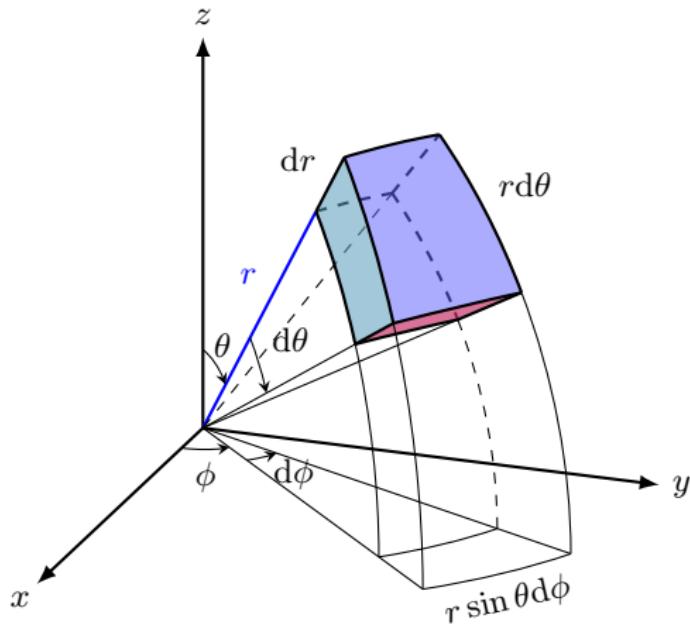
$$\theta \in [0, \pi]$$

$$\phi \in [0, 2\pi]$$

$$dV = dr(r d\phi)(r \sin \theta d\theta)$$



Volume Integral $V = \iiint dV$



$$\theta \in [0, \pi]$$

$$\phi \in [0, 2\pi]$$

$$dV = dr(r d\phi)(r \sin \theta d\theta)$$

$$V = \iiint dV$$

$$= \int_{r=0}^{R_0} \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} dr(r d\phi)(r \sin \theta d\theta)$$



Deformations from Sphericity

Constant radius, spherical volume:

$$dV = dr(r d\phi)(r \sin \theta d\theta) \quad (9)$$

$$V = \iiint dV = \int_{r=0}^{R_0} \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} dr(r d\phi)(r \sin \theta d\theta) = \frac{4}{3}\pi R_0^3 \quad (10)$$



Deformations from Sphericity

Constant radius, spherical volume:

$$dV = dr(r d\phi)(r \sin \theta d\theta) \quad (9)$$

$$V = \iiint dV = \int_{r=0}^{R_0} \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} dr(r d\phi)(r \sin \theta d\theta) = \frac{4}{3}\pi R_0^3 \quad (10)$$

Vary the radius, keep volume fixed:

$$V(\theta) = \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \int_{\phi=0}^{2\pi} \rho^2 \sin \theta d\theta d\rho d\phi \quad (11)$$



Can We Switch The Order of Integration?

$$\int_{r=0}^{R_0} \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} dr (rd\phi) (r \sin \theta d\theta) \stackrel{?}{=} \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \int_{\phi=0}^{2\pi} \rho^2 \sin \theta d\theta d\rho d\phi \quad (12)$$



Yes, We Can Switch The Order of Integration

The Fubini Theorem relates **multiple integrals** to **iterated integrals**:

Theorem (Fubini)

Given $f: \mathbb{R}^2 \mapsto \mathbb{R}$ a continuous function on the closed interval $R = [a, b] \times [c, d]$ then

$$\iint_R f(x, y) dA = \int_a^b \int_c^d f(x, y) dx dy = \int_c^d \int_a^b f(x, y) dy dx. \quad (13)$$

Proof.

See e.g. Saeed Zakeri, CUNY





First Volume Integral: ϕ integration

$$\begin{aligned} V(\theta) &= \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \int_{\phi=0}^{2\pi} \rho^2 \sin \theta \, d\theta \, d\rho \, d\phi \\ &= \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \times \phi \Big|_0^{2\pi} \\ &= 2\pi \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \, d\rho \end{aligned} \tag{14}$$



Second Volume Integral: ρ integration

$$V(\theta) = 2\pi \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \, d\rho$$



Second Volume Integral: ρ integration

$$\begin{aligned} V(\theta) &= 2\pi \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \, d\rho \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} \rho^3 \sin \theta \, d\theta \Big|_0^{r(\theta)} \end{aligned}$$



Second Volume Integral: ρ integration

$$\begin{aligned} V(\theta) &= 2\pi \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \, d\rho \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} \rho^3 \sin \theta \, d\theta \Big|_0^{r(\theta)} \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} r^3(\theta) \sin \theta \, d\theta \end{aligned}$$



Second Volume Integral: ρ integration

$$\begin{aligned} V(\theta) &= 2\pi \int_{\theta=0}^{\pi} \int_{\rho=0}^{r(\theta)} \rho^2 \sin \theta \, d\theta \, d\rho \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} \rho^3 \sin \theta \, d\theta \Big|_0^{r(\theta)} \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} r^3(\theta) \sin \theta \, d\theta \\ &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} (R_0 (1 + \alpha_0 + \alpha_2 P_2(\cos \theta)))^3 \sin \theta \, d\theta \end{aligned} \tag{15}$$



Radial Variations Must Conserve Volume

Using equation 5 for the Legendre expansion...

$$\begin{aligned}\frac{4\pi}{3}R_0^3 &= \frac{2\pi}{3} \int_{\theta=0}^{\pi} r^3(\theta) \sin \theta d\theta \\ &= \frac{2\pi}{3}R_0^3 \int_{\theta=0}^{\pi} (1 + \alpha_0 + \alpha_2 P_2(\cos \theta))^3 \sin \theta d\theta\end{aligned}\tag{16}$$

Conclusion: To conserve volume,

$$1 = \int_{\theta=0}^{\pi} (1 + \alpha_0 + \alpha_2 P_2(\cos \theta))^3 \sin \theta d\theta\tag{17}$$



Express Legendre Polynomials as Monomials.

$$\int_{\theta=0}^{\pi} (1 + \alpha_0 + \alpha_2 P_2(\cos \theta))^3 \sin \theta \, d\theta =$$
$$\int_{\theta=0}^{\pi} \left(1 + \alpha_0 + \frac{\alpha_2}{2} (3 \cos^2 \theta - 1)\right)^3 \sin \theta \, d\theta$$



Reminder: Pascal's Triangle

$n = 0:$

1

$n = 1:$

1 1

$n = 2:$

1 2 1

$n = 3:$

1 3 3 1

$n = 4:$

1 4 6 4 1



Resolve Cubic Expression

Recall Pascal's Triangle...

$$(a + b)^3 = a^3 + 3a^2b + 3ab^2 + b^3$$



Resolve Cubic Expression

Recall Pascal's Triangle...

$$(a + b)^3 = a^3 + 3a^2b + 3ab^2 + b^3$$

$$\begin{aligned} a &\rightarrow 1 + \alpha_0, \\ b &\rightarrow \frac{\alpha_2}{2} (3 \cos^2 \theta - 1) \end{aligned} \tag{18}$$

$$r^3(\theta) = R_0^3 \left(\underbrace{1 + \alpha_0}_a + \underbrace{\frac{\alpha_2}{2} (3 \cos^2 \theta - 1)}_b \right)^3 \tag{5}$$



Mathematica Moment

In[35]:= $\text{Expand}[r[\theta]^3 \sin[\theta]]$

Out[35]= $R^3 \sin[\theta] + 3 R^3 \alpha \theta \sin[\theta] + 3 R^3 \alpha \theta^2 \sin[\theta] + R^3 \alpha \theta^3 \sin[\theta] - \frac{3}{2} R^3 \alpha^2 \sin[\theta] - 3 R^3 \alpha \theta \alpha^2 \sin[\theta] - \frac{3}{2} R^3 \alpha \theta^2 \alpha^2 \sin[\theta] + \frac{3}{4} R^3 \alpha^2 \sin[\theta] + \frac{3}{4} R^3 \alpha \theta \alpha^2 \sin[\theta] - \frac{1}{8} R^3 \alpha^2 \sin[\theta] + \frac{9}{2} R^3 \alpha^2 \cos[\theta]^2 \sin[\theta] + 9 R^3 \alpha \theta \alpha^2 \cos[\theta]^2 \sin[\theta] + \frac{9}{2} R^3 \alpha \theta^2 \alpha^2 \cos[\theta]^2 \sin[\theta] - \frac{9}{2} R^3 \alpha^2 \cos[\theta]^2 \sin[\theta] - \frac{9}{2} R^3 \alpha \theta \alpha^2 \cos[\theta]^2 \sin[\theta] + \frac{9}{8} R^3 \alpha^2 \cos[\theta]^2 \sin[\theta] + \frac{27}{4} R^3 \alpha^2 \cos[\theta]^4 \sin[\theta] + \frac{27}{4} R^3 \alpha \theta \alpha^2 \cos[\theta]^4 \sin[\theta] - \frac{27}{8} R^3 \alpha^2 \cos[\theta]^4 \sin[\theta] + \frac{27}{8} R^3 \alpha^2 \cos[\theta]^6 \sin[\theta]$



Strategy: Divide and Conquer





Resolve Cubic Expression

$$\begin{aligned} r^3(\theta) &= R_0^3 (1 + \alpha_0 + \alpha_2 P_2(\cos \theta))^3 \\ &= R_0^3 \left(1 + \alpha_0 + \frac{\alpha_2}{2} (3 \cos^2 \theta - 1)\right)^3 \end{aligned} \tag{5}$$



Resolve Cubic Expression

$$\begin{aligned} r^3(\theta) &= R_0^3 (1 + \alpha_0 + \alpha_2 P_2(\cos \theta))^3 \\ &= R_0^3 \left(1 + \alpha_0 + \frac{\alpha_2}{2} (3 \cos^2 \theta - 1)\right)^3 \end{aligned} \tag{5}$$

Collect terms in **powers** of $\cos \theta$

$$\begin{aligned} r^3(\theta) &= R_0^3 (\textbf{constant} + \textbf{quadratic } \cos^2 \theta \\ &\quad + \textbf{quartic } \cos^4 \theta \\ &\quad + \textbf{sextic } \cos^6 \theta) \end{aligned} \tag{19}$$



Resolve θ Components

$$\begin{aligned}\text{constant} &= \frac{3}{24} \left(-\alpha_2^3 + 6\alpha_2^2 (1 + \alpha_0) + 12\alpha_2 (-\alpha_0^2 - 2\alpha_0 + 1) \right) \\ \text{quadratic} &= \frac{9}{8} \alpha_2 (\alpha_2 - 2\alpha_0 - 2)^2 \\ \text{quartic} &= \frac{27}{8} \alpha_2^2 (2 + 2\alpha_0 - \alpha_2) \\ \text{sextic} &= \frac{27}{8} \alpha_2^2\end{aligned}\tag{20}$$



Resolve θ Components

Let's look at the θ terms, for example,

$$\frac{27}{8} \alpha_2^3 \int \sin \theta \cos^6 \theta d\theta$$



Simple Integrals

Let $w = \cos \theta$, then $dw = -\sin \theta d\theta$ and for $k \in \mathbb{N}$

$$\int \sin \theta \cos^k \theta d\theta \Rightarrow \int w^k (-dw) \Rightarrow -\frac{1}{k+1} w^{k+1} \Rightarrow -\frac{1}{k+1} \cos^{k+1} \theta \quad (21)$$

For example...

$$\int \sin \theta \cos^2 \theta d\theta = -\frac{1}{3} \cos^3 \theta$$

$$\int \sin \theta \cos^4 \theta d\theta = -\frac{1}{5} \cos^5 \theta$$

$$\int \sin \theta \cos^6 \theta d\theta = -\frac{1}{7} \cos^7 \theta$$



Really Simple Integrals

For k even,

$$\int_0^\pi \sin \theta \cos^k \theta d\theta = -\frac{1}{k+1} \cos^{k+1} \theta \Big|_0^\pi = \frac{2}{k+1} \quad (22)$$



Basic Integrations

$$\int_0^\pi r^3(\theta) \sin \theta \, d\theta = 2R_0^3 \left(\begin{array}{l} \text{constant} \times \frac{1}{1} \\ + \quad \text{quadratic} \times \frac{1}{3} \\ + \quad \text{quartic} \times \frac{1}{5} \\ + \quad \text{sextic} \times \frac{1}{7} \end{array} \right) \quad (23)$$



Final Volume Integral: θ

$$\begin{aligned} & \int_0^\pi r^3(\theta) \sin \theta d\theta = \\ & 2R_0^3 \left(\frac{3}{24} (-\alpha_2^3 + 6\alpha_2^2(1 + \alpha_0) + 12\alpha_2(-\alpha_0^2 - 2\alpha_0 + 1)) \right. \\ & \left. + \frac{9}{8}\alpha_2(\alpha_2 - 2\alpha_0 - 2)^2 \times \frac{1}{3} \right. \\ & \left. + \frac{27}{8}\alpha_2^2(2 + 2\alpha_0 - \alpha_2) \times \frac{1}{5} \right. \\ & \left. + \frac{27}{8}\alpha_2^2 \times \frac{1}{7} \right) \end{aligned} \quad (23)$$



Integration over θ

$$\int_0^\pi r^3(\theta) \sin \theta \, d\theta = 2R_0^3 \left((1 + \alpha_0)^3 + \frac{3}{5} \alpha_2^2 (1 + \alpha_0) + \frac{2}{35} \alpha_2^3 \right) \quad (23)$$



Volume Integration

$$\int_0^{\pi} r^3(\theta) \sin \theta \, d\theta = 2R_0^3 \left((1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \frac{2}{35}\alpha_2^3 \right) \quad (23)$$



Volume is Conserved

sphere = intermediates

$$\begin{aligned}\frac{4\pi}{3}R_0^3 &= \frac{2\pi}{3} \int_0^\pi r^3(\theta) \sin \theta d\theta \\ &= \frac{4\pi}{3}R_0^3 \left((1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \frac{2}{35}\alpha_2^3 \right)\end{aligned}\tag{24}$$



Volume is Conserved

sphere = intermediates

$$\begin{aligned}\frac{4\pi}{3}R_0^3 &= \frac{2\pi}{3} \int_0^\pi r^2(\theta) \sin \theta d\theta \\ &= \frac{4\pi}{3}R_0^3 \left((1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \frac{2}{35}\alpha_2^3 \right)\end{aligned}\quad (24)$$

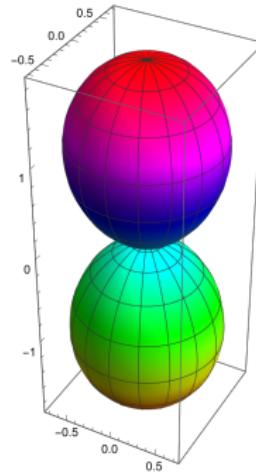
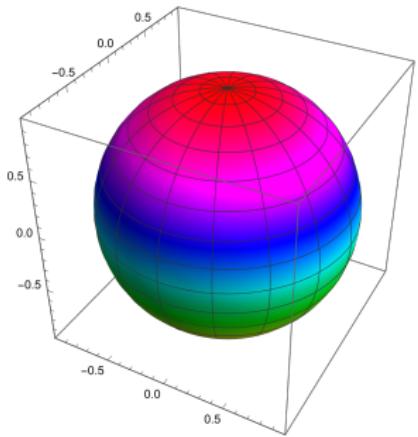
$$1 = (1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \frac{2}{35}\alpha_2^3$$



How To Compute Volume of Deformed Nucleus?

$$V = \frac{4}{3}\pi R_0^3$$

$$V = \frac{4}{3}\pi R_0^3 \times \\ (1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \frac{2}{35}\alpha_2^3$$





Recap

In order to **conserve volume**, we require

$$(1 + \alpha_0)^3 + \frac{3}{5} \alpha_2^2 (1 + \alpha_0) + \frac{2}{35} \alpha_2^3 = 1 \quad (24)$$



To Understand Variation of α_0 , α_2

To characterize the connection between α_0 and α_2 , drop the term with α_2^3 :

$$(1 + \alpha_0)^3 + \frac{3}{5}\alpha_2^2 (1 + \alpha_0) + \cancel{\frac{2}{35}\alpha_2^3} = 1 \quad (25)$$



Characterize Variation of α_0 , α_2

Solve for α_2 with basic algebra:

$$(1 + \alpha_0)^3 + \frac{3}{5} (1 + \alpha_0) \alpha_2^2 = 1 \quad (26)$$



$$\frac{3}{5} (1 + \alpha_0) \alpha_2^2 = 1 - (1 + \alpha_0)^3$$



$$\alpha_2^2 = \frac{5}{3} \frac{1 - (1 + \alpha_0)^3}{1 + \alpha_0}$$



To Understand Variation of α_0, α_2

Solve for α_2 with basic algebra:

$$\alpha_2 = \pm \sqrt{\frac{\frac{5}{3} - 3\alpha_0(1 + \alpha_0 + \alpha_0^2)}{1 + \alpha_0}}$$

$$\alpha_2 = \pm \sqrt{\frac{\frac{5}{3} - 3\alpha_0(1 + \alpha_0 + \alpha_0^2)}{1 + \alpha_0}}$$

$$\alpha_2 \approx \pm \sqrt{-5\alpha_0} + \mathcal{O}\left(\alpha_0^{5/2}\right) \quad (27)$$



Variations of α_0 , α_2 Which Conserve Volume

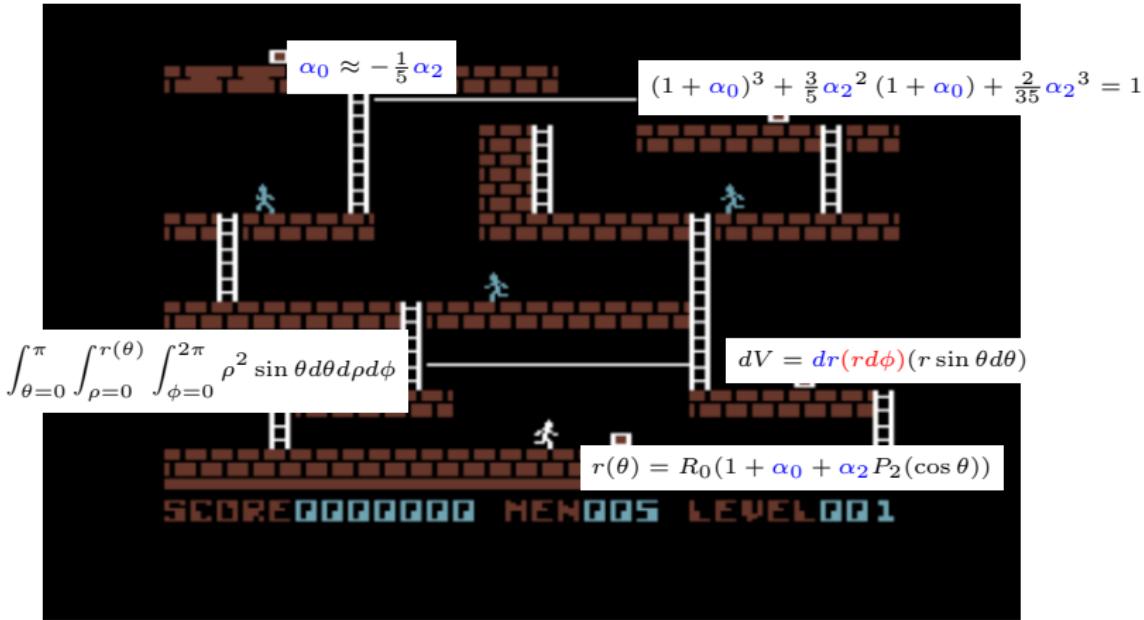
Let the drop radius vary so that **volume is conserved**.

$$\alpha_2 \approx \sqrt{-5\alpha_0} \quad (27)$$

$$\alpha_0 \approx -\frac{1}{5}\alpha_2 \quad (28)$$



Looking Back





Bohr–Wheeler Craftsmanship

$$r(\theta) = R[1 + \alpha_0 + \alpha_2 P_2(\cos \theta)$$

linear

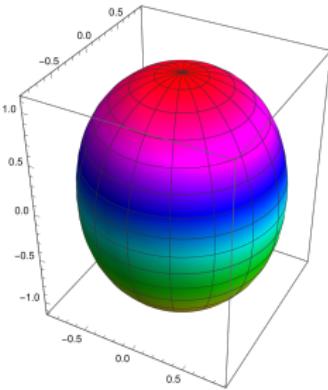
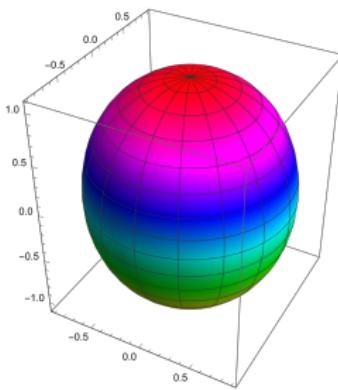
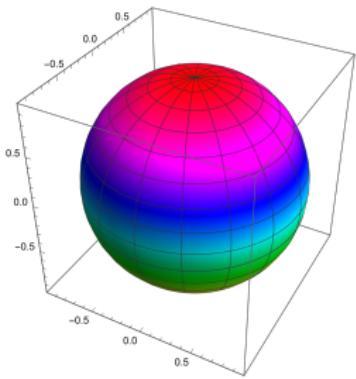
quadratic

$$+ \alpha_3 P_3(\cos \theta) + \dots], \quad (8)$$

cubic

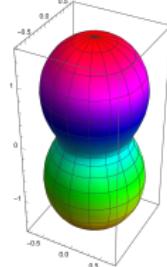
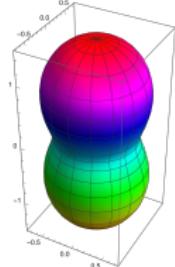
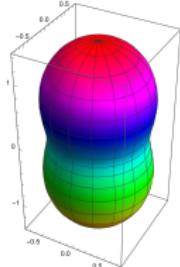
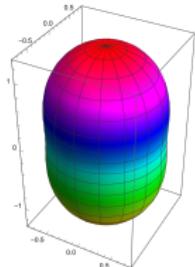
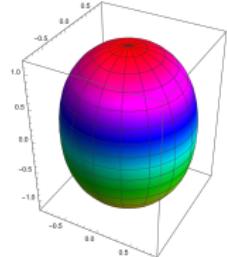
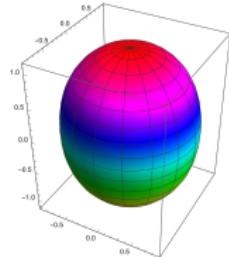
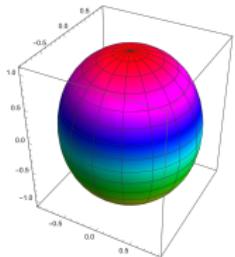
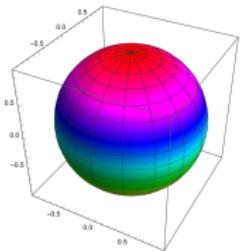


Scission Sequence



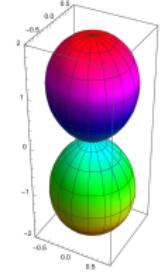
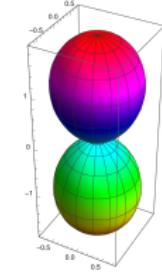
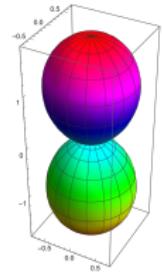
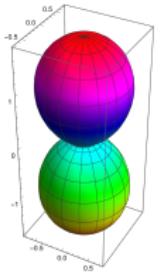
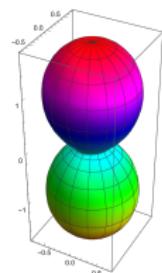
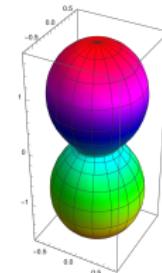
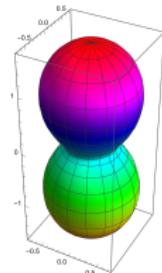
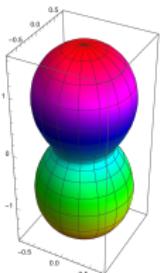


Early Scission Sequence



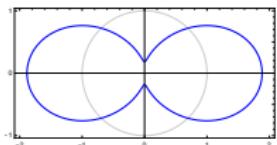
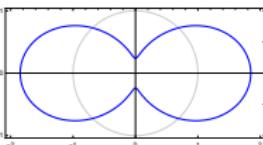
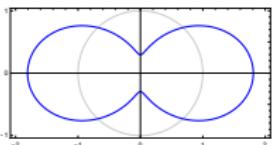
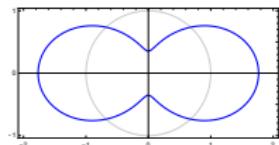
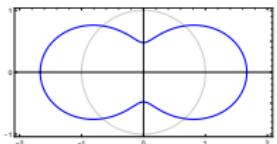
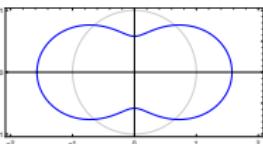
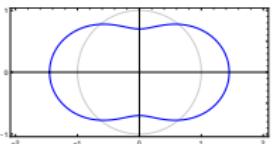
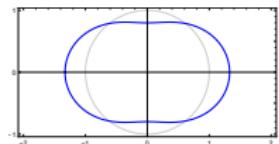
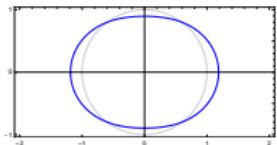
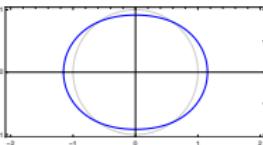
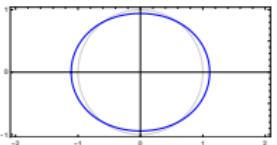
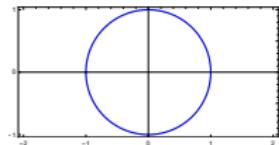


Late Scission Sequence



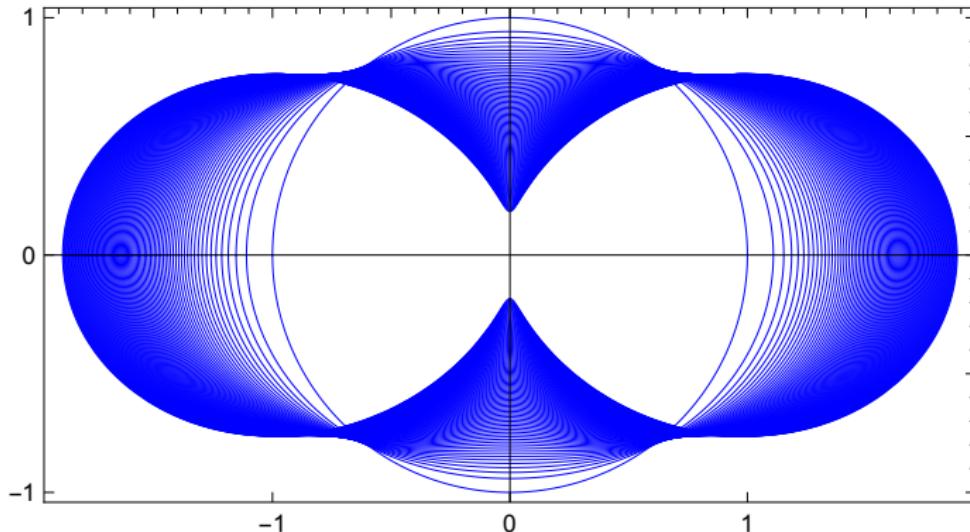


Scission Sequence Outlines



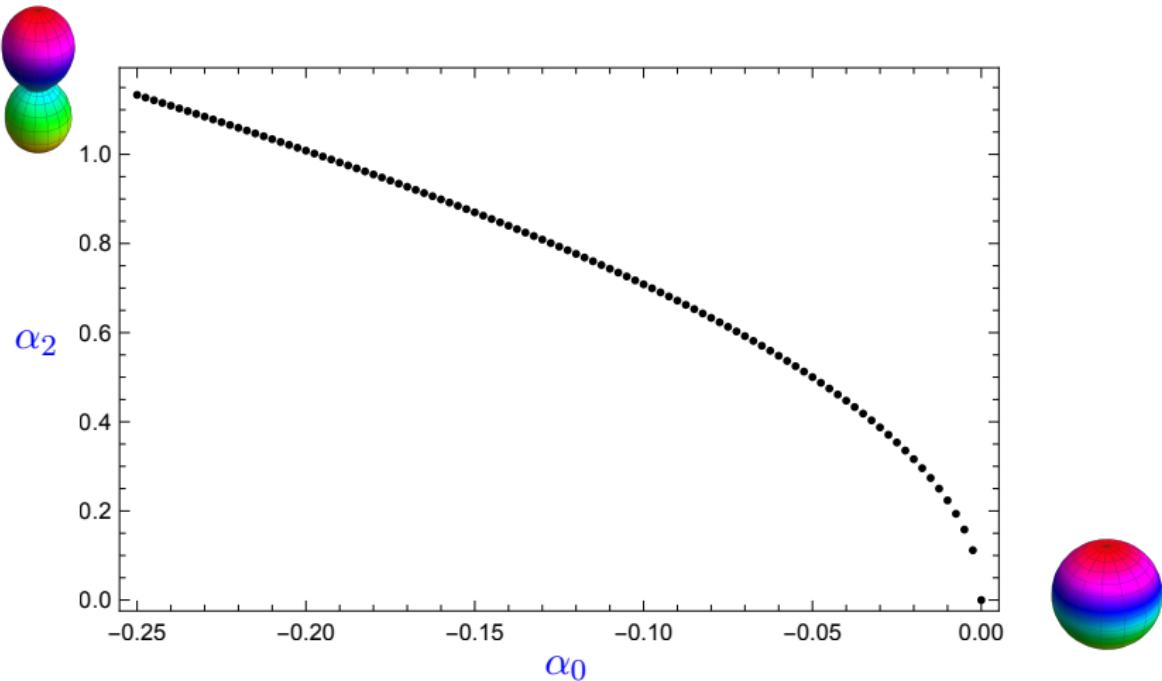


Separation Sequence



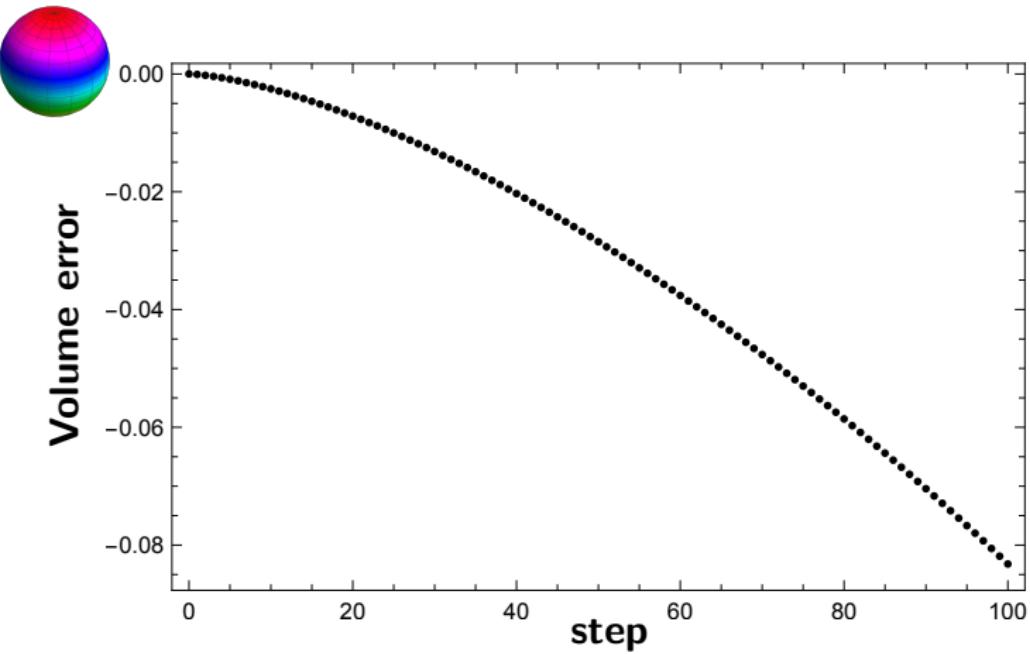


Bohr–Wheeler Deformation Sequence





Bohr–Wheeler Error in Volume





Birth of the Legendre Polynomials

“In the study of the attraction of spheroids and planetary motions” (Sansone 1959, ch. 3, p. 169)

$$\frac{1}{r} = (1 - 2\rho \cos \gamma + \rho^2)^{-\frac{1}{2}} \quad (29)$$



Birth of the Legendre Polynomials

“In the study of the attraction of spheroids and planetary motions” (Sansone 1959, ch. 3, p. 169)

$$\frac{1}{r} = (1 - 2\rho \cos \gamma + \rho^2)^{-\frac{1}{2}} \quad (29)$$

Think: Law of Cosines Again



Binomial Series

For $|\rho| < 1$

$$\begin{aligned}\frac{1}{r} &= (1 - \rho e^{i\gamma})^{-\frac{1}{2}} (1 - \rho e^{-i\gamma})^{-\frac{1}{2}} \\ &= \sum_{n=0}^{\infty} (-1)^n \binom{-\frac{1}{2}}{n} \rho^n e^{iny}\end{aligned}\tag{30}$$



Using the Binomial Series

For $|\rho| < 1$

$$\begin{aligned}\frac{1}{r} &= (1 - \rho e^{i\gamma})^{-\frac{1}{2}} \quad (1 - \rho e^{-i\gamma})^{-\frac{1}{2}} \\ &= \sum_{n=0}^{\infty} (-1)^n \binom{-\frac{1}{2}}{n} \rho^n e^{in\gamma} \quad \sum_{n=0}^{\infty} (-1)^n \binom{-\frac{1}{2}}{n} \rho^n e^{-in\gamma}\end{aligned}\tag{31}$$



Using the Cauchy Product

**Multiplication of two absolutely convergent series
when $n > 0$**

$$(f \circ g)(n) = \sum_{k=0}^n f(n)g(n-k)$$



Using the Cauchy Product

$$\frac{1}{r} = \sum_{n=0}^{\infty} \rho^n \sum_{j=0}^{[n/2]} \binom{-\frac{1}{2}}{j} \binom{-\frac{1}{2}}{n-j} \times \\ \left((-1)^j (-1)^{n-j} e^{ij\gamma} e^{-i(n-j)\gamma} + (-1)^j (-1)^{n-j} e^{-ij\gamma} e^{i(n-j)\gamma} \right)$$



Introducing the Legendre Polynomials!

$$\frac{1}{r} = (1 - 2\rho \cos \gamma + \rho^2)^{-\frac{1}{2}} = \sum_{n=0}^{\infty} \rho^n P_n(\cos \gamma) \quad (29)$$



Introducing the Legendre Polynomials!

$$\frac{1}{r} = (1 - 2\rho \cos \gamma + \rho^2)^{-\frac{1}{2}} = \sum_{n=0}^{\infty} \rho^n P_n(\cos \gamma) \quad (29)$$

$$P_n(\cos \gamma)$$

$$= (-1)^n \left(\binom{-\frac{1}{2}}{n} 2 \cos n\gamma + \binom{-\frac{1}{2}}{n-1} \binom{-\frac{1}{2}}{1} 2 \cos(n-2)\gamma + \dots \right)$$



Deriving Bonnet's Recursion Formula

Start with the generating function

$$\frac{1}{\sqrt{1 - 2xt + t^2}} = \sum_{k=0}^{\infty} P_k(x)t^k \quad (32)$$

Differentiate w.r.t.:

$$\begin{aligned} \partial_t \left(\frac{1}{\sqrt{1 - 2xt + t^2}} \right) &= \partial_t \left(\sum_{k=0}^{\infty} P_k(x)t^k \right) \\ \frac{x - t}{(1 - 2xt + t^2)^{3/2}} &= \sum_{k=1}^{\infty} kP_k(x)t^{k-1} \end{aligned} \quad (33)$$



Continuing...

Multiply both sides by the polynomial term...

$$\frac{x - t}{(1 - 2xt + t^2)^{3/2}} = \sum_{k=1}^{\infty} kP_k(x)t^{k-1}$$



Continuing...

Multiply both sides by the polynomial term...

$$\begin{aligned}\frac{x-t}{(1-2xt+t^2)^{3/2}} &= \sum_{k=1}^{\infty} kP_k(x)t^{k-1} \\ \frac{x-t}{\sqrt{1-2xt+t^2}} &= (1-2xt+t^2) \sum_{k=1}^{\infty} kP_k(x)t^{k-1}\end{aligned}\tag{34}$$



Continuing...

Multiply both sides by the polynomial term...

$$\begin{aligned}\frac{x-t}{(1-2xt+t^2)^{3/2}} &= \sum_{k=1}^{\infty} kP_k(x)t^{k-1} \\ \frac{x-t}{\sqrt{1-2xt+t^2}} &= (1-2xt+t^2) \sum_{k=1}^{\infty} kP_k(x)t^{k-1}\end{aligned}\tag{34}$$

Substitute using equation (32)

$$(x-t) \sum_{k=0}^{\infty} P_k(x)t^k = (1-2xt+t^2) \sum_{k=1}^{\infty} kP_k(x)t^{k-1}\tag{35}$$



Distribute Powers of t

$$(x - t) \sum_{k=0}^{\infty} P_k(x) t^k = (1 - 2xt + t^2) \sum_{k=1}^{\infty} k P_k(x) t^{k-1} \quad (35)$$

Distribute powers of t



Distribute Powers of t

$$(x - t) \sum_{k=0}^{\infty} P_k(x)t^k = (1 - 2xt + t^2) \sum_{k=1}^{\infty} kP_k(x)t^{k-1} \quad (35)$$

Distribute powers of t

$$\begin{aligned} \sum_{k=0}^{\infty} xP_k(x)t^{\textcolor{blue}{k}} & - \sum_{k=0}^{\infty} P_k(x)t^{\textcolor{red}{k+1}} = \\ \sum_{k=0}^{\infty} kP_k(x)t^{k-1} & - \sum_{k=0}^{\infty} 2kxP_k(x)t^{\textcolor{blue}{k}} + \sum_{k=0}^{\infty} kP_k(x)t^{\textcolor{red}{k+1}} \end{aligned} \quad (36)$$



Collect Powers of t

$$\begin{aligned} \sum_{k=0}^{\infty} x P_k(x) t^k & - \sum_{k=0}^{\infty} P_k(x) t^{k+1} = \\ \sum_{k=0}^{\infty} k P_k(x) t^{k-1} & - \sum_{k=0}^{\infty} 2x P_k(x) t^k + \sum_{k=0}^{\infty} k P_k(x) t^{k+1} \end{aligned} \tag{36}$$

Rearrangement..

$$\begin{aligned} \sum_{k=0}^{\infty} k P_k(x) t^{k+1} & + \sum_{k=0}^{\infty} P_k(x) t^{k+1} = \\ \sum_{k=0}^{\infty} 2kx P_k(x) t^k & + \sum_{k=0}^{\infty} kx P_k(x) t^k - \sum_{k=0}^{\infty} k P_k(x) t^{k-1} \end{aligned} \tag{36}$$



Collect Powers of t

$$\begin{aligned} \sum_{k=0}^{\infty} k P_k(x) t^{k+1} + \sum_{k=0}^{\infty} P_k(x) t^{k+1} &= \\ \sum_{k=0}^{\infty} 2kxP_k(x)t^k + \sum_{k=0}^{\infty} kxP_k(x)t^k - \sum_{k=0}^{\infty} kP_k(x)t^{k-1} & \end{aligned} \tag{36}$$

Group powers of t

$$\sum_{k=0}^{\infty} (k+1)P_k(x)t^{k+1} = \sum_{k=0}^{\infty} (2k+1)xP_k(x)t^k - \sum_{k=0}^{\infty} kP_k(x)t^{k-1} \tag{37}$$



When Are Series Equal?

Definition

Two summable infinite series

$$\sum_{k=0}^{\infty} a_k < \infty, \sum_{k=0}^{\infty} b_k < \infty$$

are equal iff they exhibit equality of the partial sums:

$$\sum_{k=0}^n a_k = \sum_{k=0}^n b_k \quad \forall n \geq 0$$

The series are equal term by term:

$$a_k = b_k \quad k = 0 : \infty$$



Equate Coefficients Term by Term

“Eliminate” the summations

$$\sum_{k=0}^{\infty} (k+1)P_k(x)t^{k+1} = \sum_{k=0}^{\infty} (2k+1)xP_k(x)t^k - \sum_{k=0}^{\infty} kP_k(x)t^{k-1} \quad (37)$$

Revealing Bonnet's Recursion Formula

$$(k+1)P_{k+1}(x) = (2k+1)xP_k(x) - nP_{k-1}(x) \quad (38)$$



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Nuclear Fission Using Algebra and Calculus

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