

Electron density layers

<u>Different Regions of the Ionosphere</u>

- **F** (forms F1 and F2 layers during the day, ionized by EUV 20-80 nm)
- **E** (100 120 kms, ionized by EUV 80-103 nm and X-rays 1-20 nm)
- D (70 90 kms, ionized by X-rays 0.1-1 nm)

Image: NOAA

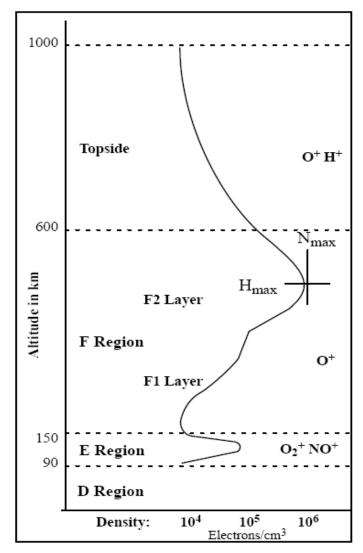
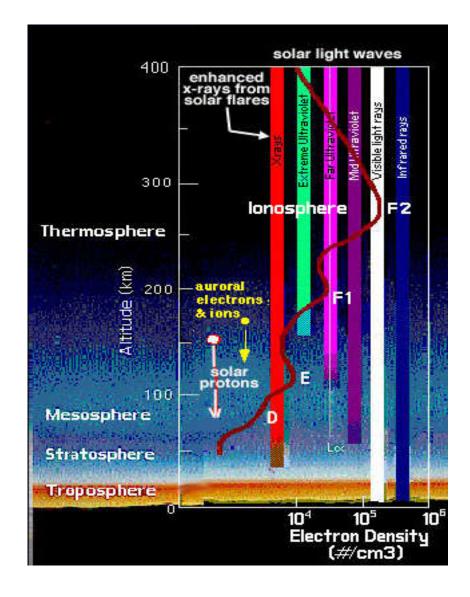


Image: NOAA



Continuity Equation

Electron density

Continuity equation for ionoized species

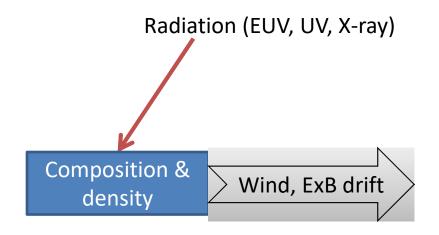
$$\frac{\partial n_j}{\partial t} = (P_j - L_j)M_j - \nabla \cdot (\rho_j V_j)$$

 n_j : Density of ionised gas species j

 M_i : Mass of j

 P_j : Production of j

 L_j : Loss von j



Ionisation/ Production

Primary Photoionisation

Secondary ionisation

Particle precipitation

Primary Photoionisation

Simple Photoionisation

$$X + Photon(\lambda \leq 100 nm) \rightarrow X^+ + e$$

$$O + Photon(\lambda \le 91nm) \rightarrow O^+ + e$$

 $N_2 + Photon(\lambda \le 80nm) \rightarrow N_2^+ + e$
 $O_2 + Photon(\lambda \le 103nm) \rightarrow O_2^+ + e$

Primary source of ionisation

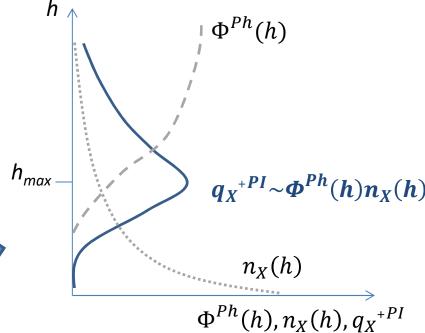
Primary Photoionisation

lonization rate = (radiation
intensity)x(ionisation cross section)x(density)

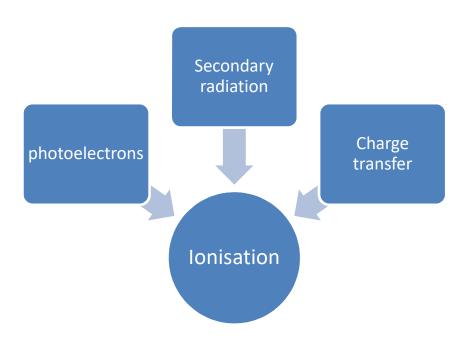
$$q_X^{+PI} = \Phi^{Ph} \sigma_X^I n_X$$

Development of ionisation production layers





Secondary ionisation



Charge transfer

• k_{X^+} reaction constant

- Same ion density
- Production of important ions like O_2^+ in the lower ionosphere and H^+ in the plasmasphere

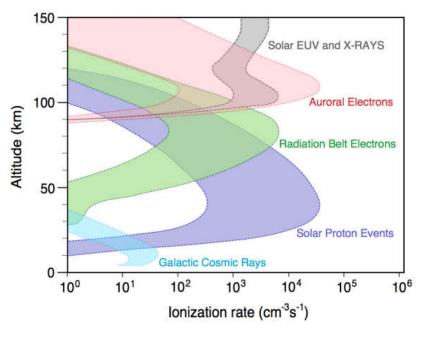
Particle Precipitation

- $X + e_p \rightarrow X^+ + e_p + e_s$
- e_p: primary electron is the precipitating electron
- e_s: secondary electron is the newly produced electron

Important process in high latitudes

- One precipitating electron has the potential to initiate numerous primary ionisation processes
- Often secondary electrons have enough energy to initiate ionisation
- Higher order electrons (tertiary etc.) also have enough energy for ionisation

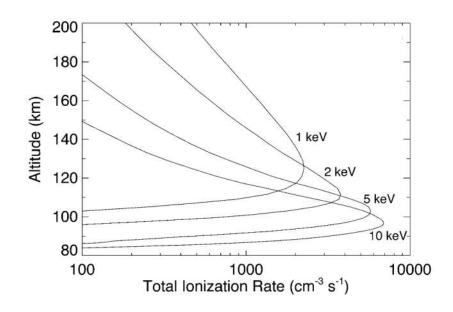
Particle Precipitation



A. Kero (2020)

Particle precipitation

Penetration Depth of Auroral Electrons Depends on Energy



Source: Solomon

Chemical recombination/ transfer processes

Dissociative recombination of molecular ions

$$\bullet XY^{+} + e \xrightarrow{k_{XY}^{+}} X^{(*)} + Y^{(*)}$$

Radiative recombination of atomic ions

$$\bullet X^+ + e \xrightarrow{k_{X^+}} X^{(*)} + h\nu$$

Charge exchange

$$\bullet X^+ + Y \xrightarrow{k_{X^+Y}} X^{(*)} + Y^+$$

Charge exchange (Ionatom interaction)

$$\bullet X^+ + YZ \xrightarrow{k_{X^+YZ}} XY^+ + Z$$

k_X+
 recombination
 constant
 * indicates
 excited state

(Sheehan & Maurice, 2004, https://doi.org/10.1029/2003JA010132)

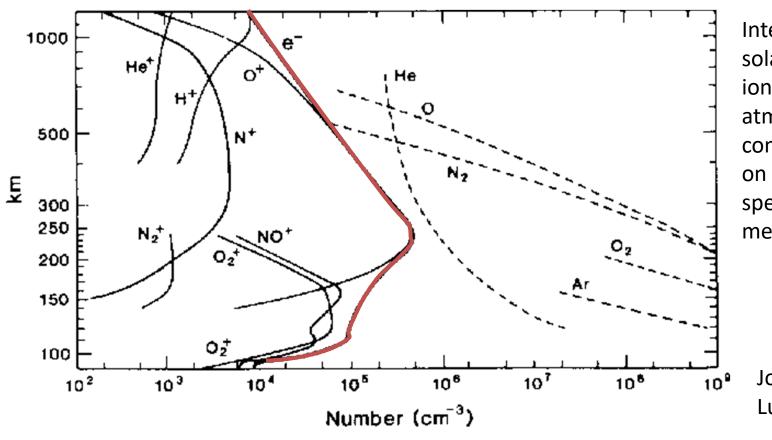
Height dependence of loss

E-region

- NO^+ , O_2^+ are dominating molecular ions
- Fastest loss through dissiciative recombination
- $L(h) \cong \alpha n^2(h)$ Loss increases like squared increase electron density

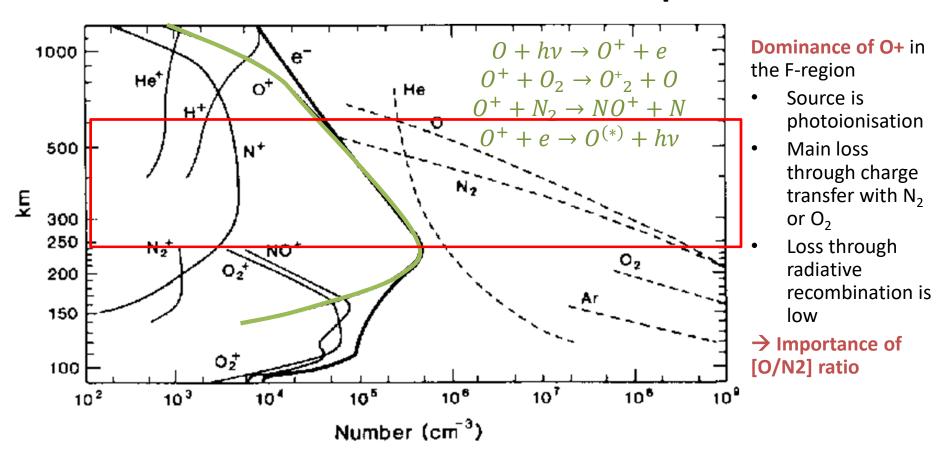
F-region

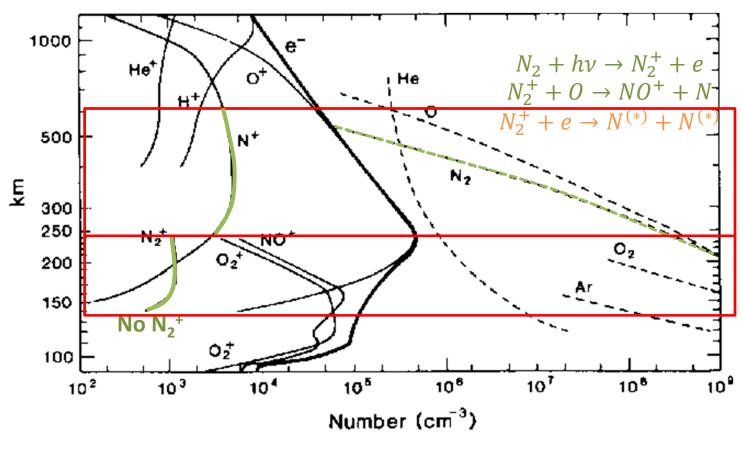
- O⁺ are dominating molecular ions
- Loss mainly trough charge transfer
- $L(h) \cong \beta(h)n(h)$ Loss coefficient β has most impact on the loss



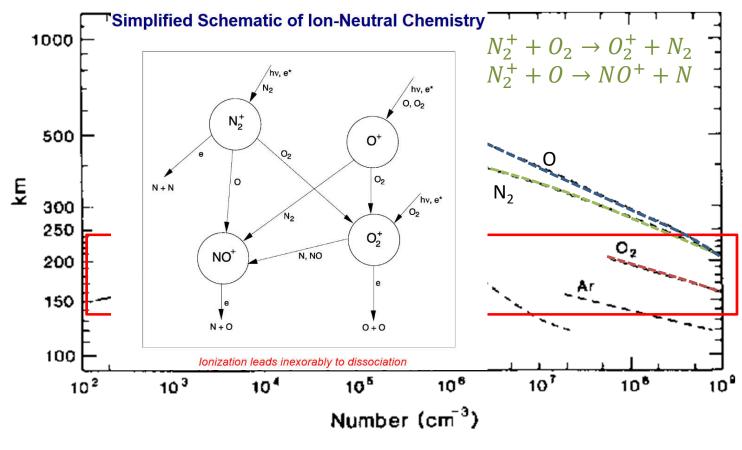
International quiet solar year daytime ionospheric and atmospheric composition based on mass spectrometer measurements.

Johnson (1969), Luhmann(1995)

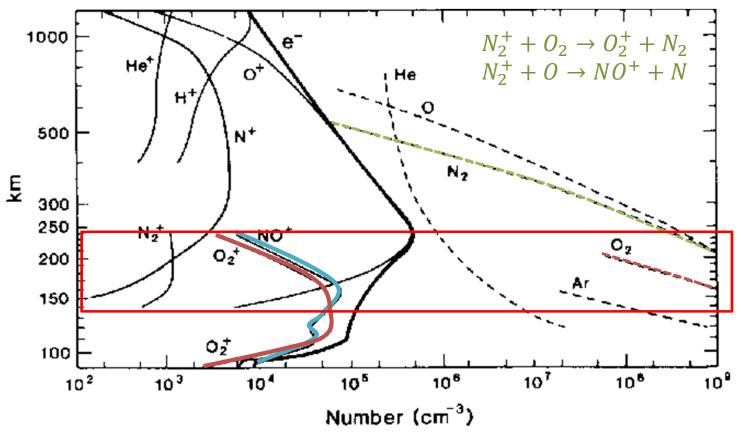




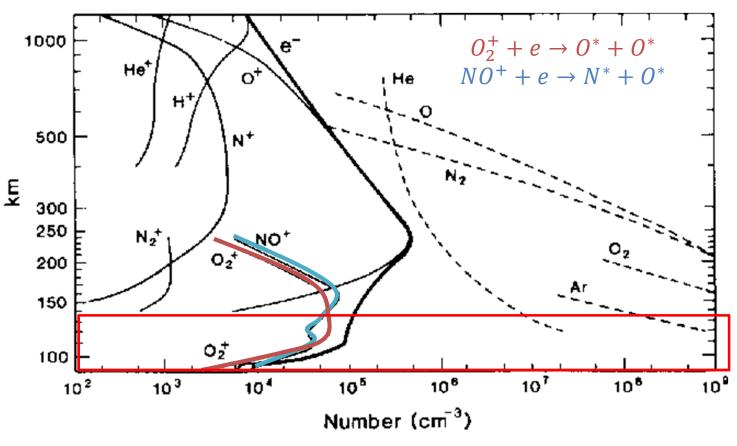
- Dissociative recobination is so fast in the topside ionosphere, that the N₂⁺ basically not exists
- In the bottom side Fregion, speed of dissociative recombination decreases



- O+ dominance decreases with decreasing altitude
- Dissociative recombination is slower
- Loss of N₂⁺ becomes dominated by charge transfer

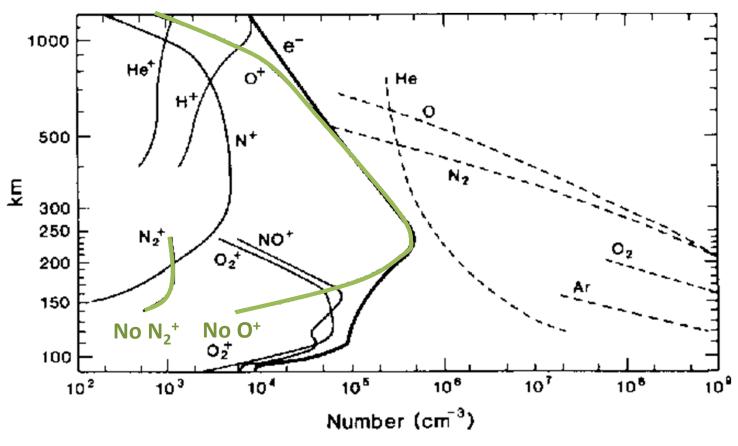


- O+ dominance decreases with decreasing altitude
- Dissociative recombination is slower
- Loss of N₂⁺ becomes dominated by charge transfer
- O₂⁺ and NO⁺ dominating ions in lower altitudes



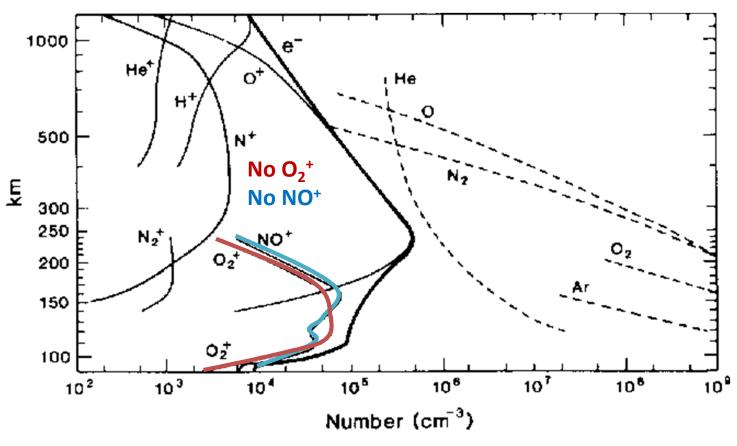
O₂ and N₂ are dominating molecules in E-region

- Very effective production of NO+ in the Eregion through charge transfer
- Photoionisation is minor, because of low density of NO
- Loss of NO+ through dissociative recombination is slower



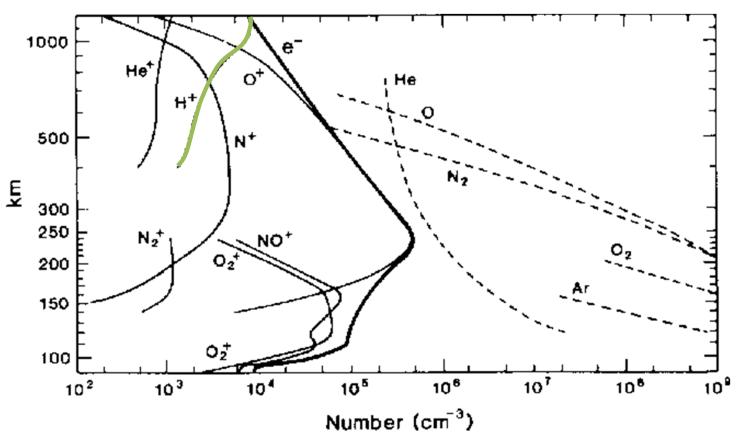
No O^+ and N_2^+ in E-region because of

- little production in that region
- Exponential increase of loss coefficient with decreasing height



No O₂⁺ and NO⁺ in the F-region because of

- decreasing production with altitude and
- Loss rates increase due to increasing electron density



Dominance of H⁺ because of

- Effectiv production through charge transfer
- At the same time there is a loss of O⁺ through the same reaction

D-Layer

- 50-95 km
- Electron density normally does not have a distinct peak
- H-Lymann alpha radiation (lambda=121.5 nm) penetrates down into the D-layer and has enough energy to ionize NO (which is found in small amounts) → production in 70-95 km
- Below 70 km, ionisation from high energetic cosmic radiation
- Contribution of solar X-rays
- All processes are height- and temperature-dependent

EXERCISES

TEC-F10 correlation

- Task 1
 - Plot F10.7 for the period 1998-2021
 - Plot local TEC (~Neustrelitz) for the same period in a comparable format
- Task 2
 - Plot local TEC for the full year 2008 and compare to F10.7 and kp or ap

Seasonal Anomaly

- Use e.g. IRI and MSIS
 - Plot local (~Neustrelitz) number density of atomic oxygen and molecular nitrogen at ~ 250 km altitude for the full year 2008
 - Plot the ratio [O/N2] for the same location and period
 - Plot TEC for the same location and period