

RAJARATA UNIVERSITY OF SRI LANKA FACULTY OF APPLIED SCIENCES

B.Sc. (General) Degree in Applied Sciences Third Year - Semester I Examination — September/ October 2019

CHE 3203 – CHEMISTRY OF POLYMERS

Time: Two (02) hours

Answer all questions.

Use of a non-programmable calculator is permitted.

Molar masses in gmol⁻¹: H = 1, C = 12, O = 16

1. a) Compare the properties of thermoplastics with the thermosetting plastics.

(20 marks)

- b) What are the factors that determine the glass transition temperature of a polymer? (15 marks)
- c) How would you distinguish elastomers, thermoplastics and engineering thermoplastics based on glass transition temperature? (15 marks)
- d) Briefly explain how the melting temperature of a polymer influences by the chemical structure. (20 marks)
- e) Give the chemical reactions for the preparation of two types phenol-formaldehyde resins. Indicate the reaction conditions appropriately. (30 marks)
- 2. a) Write the suitable chemical reactions for Initiation, Propagation and Termination of free radical vinyl polymerization in the presence of peroxides. (40 marks)
 - b) Explain the advantages of cross link polymers over linear polymers. (20 marks)
 - c) Compare the properties of crystalline polymers and amorphous polymers.

(40 marks)

Cont.

- 3. a) Explain how you determine the rate constant of the step growth polymerization for self-catalyzed systems. (Derivation of equations required) (50 marks)
 - b) The structure of a polymer X is given below.

$$\begin{array}{c|c} a & O & b \\ H_3C & b & m \\ \end{array} \begin{array}{c} O & O \\ D & m \\ \end{array} \begin{array}{c} O & O \\ C & C \\ C & C \\ \end{array}$$

Ratio of area under the curves of the peaks which are relevant to protons a, b and c are 1: 20: 16 respectively. Calculate the number average molecular weight (M_n in Dalton) of the above polymer.

(50 marks)

- 4. a) Draw the structure of the repeating unit of natural rubber. (10 marks)
 - b) Natural rubber can be described as a suspension. Briefly explain its composition.

(30 marks)

- c) Briefly explain the rubber harvesting steps. Include the chemistry appropriately. (30 marks)
- d) Write down the chemical reactions for the vulcanization of natural rubber. (30 marks)

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