

Topic: Properties of Materials – Metals (I)

Sub-topics:

1. Identification of Metals by Physical Properties
 2. Types of Metals
 3. Meaning and Examples of Alloys
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Lesson Objectives

By the end of this lesson, students should be able to:

- Identify metals based on physical properties.
 - Classify metals into ferrous and non-ferrous types.
 - Define alloys and give examples.
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1. Identification of Metals by Physical Properties

Metals can be identified using the following physical properties:

Property	Description
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Lustre	Metals are shiny when polished.
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Conductivity	Metals conduct heat and electricity well.
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Malleability	Metals can be hammered into sheets.
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Ductility	Metals can be drawn into wires.
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Hardness	Metals are generally hard and strong.
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Sonority	Metals produce a ringing sound when struck.
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Magnetism	Some metals like iron are magnetic.
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Examples:

- Iron is magnetic and hard.
- Copper is reddish and a good conductor.

- Aluminum is light and does not rust.
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2. Types of Metals

a. Ferrous Metals

These are metals that **contain iron**.

They are usually **strong, hard, and magnetic**, but they **rust easily**.

Examples:

Iron

Steel

Cast Iron

b. Non-Ferrous Metals

These **do not contain iron**. They are **lightweight, resist corrosion**, and **non-magnetic**.

Examples:

Copper

Aluminum

Zinc

Lead

Brass (an alloy)

Bronze (an alloy)

3. Alloys

Definition:

An **alloy** is a mixture of two or more metals (or a metal and a non-metal) to improve strength, corrosion resistance, or other properties.

Examples of Alloys:

Alloy	Composition	Uses
Bronze	Copper + Tin	Medals, sculptures
Brass	Copper + Zinc	Musical instruments, door handles
Steel	Iron + Carbon	Construction, machinery
Stainless Steel	Iron + Chromium + Nickel	Cutlery, kitchen equipment