

Solved selected problems of An Introduction to Quantum Theory by Keith Hannabuss

Franco Zacco

Chapter 1 - Introduction

Solution. 1.1

- (i) A frequency of $200kHz$ corresponds to an energy of

$$E = \hbar\omega = 1.0546 \times 10^{-34} \cdot 2\pi \cdot 2 \times 10^5 = 1.3252 \times 10^{-28} \text{ J}$$

- (ii) A frequency of $4.95 \times 10^{14}Hz$ corresponds to an energy of

$$E = \hbar\omega = 1.0546 \times 10^{-34} \cdot 2\pi \cdot 4.95 \times 10^{14} = 3.2799 \times 10^{-19} \text{ J}$$

- (iii) A frequency of $10^{20}Hz$ corresponds to an energy of

$$E = \hbar\omega = 1.0546 \times 10^{-34} \cdot 2\pi \cdot 10^{20} = 6.62624 \times 10^{-14} \text{ J}$$

- (iv) A frequency of $10^{23}Hz$ corresponds to an energy of

$$E = \hbar\omega = 1.0546 \times 10^{-34} \cdot 2\pi \cdot 10^{23} = 6.62624 \times 10^{-11} \text{ J}$$

□

Solution. 1.2 Let a radio station broadcast on a frequency of 200 kHz then each photon generated has an energy of

$$E = 1.0546 \times 10^{-34} \cdot 2\pi \cdot 2 \times 10^5 = 1.3252 \times 10^{-28} \text{ J}$$

So a $200\text{ kW} = 200\text{ kJ/s}$ transmitter generates

$$\frac{200 \times 10^3}{1.3252 \times 10^{-28}} = 1.5092 \times 10^{33}$$

photons per second.

Let us assume the radio station broadcasts photons in every direction then the radio station broadcasts the following number of photons per meter squared per second

$$\frac{1.5092 \times 10^{33}}{4\pi \cdot (1 \times 10^6 \text{ m})^2} = 1.2009 \times 10^{20}$$

Finally if the aerial were on a space probe at a distance of 3000 million km of the earth the number of photons per meter squared per second will be

$$\frac{1.5092 \times 10^{33}}{4\pi \cdot (3 \times 10^{12} \text{ m})^2} = 1.3344 \times 10^7$$

□