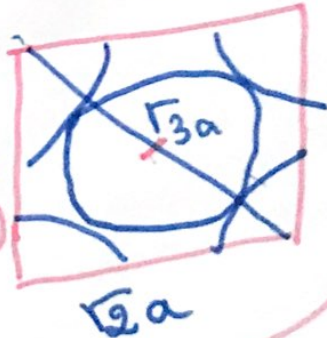
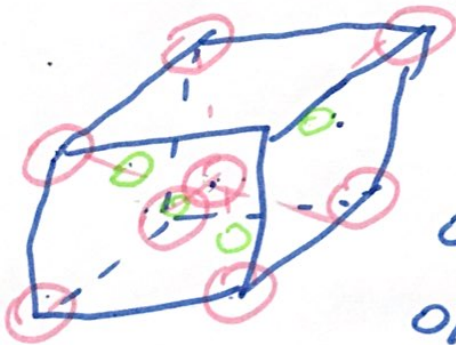


$S/S = \{12\}$ $4R = \sqrt{2}a$



soo colle Cu₂O

O²⁻ C.C. • $\underbrace{8 \times \frac{1}{8}}_{\text{somme}} + \underbrace{1 \wedge 1}_{\text{int}} = 2$
 Cu⁺ milieu des 4 des 8/2 diagonal. $\times \underbrace{4 \times 1}_{\text{int}} = 4$



Cu₂O 2 unités formulaire/maille

$$O/O = [8]$$

$$O/Cu = [4]$$

$$Cu/O = [2]$$

$$Cu/Cu = [12] \text{ (X)}$$

couplage parfait

$$\Phi_1 = L_1 i_1 + M i_2$$

$$\Phi_2 = L_2 i_2 + M i_1$$

$$\text{si } i_2 = 0 \quad \Phi_1 = L_1 i_1 \quad \Phi_2 = M i_1 \Rightarrow \frac{\Phi_2}{\Phi_1} = \frac{M}{L_1}$$

$$\text{si } i_1 = 0 \quad \Phi_1 = M i_2 \quad \Phi_2 = L_2 i_2 \Rightarrow \frac{\Phi_2}{\Phi_1} = \frac{L_2}{M}$$

on a un couplage parfait on a le m $\frac{\Phi_2}{\Phi_1}$

$$M = \sqrt{L_1 L_2}$$

$$\text{on a m } L_1 = k N_1^2$$

$$L_2 = k N_2^2$$

$$M = k N_1 N_2$$

$$k = \mu_0 \frac{S}{L} \mu_r$$

En regime sinusoidal

$$\begin{cases} u_1 = L_1 \frac{di_1}{dt} + M \frac{di_2}{dt} \\ u_2 = L_2 \frac{di_2}{dt} + M \frac{di_1}{dt} \end{cases}$$

en notat \underline{U}

$$\underline{U}_1 = j\omega L_1 \underline{I}_1 + j\omega M \underline{I}_2$$

$$\underline{U}_2 = j\omega L_2 \underline{I}_2 + j\omega M \underline{I}_1$$

$$\underline{U}_1 = j\omega [kN_1^2 \underline{I}_1 + kN_1 N_2 \underline{I}_2]$$

$$\underline{U}_2 = j\omega [kN_2^2 \underline{I}_2 + kN_1 N_2 \underline{I}_1]$$

$$\underline{U}_1 = j\omega kN_1 (N_1 \underline{I}_1 + N_2 \underline{I}_2)$$

$$\underline{U}_2 = j\omega kN_2 (N_2 \underline{I}_2 + N_1 \underline{I}_1)$$

$$\frac{\underline{U}_2}{\underline{U}_1} = \frac{N_2}{N_1}$$

$V = R_i$