



RAJARATA UNIVERSITY OF SRI LANKA  
FACULTY OF APPLIED SCIENCES

B.Sc. in Applied Sciences

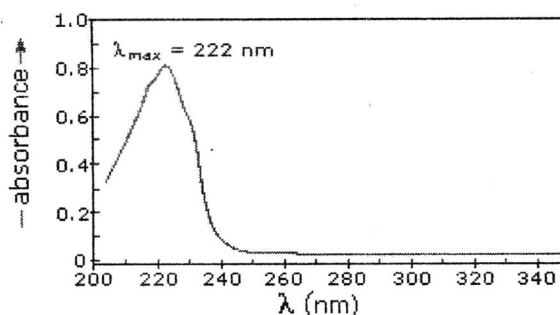
Second Year – Semester II Examination – January / February 2023

CHE 2106 SPECTROSCOPIC METHODS IN ORGANIC CHEMISTRY

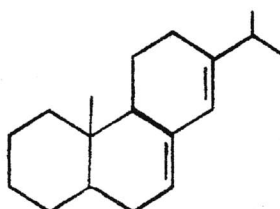
Answer ALL Questions

Time: One (01) hour.

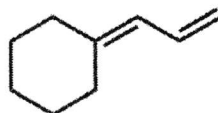
1. a) Spectroscopic techniques are useful to determine the structural features of molecules. Explain the sentence. (06 marks)
- b) The UV spectrum of 2-methyl-1,3-butadiene is given below. Discuss the electronic transitions of the compound. (08 marks)



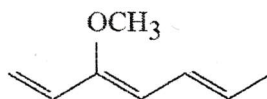
- c) Calculate the UV  $\lambda_{max}$  of the following structures using Woodward-Fischer rule. (Basic value for acyclic conjugated enone : 215 nm, heteroannular conjugated system: 214 nm, homoannular conjugated system 253 nm, homodiene component: 39 nm, extended conjugation: 30 nm, Alkyl substitution: 5 nm, Exocyclic double bond: 5 nm,  $\alpha$ -substitution: 10 nm,  $\beta$ -substitution: 12 nm,  $\gamma$ -substitution: 18 nm) (16 marks)



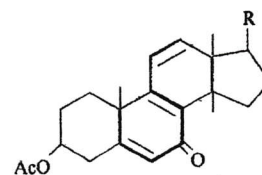
(i)



(ii)

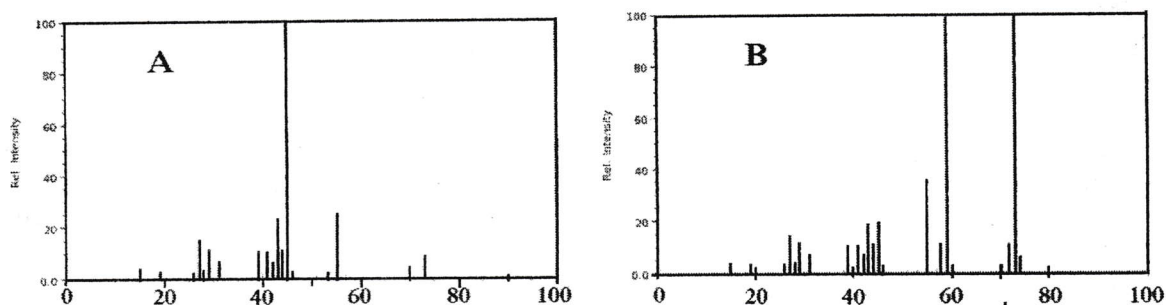


(iii)

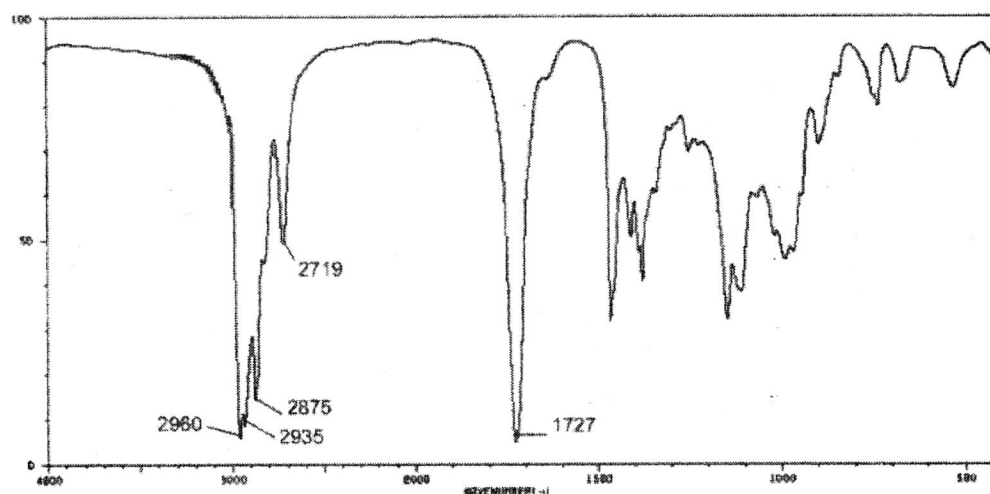


(iv)

2. a) Write a note on "Quadrupole Mass Analyzer" (10 marks)
- b) Mass spectrums of 2-methyl -2-butanol and 2-pentanol are given below. Discuss the fragmentation and identify the relevant spectra. (15 marks)



3. (a). Suggest that how infra-red spectroscopy could be used to differentiate 1-hexyne from 3-hexyne. Draw possible IR spectra for both compounds. (15 marks)
- (b). IR spectra of a compound with molecular formula  $C_5H_{10}O$  is given below. Draw possible structure/s using IR spectrums. (10 marks)



- (c). Write a short account of  $^1H$ -NMR and draw the  $^{13}C$ -NMR spectrum of ethyl acetate (20 marks)

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