

Problem Set 2

Chemistry 675, Fall 2024

Due Date: September 20, 2024

Relevant Chapters: Excerpts of Levine Chapters 1,2, 3, and 7

1. Let the initial wavefunction be defined by

$$\Psi(x, t = 0) = \frac{1}{\sqrt{2}} (\phi_1(x) - \phi_3(x)) \quad (1)$$

where $\phi_i(x)$ is the i^{th} lowest-energy eigenstate of the particle in a box.

- (a) Give the analytic expression for the probability density.
 - (b) Give the analytic expression for the expected evolution of the probability density.
 - (c) Explain how the expected evolution of the probability density relates to the expected result for a nonstationary state.
 - (d) Give the analytic expression for the average position as a function of time.
2. Let the initial wavefunction be defined as the Gaussian wavepacket

$$\Psi(x, t = 0) = \mathcal{N} e^{-a^2(x-L/2)^2/2} \quad (2)$$

where \mathcal{N} is the normalization coefficient, $L = 5 \text{ \AA}$, and $a = 4 \text{ \AA}^{-1}$.

- (a) Demonstrate analytically whether $\Psi(x, t = 0)$ is a stationary state for a particle-in-a-box potential.
- (b) Using the accompanying Jupyter notebook:
 - i. Explain what the function $f(N) = \sum_{i=1}^N c_i^* c_i$ represents.
 - ii. Plot $f(N)$ for the given expansion coefficients c_i .
 - iii. Calculate $\lim_{N \rightarrow \infty} f(N)$.
 - iv. Explain what $f(N)$ says about the number of expansion terms required to accurately simulate the wavefunction.
 - v. Plot $|\Psi(x, t)|^2$ at several choices of t to show how the wavefunction changes in time.
 - vi. Explain what $|\Psi(x, t)|^2$ tells us about the motion of a particle in a box.
 - vii. Plot $\langle x(t) \rangle$ and $\langle (x - L/2)^2(t) \rangle$.
 - viii. Explain the significance of $\langle x(t) \rangle$ and $\langle (x - L/2)^2(t) \rangle$ and their relationships to $|\Psi(x, t)|^2$.