



**RAJARATA UNIVERSITY OF SRI LANKA
FACULTY OF APPLIED SCIENCES**

**B.Sc. (4 year) Degree in Applied Sciences/ B.Sc. (Joint Major) Degree in Physics and
Chemistry
Fourth Year - Semester I Examination – January/February 2021**

PHY 4312 – STATISTICAL THERMODYNAMICS

Time: Three (3) hours

Answer All Questions

Instructions:

- This is a closed book examination. The examination paper contains 03 pages.
- You are given 3 hours to complete the examination. There are 04 questions to be answered. You may allocate your time wisely. Read the questions carefully before answering them.
- A non-programmable calculator is permitted.
- Plagiarism (copying) is considered as a punishable offence.

1. A system of two energy levels E_0 and E_1 is populated with N particles at temperature T . The particles populate the energy levels according to Maxwell-Boltzmann (classical) statistics.
 - a) Derive an expression for the average energy per particle. (5 marks)
 - b) Calculate the average energy per particle when $T \rightarrow 0$ and $T \rightarrow \infty$. (5 marks)
 - c) Derive an expression for the specific heat of the system of N particles. (5 marks)
 - d) Calculate the specific heat when $T \rightarrow 0$ and $T \rightarrow \infty$. (5 marks)

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2. A system of N distinguishable non-interacting particles, each fixed in position and carrying a magnetic moment μ is immersed in a magnetic field H . Hence, each particle exists in either $E = 0$ or $E = 2\mu H$ energy states.
- The entropy, S of the system is defined as $S = k \ln \Omega(E)$, where k is the Boltzmann constant and $\Omega(E)$ is a function of the total energy E of the system.
Provide the definition of $\Omega(E)$. (3 marks)
 - Write a formula for $S(n)$, where n is the number of distinguishable particles in the upper state. Sketch $S(n)$. (4 marks)
 - Derive Stirling's approximation for large n :
$$\ln n! = n \ln n - n$$

by approximating $\ln n!$ (5 marks)
 - Using the results in parts (b) and (c), find the value of n for which $S(n)$ is maximum. (5 marks)
 - Treating E as continuous, show that this system can have negative absolute temperature. (5 marks)
 - Although negative temperature is possible for this system why is it not possible for a gas in a box? (3 marks)
3. Suppose the energy of a particle is defined by the expression $E(z) = az^2$ where position or momentum, z can assume all values from $-\infty$ to $+\infty$. A system of such particles follows classical (Boltzmann) statistics.
- Show that the average energy per particle is $\bar{E} = kT/2$. (6 marks)
 - State the principle of equipartition of energy, and explain its relation to the above calculation. (4 marks)
4. A gas of N spinless Bose particles of mass m is enclosed in a volume V at a temperature T .
- Find an expression for the density of single-particle states $D(\varepsilon)$ as a function of the single-particle energy ε and sketch the result. (5 marks)

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- b) What is the mean occupation number of a single particle state \bar{n}_ϵ as a function of ϵ, T , and the chemical potential $\mu(T)$.
 Draw this function in the sketch drawn in part (a) for a moderately high temperature (above the Bose-Einstein transition). Show $\mu (= \epsilon)$ on the energy axis. (4 marks)
- c) Derive an expression for N/V that implicitly determines $\mu(T)$. Using the sketch drawn in part (a), determine the movement of direction of $\mu(T)$ when T decreases. (6 marks)
- d) Derive an expression for the Bose-Einstein critical temperature, T_c , below which one must have a macroscopic occupation of certain single particle states.
 (Hint: $A = \int_0^\infty \frac{x^{1/2} dx}{e^x - 1} = 1.306\sqrt{\pi}$) (6 marks)
- e) What is $\mu(T)$ for $T < T_c$?
 Describe $\bar{n}(\epsilon, T)$ for $T < T_c$. (5 marks)
- f) What is the total energy, $U(T, V)$ of the gas for $T < T_c$. (4 marks)

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